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PROJECT OFFICERS REPORT—PROJECT 5.2/5.3b

RADIOBIOLOGICAL, RADIOCHEMICAL, AND PHYSIOCHEMICAL ANALYSES -(U)

W. J. Major, Project Officer

R.A. Wessman

Tracerlab
Richmond, California

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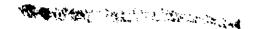
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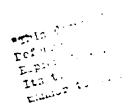
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#### **ABSTRACT**

Data on the plutonium and uranium content of biological and physical samples, collected and isolated from non-nuclear detonations of plutonium bearing weapons under various storage situations, are presented. A precision tracer (Pu-236) procedure was developed for the rapid analysis of the plutonium, which was non-uniformly distributed in these samples. A fluorimetric procedure was developed for the rapid analysis of uranium.

Measurement of the plutonium content was accomplished by equilibration of tracer with sample plutonium, radiochemical purification, tracer yielding, and alpha pulse-height analysis. This method ensured a high degree of accuracy, high sensitivity, and freedom from interference from other alpha emitters.

Over 4,000 radiobiological, radiochemical, and fluorimetric analysis are tabulated. The analyses were performed at Tracerlab's western division in three isolated laboratories plus separated counting facilities. Accurate laboratory analysis of all samples was achieved with no personnel contamination or cross-contamination of samples. Procedures for radiochemical analyses, handling of special problems, techniques of Alpha Pulse Height Analysis, quality control measures, and additional data based on radiochemical analysis and radiometric measurements are presented.

### PREFACE

We are grateful to Dr. K. Stewart of the United Kingdom and Professor R. Wilson of the University of Rochester for their helpful suggestions.

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## CHAPTER 1 INTRODUCTION

#### 1.1 OBJECTIVE

The objectives of Project 5.2b (Radiobiological Analysis was to provide accurate laboratory analysis of animal tissue, bone material, and metabolism samples for plutonium and uranium content. Plutonium analyses were performed on all samples and uranium analyses on approximately ten percent except Clean Slate II, dogs and sheep, which required uranium analyses on most samples. The uranium analyses were representative of sample and animal types.

The object of Project 5.3b (Radiochemical and Physiochemical Analysis) was to provide accurate laboratory analysis of air, deposition, water, vegetation, sticky wire, and soil samples for plutonium and uranium content. Plutonium analyses were performed on all samples and uranium on approximately 10 percent, representative of sample types.

#### 1.2 BACKGROUND

The personnel associated with Project 5.2/5.3b did not participate in the field phases of operation Roller Goaster. Reference is given therefore, to other projects for a description of operational events and sample collection for laboratory & alyses. The scope of

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this project was to provide facilities, services and materials to carry out the dual objectives listed above.

#### 1.3 FACILITIES

Laboratory: The radiobiological, radiochemical, and physiochemical analyses of the field collections for plutonium and uranium are one of the prime sources of evaluative data for Froject Roller Coaster. The samples collected represent individually, and totally, large sums of money and scientific effort. For this reason, analyses were performed with great care, attention to detail, and utmost precision. Only those techniques which resulted in unequivocal data were used. Particular attention was given to the problem of crosscontamination. Two techniques were employed. The first was sequential processing, starting with low level samples and proceeding to the higher level samples, the second involved the physical separation of high, intermediate, and low level facilities. Both techniques were used in series. The lowest of high level samples were processed initially in the intermediate level facilities. For high level samples, a special wing of the main laboratory building was employed for initial separation and aliquoting, followed by processing in the intermediate laboratory. All low level samples were processed in a separate low level laboratory.

Counting and Calculations: These facilities were located in an isolated wing of the building. Advanced counting and calculation techniques used by Tracerlab for a number of years were employed. Detailed procedures are given in the reference.

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flow, internal alpha counter to determine its range of activity. This internation was then used to preset the precision (alpha pulse height inalysis) counting time and to prevent mixup of samples. The range of counts was limited by dilutions to approximately 2,000 cpm. The range of counts in the original sample was 0 to 108. Counts per minute was converted to dpm by standard calculation procedures given in the references.

The technique of accurate alpha pulse height analysis depends on such factors as: preparation of high quality standards, calibration and maintenance of the equipment and routine background, and operational checks of counting instruments. Six Traceriab Frisch Grid Chambers were employed for plutonium alpha detection. Three Technical Measurements Corporation, multi-channel analyzers were used for readout. Four Frisch Grid Chambers were connected to one multi-channel analyzer by dividing the full range into four quadrants. The Frisch Grid Chambers were routinely operated in this manner.

The results of the alpha pulse height analysis are presented on tape. A graphical plot was made of this information where shape and resolution of alpha peaks were marginal. The various corrections and factors were applied to the data and the final result calculated as concentration per sample.

A computer program for data tabulation was developed. The program simplified new data insertion and provided for printing of results rapidly and economically. An IBM card punch, located in a room adjacent to the counting room, was employed for transcribing raw data.

Storage: All biological samples were stored in a specially built walk-in freezer located in the viewel laboratory. The unit was large enough to store all the sample freezer boxes with adequate spacing for easy access. The physical samples were stored in metal, office-type file cabinets with fabricated security locks. Mounted samples, following analyses, were stored in locked file cabinets in the counting room. Unused portions of samples were stored on shelves in a locked stockroom located adjacent to the intermediate level laboratory.

#### 1.4 SERVICES AND MATERIALS

Services and materials were provided to perform a research project consisting of plutonium and uranium analysis on the following variety of sample matrices:

Casella Impactor Discs

Casella Impactor Filters

Andersen Sampler Discs

Andersen Sampler Filters

Total Air Samples

Total Air Samples Disposable

Sequential Air Samples

Balloon Wire Swipes

Water Samples

Vegetation (Sagebrush)

**Deposition Samples** 

Soi. Samples

**Biological Samples** 

Servicus and materials were provided to assure that resultant data was most meaningful to the requirements of Project Roller

Coaster and that biweekly progress reports giving accrued results were submitted to Director, DASA. This included:

- (1)\* Use of new glassware for each analysis.
- (2)\* Isolation of personnel and facilities for varying levels of activity.
- (3) Constant monitoring of muffle furnaces, hoods, work tables, floors, etc., by trained monitors under supervision of a Certified Health Physicist.
- (4)\* Utilization of Plutonium 234 tracer techniques on all samples to assure measurable and accurate yields on all samples.
- (5)\* Analyses of all plutonium samples by alpha spectroscopy,
- (6)\* Complete dissolution of each sample prior to purification.
- (7) Establishment of reagent blanks less than 0. 1 alpha dpm  $\pm$  100% and less than 5  $\times$  10<sup>-9</sup> grams for plutonium and uranium respectively.
- (8) Laboratory monitoring of stippled plates of dissolved samples as a means of sample separation by activity level and preventing sample mixup.
- (9)\* Electrodeposition of plutonium on platinum discs as the final step of the analysis.

<sup>\*</sup> These procedural techniques were stipulated by the Roller Coaster Radiochemistry Referee Team and were conditions of the contract.

- (10) Storage of unused portions of samples and all mounted samples for a period of one year or until notification was received by the Contracting Officer, whichever came first.
- (11) Continuous Tracerlab staff evaluation monitoring of a quality control program.

#### 1.5 PERSONNEL

Tracerlab provided all the personnel for the services described under Section 1.4. These included the following and their responsibilities:

- (1) Evaluation Staff: General conduct of the project, monitoring of the quality control program, and review of periodic reports.
- (2) Project Officer (1): Supervision of the project operation, health and safety standards, review of all data, writing of periodic reports, liaison with cognizant Roller Coaster officials.
- (3) Radiobiologist (1): Operation of radiobiological laboratory and sample accountancy.
- (4) Physiochemist (1): Operation of two physiochemical labortories and sample accountancy.
- (5) Radiobiological Technicians (3): Radiobiological analyses of physical samples.
- (6) Physiochemical Technicians (3):- Physiochemical analyses of physical samples.
- (7) Uranium Technician (1): Uranium analyses and calculations.

- (8) Health Physicist (1): Routine monitoring of work areas and waste materials.
  - (9) Radiometrics Head (1): Review counting and calculations.
  - (10) Counting Technician (1): Gount plutonium samples.
- (11) Calculation Clerks (2): Calculate plutonium counting data.
- (12) Computer Data Clerk (1): Punch and proofread computer data.
- (13) Electronics Technician (1): Maintain counting instrumentation.

#### CHAPTER 2

#### **PROCEDURES**

#### 2.1 SAMPLE INVENTORY

Biological: The samples arrived by government air freight at the Alameda Air Terminal in Oakland, California, on 20 September 1963. They were contained in 12 polyfoam freezer boxes weighing approximately 50 pounds per box. All the boxes appeared to be in good condition and a receipt for the same was given to the DASA courier who had accompanied the samples from Kirtland Air Force Base, Albuquerque, New Mexico. The boxes were transported to Tracerlab by truck and placed in the walk-in freezer unit in the low level laboratory, awaiting inventory instructions from cognizant DASA security personnel. On 26 September 1963, the contents were inventoried in the presence of a DASA security officer. All samples not clearly marked were restage ged and returned to their original containers. A separate log book was established, and the following week a quality control program was iniated. Quadruplicate analyses were run on all reagent materials and a low level background established. A tracerlab code number was assigned each sample.

Physical: The samples were delivered to Tracerlab by Tracerlab's
Health Physics Officer at intervals spanning a three-week period,
starting in mid-October 1963. The samples were contained in
heavy-duty cardboard boxes, doubly wrapped. All boxes appeared

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to be in good condition. The Double Tracks samples arrived first, tollowed by Clean State I, II, and III. The samples were placed in the combination security files in the intermediate level laboratory and inventoried in the presence of the Tracerlab Security Officer. A separate log book was established and a Tracerlab code number assigned each sample. All samples were clearly marked and identification presented no problems. A quality control program was initiated shortly after arrival of the samples. This program was purposely delayed until after final sample inventory to determine if any contamination of the laboratory had occurred. Quadruplicate analyses on reagent materials and laboratory swipes were run and a low level background established.

#### 2.2 PROJECTED ACTIVITY LEVELS

Piological: Based on the results of TB-57 (Reference 1)\*, the majoraty, of tissues were expected to be low in total plutonium content. The range would be from almost undetectable to thousands of dpm of plutonium in the nasal mucosa and GI tract (Table 2.1). Because of the spread between activity levels, extreme care was taken in the preparation and processing of the low level tissue samples. As far as the receiving laboratory was concerned, these samples represented no problem in plutonium handling. Since activity in a sample might be distributed unevenly, the entire sample was always analyzed.

\* (Also, see References 2 through 8.)

TABLE 2.1 SELECTED TG-57 DATA FOR PLUTONIUM ACTIVITY

Tissue	Mean Weight (Grams)	DPM/Gm Magnitude	DPM/Qrgan
Spleen	23, 9	0, 1	2.4
GI tract plus Contents	548, 6	10	5500
Liver	309.0	0, 01	3, 1
Lung	76.6	0,5	38
Trachea	11.9	1	12
Right Femur	34.7	0, 1	3.5
Rib	4, 6	1	4.6
Hilar Lymph Node	0,45	10	4, 5
Mediastinal Lymph Nodes	0,31	10	3.1
Nasal Mucosa	Not Analyzed		

The levels of uranium at 500, 5000, 7500, and 17,500 feet animal positions corresponding to the plutonium are, based on extrapolation from TG-57 data, proportional to the ratio of uranium to plutonium in the test device. Thus in samples very low in Pu content, the uranium level was expected to be near the limits of detection of the fluorimeter.

<u>Physical</u>: The sources of plutonium and uranium from the Roller Coaster tests were soils and various types of collection devices (surface and airborne), such as filters, impactors, and sticky plates; all located at various distances from the detonation crater, based on TG-57 data. The activity levels were expected to range from 0 to 10<sup>7</sup>

dpm. Particulates of all sizes were expected to be in various matrices.

The plutonium and uranium would vary in chemical composition from metals to various alloys, oxides, and salts produced in the heat and pressure of the explosion. Projected activity levels for various sample types are given below.

- (1) Crater Samples: The crater samples will be extremely nigh level with most of the activity near the rurface. In TG-57, 80 square-inch samples were taken to a depth of two feet and sectioned into 1/4 inch increments. At the surface, this amounted to a milligram of plutonium per 25 grams of soil.
- (2) Soil Samples: The samples from soil cores and surface fallout were spread out over many square miles. Three levels of surface samples (corresponding to those observed in TG-57) were considered, i.e., 500, 40, and 2.6 μg/M<sup>2</sup> at distances of 500, 1000, and 2000 feet, respectively. An 80-square-inch (0.0515 M<sup>2</sup>) surface sample yields 25, 2, and 0.13 μg plutonium, respectively. These sample sizes were more than ample for uranium fluorescence and radiochemical analysis. Uranium in Nevada soil has been determined. The levels range from 0.1 to 6.0 μg/g of soil, which represents an appreciable normal uranium background.
- (3) Filter and Impactor Exit Filter Samples: The levels of filter activity observed in TG-57 for fallout samples of various distances were less than one microgram of Pu at 1000 feet and 0.01 µg at 25,000 feet.
- (4) Sticky Samples: The targets of the various impactor jets as well as fallout plates employ an alkyd resin surfacing to retain the

particulates as they impact or fall out on the surface. The levels of Pu activity observed on these surfaces were estimated to be of the same order of magnitude as the air filter, or in the case of sticky plates, equivalent soil samples.

#### 2.3 SAMPLE PROCESSING

Handling Techniques: Advantage was taken of information locating the field position of the sample with respect to ground zero and the gross alpha counting data supplied with each sample. Samples were screened and rough assayed to confirm the accompanying data. The handling facilities themselves were checked routinely and blank samples processed to confirm the ambient levels.

As indicated in Table 2.1, the biological samples generally constituted the lowest level samples and were processed in the low level laboratory. lowest level samples and were processed in the low level laboratory. Those biological samples having a probable higher Pu content were processed in an isolated section of the low level laboratory. In contrast, the debris samples, potentially several orders of magnitude higher in plutonium content even at the 250,000-foot distance, were treated as high level samples and processed separately until their Pu content was established.

Techniques gained through experience for prevention of cross contamination and mixing in the laboratory were employed. In particular, these techniques included proper recording and marking of each sample by the analyst at every stage in the process. All reagents used in this work were made up fresh in new containers and designated accordingly. New glassware was used for every analy-

RFP 3-63. When muffle furnaces were employed, each sample was covered to remove the possibility of flake out. Hoods, laboratory surfaces, and other exposed areas were cleaned routinely and monitored for possible alpha contamination. The unused portions of samples were returned to the original containers when possible, checked for proper labeling, and stored in a locked cabinet until notification of disposition was received from the contracting officer. Similarly, both the counting discs and planchets (stored in individual envelopes or pillboxes) and the original counting data were retained until notified by the contracting officer. Blanks, spikes, duplicates, and actual standards as required were analyzed and furnished to referees or the agency designated by DASA.

Security and Accountability: Several means of assuring adequate security and accountability of all biological and physical samples were investigated. The methods given below offered the most effective operation.

- (1) Biological and physical samples inventoried by two persons and sample numbers initiale! by each on inventory sheets.
- (2) Active Inventory: Biological and physical samples withdrawn from inventory (by convenience, activity level, and event) recorded in separate logs, assigned a consecutive Tracerlab number, and initialed by custodian.
- (3) Each sample assigned a card at start of analysis, card initialed by analyst at start of every major step (preparation, tracer addition, dissolution, and purification).

- (4) Sample card information dup icated in sample log.
- (5) After decontamination sample assigned a data sheet, pertinent information entered, and sample transferred to counting room.
- (6) Sample recorded and initialed by counting room custodien in a special counting room log with a consecutive number matching that in the chemistry log.
- (7) Sample stored in security file and final calculations processed by Traceriab's normal red dot doublecheck.
- (8) Final data reviewed by counting room supervisor, project officer, and department manager.
- (9) Following final review of the raw data, a tabulation of all values was made by animal numbers (for biological samples) and arc location (for physical samples).

<u>Production Operation</u>: The following procedures were employed to achieve maximum production.

- (1) Different analyst assigned to each phase of the production operation on a rotation basis.
  - (2) Samples prepared for dissolution in batch type operation.
  - (3) Tracer added in batch type operation.
  - (4) Samples solubilized in batch /pe operation.
- (5) Samples decontaminated in continuous operation (samples were processed by pairs. Thus, it was possible to precipitate, centrifuge, extract, etc., several sets of samples in a continuous operation eliminating dead time).
  - (6) Plated samples monitored for approximate yield.

(7) Plated samples inspected and data sheet prepared by custodian.

à

- (8) Samples 2 TT counted to estimate requir J alpha pulse height analysis time and to provide a double check on final result.
- (9) Production schedules set for each analyst and weekly results posted.

Preparation: The treatment of the physical samples varies according to the nature of the collection media. These divide into three types: a heterogeneous mixture of soil and debris, air and surface fallout, and resuspended particulates collected on organic filters and similar material adhering to sticky plates. Because of the high levels of plutonium alpha activity expected, all field and laboratory procedures were reviewed with regard to preventing cross contamination and preserving the integrity of the samples. The level of uranium was not hazardous but similar precautions apply.

The samples for radiochemical and uranium fluorimetric analysis were categorized at the time of receipt of the samples. The samples were already grouped at the test site. Similar levels of alpha activity were handled together and precautions against cross contamination were maintained. Because of the relatively high levels of alpha activity, some samples were processed in glove boxes.

All samples were ultimately reduced to the levels required for laboratory operations and stored or processed as scheduling permitted.

Dilutions were made such that no aliquot contained more than 4000 dpm.

Sample Control: Strict sample control was Instituted in order to be sure that samples were not misplaced, delayed, or processed with samples of different activity magnitude. As mentioned earlier, many samples received for processing contained designations as to general activity level and sampling location. This information was useful in routing the sample to the proper dissolution laboratory. Locations of particulate samplers consisting of six-stage Andersen impactors, five-stage Casella impactors, total air samplers, and sticky cylinders, are given for reference in Figures A. 2 through A. 8.

Figure A. 1 shows Clean State igloo dimensions.

For each lot of samples received, a sample check-off sheet was initiated containing sample numbers and due dates. It was reproduced and distributed to the key personnel along the processing route. These personnel checked off the samples as they were processed, and thus continuous check on sample status was maintained. In addition, a sample processing card and sheet (mentioned in Section 2.3) was initiated for each sample. The card contained pertinent chemical information and followed the sample through dissolution and decontamination. The processing sheet, in addition to the card information, contained spaces for recording all data needed to calculate the analytical results. Spaces were provided for sample size, sample aliquots, tracer aliquots, etc. It also contained sample routing instructions and followed the sample from decontamination through calculations. A typical processing sheet is included in Appendix C. When final calculations were completed and reviewed by the project officer, the data was transcribed to IBM cards for the preparation of a computer report. Quality Control: In order to maintain high quality standards, the analytical work of this laboratory was closely controlled. This control was maintained in the laboratory and in the counting room, Blank samples were processed completely through chemistry and counting to determine if there was any laboratory contamination. Known plutonium samples (standards prepared by adding a Pu-239 spike to a matrix material similar to those being processed) were similarly cycled through the laboratory to check on procedures and counting geometry factors. In addition, standard and calibration samples furnished by DASA referee personnel were processed to assure results of all Roller Coaster contractor laboratories were comparable. Beta gamma, and alpha activities of electrodeposited alpha samples of solutions of pure Pu<sup>239</sup> and mixed Pu<sup>239</sup> were cross-counted among Air Force and AEC laboratories during the Roller Coaster analyses period. Periodic blank samples were also cross-counted.

The techniques used in maintaining high quality standards for uranium analysis are specified in the detailed Fluorimetric Determination Procedure included in Appendix B. Blanks, standard solutions, and spiked unknowns were analyzed with each batch of 20 samples. A routine review of the results of all samples processed was performed by the project leader or his delegated assistants. All counting data was reviewed to asertain that alpha spectrometer runs were good as to alpha peak resolution, that the alpha peaks were properly shaped without undue tailing of one peak into the next, and that the base area of the peak was properly defined. Other aspects of the sample

run were checked such as adequate chemical yield, clean weightless electrodeposits on Pu sample plates, etc. Any discrepancies
were reported to the project leader on the QUALITY CONTROL - SAMPLE DEFICIENCY REPORT form for appropriate action. A copy of this
form is attached in Appendix C.

#### 2.4 GENERAL LABORATORY METHODS

Discussion: The material presented in this section describes, in general, radiochemical techniques used by Traceriab for a number of years for determining Pu-239 and uranium in various sample types.

In particular, Pu-236 tracer yielding for all pluton' an enalyses was employed. Radiochemical techniques were such that yields greater than 60% were generally obtained. Separation of uranium and pluton-tum was carried out on every sample. Furthermo: all plutonium samples were measured by pulse height analysis techniques, with high sensitivity and assurance of no interference by other alpha emitters. The counting error had a precision of 10% for the low counting samples when economically feasible and 3% or better for the more active samples.

The recovery efficiency for uranium was checked on each analysis according to the techniques described in "SCTM 369-59(51) Test Group 57 Radiochemistry", by R. J. Everett and R. W. Drake. All samples were completely dissolved. An aliquot, usually 10% of the total dissolved solution, was taken from samples requiring uranium analysis. When required, an extraction was performed to remove quenching agents suc 12 copper and iron cations. The total uranium was determined by a fluorescence technique in a Jarrell-Ash

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Fluorimeter.

The biologicals presented some unique challenges both in dissolution and purification. Prior to the start of the project, an exhaustive literature survey was made and consultations held with individuals learned in the bioassay field on the problem subject. Most
of the information and procedures offered, however, dealt with organics less than 50 grams in weight. Very little was known about analyzing pound size samples for plutonium and uranium. It was decided,
therefore, to combine modified Tracerlab biological procedures with
those in the literature, to fit the situation. A lengthy development program produced the detailed radiochemical and fluct imetric procedures
given in Appendix B.

A technical paper, "Routine Determination of Plutonium by Tracer Techniques in Large Biological Samples," based on our biological development work, was presented at the Hanford Symposium on "Inhaled Radioactive Particles and Gases" and at the 9th annual Health Physics Society in Cincinnati, Ohio. A copy of the paper is given in Appendix D.

The biological samples were processed concurrently with the physical samples, which were arranged in order of Double Tracks, Clean Slate I, II, and III. The techniques for dissolution, separation, purification, are described in the following paragraphs. It should be mentioned that a given procedure does not necessarily cover every chemical or counting situation. Often one or more samples among identical types required special treatment.

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Biological Sample Dissolution: The biological dissolution procedures employed resulted in the least loss of sample,. Wet versus dry dissolution was experimentally compared. Dry ashing of most biological samples, although convenient and inexpensive, results in loss of sample by spattering, mechanical entrainment, and polymerization and formation of insoluble oxides of Pu. This is especially true of samples which have high organic-to-ash ratios. In these cases it may be difficult or impossible to recover all of the sample plutonium from the walls of the ashing container. Loss occurring at this stage may result in inaccurate sample yielding. Bone samples have a low organic-to-ash ratio, and the ash serves as a carrier to prevent loss of Pu during dry ashing. The bone ash is bulky and easily removed from the ashing container by dissolving in acid and then is equilibrated with Pu tracer. Wet dissolution in the presence of Pu-236 tracer allows exchange of sample Pu with tracer and control over excessive temperatures, thus preventing formation of insoluble oxides and polymers. Wet dissolution was routinely performed, using a refluxing apparatus (Figures F.1 through F.3) or open beakers, by an experienced chemist. All samples processed for plutonium analyses employed Puns tracer for yielding. The tracer activity was normally aliquoted such that it was within a factor of five of the expected sample activity but a minimum of 15 dpm. The tracer was always added at the start of dissolution except for bone samples. Dissolution of the biological samples varied with the tissue or metabolism type and size. A brief discussion of each is given below.

(1) Small tissues (<2 ounces): Samples were placed in ap-

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propriate sized beakers (250 to 400 ml), organ activity was estimated from the data in TG-57, and appropriate tracer, added. Each sample was covered with HNO3 and the mixture boiled to low volume. Fuming HNO3 and HClO4 were added and the mixture again boiled to low volume. When the HNO3 was driven off and the HClO4 concentrated by boiling, an exothermic reaction took place and moderate foam swelled up inside the beaker. The reaction could be controlled by the addition of HNO3, but with small organics this was not usually necessary. Further boiling produced a clear solution containing only minor amounts of salts which were solubilized on dilution. Care was taken to avoid evaporation to dryness since formation of explosive perchlorate salts would result.

in appropriate sized beakers (1 to 4 liters) and tracer added. Each sample was covered with HNO3 and boiled to low volume. Sulphuric acid was added to char the organic and the mixture fumed to low volume twice until a deep red solution was obtained. Fuming HNO3 was added and the solution boiled to low volume. Nitric acid and HClO4 were added in that order and the solution boiled to low volume. Sulphuric acid was again added and the solution boiled to low volume. Sulphuric acid was again added and the solution boiled to low volume to drive off all HClO4 which forms explosive mixtures with the cupferron—CHCl3 reagent added later to extract plutonium and uranium. The dissolution of the samples in this category was done by the HClO4 method rather than the H2SO4 reflux method since the former was much faster. Also the amount of organic present at the time of HClO4 addition was small and any exothermic reaction (occurring

when hot concentrated HClO<sub>4</sub> is mixed with organic matter) was mini-

Large Tissues (>3 pounds): Most of the samples in this category were dissolved in four-liter beaker; a few tissues had to be divided into two or more sections to fit. To eliminate the thawing process, an electric knife was used for the division. Tracer.  $K_2SO_x$ , Hg catalyst, and antifcam agent was added to each section. Enough H2SO4 was added to cover the sample, and an inverted 6-inch funnel, held by a ring stand, was placed inside the beaker. The sample was placed on an individual hot plate (covered with asbestos to avoid cracking of beaker at ensuing high temperatures) and heated at low temperature until a black tar mixture was obtained. The heat was increased until the tar turned to, in order, black jelly, black liquid, red liquid, clear solution. During high-temperature heating the  $\mathrm{H}_{200_4}$  refluxed and the inverted funnel was raised or lowered to control the action. Asbestos wrapped around the beaker increased the temperature and reflux conditions. A trace of carbonaceous material left on the peaker and funnel walls after refluxing was removed by  $\mathrm{HClO}_{\scriptscriptstyle{A}}$  cleansing and boiling. However, the formation of metal organic salts from the  $\mathrm{HClO}_4$  precluded good decontamination and  $\mathrm{H_2SO}_4$ washings were substituted. The remaining H\_SO\_4 was finally fumed to wet dryness. Since many samples were processed simultaneously the billows of heavy, toxic SO3, HClO4, and nitrous oxide fumes outside the lab created a potential health problem. A multi-vacuum apparatus, leading to a large polyethylene carboy filled with a dilute base, trapped most of the fumes. The balance were solubilized in a waterscrubber apparatus attached to the outside mood vents. Distillation and condensation of the acid fumes was also tried inside the hood and found to be effective. During H<sub>2</sub>SO<sub>4</sub> evaporation large amounts of inportant salts (from the combination of the acid radical and the minerals in the animal organ) precipitated out of the acid solution. Since these salts interfered with later decontamination of Pu and U, a procedure was developed in which the heavy elements were reduced with NH<sub>2</sub>CH-HCl and extracted from the bulk salts with cupierron and CHCL<sub>3</sub>. As a result of this modification clear plates and good yields were obtained.

Bone Samples (All Sizes): Al. bone samples were dried in (4) a drying oven overnight to reduce smoking and popping during the ashing operation. Following the drying process the bones were cut as required and ashed at 500°C overnight in Corningware (Corningware is glazed and eliminates ash sticking to the walls). The ash, salts, and low smoke content of bones obviated swelling and entrainment loss of Pu (as contrasted to animal organ samples). The ash was transfered quantitatively to a beaker and dissolved in HCl. Tracer was added at this point rather than at the start to assure equilibration and accurate yield. Losses of U as well as Pu in ashed bone samples are prevented by the heavy ash content. The solution was boiled to low volume and the plutonium and uranium extracted from the large amounts of salts by the cupterron - CHCl3 method mentioned earlier. The extracted material was boiled to low volume and reboiled with  $\mathrm{HClO}_4$  to wet dryness. in the larger bone samples, a white residue appeared at this point and a second extraction was necessary. A few ml of HCl was added to the final wet dry HClO4 mixture prior to the second extraction. An attempt was made to remove Pu from an acid solution on a  $\operatorname{Zr}_3(PO_4)_4$ 

precipitate. However, the yields were lower and plates extremely dirty, distorting the alpha energy spectrum. This procedure was discarded early for the extraction.

ples were analyzed similarly except urine samples were first evaporated to wet dryness (after tracer addition). Both types of samples were then covered with HNO3 and boiled to low volume. Fuming HNO3 was added and the solution cautiously evaporated to dryness. When the samples were near dryness, ignition occurred and the residue burned slowly with the evolution of nitrous oxide fumes. After the pyrotechnic flame had subsided the residue was taken up with HNO3 and HClO4 and boiled to low volume as in the medium tissue procedure. Sulphuric acid was added and the mixture boiled to drive off HClO4 prior to the extraction with cupferron-CHCl4.

Biological Sample Purification: After dissolution, the sample was purified. Purification is necessary to decontaminate the sample from other radionuclides present and secondly to separate plutonium from macro amounts of all other elements. The final product is a weightless, contaminant-free invisible deposit of plutonium. The preparation of a weightless deposit yields sharp, well-resolved alpha peaks. The procedure is simple, well-established, and with normal care results in high yield. The basic steps in the procedure are Fe (OH) precipitation, ion-exchange separation, and electroplating onto a polished platinum disc. The plated sample is placed in a small labeled metal contains, and is counted by alpha pulse height analysis.

The purification of the larger biological and bone samples pre-

sented some special problems. Off color (white) Fe(OH)<sub>3</sub> precipitates, acid dissolution residues, and violent NaBrO<sub>3</sub> oxidation reactions often occurred if a hexone extraction was employed.

Removal of salts in the cupierron - CHCl<sub>3</sub> extraction immediately following dissolution eliminated most of the problems described.

As mentioned earlier, some bone samples required two additional cupierron - CHCl<sub>3</sub> extractions to prevent large CaOH-CaPO<sub>4</sub> precipitates occurring in the first step in purification. Purification of a biological sample sometimes required large volumes of CHCl<sub>3</sub> and several days of an analyst's time.

Physical Sample Dissolution: All physical samples processed for plutonium analysis employed Pu-236 tracer for yielding. Tracer additions were similar to those of the biological analyses except where large dilutions were necessary. Heavy soil samples were set aside pending investigation of a partial dissolution procedure. Wet chemistry techniques (using HF) were employed on samples containing small amounts of soil. Filter and sticky film samples were treated with fuming nitric and perchloric to destray organic matter and then with HF to dissolve any silicates present. Physical samples, as biological, were treated with H<sub>2</sub>SO<sub>4</sub>-HClO<sub>4</sub> to assure equilibration of tracer and sample Pu.

Dissolution of the physical samples varied with sample type and size. Generally HF treatment was required to remove silicates and all of the dissolutions were started or transferred to teflon beakers. A brief discussion of those types giving dissolution

problems are listed below (HF disc boiling avoided, prevent U pickup).

- (1) Cassella and Andersen Discs: No special obstacles were encountered until the end of the normal dissolution procedure. A white residue was observed on the surface of some of the discs after removal from the acid dissolver solution and subsequent air drying. The residue was checked for activity but none was apparent. As a precaution, the glass disc was rinsed into the original beaker with a 1N HF-HNO<sub>2</sub> solution which removed all traces of the residue.
- (2) Sticky Films, Method No. 1: The sample was covered with fuming HNO<sub>3</sub> and boiled to low volume. The procedure was repeated until the solution turned from black to a dull red (usually required approximately one liter of fuming HNO<sub>3</sub>). Perchloric acid was added and the mixture boiled to low volume. An exothermic reaction occurring at this point was allowed to go to completion. Further boiling produced a clear solution.
- with fuming HNO<sub>3</sub> and boiled to low volume. The process was repeated 2 or 3 times and the mixture allowed to dry and ignite on the last time. Ignition was encouraged by dropwise addition of fuming HNO<sub>3</sub> and heat. Final dissolution of the carbon black sample was accomplished with addition of HClO<sub>4</sub>. Limited exothermic reaction occurred in this step.
- (4) Sticky Films: Method No. 3: Approximately 5 ml of CH<sub>3</sub>OH was added and the sample ignited with a Fischer burner. Furning HNO<sub>3</sub> and HClO<sub>4</sub> were added after ignition was completed and mixture boiled to low volume. If the sample contained appreciable

amounts of dirt, bumping occurred. Addition of 3 to 5 ml of H<sub>2</sub>SO<sub>4</sub> climinated this bumping in boiling to low volume (some foaming occurred at this point). Finally, HF was added to effect complete discolution of the dirt. This method proved to be the most economical timewise and in consumption of reagent. No loss of sample was evident by activity measurements of filter collections of the fumes.

- (5) Total Air (TAS): These samples were usually dissolved in fuming HNO<sub>3</sub> and charred, followed by a fuming HNO<sub>3</sub>-HClO<sub>4</sub> dissolution. Samples with appreciable amounts of dirt required HF treatments.
- (6) Total Air Disposable (TASD): Samples were relatively bilky and required several acid dissolutions. The samples were treated with fuming  $\mathrm{HNO_3}$  and charred. A sticky ring remained on the wall of the teflon beaker which was dissolved by boiling with  $\mathrm{H_2SO_4}$  and  $\mathrm{HClO_4}$ . Nitric acid and HF were added after the last  $\mathrm{HClO_4}$  reaction and the sample boiled to low volume.
- done almost entirely in teflon beakers due to required HF treatments. Fuming HNO<sub>3</sub> was first added to cover the sample and the sample boiled until yellow fumes were no longer evident (bumping occurs at too low a volume). Perchloric acid was added and the mixture boiled to wet dryness. Little reaction occurs during this step. Fuming HNO<sub>3</sub> was added to cool the mixture and HF added (15 to 25 ml for each 5 gms of soil, 1 ml at a time to control reaction) until low foam is rection subsided. The mixture was boiled until a clear solution was

<sup>\*</sup> Not normal, but backing stuck to filter.

obtained. All the sample except water soluble salts and a trace of hard silica (the latter showed no measurable activity) went into solution after acid boiling. Dilution of the acid solution (250 ml for each 10 cms of soil) dissolved all residues but the hard silica.

(8) Large Soil Samples (>5 gms): The physical samples were of greater variety but generally easier to dissolve than the biologicals with the exception of those containing heavy dirt. All of the soil samples, (approximately 60) received for analyses contained 1 to 6 pounds of dirt and sand. Samples this size can be dissolved with large quantities of acids and a lengthy digestion period. The complicating factor, however, is dissolution of the water soluble salts which precipitate during the acid digestion (salting out process). For example, dissolution of 10 grams (453, 6 gms equal 1 pound of soil) requires an ultimate dilution of 250 ml to dissolve the water-soluble salts. A proportionately greater dilution is needed for larger samples. Obviously, the Pu-239 activity in a workable aliquot of an infinitely large dilution would be barely detectable even on hot samples. It seemed destrable therefore. To develop a method for separating the Pu compounds from the hulk of the soil. Flotation agents were tried on the premise the heavter plutonium bearing particles would separate from bulk soil by gravity centrifugation. The so-called soils, however, apparently had components equal or greater in density to plutonium compounds since 95% of the material was deposited in the bottom of the centrifuge cone. Pernaps the soils were heavy dense volcanic ash,

Following the flotation procedure a tracer and tracer-free partial dissolution of the soils was attempted and the results were highly suc-

cessful. In this method, eight 50 gram samples from Clean Siste II and III soil throwout collections were treated in a manner similar to total dissolutions of small soil samples except the reaction was stopped after approximately one fourth (30 minutes of dissolution time required) of the soil was dissolved. Approximately 50 ml of f-HNO<sub>3</sub> and saturated  $H_3$  BO $_4$  was added and the mixture boiled to wet dryness. Hydrochloric acid additions with boiling were repeated until the HNO, was destroyed. Care was taken to avoid excessive foaming and swelling of the heavy scum which appeared at low volume. The mixture was transferred to a large sized poly bottle with HCl washin' and diluted to the half full mark with H2O. Hydroxylamine -HCl, CHCl<sub>3</sub> and cupierron reagent were added, the mixture stirred vigorously, and centrifuged to separate the phases. Approximately 95% of the now dark CH<sub>3</sub>Cl<sub>3</sub> layer was removed with a transfer pipet, care being taken not to disturb the interface scum. The extraction was repeated until the CHCl3 layer was colorless (usually required 4 to 6 extractions). The extracted collections were evaporated at low heat to a heavy sludge (light flaming occasionally occurred in the sludge). Dilute  $HNO_{\chi}$  (6N) was added to the sludge and the mixture boiled to a heavy black tar. If bumping occurred HCl was added. Nitric acid was added, the mixture boiled to wet dryness, and the procedure repeated with f-HNO, until the tar turned a black liquid. Perchloric acid was added and a resulting slow exothermic reaction allowed to go to completion. The solution was boiled until perchlorate salts precipitated. Most of the salts were dissolved by repeated boiling with aqua regia. Remaining salts were washed with

 $\rm H_2O$  and boiled in fuming HNO $_3$  -HCl. The solutions were combined and diluted to the mark in a volumetric flask with fuming HNO $_3$  and H<sub>2</sub>O.

The residue from each of the partial dissolutions was completely dissolved in a manner similar to that for the small soil samples. Results of the eight samples showed approximately 95% of the plutonium was extracted. Subsequent experiments, however, show that partial dissolution must be restricted to sample sections of 200 gms or less because of dilution and extraction limitations.

bottles. The cap had been sealed with tape but most of the bottles had leaked rather badly. The pH of each sample was determined with a Beckman pH meter. The volume of the sample was measured in a graduated cylinder. All of the samples contained appreciable amounts of algae and dirt. A suspension aliquot of each sample was centrifuged and a stippled plate activity measurement made of each supernate. Aliquots from samples showing activity in the centrifuge supernate were filtered through a millipore filter and the filtrate analyzed by alpha pulse height analysis. The millipore filter was leached with successive additions of 0.1N HCl over a 48-hour period. Each leach was filtered through a new millipore and the filtrate analyzed by 2 TI counting of a stippled plate. The millipore filters from some of the leached samples were dissolved, purified, and analyzed by alpha pulse height analysis.

A separate suspension aliquot of several of the samples was ana—
lyzed by cupferron - CHCl<sub>3</sub> extraction at neutral pH and counted by
alpha spectroscopy.

All the analyses were for plutonium content and some for uranium.

Physical Sample Purification: The majority of the physical samples were purified, following dissolution, in a routine manner by the purification procedure given in Appendix B. Physical Samples with heavy dirt, however, required several cupferron-CHCl<sub>3</sub> extractions and alternate NaOH-Na<sub>2</sub>CO<sub>3</sub> NH<sub>4</sub>OH precipitations to free the plutonium compounds from excessive salt concentrations. In samples containing large amounts of Fe, a brown residue appeared on the resin purification column. This residue was dissolved during HCl elution. A trace of insoluble salts sometimes formed in the purified solution (tentatively identified as aluminum and titanium oxides) but contained no activity.

Electroplating: A rapid electrodeposition procedure was used to obtain from the purified sample a weightless, invisible deposit of plutonium on a platinum disc. A plating time of 10 minutes was usually required. The disc was 5 mils thick with a mirror finish, precut to 2.2 cm in diameter. The electrodeposition cell, designed by our laboratory, limited the plating solution exposure to a glass tower, teflon gasket, and platinum disc.

An excess of solution during the plating operation can result in as much as 70% loss of activity. The optimum volume of the plating solution was found to be approximately 4 ml which represents about 1/4

inch of liquid in the plating cell. In general, those samples having heavy dirt at first produced dirty plates. Changing lab reagents, plating solutions, and re-extracting the sample with hexone just prior to plating did little to improve plate quality. Cupferron -CHCl<sub>3</sub> extraction, baking the water extractant with successive additions of aqua regia, and resin column purification produced clean plates. Dirty plates also occurred when any residual organic material was not destroyed or when extraction was incomplete. Rinsing the plates with distilled water and flaming improved plate quality.

Stippled Plates: All samples with field monitor activity levels above a certain range were dissolved tracer free and a stippled aliquot measured for approximate activity. If the high activity value was confirmed, a dilution was made and tracer added to the aliquot. This prevented mis-match of Pu-239 activity and tracer so that one alpha peak was not swamped by the other in the pulse height spectra.

Odors: Last but not least was the problem of nefarious odors emanating from the dissolved large tissues and especially metabolism samples. Fortunately, the odors from tissues were all but eliminated by HClO<sub>4</sub> type dissolution and H<sub>2</sub>SO<sub>4</sub> refluxing methods. Boil-downs of urine and feces samples, though often produced a pungent odor in the lab. A resourceful chemist purchased a Buddha incense burner, and the resulting atmosphere was satisfactory to everyone's olfactory senses.

<u>Uranium Separation</u>: A uranium separation from plutonium was made in each plutonium analysis by a basic carbonate precipitation which carries plutonium. The uranium carbonate complex is soluble

under these conditions. However, if due to sample impurities, the uranium does not solubilize completely, it will not interfere with the measurement of Pu-239 (5.0 to 5.2 Mev integration limits) in an alpha pulse height analysis since uranium alphas fall at a lower energy. In the procedures outlined, the uranium separated in the plutonium procedure was not used for analysis. The uranium analysis was performed with sufficient sensitivity on another aliquot of the dissolved sample.

Assuming 1  $\mu$ g of natural uranium to be present in a sample containing 10 dpm Pu-239, an unexpectedly high ratio, the following sample activities can be expected:

Nuclide	cpm	Alpha Energy	
U-238	0.26	4,18 Mev	
U-235	0.01	4,40 "	
U-234	0.25	4.75 "	
Pu-439	3.47	5.14 <sup>#</sup>	
Pu-238	-	5,48 "	
Pu-236	-	5.75	

The closest Pu and U alphas, as is evident, are sufficiently separated in energy.

# 2.5 PREPARATION OF TRACER

The tracer employed in yielding plutonium isotopes is Pu-236 (made by the d, n reaction on highly purified U-235). The Pu-236 was prepared in a cyclotron irradiation and chemically purified at Traceriab. Approximately 20,000 dpm was aliquoted and pulse-height analyzed to determine isotopic purity and percent Pu-239, 240 pres-

ent, if any. Conections to subsequent samples were applied if any Pu-239, 240 was found in the tracer. Previous experience has shown that on high purity Pu-236, the ratio of Pu-239/Pu-236 is about  $1 \times 10^{-5}$ . The importance of any correction depends on the Pu-239, 240 activity in the sample analyzed. Once the purity of the tracer had been established, two stock solutions were standardized at about 400 and 20 dpm/ml. A choice of stock for each enalysis depended on the anticipated activity of the individual sample.

Ionic Pu-236 tracer has shown a tendency to polymerize and/or form oxides on standing or in the presence of trace quantities of organics. This can result in incomplete equilibration with other Pu radioisotopes and loss of yield. Preparation of Pu-236 standards from a concentrated stock solution, therefore, included an HClO treatment to destroy organics and solubilize all tracer activity.

Residual amounts of the acid were left in the standard solutions to hold the tracer in a soluble form.

The tracer was standardized by isotopic dilution and exhaustive electrodeposition. In both methods, a suitable aliquot was withdrawn from stock, electrodeposited on a platinum disc, and counted. In exhaustive electrodeposition the plating solution was reduced in volume and any remaining plutonium again electrodeposited. This process was repeated until further electrodeposition produced no change in disc activity. Summation of the electrodepositions gave the tracer concentration. Four to eight determinations were normally averaged to yield a final value. Concentrations were usually determined to plus or minus 2%.

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In isotopic dilution a spike of National Bureau of Standards stock solution\* (99.97% pure) was added, for yielding, to nine aliquots of the purified Pu-236 stock solution. The spike and tracer were equilibrated by evaporation with  $\rm H_2SO_4$  and  $\rm HCIO_4$  and electrodeposited on the platinum disc. The plated samples were counted and the Pu-236 concentration calculated after Pu-239 yielding.

Exhaustive electrodeposition gave an average concentration of  $25.0 \pm 0.38$  dpm/ml Pu-236 for four aliquots. Isotopic dilution gave an average of  $25.7 \pm 0.26$  dpm/ml for nine aliquots. Experience has shown that the first method is susceptible to low results due to sequential handling losses. This point has been confirmed by standardization of the tracer using a combination of both techniques on the same aliquots of tracer.

It was anticipated that the Pu-236 tracer might change concentration over a period of time due to a combination of factors, primarily evaporation of the media and/or deposition of the tracer on the walls of the polyethylene storage bottle. To minimize this error, aliquots of the standardized stock solutions were added to several small polyethylene storage bottles and acidified with 6N HCl.

To insure that the accuracy of the tracer standardization was maintained, a set of two exhaustive electrodepositions was performed after five months. (See Tracer Standardization Procedure in Appendix B for method). The set of two platings had to agree within 2.5% and their average within 2% of the previous standardization.

or further platings and/or complete restandardization was necessary.

<sup>\*</sup>An analysis of the NBS standard (listed as 99.97% pure) on our Mass Spectrometer gave the following isotopic composition; 94.386 weight % Pu-239, 5.271 weight % Pu-240, and 0.343 weight % Pu-241. The Pu-239, Pu-240 alpha disintegration rate of the solution was calculated from this data.

# 2.6 ACTIVITY MEASUREMENTS

Counting: Each plutonium sample was electroplated on a S-mil platinum disc (for best peak resolution) and counted on an alpha pulsa height analyzer. The disc was ignited to remove any residual deposit, since resolution decreases proportionately with an increase in deposit thickness. In order to utilize existing equipment, the outputs from four Frisch Grid Chambers were connected to one multichannel analyzer by dividing the full range of the analyzer (255 channels) into quadrants of 64 channels each. The instrument controls were adjusted so that the sixty four channels covered the entire energy range of the plutonium isotopes. The amplifier controls were adjusted to cover the range 4.5 to 6.0 Mev which included Pu-238 (5.49 Mev), Pu-239 (5.15 Mev), and Pu-240 (5.15 Mev) and Pu-236 (5.75 Mev) tracer. This amplifier gain setting gave a scale factor of approximately 37 Kev per channel, and each isotope present was registered over a spread of about ten channels. Optimum gain settings discriminated against activity energies outside the Pu-236, 239 energy region. The result was a pure spectra of Pu-236, 238, 239. Even slightly dirty plates showed minimum straggling in the valley region of the spectrum. Occasionally a small alpha 1 tak from the U-232 decay daughter of the tracer was seen if a sample was recounted several weeks after chemical decontamination. However, the U-232 peak, located at an independent energy, in no way interfered with the analysis. Samples with low yields or poor spectra were reworked.

A disposable metal collimating ring, surrounding each sample

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disc, was used with each sample to preclude the counting of degraded alpha particles. Some loss in counting efficiency resulted but was offset by improved peak contours and distinct separation of alpha energy peaks. The resolution (full width at half-maximum) of the four Frisch-Grid of ambers including disp collimation was 0,88% at 5,15 Mev. The alpha peak counting officiency was approximately 35%.

The counting time for an unknown sample was determined by the isotope having the lowest activity. A total collected count of this isotope which gave a standard statistical counting error of 10% was considered satisfactory. However, if possible, a total collected count giving a standard error of 3% was obtained. The tracer yield was determined to an error of 3%.

Table 2.2 indicates the variables involved in the choice of counting time. Counting times of much greater than 1000 minutes were not economically justified. Also samples with adequate yields snowing no Pu-239activity after 40 minutes of counting time were reported as such with a standard error for the background count of the instrument.

TABLE 2.2 TOTAL COUNTING TIME REQUIRED TO GIVE LISTED

1	ERROR*		٠.	
·	3%	5%	10%	
Activity	Error	Error	Error	
10 cpm	111	40	10	
1 "	1110	400	100	
0,1 *	11100	4000	1000	

<sup>\*</sup>Background for alpha pulse analyzer is virtually negligible, ranging from 0.006 to 0.01 cpm.

Determinition of Geometry: A geometry factor is used to convert the observed counting rate of an unknown sample to absolute disintegrations per minute (dpm). The geometry factor is defined as the observed corrected counting rate divided by the absolute disintegration rate of a calibrated plutonium standard source. The observed counting rate of a sample always contains the following inherent losses.

- (1) 2 TT Geometry: The geometry is restricted to 2 TT steradians by virtue of a flat disc mounting.
  - 2) Collimation Loss: Described in counting section.

Analysis of Pulse Height Data: A short run of a calibrated plutonium standard source was made before and after the analysis of an unknown plutonium sample. This procedure gave an evaluation of the counting geometry and resolution of the instrument including any channel shirting of the alpha peak, in this interval of time.

After counting a sample, a Pulse Height Graph sheet was used to make a graphical plot of the data as needed. Channel counts on the ordinate were plotted versus channel number and/or energy on the abscissa. From this graph, together with the data tape, an analysis of the isotope peaks was carried out utilizing the attached processing sheet. In selecting the group of channels representing each isotope peak, the following points were considered:

(1) Width of Isotope Peak Base: Since each sotope peak which represents a single alpha particle energy is theoretically of the same contour, its base will cover the same number of channels with only the neight of the peak differing in each base. The peak

contour may be represented approximately by a Gaussian distribution curve as described in the references.

- (2) Low Energy Tail: The low energy tail of each isotope peak will continue down to zero energy. However, no counts less than 1% of the peak height are added to the totalized peak count. The totalized count of the calibrated plutonium standard source is evaluated in the same manner.
- (3) Background: On low counting samples, it is necessary to correct for background. This correction is compiled from a statistical summation of consecutive background determinations and is subtracted empirically in the peak energy region, from a knowledge of the isotopes present and of their peak contours.
- (4) Peak Resolution: The resolution is determined by the width of the peak contour and will determine the possibility of detecting isotope peaks in close proximity. Resolution may be mathematically denoted as the peak width at its mid-height divided by the peak energy, each value expressed in the same energy units. The resolution for selected analysis is calculated in order to determine the amount of instrument drift. The desired resolution was always better than 1.5 percent.
- in peak resolution during an analysis. It is evidenced by a broadening of the peak contour and, in extreme cases, will give rise to excessive peak overlap. The amount of drift will indicate the degree
  of instrument stability during an extended analysis. Repeat analyses
  will be indicated if the amount of drift impairs good peak resolution.

Alpha Spectroscopy Quality Control: The reliability of the sipha pulse height analysis system must be checked periodically regardless of the observed reliability. Day to day standards of instrument of articles are measured by observing the before and after runs of the standards (see counting reference for details). The width and energy location of the standard and sample isotope peaks base must correspond. A drift of more than 1 percent at five Mev between the two is cause for instrument repair.

In addition, a background spectrum must be taken at least once per month (and more often if contamination is suspected). The background, taken in the region of 4.0 to 6.2 MeV, must match the sample isotope peak base, and shall not exceed five counts per hour.

Any excess is reason for determination of the cause of the background and removal of the source.

The counting efficiency of the instrument is checked monthly by counting a caribrated three-peak alpha source containing Pu-237, Am-241 and U-233. The source is counted in the energy region of 4.5 to 5.7 MeV.

The individual peak efficiency is checked by integrating the peaks and comparing with the assigned isotopic dpm values of the source. A divergence of more than 1 percent from the assigned values on subsequent efficiency checks is cause for further investigation and/or maintenance as necessary.

# 2.7 CALCULATIONS

<u>Plutonium Isotopes</u>: The results of the alpha pulse-height analyses are presented on printed tape. A graphical plot of a typical

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spectrum is illustrated in Figure F.4. The energy calibration line was calculated from the pre-and post-counting energy calibrations of the counting chamber. A summation was made of counts under each isotope peak present. These counts were corrected for low energy tail, background, peak resolution, and instrument drift. The plutonium content of the sample was calculated by:

Plutonium-239 and Pu-240 could not be calculated separately as their alpha energies were too close to resolve with a Frisch-Grid Chamber.

The counting efficiency of each Frisch-Grid Chamber was measured, using a high precision alpha standard, and it was not necessary to calculate a yield separately to determine the plutonium content. However, he yield was always determined as a quality control measure in order to assess the efficiency of the chemistry procedure. The yield was calculated by:

<u>Uranium</u>: The results of the fluorimetric analyses are presented as milliamps on the Jarrell-Ash fluorimeter. Milliamps are converted to  $\mu g \ U_3 O_8 / total$  sample by:

$$U_3O_8$$
 (µg/ total sample) = (Ma sample - Ma bkg.) x Cf % yield x aliquot factor

# CHAPTER 3

# DATA PRESENTATION

# 3.1 DISCUSSION

Tables E.1 through E.13 contain the plutonium and uranium data for the Physical, Biological, and Quality Control Samples. Also included are counting time, yield, and rework information. The tables are a summation of all the bi-waekly reports plus new data generated since the last bi-weekly report. New data are starred. A key to the sample types precedes the tables.

# 3.2 BIOLOGICAL DATA

Tables E.5 through E.7 contain the plutonium and uranium data for the Biological samples. Tables are listed by animal type. The data are listed by tissue type and number. All plutonium values are reported as dpm Pu-239, 240/total sample. All uranium values are reported as  $\mu g U_3 O_8$  total sample. A counting error in dpm was assigned each Pu-239, 240 value and the data presented in orders of magnitude. The error assignment as well as the base value is in terms of the given power. Pertinent information relative to the analysis appears in the remarks column. Zero or negative values were included, accompanied by the counting error; but never was the positive numerical value of the latter less than the negative value. In most instances, the statistical precision of the data meet requirements set by the referee team. In general, most of the yields exceeded 50%. A few samples with low yields were either reworked

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or counted for longer periods to assure good statistical accuracy.

Uranium data were derived from 10% aliquots and are listed opposite
the Pu-239, 240 value for a given sample.

The dog and sheep data are given for the Double Tracks or Clean Slate

II events, the burro on Double Tracks only. Since exact animal location was not available, no attempt was made to correlate laboratory analytical data with field information. Comparisons were made among tissue types, metabolism samples, animals, and events for Pu and U content as follows:

- (1) Dog Tissues: Most of the Pu was concentrated in the G.I. Tract, lung, trachea, and nasal mucosa. On the average, the G.I. Tract was a factor of approximately  $10^2$  greater than any of the other three tissues. The activity in the four tissues ranged from <1 to1.63  $\times$   $10^4$  dpm. The U concentration in most tissues was less than 1  $\mu$ g per sample and showed little tendency to follow the Pu. For example, the tissue with the highest U value, 2.88 $\mu$ g in the lung of Animal 1020, contained only 3.36 dpm of Pu.
- (2) Sheep Tissues: The Pu was found principally in the trachea, G.I. Tract, Lung, and Nasal Mucosa. In general, where the Pu was distributed among these tissues, it was divided fairly equally. However, more of the lung samples contained appreciable amounts of Pu. Activity values in the sheep tissues mentioned ranged from < 1 to 5.45 x 10<sup>2</sup>. Similarly to the dog samples. U concentration was usually less than 1 μg and did not necessarily follow the Pu. The tissues of three samples, 2050, 2052, 2127, were the highest in uranium but low in plutonium.

- (3) Burro Tissues: Most of the Pu was concentrated in the lungs but appreciable amounts were detected in some liver, hilar node, and G.I. tract samples. Also one bone, kidney, and trachea showed appreciable Pu. The lung-to-liver ratio ranged from near unity in a few samples to 55 in Sample 3043. Very few uran-tum analyses were run on burro tissues. All were less than 1 ug.
- (4) Metabolism samples: All the analyses in this category were performed on sheep eliminations. Most of the metabolism data showed high values relative to the animal tissues. The data ranged from approximately  $10^1$  to  $10^4$  dpm per sample. Neither urine nor feces values necessarily dominated the Pu content of a given animal sample. Uranium analyses were not required of metabolism samples.
- (5) Arimals: The G.I. tract of the dog samples and burro lungs dominated the Pu content of all animal tissues. Uranium data was too low, in most instances, to make a significant comparison among animals. Average values for tissues containing the largest amounts of Pu are given below in Table 3.1 in dpm per sample.
- (6) Event: Only three sheep lung samples from Clean Slate II contained appreciable amounts of Pu. All other large values were in Double Tracks data.

# 3.3 PHYSICAL DATA

Tables E.1 through E.4 contain the plutonium and uranium data for the physical samples. Tables are listed by event. The data are listed by sample position in the test pattern with corresponding

TABLE 3.1 ANIMAL TISSUES WITH GREATEST PU CONTENT

	<u>Do</u> g	Sheep	Burro 2.0 x 10 1 *
Bonc			
Kidney			$3.6 \times 10^{1*}$
Liver			$5.3 \times 10^1$
Lung	$4.8 \times 10^{1}$	$9.0 \times 10^1$	$5.52 \times 10^2$
Hilar Node			5.6 x 10 <sup>1*</sup>
Trachea	$2.4 \times 10^{2*}$	$2.2 \times 10^{1*}$	$1.7 \times 10^{14}$
G.I. Tract	$2.7 \times 10^{3}$	$2.4\times10^2$	$3.2 \times 10^{14}$
Nasal Mucosa	$9.8 \times 10^{1}$	$1.6 \times 10^{2*}$	

<sup>\*</sup>Based on one or two analyses only. All other values average of several analyses.

TLW collection and analysis number. Pretest and offsite data appear at the end of each table. The statements concerning data reporting and counting precision under the previous section apply to the physical data. Yields were rarely below 60% except for a few samples with heavy dirt. Uranium analyses were performed on aliquots to 10% of the sample and are listed opposite their Pu-239, 240 counterparts. Pertinent information relative to the analysis is footnoted. The ratio of Pu-239, 240 by radiochemical analysis to the field monitor value is given in the last column. To be meaningful, any radiochemical or field monitor value from 0-to-1 dpm/total sample was arbitrarily assigned a 1-dpm value for the ratio calculation. In such instances, the ratio was preceded with a computer approximate sign (CA).

The field positions of physical samples were well documented and an attempt was made to correlate some of the data by event as follows:

- (1) Doubletracks: In general, deposition contours determined from radiochemical analyses of aluminum collectors and deposition films agreed with those established by alpha field surveys. A moderate amount of activity was detected, however, outside the contour of the P and R arcs as far as Station 068 on the right side. The Casella and Andersen disc numbers showed mixed results with regard to their internal system. In some instances, the plutonium content decreases progressively with successive stages, but often the second and third stages have values higher than the first. Uranium-toplutonium ratios were somewhat erratic and were higher than expected The uranium content does not follow the plutonium in many cases, particularly in values from far out arc locations. Radiochemical-tofield monitor plutonium ratios were within a factor of unity in general with deviation orders of magnitude in either direction for many analyses. Particularly noticeable were the high ratios of some aluminum collectors and deposition films.
- deposition films and agreed in general with contours from alpha survey readings. Moderate activity appeared outside the contour on Arc H as far as Stations 024 on the left and 038 on the right. The Casella's and Andersen's showed most of the activity to be concentrated in the first impactor stage followed by a decrease of activity with successive stages. Uranium-to-plutonium ratios were erratic, but, in general, lower than those in the Double Tracks event. Radio-chemical-to-field monitor plutonium ratios were usually within a factor of unity. A few exceptions were apparent, but orders of

magnitude deviations were much less frequent than in Double Tracks.

- (3) Clean Slate II: Deposition contours determined from Arcs B to L deposition films and aluminum collectors are consistent with alpha survey readings. Deposition data from Arcs E and F showed moderate activity outside the contour at stations 014 and 090 respectively. Activity data from the Casella's and Andersen's resembled that of the Clean Slate I event. Also uranium-to-plutonium ratios were similar to those of Clean Slate I. Radiochemical-to-field monitor plutonium ratios were similar to Clean Slate I.
  - (4) Clean Slate III: Deposition contours were determined from Arcs B to L deposition films and were consistent with alpha survey readings. Moderate activity appeared outside the contour on Arc B as far as Station 100. The Casella and Andersen data followed the pattern of Clean Slate I and II.

Uranium-to-plutonium ratios were similar to those of Clean Slate I and II. Radiochemical-to-field monitor plutonium ratios were consistent within a factor of five of unity, except in Arcs E to L and some soil fractions, both of which contained values ranging from 2 to 500.

#### 3.4 MISCELLANEOUS DATA

Estimated Activity Expenditure: Table E.8 contains a data listing, by arc location, of estimated plutonium activity losses of project 2.6C "A" samples. "A" samples refer to those Casella's and Andersen's whose first and second stages were combined for particulate analyses—and later transferred to this project for radiochemical analysis. Therefore, to obtain a better value for the

"A" samples, each value in Table E.8 should be added to its counterpart in Tables E.1 to E.5.

Distilled Water Samples: Tables E.9 through E.11 contain the plutonium and uranium data for the distilled water samples. Tables are listed by event and data by arc location. All Pu values are given as dpm/total sample volume. In general, leach filtrate values decreased with successive leaches except for the last leach which spanned a greater time period. Particularly interesting, with respect to plutonium solubility, are the high values for aliquots in which Pu was extracted at neutral pH. The residues of five glass bottles in which water samples were stored were found to cor.sin 1 to 22% of the activity of the original contents.

Tracer Standardization: Table F.12 lists the results of isotopic dilution and exhaustive plating analyses of solutions containing Pu-239 and/or Pu-236. Good agreement among analyses is apparent from the standard deviation column.

#### 3.5 CONTROL DATA

Biological: Table E.13 contains Roller Coaster plutonium quality control data listed by Rochester collection number with corresponding TLW analysis number. All the data are reported as dpm

Pu-239, 240/total sample. Also included are yield and counting time for each analysis. The samples were blanks or spiked samples, and the data indicate the latter since few show less than 99 dpm/total sample.

Table E.14 contains TLW internal plutonium and uranium quality control data listed by TLW analysis number and sample type.

All the data, excepting three Pu values, are near the detection limits of the measuring instrument. The base values for the three

exceptions show less than 1 cpm and are not considered significant.

Simulated blanks of beef liver and hamburger were analyzed, early
in the program, for plutonium content and found to contain less
than 1 dpm. The data was not recorded since it provided little
useful information.

Physical: Table E.15 contains Roller Coaster plutonium control data listed by arc location with corresponding TLW collection and analysis number. All values are the results of investigation of sample aliquoting by partial dissolution—extraction methods or analyses of solutions forwarded by the Roller Coaster analyses team. All the data are reported as dpm Pu-239, 240/total sample for soil samples and dpm/ml for solutions. Partial dissolution would appear to a valid procedure based on the small amounts of activity left in the residue. The solution activity range from 0.01 to 4.84 x 10<sup>3</sup> dpm/ml.

Table E.16 contains TLW internal plutonium and uranium quality control data listed by TLW analysis number and sample type.

All the laboratory blanks were near the detection limits of the measuring instrument. Analysis of a sample (previously analyzed in our mass spectrometer) for Pu-239, 240 content, using our low and high level Pu-236 standards, reconfirmed the Pu-236 standardization values. Mass spec values are given for comparison in the remarks column.

## 3.6 DATA SUMMATION

Biological: Table E.17 contains a tabulation of all the biologi-

cal analyses. Data are listed by animal and tissue type. A total of 744 plutonium and 87 uranium analyses had been performed at the conclusion of the project.

Physical: Table E.18 contains a tabulation of all the physical analyses. Data are listed by event and sample type. A total of 2607 plutonium and 598 uranium analyses had been performed at the conclusion of the project.

## CHAPTER 4

## CONCLUSIONS AND LABORATORY RESULTS

# 4.1 CONCLUSIONS

The conclusions on the data generated in 5.2/5.3b projects are limited to sample processing and surface inspection of the results, since, in regard to the latter, it is the function of the evaluation team to inverpret the significance of the data. The following are applicable to our projects:

- (1) All samples received for analysis have been inventoried and accounted for.
- (2) Adequate procedures for sample processing and accurate analysis were developed.
- (3) Facilities and personnel were fully utilized to maintain the desired production schedule.
- (4) The requirements of the referee team were not unduly restrictive and have been met in most instances.
- (5) Tracer techniques were employed and found to be highly satisfactory.
- (6) Most of the plutonium in the tissue biological samples was concentrated in the G. I. Tracts of the dogs and lungs of the burros. Metabolism samples from sheep eliminations ranged from approximately  $10^1$  to  $10^4$  dpm per sample.
  - (7) Double Tracks animals showed higher Pu content than Clean

Slate II animals. Only three sheep lung samples from Clean Slate II contained appreciable amounts of plutonium.

- (8) Plutonium data of the physical samples is fairly consistent with that of alpha field surveys and field monitor values with some deviations noted.
- (9) Uranium values are erratic in some instances and do not necessarily follow the plutonium values.
- (10) Casella and Andersen samples are fairly consistent in showing an activity decrease with successive stages. Deviations are occasionally noted in the second and third stages where values are higher than expected.

## 4.2 LABORATORY RESULTS

The results listed below pertain to our laboratory experiment with the biological and physical samples. Several improvements were made and others are suggested.

- (1) Biological samples were difficult to identify since fluids from the animal had often obscured the writing on the paper-type tag. The sample should be doubly wrapped in poly bags by sealing the sample in one bag and covering with another. The outside bag should then be labeled with a Dymo punch. The double bag would also prevent samples from freezing together, necessitating complete thawing before processing.
- (2) Many of the samples were heavy in iron content and a rapid method for removing this element was needed. Experiments with a nitrated ion-exchange column resulted in a procedure superior to hexone extraction in all phases.

- (3) The large biological samples usually required a lengthy thawing period before they could be cut, even with an electric knife. It is desirable to eliminate the thawing process entirely since it consumes an analyst's time; a special heating setup is required, spread of contamination is a risk, and overpowering odors develop. An electrically heated knife is now on the market and all reports indicate it would cut the frozen samples easily.
- (4) The urine samples were collected on Kimpac, which contained large amounts of dirt. The dirt seems to have an affinity for the plutonium in the urine, and extraction procedures had to be employed to obtain a good yield. A method of collection to eliminate the dirt would be desirable.
- (5) Several samples had to be reworked to obtain better yields. At first, all the tailings were scavenged for missing plutonium; however, experience showed 95% of this activity was always in the aqueous discard of the first extraction. It is not economically feasible to spend time scrounging for the remaining 5%.
- (6) The biological samples occupied relatively large amounts of space and had to be kept in a frozen state over a long period of time. An oversized, walk-in freezer, adequately lighted, is recommended for easy access to samples and sequential handling.
- (7) Fluorimetric analysis for uranium may be performed directly on the dissolved sample. However, experience has shown quenching occurs if the sample is not chemically pure. Three extractions at the start, to remove interfering ions, is recommended.

- (8) Uranium, as plutonium, is lost during chemical processing and should be yielded. Use of U-233 tracer, or analysis, in duplicate with one part containing a spike of uranium standard is suggested. The latter was employed in this project.
- (9) Acid dissolutions in large volume are extremely corrosive to all types of metal hoods and exhaust systems. Even coating the metal with an acid resistant paint is only a temporary cure. Several other types of hoods and blowers have been investigated since the start of the project, and a polypropylene system with an internal scrubber appears to offer the best performance. Resistance to normal hot plate temperatures and all acids, including HClO<sub>4</sub>, is touted by vendors of these hoods and blowers. The scrubber system is needed to remove noxious acid fumes and to wash out potentially explosive collections of nitrate and perchlorate dust mixtures. A movable safety shield should be installed in each hood, wit above normal ventilation, to remove the copious quantities of acid fumes being generated.
- (10) Near the midpoint of the project a computer program was developed to incorporate additional information such as samples, yields, counting times, etc. This method of reporting provided for rapid transscription of new data to tabular form, reducing the delays of typing, proofreading, and copying. It is recommended to institute the computer program at the start to save the typing effort.
- (11) The bonus benefit of the cupferron CHCl<sub>3</sub> portion of partial dissolution procedures may be its application to water and urine analyses. The cupferron CHCl<sub>3</sub> extraction process, primarily independent of sample pH, may be the best method for determining soluble plutonium content of a water-algae-dirt solution.

- the validity of the data. A plate which has a scum on it, is bent, scordined, or scratched may distort the alpha spectrum to a point where the results are marginal. In addition to the radiochemical purification techniques mentioned earlier, it was determined that good plate quality most often resulted from proper flaming. The plate should be thoroughly washed with triple distilled H<sub>2</sub>O and flamed at high temperature over a Fischer burner for two minutes and the process repeated once. Flaming with methanol is not desirable, as it will sometimes produce plates with a white scum.
- (13) Dissolution of the biological samples as rapidly as possible is recommended since freezer failure is always a possibility. Samples can be stored for purification at a later date.
- (14) In future projects of this type, it might be expedient to analyze biological samples in order of animal number and physical samples by arc location. This should reduce the many man hours that were expended in cross referencing animal numbers and arc locations with TLW field and analysis numbers.
- (15) Near the end of the project an opportunity arose to compare plutonium spectra of platinum with stainless-steel-mounted samples. Two hundred stainless steel plutonium mounts were counted, using collimation and a plutonium standard mounted on stainless. In general, the stainless mounts showed well defined peaks but a broader base, indicating a loss of resolution. Also, the Pu-236 and Pu-239 alpha peaks of the stainless samples occurred at a lower energy, showing a 3 to 7 channel shift downward for each isotope.

(16) The final measurement in the radiochemical analysis for Pu-239 is taken from the alpha spectrum of the electrodeposited sample. The sample Pu-239 and Pu-238 activity must be matched to the Pu-236 tracer to prevent interference between peaks.

In tracer techniques for Pu-239 analysis, the accuracy of the results are only as accurate as the standardization of the Pu-236 tracer. It is recommended that the value of cpm per dpm per unit volume of tracer be accurately and precisely determined, for the detector to be used in obtaining alpha spectra of the final sample plates. It is also recommended that specifications be set up for quality control and preventative maintenance procedures for the detector and electronic equipment; and methods of spectra interpretation. These should be rigidly designed to assure good resolution of the alpha puaks. Detector backgrounds should be rigidly controlled at a predetermined level by frequent detector background runs and limiting the total amount of activity allowed in the detector. Recounting or rework of the sample should be done, as required, to adhere to the specifications.

# APPENDIX A DETAILED LOCATION OF PHYSICAL STATIONS

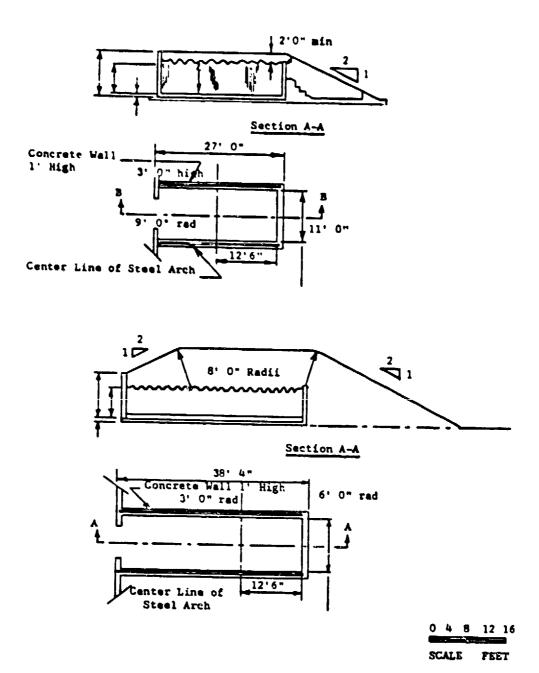
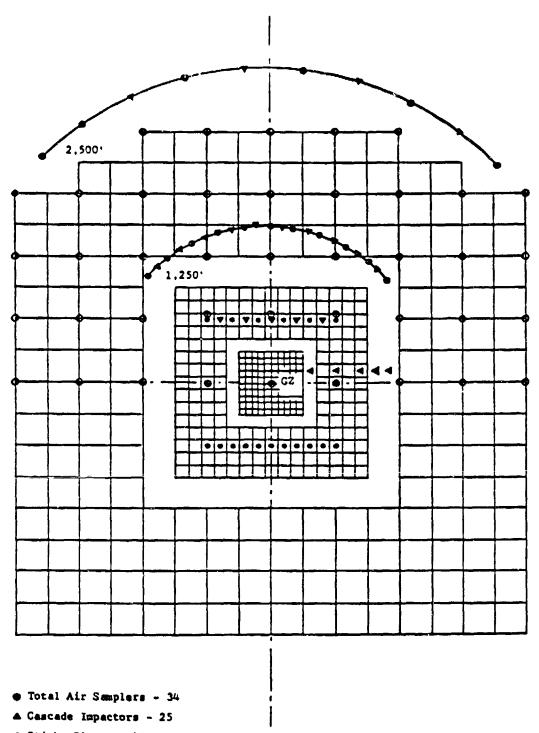


Figure A.1 Clean State igloos.

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Sticky Plates - 40

Figure A.2 Fixed syrface instrument array (-2,000 to +2,500 feet).

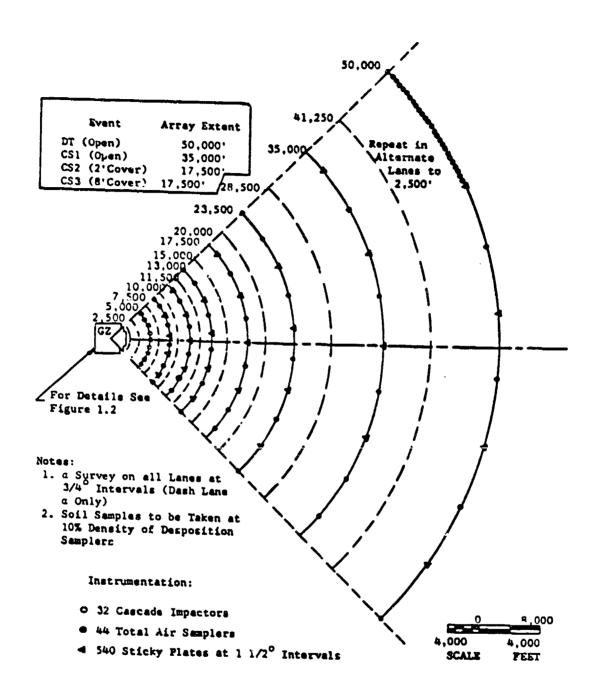


Figure A.3 Fixed surface instrument array (+ 2,500 to + 50,000 feet).

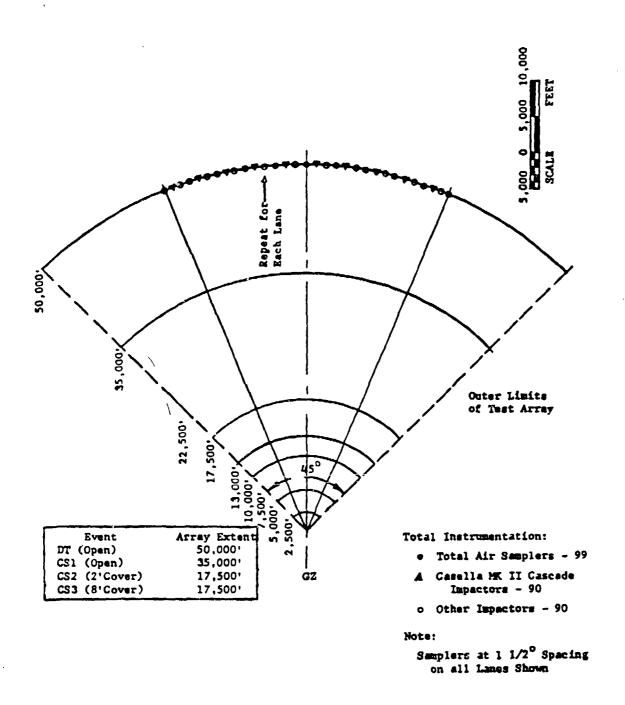


Figure A.4 Movable surface instrument array (+ 2,500 to + 50,000 feet).

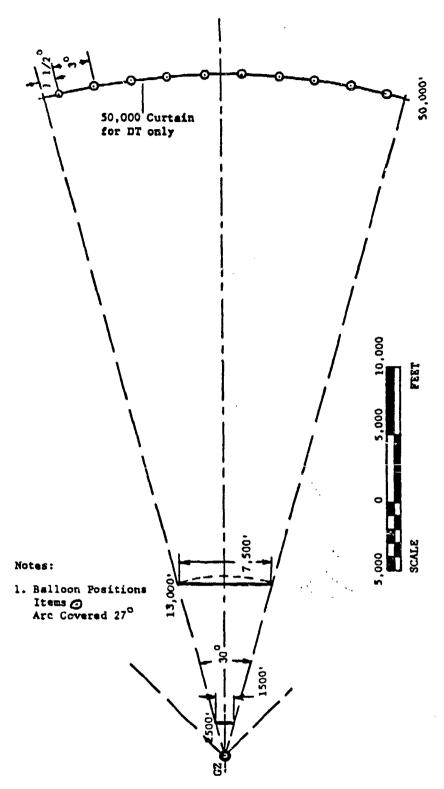


Figure A.5 Movable balloon-supported instrument array.

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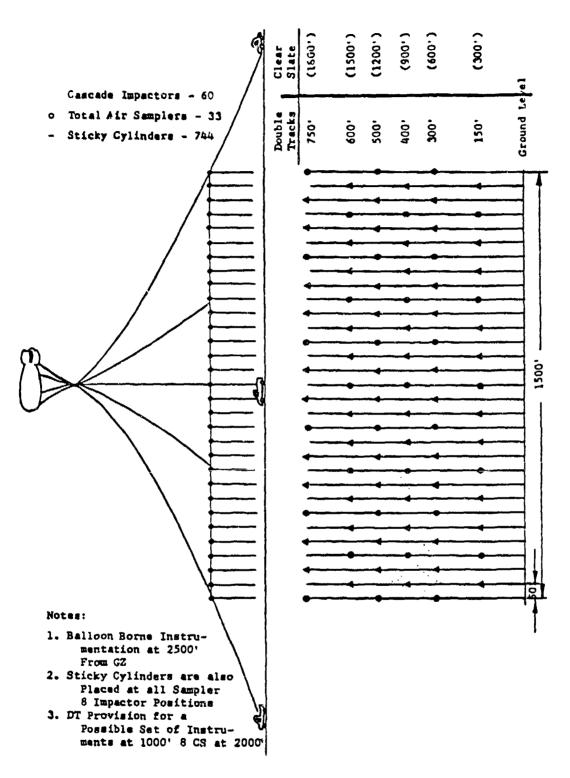


Figure A.6 Details of balloon curtain at 2,000-foot radius.

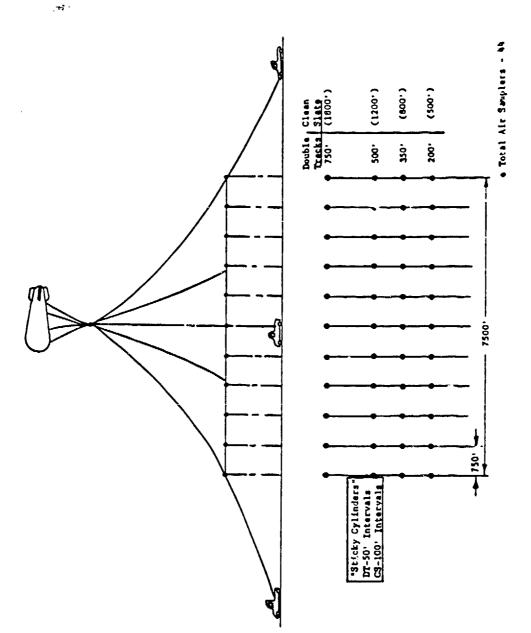


Figure A.7 Details of balloon curtain at 13,000-foot radius (never operable).

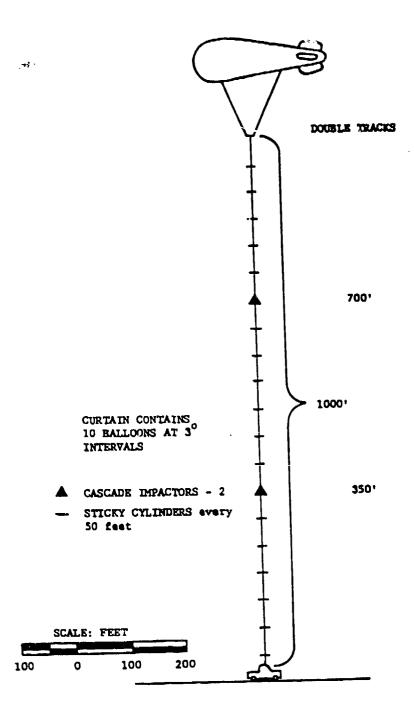


Figure A.8 Details of balloon curtain at 50,000-foct radius.

# APPENDIX B RADIOBIOLOGICAL, RADIOCHEMICAL, AND PHYSIOCHEMICAL PROCEDURES FOR PU<sup>239</sup>, PU<sup>240</sup>, AND URANIUM IN VARIOUS SAMPLES

## Pu-239 DISSOLUTION PROCEDURE

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- l. Place Casella impactor glass disc in an appropriate size teflon beaker (Note a).
  - a. If the sample monitors <4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6NHNO3, a small aliquot is pipetted into a 40ml centrifuge cone, and an appropriate amount of tracer is added.
- 2. Add enough furning HNO3 to wet all of the sample. Heat on a hot plate until the sample has dissolved.
- 3. Remove from hot plate and add about 6 ml 78 percent HClO4 for every 100 ml fuming HNO3 added in step 2. Heat on hot plate until exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of fuming HNO3, pouring acid carefully down wall of beaker (note b).
  - h. At times the reaction ceases and the solution turns black. This is caused by the supply of fuming HNO<sub>3</sub> becoming exhausted and is remedied by addition of more acid.
- 4. Remove the glass disc with teflon forceps and rinse with  $1N \ HNO_3$   $1N \ HF$  adding washings to teflon beaker (note c).
  - c. If a white residue remains on the disc, rinse twice more with the <u>IN HNO<sub>3</sub>-IN HF</u> solution adding washings to teflon beaker. If the sample does not contain any insoluble material at this point omit steps 5 through 8.
- 5. Add 10 ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable

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amounts of dirt, repeat HF treatment at least once.

- 6. Add 4 ml saturated H<sub>3</sub>BO<sub>3</sub> and boil for 3 minutes.
- 7. If residue remains, wash with portions of warm 6N HNO, until it dissolves.
- 8. Transfer any undissolved residue to the teflon beaker quantitatively with HNO<sub>3</sub> washes and repeat steps 5, 6, and 7.
- Transfer the solution to a 40-ml centrifuge cone and proceed with step
   1 Pu-239 Purification Procedure.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 Purification Procedure.

FILTER\*

- Place the sample or aliquot in a suitably sized pyrex beaker (note a)
  or teflon beaker if sample is small.
  - a. If the sample monitors < 4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6N HNO<sub>3</sub> a small aliquot is pipetted into a 40 ml centrifuge cone, and an appropriate amount of tracer is added.
- Add enough furning HNO<sub>3</sub> to wet all of the sample. Heat on a hot plate until the sample has dissolved.
- 3. Remove and add about 10 ml 78 percent HClO<sub>4</sub> for every 100 ml fuming HNO<sub>3</sub> added in step 2. Heat on hot plate until an exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of fuming HNO<sub>3</sub>, pouring acid carefully down wall of beaker (note b).
  - b. At times the reaction ceases and the solution turns black. This is caused by the supply of fuming HNO<sub>3</sub> becoming exhausted and is remedied by addition of more acid.
- 4. Transfer the contents of the beaker to a teflon beaker (note c) by means of a transfer pipet. Wash the beaker with several 6N HNO<sub>3</sub> washes, scrubbing the sides and bottom with a polyethylene policeman. Perform at least two washes with 3-ml aliquots of 1N HNO<sub>3</sub> ~ 1N HF.
  - c. If started in teflon, omit step 4 but add a few ml HNO<sub>3</sub>. If the sample does not contain any insoluble material at this point, omit steps 5 through 8.
- 5. Add 10 ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable amounts of dirt, repeat HF treatment at least once.
- 6. Add 4 ml saturated H<sub>2</sub>BO<sub>2</sub> and 8 ml HNO<sub>3</sub> and boil for 3 minutes.
- 7. If residue remains, wash with portions of warm 6 N HNO3 until it dissolves.

- 8. Transfer any undissolved residue to the teflon beaker quantitatively with HNO<sub>3</sub> washes and repeat steps 5, 6, and 7.
- 9. Transfer the solution to a 40-ml centrifuge cone and proceed with step 1
  Pu-239 Purification Procedure.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-Z39 Purification Procedure.

SAMPLER DISC\*

- Place Andersen sampler glass disc in an appropriate size teflon braker (Note a).
  - a. If the sample monitors <4000 alpha cpm, add Pu-239 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6N HNO<sub>3</sub>, a small aliquot is pipetted into a 40-ml centrifuge cone, and an appropriate amount of tracer is added.
- Add enough fuming 103 to wet all of the sample. Heat on a hot plate until the sample has dissolved.
- 3. Remove from hot plate and add about 6 ml 78 percent HClO<sub>4</sub> for every 100-ml furning HNO<sub>3</sub> added in step 2. Heat on hot plate until exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of furning HNO<sub>3</sub>, pouring acid carefully down wall of beaker (note b).
  - b. At times the reaction ceases and the solution turns black. This is caused by the supply of fuming HNO<sub>3</sub> becoming exhausted and is remedied by addition of more acid.
- Remove the glass disc with teflon forceps and rinse with 1N HNO<sub>3</sub> 1N HF adding washings to teflon beaker (note c).
  - c. If a white residue remains on the disc, rinse twice more with the lN HNO<sub>3</sub> - lN HF solution adding washings to teflon beaker. If the sample does not contain any insoluble material at this point omit steps 5 through 8.
- 5. Add 10-ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable amounts of dirt, repeat HF treatment at least once.
- 6. Add 4 ml saturated H3BO3 and 8 ml HNO3 and boil for three minutes.
- 7. If residue remains, wash with portions of warm Aqua Regia until it dissolves.

- ð. Transfer any undissolved residue to the teflon beaker quantitatively with  $\mathrm{HNO}_3$  washes and repeat steps 5, 6, and 7.
- 9. Transfer the solution to a 40 ml centifuge cone and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- Place the sample or aliquot in a suitably sized pyrex beaker (note a)
  or teflon beaker if sample is small.
  - a. If the sample monitors <4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6N HNO3 a small aliquot is pipetted into a 40-ml centrifuge cone, and an appropriate amount of tracer is added.
- Add 1/3 volume furning HNO<sub>3</sub>, boil to dryness, and char. Repeat until
  only small amount of carbon is left (the sample will dissolve but not decompose in furning HNO<sub>2</sub>).
- 3. Remove from hot plate and add about 6 ml 78 percent HClO<sub>4</sub> for every 100-ml furning HNO<sub>3</sub> added in step 2. Heat on hot plate until exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of furning HNO<sub>3</sub>, pouring acid carefully down wall of beaker (note b).
  - b. At times the reaction ceases and the solution turns black. This is caused by the supply of furning HNO<sub>3</sub> becoming exhausted and is remedied by addition of more acid.
- 4. Transfer the contents of the beaker to a tellon beaker (note c) by means of a transfer pipet. Wash the beaker with several 6N HNO washes, scrubbing the sides and bottom with a polyethylene policeman. Perform at least two washes with 3-ml aliquots of 1N HNO = 1N HF
  - c. If started in teflon, omit step 4 but add a few ml HNO<sub>3</sub>. If the sample does not contain any insoluble material at this point, omit steps 5 through 8.

- 5. Add 10 ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable amounts of dirt, repeat HF treatment at least once.
- 6. Add 4 ml saturated  $H_3BO_3$  and 8 ml  $HNO_3$  and boil for 3 minutes.
- 7. If residue remains, wash with portions of warm 6N HNO3 until it dissolves.
- 8. Transfer any undissolved residue to the teflon beaker quantitatively with HNO<sub>3</sub> washes and repeat steps 5, 6, and 7.
- 9. Transfer the solution to a 40-ml centrifuge cone and proceed with step 1 Pur239 Purification Procedure.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H2O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-29 Purification Procedure.

- Place the sample or aliquot in a suitably sized pyrex beaker (note a) or teflon beaker if sample is small.
  - a. If the sample monitors < 4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6 N HNO<sub>3</sub> a small aliquot is pipetted into a 40-ml centrifuge cone, and an appropriate amount of tracer is added.
- Add 1/3 volume furning HNO<sub>3</sub>. Boil on a hot plate to dryness and char.
   Repeat once and take up in 1/3 volume furning HNO<sub>3</sub>.
- Remove from hot plate and add about 10 ml 78 percent HClO<sub>4</sub> for every 100 ml fuming HNO<sub>3</sub>. Heat on hot plate until an exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of fuming HNO<sub>3</sub>, pouring acid carefully down wall of beaker (note b).
  - b. At times the reaction ceases and the solution turns black. This is caused by the supply of furning HNO<sub>3</sub> becoming exhausted and can be remedied by addition of more acid.
- 4. After the reaction has subsided, add 5 ml H<sub>Z</sub>SO<sub>4</sub> and 10 ml HNO<sub>3</sub>.

  Boil to wet dryness and repeat. Take up with 10 ml furning HNO<sub>3</sub>.

  Repeat the evaporation and take up with another 10 ml of furning HNO<sub>3</sub>. Boil solution to approximately 5 ml (note c).
  - c. If insoluble sulfates are present transfer solution to a centrifuge cone and centrifuge. Save the supernate and wash the residue with 6N HCl. Add the washings to the supernate. Discard the residue.
- 5. Transfer the contents of the beaker to a teflon beaker (note d) by means of a transfer pipet. Wash the beaker with several 6N HNO<sub>3</sub> washes, scrubbing the sides and bottom with a polyethylene policeman. Perform at least two washes with 3 ml aliquots of 1N HNO<sub>3</sub> 1N HF.

- d. If started in teflon, omit step 5 but add a few ml HNO<sub>3</sub>.
  If the sample does not contain any insoluble material at this point, omit steps 6 through 9.
- 6. Add 10 ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable amounts of dirt, repeat HF treatment at least once.
- 7. Add 4 ml saturated H<sub>3</sub>BO<sub>3</sub> and 8 ml HNO<sub>3</sub> and boil for 3 minutes.
- 8. If residue remains, wash with portions of warm 6N HNO<sub>3</sub> until it dissolves.
- 9. Transfer any undissolved residue to the teflon beaker quantitatively with HNO<sub>2</sub> washes and repeat steps 6, 7, and 8.
- 10. Transfer the solution to a 40-ml centrifuge cone and proceed with step 1 Pu-239 Purification Procedure.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-259 Purification Procedure.

- Place the sample or aliquot in a suitably sized pyrex beaker (note a)
  or teflon beaker if sample is small.
  - a. If the sample monitors < 4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6№ HNO<sub>3</sub> a small aliquot is pipetted into a 40-ml centrifuge cone, and an appropriate amount of tracer is added.
- Add enough furning HNO<sub>3</sub> to wet all of the sample. Heat on a hot plate until the sample has dissolved.
- 3. Remove from hot plate and add about 6 ml 78 per cent HClO<sub>4</sub> for every 100 ml furning HNO<sub>3</sub> added in step 2. Heat on hot plate until exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of furning HNO<sub>3</sub>, pouring acid carefully down wall of beaker (note b).
  - b. At times the reaction ceases and the solution turns black. This is caused by the supply of fuming HNO<sub>3</sub> becoming exhausted and is remedied by addition of more acid.
- 4. Transfer the contents of the beaker to a teflon beaker (note c) by means of a transfer pipet. Wash the beaker with several 6N HNO3 washes, scrubbing the sides and bottom with a polyethylene policeman. Perform at least two washes with 3 ml aliquots of 1N HNO3 1N HE.
  - c. If started in teflon, omit step 4 but add a few ml HNO<sub>3</sub>. If the sample does not contain any insoluble material at this point, omit steps 5 through 8.
- Add 10-ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable

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amounts of dirt, repeat HF treatment at least once.

- 6. Add 4 ml saturated H<sub>3</sub>BO<sub>3</sub> and 8 ml HNO<sub>3</sub> and boil for 3 minutes.
- If residue remains, wash with portions of warm 6N HNO<sub>3</sub> until it dissolves.
- 8. Transfer any undissolved residue to the teflon beaker quantitatively with HNO<sub>3</sub> washes and repeat steps 5, 6, and 7.
- Transfer the solution to a 40-ml centrifuge cone and proceed with step
   Pu-239 Purification Procedure.

<sup>\*</sup>If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 Purification Procedure.

<sup>\*\*</sup>The sequential tape is cut into equal sections. In order to monitor the various sections, unroll the tape carefully and pass the exposed side under a sensitive lab detector. Record on the chemical processing form activity levels and/or physical spots on the tape. Cut the tape into appropriate sections.

- Place the sample or aliquot in a suitably sized pyrex beaker (note a) or teflon beaker if sample is small.
  - a. If the sample monitors <4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6N HNO<sub>3</sub>, a small aliquot is pipetted into a 40-ml centrifuge cone and an appropriate amount of tracer added.
- Add 3 ml CH<sub>3</sub>OH, ignite, and cover be ker with a speedy vap. After burning is completed, cover residue with fuming HNO<sub>3</sub> and boil to wet dryness.
   Repeat the fuming HNO<sub>3</sub> evaporation step. Take up in approximately
   1/4 volume fuming HNO<sub>3</sub>.
- 3. Remove and add about 10 ml 78 percent HClO<sub>4</sub> for every 100 ml fuming HNO<sub>3</sub> added in step 2. Heat on hot plate until an exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of fuming HNO<sub>3</sub>, pouring acid carefully down wall of beaker (note b).
  - b. At times the reaction ceases and the solution turns black. This is caused by the supply of fuming HNO<sub>3</sub> becoming exhausted and is remedied by addition of more acid.
- 4. Transfer the contents of the beaker to a teflon beaker (note c) by means of a transfer pipet. Wash the beaker with several 6N HNO<sub>3</sub> washes, scrubbing the sides and bottom with a polyethylene policeman. Perform at least two washes with 3 ml aliquots of 1N HNO<sub>3</sub> 1N HF and heat on hot plate.
  - c. If started in teflon, omit step 4 but add a few ml HNO<sub>3</sub>. If sample does not contain any insoluble material at this point, omit steps 5 through 8.

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- 5. Add 10 ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable amounts of dirt, repeat HF treatment at least once.
- 6. Add 4 ml saturated H<sub>3</sub>BO<sub>3</sub> and 8 ml HNO<sub>3</sub> and boil for 3 minutes.
- 7. If residue remains, wash with portions of warm 6N HNO<sub>3</sub> until it dissolves.
- 8. Transfer any undissolved residue to the teflon beaker quantitatively with HNO<sub>3</sub> washes and repeat steps 5, 6 and 7.
- Transfer the solution to a 40-ml centrifuge cone and proceed with step
   Pu-239 Purification Procedure.



<sup>\*</sup> if uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 Purification Procedure

#### A-Filtered Aliquot

- 1. Determine the pH of the sample with a Beckman pH meter.
- 2. Pipet a 25-ml aliquot of the clear liquid onto a millipore filter and allow the solution to drain thoroughly into an appropriate container. Do not wash the filter.
- 3. Pipet 1 ml of the filtrate into a 40-ml centrifuge cone and save, for the determination of uranium.
- 4. Add an appropriate amount of tracer to remaining filtrate.
- 5. Add 10 ml HNO<sub>3</sub> and 1 ml HClO<sub>4</sub> and boil to HClO<sub>4</sub> fumes. Cool and transfer to a 40-ml tube. Proceed with step 1 of Pu-239 Purification Procedure.

#### B-Excepted Aliquot

- 1. Stir the sample and pipet a representative 25-ml alique into a 16-ounce plastic bottle. Adjust the pH slightly with NH<sub>4</sub>OH to offset the acidity (6N) of the tracer in the next step.
- 2. Add an appropriate amount of tracer to the sample aliquot.
- 3. Proceed with step 1 of Extraction Procedure using 5 ml neutralized NH<sub>2</sub>OH· HCl and 25 ml CHCl<sub>3</sub> portions for extractions.

#### C-Total Sample

 Pour sample into a large teflon beaker and wash the container with H<sub>2</sub>O adding washing to the beaker. Add an appropriate amount of tracer.

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- Boil to low volume and add 150 ml fuming HNO<sub>3</sub> and 25 ml HClO<sub>4</sub>.
   Boil to HClO<sub>4</sub> fumes.
- 3. Add 50 ml fuming HNO<sub>3</sub> and 10 ml HF. Boil to low volume and add
  1 to 2 ml sat. H<sub>3</sub>BO<sub>3</sub> and 10 ml HNO<sub>3</sub>. Boil to approximately 5 ml and
  transfer to 40-ml centrifuge cone. Proceed to step 1 of Pu-239
  Purification Procedure.

### D-Glass Bottle Decontamination

- Rinse the container from Part C, above, three times with hot Aqua Regia and pour the washing into a large teflon beaker.
- 2. Rinse the bottle with 1 N HNO<sub>3</sub> 1 N HF adding the rinse to the Aqua Regia wastes. Rinse with H<sub>2</sub>O and add washes to beaker.
- Add an appropriate amount of tracer, then proceed with step 2, part C
   above.

#### E-Millipore Filter

- 1. Remove the millipore filter from Part A above, or Part G below, carefully with forceps and place in a small teflon beaker. Add an appropriate amount of tracer.
- Add 75 ml of furning HNO<sub>3</sub> and 15 ml HClO<sub>4</sub>. Boil to HClO<sub>4</sub> fumes and proceed with step 3, Part C above.

#### F-Centrifuge Supernate

1. Stir the sample and pipet approximately 25 mlinto a 40-ml centrifuge cone.

- Centrifuge and pipet 1 ml of the supernate onto a labelled stainless
  steel disc and evaporate to dryness under a heat lamp.
- 3. Place in metal container and submit for 2 T counting.

## G-Leached Supernate

- 1. Stir the sample and quickly pipet a 25-ml aliquot onto a millipore filter and allow the supernate to drain thoroughly into an appropriate container.
- 2. Remove the filter with forceps, place in a beaker containing a measured volume of 0. 1N HCl. Stir intermittently for a measured period and pour the solution onto a fresh filter and catch the filtrate in another container.
- 3. Repeat step 2 combining filters for measured periods up to 48 hours.
- 4. Pipet a 250 haliquot from each filtrate fraction onto a labelled stainless steel disc and evaporate to dryness under a heat lamp.
- 5. Place in a metal container and submit for 2 T counting.
- 6. Pipet a 1-ml aliquot, from selected filtrate fractions, into a 40-ml centrifuge cone and save for uranium analysis.

<sup>\*</sup>If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone and proceed with step 1 Pu-239 Purification Procedure.

- Add sample to a suitably sized teflor beaker (note a). Add 30 ml HF and
  10 ml HNO<sub>3</sub>. After initial exothermic reaction has ceased, boil to dryness
  (or until spattering starts).
  - a. If the sample monitors < 4000 alpha cpm, add Pu<sup>236</sup> tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6NHNO<sub>3</sub> a small aliquot is pipetted into a 40-ml centrifuge cone, and an appropriate amount of tracer is added.
- 2. Repeat HF treatment until no change in the sample crud is perceived.
- 3. Add 30 ml HClO<sub>4</sub>, 10 ml HNO<sub>3</sub>, and 10 ml HF. Boil to strong fumes of HClO<sub>4</sub>. Remove from hot plate and cool.
- 4. Rinse down the sides of the beaker with 6NHNO3, add 3 ml H3BO3 and boil to low volume. Take up in 6NHNO3 and heat gently.
- 5. If a residue is still present, centrifuge, add 20-ml portions of 6NHNO3 to residue and warm. (Watch for bumping!) Combine washings and supernate if all residue has dissolved (note b).
  - b. A residue which persists will sometimes dissolve with repeated Hot Aqua Regia treatment (maximum 3). If this treatment fails, put sample into a flask and add 1/3 volume furning HNO<sub>3</sub>. Add H<sub>2</sub>O and shake vigorously, venting flask periodically.
- If residue still remains, centrifuge, and repeat steps 2 through 5 on the residue (note c).
  - c. A trace of hard silica final remains for some samples. The amount of activity associated with this residue was found to be insignificant.

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7. Transfer the solution from step 5 to a centrifuge cone and proceed with step 1 of Pu-239 PURIFICATION PROCEDURE.

\*If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H.O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transger to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

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- Add sample to a suitably sized teflon beaker. Rinse container with IN

  HNO<sub>3</sub> HF and add to sample (note a.) Add 20 ml HF for each 5 grams of soil.

  After initial exothermic reaction has ceased, boil to dryness (or until spattering starts).
  - a. If the sample monitors <4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alph cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6N HNO<sub>3</sub>, a small aliquot is pipetted into a 40 ml centrifuge cone, and an appropriate amount of tracer is added.
- 2. Repeat HF treatment until no change in the sample crud is perceived.
- 3. Add 30 ml  $\rm HClO_4$ , 10 ml  $\rm HNO_3$ , and 10 ml HF. Boil to strong fumes of  $\rm HClO_4$ . Remove from hot plate and cool.
- 4. Rinse down the wall of the beaker with 6N HNO<sub>3</sub> and Sat.  $H_3BO_3$  and boil to low volume. Take up with 50 ml HCl and boil with repeated additions until HNO<sub>3</sub> is gone (note b).
  - b. Avoid low volume, as excessive foaming and swelling will occur.
- 5. Cool and transfer the solution to a poly bottle (250 to 500 ml depending on the sample size) and proceed with step 1 Pu-239 EXTRACTION PROCEDURE.

<sup>\*</sup> If uranium is required, withdraw a representative aliquot after dissolution and spike it with a known amount of standardized uranium (duplicate the extraction procedure with this aliquot). Transfer the remaining sample into a volumetric flask after extraction, and dilute to mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- 1. Remove the sample from its polyethylene bag and place in a 600 to 800-ml beaker. Rinse the plastic bag with HNO3 and add the washings to the beaker.
- Pipet an appropriate amount of Pu-236 tracer and add enough HNO<sub>3</sub> to cover the sample. Heat gently and boil the solution to a small volume.
- 3. Gool the solution and add 50 ml fuming nitric and 50 ml HGlO<sub>4</sub>.
  Heat gently until the vigorous exothermic HGlO<sub>4</sub> reaction starts.
  Remove the beaker from the hot plate and allow the reaction to go to completion.
- 4. Fume the solution to a small column and transfer to a centrifuge cone.
- 5. Centrifuge and decant the supernate into the dissolution beaker.

  Leach and decant the residual sand several times with hot HNO<sub>3</sub>.

  Centrifuge each leach, combining supernates (note a).\*\*
  - a. Save the residue for extraction of residual plutonium in the event of a low sample yield.

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- 6. Evaporate the combined supernates to a small volume and continue with step 1 Pu-239 PURIFICATION PROCEDURE (note b).
  - If heavy insoluble salts occur after evaporation proceed with step 1 Pu-239 EXTRACTION PROCEDURE \*\*.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution into an appropriate volumetric flask, and dilute carefully to the mark with H,O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 Purification Procedure.

<sup>\*\*</sup> If uranium is required, withdraw a representative aliquot after dissolution and spike it with a known amount of standardized uranium (duplicate the extraction procedure with aliquot). Transfer the remaining sample into a volumetric flask after extraction, and dilute to mark with H2C. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- 1. Add sample to a suitably sized tellon beaker (note a). Add 30 ml HF and 10 ml HNO3. After initial exothermic reaction has ceased, boil to dryness (or until spattering starts).
  - a. If the sample monitors < 4000 alpha cpm, add Pu<sup>236</sup> tracer aliquot before adding sample. If > 4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6NHNO<sub>3</sub> a small aliquot is pipetted into a 40-ml centrifuge cone, and an appropriate amount of trace. is added.
- 2. Repeat HF treatment until no change in the sample crud is perceived.
- Add 30 ml HClO<sub>4</sub>, 10 ml HNO<sub>3</sub>, and 10 ml HF. Boil to strong fumes
  of HClO<sub>4</sub>. Remove from hot plate and cool.
- 4. Rinse down the sides of the beaker with on HNO3, add 3 ml H3BO3 and boil to low volume. Take up in 6N HNO3 and heat gently.
- 5. If a residue is still present, centrifuge, add 20-ml portions of 6NHNO<sub>3</sub> to residue and warm. (Watch for bumping!) Combine washings and supernate if all residue has dissolved (note b).
  - b. A residue which persists will sometimes dissolve with repeated hot Aqua Regia treatment (maximum 3). If this treatment fails, put sample into a flask and add 1/3 volume fuming HNO<sub>3</sub>. Add H<sub>2</sub>O and shake vigorously, venting flask periodically.
- If residue still remains, centrifuge, and repeat steps 2 through 5 on the residue (note c).
  - c. A trace of hard silics final remains for some samples. The amount of activity associated with this residue was found to be insignificant.

7. Transfer the solution from step 5 to a centrifuge cone and proceed with step 1 of Pu-239 PURIFICATION PROCEDURE.

\*If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- Add sample to a suitably sized tation beaker. Rinse container with 1N  $_{7}HNO_{3}$ - HF and add to sample (note a.). Add 20 ml HF-HNO $_{3}$  for each 5 gms of soil. After initial exothermic reaction has ceased, boil to cryness (or until spattering starts).
  - If the sample monitors <4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6N HNO a small aliquot is pipetted into a 40-ml centrifuge cone, and an appropriate amount of tracer is added.
- Repeat HF treatment until no change in the sample crud is perceived.
- 3. Add 30 ml HClO<sub>4</sub>, 10 ml HNO<sub>2</sub>, and 10 ml HF. Boil to strong fumes of HClO<sub>4</sub>. Remove from hot plate and cool.
- Rinse down the sides of the beaker with  $6N \text{ HNO}_3$  and Sat.  $H_3BO_3$  and boil to low volume. Take up with 50 ml HCl and boil with repeated additions until HNO<sub>3</sub> is gone (note b.).
  - b. Avoid low volume as excessive foaming and swelling will occur.
- Cool and transfer the solution to a poly bottle (250 to 500 ml depending on sample size) and proceed with step 1 Pu-239 EXTRACTION PROCEDURE.

<sup>\*</sup> If uranium is required, withdraw a representative aliquot after dissolution and spike it with a known amount of standardized uranium (duplicate the extraction procedure with this aliquot). Transfer the remaining sample into a volumetric flask after extraction, and dilute to mark with H2O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- 1. Place the sample or aliquot in a suitably sized pyrex beaker (note a) or teflor beaker if sample is small.
  - a. If the sample monitors < 4000 alpha cpm, add Pu-236 tracer aliquot before adding sample. If >4000 alpha cpm, the dissolution is done tracer free, the solution diluted to an accurate volume to obtain a final acidity of 6N HNO<sub>3</sub> a small aliquot is pipetted into a 40 ml centrifuge cone, and an appropriate amount of tracer is added.
- 2. Add enough furning HNO<sub>3</sub> to wet all of the sample. Heat on a hot plate until the sample has dissolved.
- 3. Remove and add about 10 ml 78 percent HClO<sub>4</sub> for every 100 ml fuming HNO<sub>3</sub> added in step 2. Heat on hot plate until an exothermic reaction begins. Remove beaker from hot plate and allow reaction to proceed, controlling it by the addition of 1 to 10 ml portions of fuming HNO<sub>3</sub> pouring acid carefully down wall of beaker (note b).
  - b. At times the reaction ceases and the solution turns black. This is caused by the supply of fuming HNO<sub>3</sub> becoming exhausted and is remedied by addition of more acid.
- 4. Transfer the contents of the beaker to a teflon beaker (note c) by means of a transfer pipet. Wash the beaker with several 6N HNO<sub>3</sub> washes, scrubbing the sides and bottom with a polyethylene policeman. Perform at least two washes with 3 ml aliquots of 1N HNO<sub>3</sub> 1N HF.
  - c. If started in tellon, omit step 4 but add a few ml HNO<sub>3</sub>. If the sample does not contain any insoluble material at this point, omit steps 5 through 8.
- 5. Add 10 ml HF and evaporate to wet dryness. Do not allow sample to bake dry at any time during the procedure. If sample contains appreciable amounts of dirt, repeat HF treatment at least once.
- 6. Add 4 ml saturated  $H_3BO_3$  and 8 ml  $HNO_3$  and boil for 3 minutes.
- 7. If residue remains, wash with portions of warm 6N HNO, until it dissolves.

CONFIDENTIAL

WIRE SWIPE \*

- 8. Transfer any undissolved residue to the teilon beaker quantitatively with HNO<sub>3</sub> washes and repeat steps 5, 6, and 7.
- 9. Transfer the solution to a 40-ml centrifuge cone and proceed with step 1 Pu-239 Purification Procedure.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 PU-239 Purification Procedure.

- Flace the sample in an appropriate corningware oven dish, cover and dry at 1100 C overnight. Transfer dish to a muffle furnace and ash at 600°C overnight.
- Remove, cool, and grind the bone ash with a glass stirring rod or pettle.
- Dissolve the pulverized ash in concentrated HCl at low heat on the hot 3. plate (note a).
  - If more than a trace of insoluble material is present, the following \*tep\* must be performed.
    - (1) Decant solution into a beaker. Transfer solid residue to a platinum dish. Evaporate to dryness under a heat lamp.
    - (2) Add at least 3 times the amount of residue) solid Na CO. Fuse at .900°C in a muffle furnace for 10 minutes.
    - (3) Dissolve in HCl and transfer to sample beaker. Continue with step. 5.
- 4. Transfer the solution to a poly bottle(2 liter acid bottle for large bones). with a transfer pipet.
- Wash the crucible with hot concentrated HCl and add washings into the 5. bottle.
- 6. Proceed with step 1 of the Pu-239 EXTRACTION PROCEDURE.

<sup>\*</sup> If uranium is required, withdraw a representative aliquot after dissolution and spike it with a known amount of standardized uranium (duplicate the extraction procedure with this aliquot). Transfer the remaining sample into a volumetric flask after extraction, and dilute to mark with H.O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- Remove the frozen sample from its polyethylene bag and place in a small (250 to 400 ml) beaker. Rinse the plastic bag with HNO<sub>3</sub> and add the washings to the beaker.
- Pipet an appropriate amount of Pu-Z36 tracer and add enough HNO<sub>3</sub> to cover the sample. Heat gently and boil the solution to a small volume.
- 3. Cool the solution and add 50 ml fuming  $\mathrm{HNO}_3$  and 50 ml  $\mathrm{HClO}_4$ . Heat gently until the vigorous exothermic  $\mathrm{HClO}_4$  reaction starts. Remove the beaker from the hot plate and allow the reaction to go to completion.
- 4. Fume the solution to a small volume and transfer to a centrifuge come.
- 5. Proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

<sup>\*</sup> If uranium analysis is required, transfer the sample, after dissolution, into an appropriate volumetric flask and dilute carefully to the mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrique cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- Remove the frozen sample from its polyethylene bag and place in an appropriate size beaker. Rinse the plastic bag with HNO<sub>3</sub> and add the washings to the beaker.
- Add an appropriate amount of Pu-236 tracer and enough HNO<sub>3</sub> to cover the sample. Heat gently and boil the solution to a low volume.
- 3. Cool and add enough H<sub>2</sub>SO<sub>4</sub> to raise the level of solution in the beaker to approximately 1 inch. Heat gently until a vigorous reaction starts, then remove from the hot plate until the reaction subsides.
- 4. Fume this solution (black liquid) to a small volume and heat with HNO<sub>3</sub> until the solution turns red and finally clears. Add fuming HNO<sub>3</sub> and HClO<sub>4</sub> and fume to a small volume. Add H<sub>2</sub>SO<sub>4</sub> and fume to low volume to drive off the HClO<sub>4</sub> (note a).
  - a. HClQ forms explosive mixture with cupferron-CHCl3 reagent, added later to extract plutonium and uranium from precipitated salts, and must be removed.
- 5. Cool, transfer the solution to a poly bottle (250 to 500 ml depending on sample size), and proceed with step 1 of the Pu-239 Extraction Procedure.

<sup>\*</sup> If uranium is required, withdraw a representative aliquot after dissolution and spike it with a known amount of standardized uranium (duplicate the extraction procedure with this aliquot). Transfer the remaining sample into a volumetric flask after extraction, and dilute to mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-m1 centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

ORGANIC TISSUE .

- 1. Remove the frozen sample from its polyethylene bag and allow it to thaw for a few minutes under a heat lamp. If the sample shape or size is such that it will not fit in the bottom half of a four-liter beaker, cut the sample into appropriate sections and place each section in a separate beaker. Rinse the plastic bag with HNO<sub>3</sub> and add the washings to the beaker.
- 2. Add enough H<sub>2</sub>SO<sub>4</sub> to completely cover the sample. Pipet an appropriate amount of Pu-236 tracer (within a factor of five of the expected sample activity but a minimum of 15 dpm) into each beaker and add approximately 5 grams K<sub>2</sub>SO<sub>4</sub> and 2-3 drops Hg metal. Spray the sample with Dow-Corning Anti-foam A silicone defoamer.
- 5. Attach the stem of an inverted 6-inch funnel to a ring stand and clamp and lower the funnel mouth into the beaker. Secure a few inches above the sample.
- 4. Digest the sample gently with low heat until a black tarry mixture is obtained. Increase the heat gradually and reflux until the mixture is a clear solution. The tarry mixture will turn to a black jelly, black liquid, red liquid, and finally, a clear solution. Raise or lower the funnel during dissolution to control the reflux action. Wash down any carbonaceous material on the beaker and funnel walls with H<sub>2</sub>SO<sub>4</sub>.
- 5. Evaporate the  $\rm H_2SO_4$  until salts start forming. Remove the funnel and add  $\rm HNO_3$  cautiously to cool the solution. If the sample had been divided into

sections for the dissolution, combine the sections into one beaker and evaporate to low volume. Cool, transfer the sample with water to a 2-liter acid bottle, and proceed with step 1 Pu-239 EXTRACTION PROCEDURE.

<sup>\*</sup> If uranium is required, withdraw a representative aliquot after dissolution and spike it with a known amount of standardized uranium (duplicate the extraction procedure with this aliquot). Transfer the remaining sample into a volumetric flask after extraction, and dilute to mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40-ml centrifuge cone, and proceed with step 1 Pu-239 PURIFICATION PROCEDURE.

- l. Place the sample in a 4-liter beaker, rinsing the container with H<sub>2</sub>O and HNO<sub>3</sub>. Add an appropriate amount of Pu-236 tracer.
- Cover the beaker with a speedy-vap and boil to wet-dryness on a hot plate. Cover the sample with HNO<sub>q</sub> and boil to low volume.
- 3. Add 100 to 200 ml furning HNO<sub>3</sub> and cautiously evaporate the solution to wet dryness (note a).
  - a. At near dryness, ignition occurs and the residue carbonises.
- 4. Cool and rinse the speedy-vap and the sides of the beaker with approximately 100 ml HNO<sub>3</sub>. Add 75 ml and HGlO<sub>4</sub> and fume the mixture to dense HGlO<sub>4</sub> fumes to destroy residual organic matter.
- 5. Add 200 ml H<sub>2</sub>SO<sub>4</sub> and fume the mixture to low volume to drive off all the HClO<sub>4</sub>. Gool and transfer to a 2-liter acid bottle.
- 6. Continue with step 1 Pu-239 Extraction Procedure.

<sup>\*</sup> If Uranium is required, withdraw a representative aliquot after dissolution and spike it with a known amount of standardized Uranium (duplicate the extraction procedure with this aliquot). Transfer the remaining sample into a volumetric flask after extraction, and dilute to mark with H<sub>2</sub>O. Pipet an appropriate aliquot into a centrifuge cone and save for uranium analysis. Evaporate the remainder of the volumetric solution to low volume, transfer to a 40 ml centrifuge cone, and proceed with step 1 Pu-239 Purification Procedure.

- Transfer the mounted Plutonium sample to a small teflon beaker. Add
   ml fuming HNO<sub>3</sub> and 10 ml HF and boil to low volume.
- 2. Remove disc with tesson forceps and rinse with 1 N HNO<sub>3</sub> 1 N HF adding washes to beaker.
- Dry the disc under a heat lamp and check for residual activity. Repeat stripping process if significant activity is detected.
- 4. Boil solution to wet dryness, add 10 drops of H<sub>3</sub>BO<sub>3</sub> and 2 ml HNO<sub>3</sub>, and boil to wet dryness.
- 5. Transfer the sample to a glass beaker with 6 N HNO3 and proceed with step 4 of the Pu-239 Purification Procedure.

- 1. Dilute the sample (note a) contained in poly or acid bottle, to 2/3 volume with H<sub>2</sub>O. Add an appropriate amount of Sat. NH<sub>2</sub> OH. HCl and CHCl<sub>3</sub> (note b). Stir at high speed with a mechanical stirrer for a few minutes. Add 50 to 100 ml 6% cuplerron reagent and stir again at high speed for 5 minutes.
  - a. The sample solution must be in approximately 1N HCl free of  $NO_3$ . Boil solution in HCl if necessary and dilute with  $H_2O$ .
  - b. Large tissues 100 ml Sat. NH<sub>2</sub> OH · HCl, 150 ml. CHCl<sub>3</sub>

    Medium tissues 50" " " " 100 ml. "

    Large Soils 10" " " 50 ml. "

    Small Soils 5" " " " 25 ml. "
- 2. Centrifuge to separate the phases. Add a few drops of aerosol solution to reduce foaming between layers. Transfer the CHCl<sub>3</sub> phase, using a transfer pipet to a 400-ml beaker. Repeat the extraction, without the addition of more cupferron, until the CHCl<sub>3</sub> phases are colorless.
- 3. Boil the CHCl<sub>3</sub> collections to low volume, (approximately 3 ml) and allow the contents to cool. Rinse the walls of the beaker with HNO<sub>3</sub> and boil to approximately 3 ml. Repeat the rinse with fuming HNO<sub>3</sub> and boil to low volume (avoid dryness) (note c).
  - c. Repeat furning HNO<sub>3</sub> cycle for soil samples until solution turns a clear red color.

- 4. Add 25 ml furning HNO<sub>3</sub> z 25 ml HClO<sub>4</sub>. Cautiously heat until an exothermic HClO<sub>4</sub> reaction begins. Remove the beaker from the hot plate and allow the reaction to go to completion.
- 5. Fume the HClO<sub>4</sub> to low volume (note d). Gool the solution and transfer with water washes to a centrifuge cone.
  - d. A white residue often appears at this point in large bone samples. Repeat the extraction in this event.
- 6. Continue with Step 1 Pu-239 PURIFICATION PROCEDURE.

- 1. To the sample contained in a centrifuge cone, add 10 mg Fe<sup>+ 3</sup> unless the sample is known to contain that much. Digest in a hot water bath for 10 minutes and carefully add 19N NaOH (pellets may be required if the volume is too large) until the solution is basic (note a). Add 3 ml saturated Na<sub>2</sub>GO<sub>3</sub> and digest in a hot water bath 10 minutes. Centrifuge and decant supernate to waste. Dissolve the precipitate in HNO<sub>3</sub> and dilute to approximately 15 ml.
  - a. Do not make the solution too basic, as Fe is amphoteric.
- Make the solution basic with NH<sub>4</sub>OH and digest the precipitate in a hot water bath for 10 minutes. Centrifuge and decant the supernate to waste. Wash the precipitate twice with 10-ml portions of H<sub>2</sub>O containing 1 drop NH<sub>4</sub>OH.
- Dissolve the precipitate in a minimum of HNO<sub>3</sub> and add 5 ml 6N HNO<sub>3</sub> (note b).
  - b. An insoluble brown precipitate sometimes persists if iron is present in excess. However, during the HCl column additions in steps 5 and 6 this precipitate dissolves, changing from brown to blue green. Addition of more HCl finally destroys the blue green color.
- 4. Prepare a 100 to 200 mesh Dowex 1-X 10 resin column by adding approximately 1/2 inch of resin to a tubulated glass column, 12 mm I.D. and 85 mm in length, containing a Dacron wool plug at the bottom. Insert another plug at the top and precondition the column with 10 ml 6N HNO3.
- 5. Pour the solution from step 3 onto the column. Wash the centrifuge tube with 20 ml 6N HNO3, followed by 10 ml HCl and add the washings to the column.
- 6. Allow the column to drain and elute the plutonium into a 50-ml beaker with 30 ml of freshly prepared HCl-NH<sub>4</sub>I (approximately 50 mg NH<sub>4</sub>I per 30 ml HCl).

- 7. Evaporate the elute to approximately 2 ml and add 2 ml HNO<sub>3</sub> to destroy I.
  Add 3 ml HClO<sub>4</sub> and evaporate to wet dryness. Repeat with 5 ml Aqua
  Regia.
- 8. Take up the residue in 1 ml HCl and evaporate to dryness. Do not bake. Rotate the beaker to insure complete dryness. Add 2 ml HCl, boil to 1 ml and transfer to a prepared electroplating cell (note c). Rinse the beaker with two 1/2-ml HCl washes and one 1/2-ml water wash. Transfer each wash to the plating cell (note d) and proceed with step 1 of Pu-239 ELECTROPLATING PROCEDURE.
  - film by rinsing several times with acetone and alcohol.

    Write the sample identification on the back of the disc.

    Ignite to red heat in a Fisher burner flame. The electroplating cell must be clean and free of any foreign material. Check for leakage before use.
  - d. Keep the plating solution at minimum volume during this transfer and also during the titration.

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- Add drop methyl red indicator. Add NH<sub>4</sub>OH dropwise until the indicator shows the solution to be basic (yellow). Add 2N HCl dropwise until the solution is just acid. Add 1 drop in excess.
- 2. Place the sample on a Sargent-Slomin electrolytic analyzer. Adjust the rotating anode to approximately 1/4 inch above the platinum disc. Plate for 20 minutes at a starting current of 2.5 amp and approximately 5 volts. The current may fluctuate during the plating period. Check occasionally and adjust the current to maintain 2.6 amp throughout the plating operation.
- 3. At the end of the electroplating period, add 1 ml NH<sub>4</sub>OH. Stir for 15 seconds. Turn off the current and stirrer. Remove the anode from the plating solution.
- 4. Immediately transfer the plating solution into the beaker used for evaporation. Rinse the inside of the plating cell 3 times with water washes. Combine the washes with the plating solution in the beaker.
- 5. Dismantle the plating cell and remove the platinum disc. Rinse with alcohol and ignite the disc to red heat.
- Place disc in a lined and labeled tin box and submit for alpha pulse height analysis.

- 1. Divide the aliquot set aside for uranium determination into two equal portions, and transfer to an appropriate glass beaker. Add an appropriate uranium spike (for yielding) to one portion (note a).
  - a. If the aliquot is taken from a cupferron-CHCL<sub>3</sub> extraction sample, do not spike or divide it. Analyze concurrently with the aliquot spike prior to extraction.
- Evaporate the solution to low volume and dilute to 5 ml with 2N HNO3:
   Transfer to a centrifuge cone and saturate with NH4NO3 crystals.
- Add 10 ml hexone and stir at high speed for 5 minutes. Transfer the hexone layer to a fresh centrifuge cone.
- 4. Repeat step 3 twice with 5-ml additions of hexone and combine organic phases.
- Scrub the hexone phase twice with a saturated solution of NH<sub>4</sub>NO<sub>3</sub> to remove Pu and other heavy elements and discard aqueous phase.
- 6. Back extract the uranium twice with two 5-ml additions of water and transfer the aqueous phases to a 50-ml glass beaker. Boil to wet dryness, add 5 ml Aqua Regia, and boil to wet dryness. Repeat Aqua Regia step (note b).
  - b. Aqua Regia destroys residual hexone and NH<sub>4</sub>NO<sub>3</sub> which may be carried through the back extraction.
- Take up solution in 6N HNO<sub>3</sub> and proceed with step 1 Fluorimetric Determination of Uranium (note c).
  - c. If the fluorimetric analyses is delayed, store the solution in concentrated HNO<sub>3</sub>.

- 1. Evaporate the sample to 1 ml and transfer to a platinum fusion dish

  (note a ) resting on a Nichrome wire screen-ring holder. Evaporate
  the sample to dryness under a heat lamp.
  - a. The fusion dishes are formed from satin finish 90% Pt-10% Ir alloy discs (0.015 inch thick by 0.748 ± 0.001 inch diameter) in a special forming die (0.750 inch diameter). The new dishes are cleaned by boiling in a 1-to-1 mixture of H<sub>2</sub>SO<sub>4</sub> and HNO<sub>3</sub> and then rinsed thoroughly in water and distilled water. They are then fused twice with NaF-LiF flux and washed before their initial use.
- 2. Start a blast burner and by regulation of the air supply, stabilize the flame to 800°C. Position the dish holder above the flame, add 1/2 gm NaF-LiF Flux (note b) to the dish and ignite to a bright red heat for 3 minutes (note c). Allow to cool in place for 15 minutes and transfer to a 12-hole uranium dish container. Transfer the dish to a calibrated Jarrell-Ash fluorimeter (note d), within 2 hours and record the milliamps. Convert reading to μ gram per total sample using a calibration curve (note e) and calculation factors.
  - b. The flux powder from which pellets are formed for fusion is composed of 98% NaF-2% LiF mixed intimately. This is made up in one-pound batches in order to insure uniformity in day-to-day use. A satisfactory bath of flux will show 0.010 µg or less per 0.5-gram pellet. The purest NaF available gives 0.003 µg. The pellets are formed in a TLW pellet-maker, fabricated from

pyrex glass, and adjusted to deliver a 0.50 ± 0.01 grain pellet. The pellet-maker is gently pressed for about ten times into the flux powder container, and a spatula is used to flatten the bottom of the pellet.

- c. The flux will melt within one minute if the flame is correctly adjusted and the fusion is continued for three minutes after the last of the flux has melted. At the end of the fusion, turn off the air and gas simultaneously.
- d. The Jarrell-Ash fluorimeter is operated and maintained according to the supplied manual. To prepare for a series of readings, push the receptacle slide to the front stop. The instrument reference source is now under the ultraviolet light and the meter circuit is switched to the 0.001 scale. The voltage is adjusted so as to give a reading of 100 divisions. Adjust the reference to zero and enter blank and read. Remove blank, enter sample, and record reading. Check reference reading between each sample and zero if necessary. Record all necessary data.
- e. The fluxometer is calibrated prior to sample analysis and daily thereafter. The initial calibration is performed by analyzing 100  $\lambda$  spikes of standardized uranium solutions. The standard uranium solution used for spiking and standard measurement ranges from 0.050 to 8.24  $\mu$ g U<sub>3</sub>O<sub>8</sub> ml. These concentrations are alrequoted to allow readings on the fluorimeter to be made on the 0.001-1 scale. A calibration curve is plotted from the scale

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readings. Blanks with the multiplier phototube operating at about 400 volts. show a typical reading as follows:

Scale

0.001

Instrument standard set on 100 div.

Blank:

0.019 divisions or

0.006 ME

Daily calibrations are performed by analyzing 100 % spikes of 0.973 and 1.46  $\mu$ g  $U_3O_8$  per ml standardized granium solutions. If the initial and daily calibration curves are mismatched a new calibration curve is prepared as in Step 2.

3. Remove the dish from the fluorimeter and discard the fluoride pellet.

Clean the disc by boiling once in 0.1N HCl, twice in H<sub>2</sub>O<sub>1</sub> and fusing with NaF-LiF flux. Repeat acid and we er treatment, rinse with H<sub>2</sub>O<sub>2</sub> and place disc face down on a clean paper towel to dry. The disc is now ready for the next sample.

<sup>\*</sup> The sensitivity of this method is 0.001 µg of UpO2. Reproducibility is approximately 5%.

The production of Pu-236 is accomplished by deuteron bombardment of U-235 in an accelerator by the following reaction:

U-235 (d, n) Np-236

Np-236 22 hr Pu-236

In order to obtain Pu-236 free of Pu-238 and Pu-239, the U-235 must be 99+ percent pure.

The deuteron energy should be between 12 to 17 mev. The target is

U-235 foil about 200 mg/cm<sup>2</sup> thick and with an area of approximately 1 inch
square, depending on beam size. This is about one gram U-235.

After bombardment, the Np-236 is allowed to decay and Pu-236 milked off by appropriate chemical procedures.

Since fissionable material is being irradiated, the target must be sandwiched with aluminum foil. It is desirable to have a scaled target containing an inert atmosphere and naving its own water cooling lines.

The surrounding foil means a deuteron energy of about 25 mev is required, this being degraded by the foil to the 12 to 17 mev range.

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- 1. When a new stock of Pu<sup>236</sup> is received, transfer the solution to a 50-ml lusteroid cone and add ~5 mg La and ~5 mg Fe .
- 2. Make the solution basic with NH4OH. Let stand 3 minutes and centrifuge. Wash the hydroxides with 10 ml water. Discard the supernate and wash. Dissolve the hydroxide in 3 drops of HNO3.
- 3. Dilute to 10 ml with water. Heat the solution for 3 minutes on a 75° C water bath. Add 20 mg NaHSO3 a little at a time, to insure complete reduction. Continue to heat for 5 minutes. Add 10 drops HF with stirring, and heat for a few minutes. Cool and centrifuge. Wash the LaF3 with 2 ml 1N HC1-1N HF. Discard the supernate and wash.
- 4. Slurry the LaF3 in 1 ml saturated H3BO3 and heat on a 750 C water bath for a few minutes. Add I ml HCl and I ml water and continue to heat on the water bath to obtain a clear solution. Dilute to 10 ml with water, Add~2 mg Fe<sup>+3</sup>.
- 5. Repeat steps 2, 3, and 4. Do not add Fe to the repeated step 4.
- 6. Add NH4OH to precipitate La(OH) . Digest in a hot water bath for a few minutes. Centrifuge, and wash the precipitate with 5 ml water containing I drop NH4OH. Discard the supernate and wash.
- 7. Dissolve the La (OH)3 in 1 ml HCl and 2 drops HNC3. Heat the solution for 3 minutes in a hot water bath. Cool the solution in an ice bath, and saturate with HCl gas. Allow to come to room temperature.
- 8. Transfer the solution to a prepared Dowex AG 1-X8 (100 to 200 mesh) column. Prepare an eluting solution containing 15 ml HCl and 1/2-ml HNO3. Rinse the tube with several 1-ml portions of this solution. Transfer these washes to the column. Wash the column with the

- remaining solution in 2-ml portions. Wash with 15 ml HCl in 2-ml portions. Discard the effluents and washes.
- 9. Prepare an eluting solution containing 20 ml HCl and 75 mg NH<sub>4</sub>1. Elute the Pu from the column into a 50-ml beaker with 2-ml portions of this solution, allowing the first 2-ml portion to pass through. Add the second 2-ml portion and plug the top of the column with a piece of pressure-sensitive tape for 5 minutes. Remove the tape and continue to elute in 2-ml portions. Pass through 6 ml of HCl in 2-ml portions.
- 10. Evaporate the solution in the 50-ml beaker just to dryness with addition of HNO, in order to drive oif all iodine. Take up the activity in 6N HCl. Transfer the activity to a polyethylene bottle using 6N HCl washes. Add sufficient 6N HCl to give a concentration of ≈3000 dpm per ml.
- 11. Transfer the contents of the 3000 dpm per ml concentrated stock solution to an appropriate size glass beaker. Add 10 mg Fe 3, 4 ml H<sub>2</sub>SO<sub>4</sub>, and ~2 ml HClO<sub>4</sub>. Evaporate to SO<sub>3</sub> fumes.
- 12. Wash the sides of the beaker with HCl and evaporate to near dryness. Take up in 1-2 ml HCl and transfer to a 40-ml centrifuge cone with H,O+HCl washes.
- 13. Ppt Fe(OH), with NH4OH and centrifuge. Wash ppt with H2O containing a few drops NH4OH and centrifuge. Dissolve the Fe(OH)3 in a few drops HNO3 and dilute to 5 ml with 6N HNO3. Add ~1/2-ml saturated NaBrO3 solution.
- 14. Warm on hot water bath a few minutes. Saturate the solution with NH<sub>4</sub>NO<sub>3</sub> crystals, add 5 ml hexone and stir with a mechanical stirrer 3 minutes. Repeat hexone extraction twice adding Sat. NaBrO3 and more NH4NO3 crystals as necessary.

- 15. Wash the hexone phases by stirring with 5-ml 6N HNO3 for 1 minute and discard the washes. Back extract the Pu with three 5-ml additions of 0.1N HNO3. Transfer the aqueous phase to a 50-ml glass beaker.
- 16. Add 10-ml HCl and boil to wet dryness. Repeat HCl addition and evaporation twice. Take up with 10-ml 6N HCl. Transfer contents to 250-ml poly buttle, and dilute to 250 ml with 6N HCl.
- 17. Add one ml of Con HClO<sub>4</sub> and cap tightly. The activity value should be~4.4 x 10<sup>3</sup> dpm/ml. Label R. C. Pu<sup>236</sup> Master Stock Solution.
- 18. Pipet exactly 10 ml of the R.C. Pu<sup>236</sup> Master Stock Solution into a 2000-ml volumetric flask and add 10 ml Con HClO<sub>4</sub>. Dilute to the mark with 6N HCl.
- 19. Transfer the solution (not quantitatively) into eight clean, dry 250-ml poly bottles and cap tightly. The activity value should be 22 dpm/ml in each poly bottle. Label R.C. Pu<sup>236</sup> Low Level Stock Solution. Circle caps and bottles with green label on tape.
- 20. Pipet exactly 200 ml of the R.C. Pu<sup>236</sup> Master Stock Solution into a 2000-ml volumetric flask and add 10 ml HClO<sub>4</sub>. Dilute to the mark with 6N HCl.
- 21. Transfer the solution (not quantitatively) into eight clean, dry 250-ml poly bottles and cap tightly. The activity value should bew440 dpm/ml. Label R. C. Pu<sup>236</sup> High Level Stock Solution. Circle caps and bottles with red label on tape.

- Pipet 1 ml each of R.C. Pu<sup>236</sup> Low Level and High Level Stock
   Solution into an electroplating cell (usually process duplicate aliquots).
   Add 1/2 ml HCl.
- 2. Add 1 drop methyl red indicator. Add NH<sub>4</sub>OH dropwise until the indicator shows the solution to be basic (yellow). Add 2N HCl dropwise until the solution is just acid. Add 1 drop in excess.
- 3. Place the sample on the Sargent-Slomin electroplater. Adjust the platinum anode (note a) to approximately 1/4-inch above the platinum disc (Note b). Plate for 10 minutes at a starting current of 2.5 amps and about 5 volts. The current may fluctuate during the plating period. Check occasionally and adjust the current to maintain 2.6 amps throughout the plating period.
  - a. The same platinum anode, glass tower, and washer is used through three successive platings of a given aliquot.
  - b. The platinum disc and anode must be freed of any grease film by rinsing several times with acetone and alcohol. Write the sample identification on the back of the disc. Ignite to red heat in a Fisher burner flame. The electroplating cell must be clean and free of any foreign material. Check for leakage before use.
- At the end of the electroplating period, add 1 ml NH<sub>4</sub>OH. Stir for 15 seconds. Turn off the current and stirrer. Remove the anode from plating solution.
- 5. Immediately transfer the plating solution into a 50-ml beaker. Rinse the inside of the plating cell three times with water washes. Combine the washes with the plating solution in the beaker.

- 7. Place sample in a lined and labeled tin box and submit for counting analysis.
- Evaporate the solution to approximately 3 ml. Add 3 ml HNO<sub>3</sub> and l ml HCl. Evaporate to wet dryness. Repeat the HNO<sub>3</sub>-HCl treatment twice.
- 9. Pick up in 1 ml HCl and take to dryness. Do not bake. Rotate the beaker to insure complete dryness. Add 2 ml HCl, boil to 1 ml, and transfer to the same electroplating cell using a new platinum disc as the cathode. Rinse the beaker with two 1/2-ml HCl washes and one 1/2-ml water wash. Transfer each wash to the plating cell (note c).
  - c. Keep the plating solution at minimum volume during this transfer and also during the titration.
- 10. Repeat steps 2 through 9 to obtain second plate.
- 11. Repeat steps 2 through 7 to obtain third plate.
- 12. The three successive platings from each aliquot are counted on a calibrated alpha spectrometer. The total Pu<sup>236</sup> dpm on each plate is added to determine the average Pu<sup>236</sup> tracer concentration. Calculate the tracer stock concentration as of January 1 of the current year.

- 2. Pipet 1 ml each of R.C. Pu<sup>236</sup> Low Level and High Level Stock Solution and 1 ml each of Pu<sup>239</sup> Stock Solution (a) 90 dpm) into two 50-ml beakers.
- Add 10 ml of H<sub>2</sub>O, 1 ml Con. HClO<sub>4</sub>, 2 ml Con. H<sub>2</sub>SO<sub>4</sub>, 10 mg Fe<sup>† 3</sup>
   and evaporate to dryness.
- 3. Wash the sides of the beaker with HCl and evaporate to near dryness.

  Take up in 1-2 ml HCl and transfer to a 40-ml centrifuge cone with H2O and HCl washes.
- 4. Ppt Fe(OH)<sub>3</sub> with NH<sub>4</sub>OH and centrifuge. Wash ppt with H<sub>2</sub>O containing a few drops NH<sub>4</sub>OH and centrifuge. Dissolve the Fe(UH)<sub>3</sub> in a few drops HNO<sub>3</sub> and dilute to 5 ml with 6N HNO<sub>3</sub>. Add~1/2 ml saturated NaBrO<sub>3</sub> solution.
- 5. Warm on hot water bath a few minutes. Saturate the solution with NH<sub>4</sub>NO<sub>3</sub> crystals, add 5 ml hexone and stir with a mechanical stirrer 3 min. Repeat hexone extraction twice adding Sat. NaBrO<sub>3</sub> and more NH<sub>4</sub>NO<sub>3</sub> crystals as necessary.
- 6. Wash the hexone phases by stirring with 5 ml 6N HNO<sub>3</sub> for 1 min and discard the washes. Back extract the Pu with three 5 ml additions of 0.1N HNO<sub>3</sub>. Transfer the aqueous phase to a 50-ml glass beaker.
- 7. Evaporate the solution containing the heavy element tracer and activity to approximately 1 ml. Add 1 ml HNO3 and 1 ml HCl. Evaporate to wet dryness. Repeat the HNO3 HCl treatment twice (Note a).
  - a. Repetition of HCl-HNO<sub>3</sub> treatment is not necessary for the plating of uranium.

with PuZ39 Tracer

- 8. Pick up in 1 ml HCl and take to dryness. Do not bake. Rotate the beaker to insure complete dryness. Add 2 ml HCl, boil to 1 ml, and transfer to a prepared electroplating cell (note b). Rinse the beaker with two 1/2-ml HCl washes and one 1/2-ml water wash. Transfer each wash to the plating cell (note c).
  - b. The platinum disc and anode must be freed of any grease film
    by ringing several times with acetone and alcohol. Write the
    sample identification on the back of the disc. Ignite to red heat
    in a Fisher burner flame. The electroplating cell must be clean
    and free of any foreign material. Check for leakage before use.
  - c. Keep the plating solution at minimum volume during this transfer and also during the titration.
- Add 1 drop methy<sup>1</sup> red indicator. Add NH<sub>4</sub>OH dropwise until the indicator shows the solution to be basic (yellow). Add 2N HCl dropwise until the solution is just acid. Add 1 drop in excess.
- 10. Place the sample on the Sargent-Slomin electroplater. Adjust the anode to not more than 1/4 inch above the platinum disc. Plate for 10 to 15 minutes at a starting current of 2.5 amps and about 5 volts.

  The current may fluctuate during the plating period. Check occasionally and adjust the current to maintain 2.6 amps throughout the plating period (notes d and e).
  - d. Fifteen minutes plating time is required for Pa.
  - e. Twenty minutes plating time is required for T.P. (Am-Cm) samples.

with Pu<sup>239</sup> Tracer

- 11. At the end of the electroplating period, add 1 ml NH<sub>4</sub>OH. Stir for 15 seconds. Turn off the current and stirrer. Remove the anode from plating solution.
- 12. Immediately transfer the plating solution into the beaker used for evaporation. Rinse the inside of the plating cell 3 times with water washes. Combine the washes with the plating solution in the beaker.
- 13. Dismantle the plating cell and remove the platinum disc. Rinse with alcohol and ignite the disc to red heat; cool. Check yield on laboratory alpha counter before submitting sample to the counting room.
- 14. Place sample in a lined and labeled tin box and submit for counting analysis.
- 15. Determine the Pu<sup>236</sup> to Pu<sup>238</sup> and Pu<sup>239</sup> ratios by alpha pulse height analysis. A correction factor for the very small amount of Pu<sup>238</sup> and Pu<sup>239</sup> added to each sample is determined from this data.

# APPENDIX C SAMPLE DATA SHEETS

QUALITY CONTROL - SAMPLE DEFICIENCY REPORT	Sample No.
To be initiated by first employee to become aware of deficiency and then reported to supervisor:  (Fill out in pencil)	deficiency and then reported to supervisor:
1. Replicate nonagreement, 6 · 1.	. Sample contaminated Sample cross contaminated
r. ble checked in error on mixup.	
7. Computer processing.	
RESPONSIBILITY (List person(s) who will initial form when notified)	n notified)
1. Calculation 4.	Decontaining tion
3. Electronics 6.	Activity or carrier aliquoting Other
REASON FOR DEFICIENCY	
DISPOSITION	
1. Recount (circle) LBG, MEW, Gross a, a spac, P-20, y-spec. 2. Rework (indicate method) 3. Rewelgh (indicate method) 4. New Allquot	2G, γ- • p• ε.
OVAL AND ACTION (Circle applied	ible people) Cher

er, Sample Desig Kit Just I	
WITH MEASUREAGE	CHEMISTICY
TYPICAL DATA RESULTS FOR TISSUE WITH MEASUREALER. Sample Design	PU-239 CONTENT

Est Yield 92%	Sample Frac / 0	Total Tracer 25. 74 doing	scorched acratched		
Tree by D. O. Bale 24/13 Tracer by D.B. Bate 19/1/13 Dies in by Jeff. Bate 19/1/13 Dies Alle 92 %	Minister Court 27 AL Sample Frac 1.0		Cont. Lines Alexander beave dire	Diss'n Type TEME Plate Qual dented Glean st. dirty mod dirty	
1/2 Tracer by Duly Buth 1911	to nor Heat's Asset por	Dilution 500 Mb Uranian Aliq 50 Mb	Vol. Tracer/Suct. J.P. elle.	Plate Qual dented CLICAD 41	
Pres by Dad. Bally	Farinated 14" 20 = 30 days nor leg's seed of	Dilution 500 HL	Sample Sections	Diss'n Type TEMES	Stip. ActStyle

20 Total Counts 67/17  * 1.8 1.8  Net ents [Pul <sup>34</sup> Ful <sup>37</sup> ]/Geoxtime : 39 3 201	LYSIS FOR PU Min. Run 2760 Date 342-240-63	Puin  Tot. counts in channels	94.72    0.747    4.0   " 63.27    time x Geo x yield x aliq.   Net Counts   B	Chemistry Siepls of - 10318 (Bull Luse)
Date 3 57, 8 34-63 Minutes Run  2, 52  2, 52  Net cpm  Net	SHT ANA		0.747	06-06
Inst 2 T 3 Gross cpm	7	Puin Tracer Total counts in channels 15-56 : 1281 Background Counts from Met Counts . 1280	1 84.72)( 23.74 )( 0.76k time x Geo tracer(dpm) corr Net counts/A	TRACER FRACTIONS

Counting SECTER WILLE

90-90 2

TRACER FRACTIONS Sample Activity (dpm) 0-50

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00

Calculations

2000-4000

Sample Activity (Apr., 90-2000

Tracer Addition (dpm)

H. Tracer Addition (dpm) 400 ... 5 4000 requires aliquol

PICAL DATA RESULTS FOR TISSUE  TH UNDETECTABLE PU-239  CHEMISTRY  DAILY AND Tracer by Day B. Ministry  Daily A. Tracer by Day B. Minister Count  Lodge I are Regid 20 down Minister Minister 20 down  Cologn Minister Minis	20 Total Counts 270  2 / 938 S.F. s / 200	10n FOR PU  10n 600 Date 091. 45-45	Error (dpm) : 180.3 **Total Error : 62.7	3 43 '
TA RESULTS FOR TISSUE  CHEMISTRY  CHEMISTRY  CHEMISTRY  CHANGE BLE PU-239  CHEMISTRY  CHANGE BY AND BOTH BOTH BOTH BY AND BOTH BY AND BOTH BY AND BY AND BOTH BY AND BY AN	Date OST, BOS - Get Minutes Run 50 23 1 Net com Net com Net com	ALPHA PULSE HEIGHT ANALYSIS FOR PU  Pulm  Pulm  Tot. counts in channels  Background  Counts from  Net Coat	2.857 yield x Geo x yield x sliq 2.857 yield Ne	S 50-90 Chemistry States  Bu Counting AB AC  Austr-1000 Catculations  800 • REVIEWED BY  C.1
TYPICAL DATA RESULTS FOR TISSUE WITH UNDETECTABLE PU-239 CONTENT CONTENT CONTENT District Pu'' O-20 dpd 1 1111 Req'd 20 dpd Mnnil Dilution Abasé Sample Sections Vol. Tracer/Sact. 1011 District Type 7262 Plate Qual. dented (clean) al. dirty Stip. Act. Abasé	9.00 Date 082.4 Grass cpm Bad. cpm	STC  Julik Tracer  Jolal counts in channels  Jackground  Jounts from  Net Counts	2)/.8)( 20.18 )( 0.942 ) .  me x Geo tracer(dpm) corr  Net counts/A	TRACER FRACTIONS  L. Sample Activity (dpm) 0-50  L. Tracer Addition (dpm) 10  J. Tracer Addition (dpm) 400  L. Tracer Addition (dpm) 400

## APPENDIX D TECHNICAL PAPER FOR HANFORD SYMPOSIUM AND 9TH ANNUAL MEETING HEALTH PHYSICS SOCIETY

ROUTINE DETERMINATION OF PLUTONIUM BY TRACER TECHNIQUES IN LARGE BIOLOGICAL SAMPLES \*

by

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#### ABSTRACT

A precision tracer procedure was developed for the rapid analysis of non-uniformly distributed plutonium in large biological somples. Liver, lung, kidney, lymph node, trachea, gastrointestinal tract, masal mucosa, pharyngeal mucosa, bone, urine and feces samples from burros, sheep, and dogs (exposed to plutonium aerosols) were assayed for plutonium. The chemical procedure consisted of: equilibration of sample plutonium with <sup>236</sup>Pu tracer, wet ashing by refluxing with  $H_2SO_4$  and a catalyst; extraction of plutonium from bulk salts with cupferron-chloroform; purification with ion-exchange resins; and electrodeposition on platinum. These procedures minimized the requisite volume of acids and avoided the violent exothermic reactions of some wet ashing procedures. Problems associated with dry ashing, such as loss of the radioisotope by entrainment in solid carbon particles, and formation of insoluble oxides of plutonium, were avoided. Also, the need for the large ashing furnaces was obviated.

This report is based on work performed under Contract DA-49-146-XZ-192, Mod 4, between the Defense Atomic Support Agency and Tracerlab, A Division of Laboratory for Electronics, Inc.

Measurement of the plutonium content was accomplished by tracer yielding and alpha pulse-height analysis. This method ensured a high degree of accuracy, high sensitivity, and freedom from interference from other alpha emitters. A typical chemical yield was 55%, and the counting precision was within 3%. Limits of detection were approximately 0.05 dpm for a thousand-minute count.

#### INTRODUCTION

Procedures for the determination of picocurie amounts of plutonium in biological materials are abundant in the literature.

Inherent drawbacks to these procedures are: procedures are limited to specific types and amounts of samples; dissolution and purification losses are erratic, usually requiring calibration of (Reference 9) chemists /; and the chemical procedure must often be revised for each sample. Weiss and Shipman / have used Pu tracer to follow procedural steps in urinalysis. Painter / has used Pu tracer to follow activity distribution in dogs. In neither case, however, was the tracer used to yield another isotope of plutonium.

Attempts to minimize plutonium losses sometimes result in a product which contains interfering alpha emitters such as the 241 Am daughter of 241 Pu, present in many plutonium samples. (Reference 12) Hollstein / mentions contamination from uranium and Weiss and Shipman mention 232 Th, 238 U, 231 Pa, and 237 Np contamination in (Reference 13) urine samples. Other workers such as Sanders and Leidt / have expressed dissatisfaction with processing methods. Kooi and (Reference 14) Hollstein / were dissatisfied with erratic plutonium recovery and the inapplicability of available procedures to water containing large concentrations of iron or calcium.

By means of a 236 Pu tracer technique, we have successfully determined sub micro amounts of plutonium in nuclear debris,

atmospheric filters, soil, dry fallout, rainwater, food, environmental samples, and neutron-activated materials for the past twelve years. In this procedure a known quantity of 236 Pu is equilibrated with sample plutonium, and 236 Pu as well as the heavier plutonium isotopes are later determined by alpha spectrometry. Positive identification of all isotopes and an accurate correction for processing losses are assured. The method is adaptable to most samples containing plutonium, since 236 Pu is usually absent.

This tracer method has been combined with an appropriate bioassay procedure for routine determination of non-uniformly distributed plutonium in large biological samples. In a recent program burros, sheep, and dogs were exposed to plutonium aerosols to simulate uptake, deposition, retention, and translocation of plutonium in humans exposed to plutonium aerosols from a non-nuclear detonation. Over 600 liver, lung, kidney, lymph node, traches, gastrointestinal tract, nasal mucosa, pharyngeal mucosa, bone, urine, and feces samples from the animals were analyzed by our lab for plutonium content. These samples ranged from a few ounces to 15 pounds and contained from 0 to 60.00 dpm of 239, 240 Pu.

#### EXPERIMENTAL

### Preparation of 236 Pu Tracer

The <sup>236</sup>Pu tracer was prepared in a cyclotron irradiation and chemically purified at Tracerlab. No measureable <sup>239,240</sup>Pu or <sup>238</sup>Pu was apparent after purification. Both exhaustive electrodeposition and isotopic dilution methods were used to standardize the tracer.

In exhaustive electrodeposition, four aliquots were withdrawn from the purified stock and the tracer electrodeposited on a platinum disc. The plates were counted without collimators in our Frisch grid chambers and tracer activity corrected for chamber counting efficiency. Electrodeposition and counting were repeated on the plating supernate until a plate relatively free of activity was obtained. Three platings were usually required. Summation of the electrodepositions gave the tracer concentration.

In isotopic dilution a spike of National Bureau of Standards 239, 240 Pu stock solution\* (99, 97% pure) was added, for yielding, to nine aliquots of the purified 236 Pu stock solution. The spike and tracer were equilibrated by evaporation with H<sub>2</sub>SO<sub>4</sub> and HClO<sub>4</sub> and isolated by the procedure described below. The plated samples were counted and the 236 Pu concentration calculated after 239 Pu yielding.

Exhaustive electrodeposition gave an average concentration of  $25.0 \pm 0.38 \, \mathrm{dpm/ml}$  236 Pu for the four aliquots. Isotopic dilution gave an average 25.7  $\pm$ 0.26 dpm/ml for the nine aliquots. Experience has shown that the first method is susceptible to low results due to sequential handling losses. This point has been confirmed by standardization of the tracer, using a combination of both techniques on the same aliquots of tracer.

#### Sample Dissolution

The choice of dry or wet ashing is often a matter of personal pre(Reference 15)
ference or of facilities available. Comar / discusses the relative
(Reference 16)
merits of each and usually favors dry ashing. Greenberg / recommends the convenience of dry ashing. Wet-versus-dry ashing was

<sup>\*</sup>An analysis of the NBS standard (listed as 99.97% pure) on our Mass Spectrometer gave the following isotopic composition: 94,386 weight % <sup>239</sup>Pu; 5,271 weight % <sup>240</sup>Pu; and 0,343 weight % <sup>241</sup>Pu. The <sup>239</sup>Pu, <sup>240</sup>Pu alpha disintegration rate of the solution was calculated from this data.

experimentally compared. Dry ashing resulted in losses from spattering and physical entrainment of plutonium in solid carbon particles. It was assumed the greatest losses would occur in samples having high organic-to-ash ratios, such as soft tissues. Losses from the formation of difficultly soluble compounds of plutonium at high ashing temperatures (Reference 17) were also suspected. Toribara and Predmore / found urine, bone, and soft-tissue ash readily soluble in strong HCl, while fecal and food ash contained a difficultly soluble residue which sometimes trapped 97% of the plutonium, requiring a drastic leaching and fusion for dissolution.

In dry ashing, sample plutonium and tracer cannot be equilibrated until after the ashing process; and losses occurring at this stage result in inaccurate sample yielding. Wet ashing of tissue and fluid samples in the presence of 236 Pu tracer was tried and gave the desired results. The procedure, and adaption of the Kjeldahl method, utilized the oxidizing power of concentrated sulfuric acid, a mercury catalyst, and refluxing at high temperatures. The apparatus consisted simply of large beakers covered with inverted funnels (Figure D.1). Unaccountable sample plutonium losses were eliminated by addition of the tracer at the start. Equilibration of sample plutonium and tracer was assured by dissolution of the sample in the strongly oxidizing media. Requisite volumes of acids and violent exothermic reactions of some wet ashing procedures were minimized. After a clear solution was obtained, the excess acid was evaporated.

Bone samples were dry ashed, since they have a low organic-toash ratio, and the ash serves as a carrier to prevent loss of plutonium during high-temperature ashing. The bone ash was easily removed from the ashing container by acid dissolution and then equilibrated with plutonium tracer.

#### Plutonium Isolation

The residue from wet asking usually contained large amounts of inorganic salts. Most of the bulk salts were separated by extracting the plutonium, reduced to the trivalent state with hydroxlamine hydrochloride, into a cupierron-chloroform solution. The extraction was a (Reference 9) scaled-up version of the method outlined by Beaufait and Lukens/ (Reference 18) and Langham. The residual salts were further reduced by co-precipitation of plutonium with Fe(OH) 3 from a hot basic carbonate solution and then from an ammoniacal solution.

The plutonium was separated from iron and traces of other metallic sons by ion-exchange reson-absorption and slutium with (Reference 19)

HCl-NH<sub>4</sub>I. The eluent was evaporated to low volume and treated with HNO<sub>3</sub>, HClO<sub>4</sub>, and HCl to remove iodine and residual sesin particles. If a residue persisted, the sample was recycled through the chemistry procedure, as a visible residue at this point would result in a dirty plate in the procedure below.

(Reference 20)
A rapid electrodeposition procedure reported by Mitchell/
was used to obtain from the clear solution a weightless, invisible deposit
of plutonium on a platinum disc with a plating time of 10 minutes. The
disc was 5 mils thick, with a mirror finish, pre-cut to 2, 2 cm in diameter.
The electrodeposition cell, designed by our laboratory, limited the
plating solution exposure to the glass tower, teflon gasket, and platinum
disc.

Sequential outlines of the sample processing are given (Figures D.2 and D.3, and a detailed chemical procedure is appended.

#### Activity Measurements

Each electrodeposited plutonium sample was counted on an alpha pulse-height analyzer. For this purpose, the outputs from four Frisch grid chambers (Tracerlab Model RLD-1) were connected

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to one multi-channel analyzer (Technical Measurements Corp., Model CN-116) by dividing the full range of 256 channels into quadrants of o4 channels each.

The chamber was calibrated for alpha energy using a multi-peak alpha source containing 239 Pu, 241 Am, and 236 Pu (Tracerlab R-37 Alpha Spectrometer Kit). The calibration was made immediately before and after each sample was counted. This provided a check for instrument drift during the sample counting interval. The instrument controls were adjusted so that the 64 channels covered an energy range of 4.5 to 6.0 Mev. This range included the isotope energies of 239 Pu (5.14 Mev), 240 Pu (5.16 Mev), 238 Pu (5.49 Mev), and 236 Pu (5.75 Mev). The amplifier gain setting gave a scale factor of 37 Kev per channel. Each isotope present was registered over a spread of approximately ten channels.

A disposable metal collimating ring, surrounding each sample disc, was used with each sample to preclude the counting of degraded alpha particles. Some loss in counting efficiency resulted but was off-set by improved peak contours and distinct separation of alpha energy peaks. The resolution (full width at half-maximum) of the four Frisch Gridchambers, including disc collimation, was 0.88% at 5.15 Mev. The alpha peak counting efficiency was approximately 35%.

The counting time for an unknown sample was determined by the isotope having the lowest activity. If activity levels permitted, the lowest activity peak was counted to a standard error of within 3%.

The maximum counting time for any sample was limited to 1000 minutes.

239, 240

Calculation of Pu

The results of the alpha pulse-height analyses are presented on printed tape. A graphical plot of a typical spectrum for a tissue con-

taining a moderate amount of <sup>239</sup>Pu is illustrated in Figure D.4. The energy calibration line was calculated from the pre-and postcounting energy calibrations of the counting chamber. A summation was made of counts under each isotope peak present. These counts were corrected for low energy tail, background, peak resolution, and instrument drift. The plutonium content of the sample was c loulated by:

Pu and Pu could not be calculated separately, as their alpha energies were too close to resolve with a Frisch-Grid Chamber.

The counting efficiency of each Friech-Grid Chamber was measured, using a high precision alpha standard, and it was not necessary to calculate a yield to determine the plutonium content. However, the yield was usually determined as a quality control measure in order to assess the efficiency of the chemistry procedure.

Quality\_Control

The biological specimens, received in the form of frozen samples sealed in plastic bags, were stored in a walk-in freezer prior to processing. Samples were analyzed in a low-level radiochemical laboratory to prevent contamination from outside sources and processed with adequate spacing using all new glassware to prevent cross-contamination. The few samples which were expected to be higher in plutonium content, by virtue of their exposure, were processed separately.

All reagents were made up fresh and analyzed for any laboratory blank. Simulated samples (prepared by adding a 236 Pu spike to a matrix similar to the animal specimens) were analyzed to check on procedures. The laboratory blanks were effectively background. For an optimum

sample counting time of 1,000 minutes, the blanks were found to be in the region of 0.05 dpm, which is the limit of detection for the instrument.

All activity measurements were performed in an isolated and countrolled area. The background count rate and counting efficiency of each Frisch-Grid Chamber was checked periodically. In the 239, 240 Pu energy region, the background count was approximately 0.001 cpm.

#### RESULTS

Based on the samples analyzed, the reflux wet ashing-tracer procedure has been found to be successful. It was moderately time-consuming and expensive, however. Ordinarily 3 to 5 days were required to dissolve 16 to 20 large tissue samples. A combination of  $NO_3$ ,  $H_2SO_4$ , fuming-HNO<sub>3</sub> (Reference 21) and HClO<sub>4</sub> dissolutions in open beakers was substituted for smaller samples./

A tabulation of radiochemical yields are listed in Table D.1. Generally, samples 0.1 to 4 ounces and 4 to 14 ounces averaged 55% and 48%, respectively. Samples 1 to 9 pounds averaged approximately 50%. Yields from samples up to 15 pounds are still to be determined. Averages reported include some low yield samples which were processed before our procedures had been fully developed. Recent data show improved yields in all sample categories.

The use of tracer did not lessen the difficulties of recovering such minute amounts as 0.1 dpm  $(7 \times 10^{-13} \text{ grams})$  of plutonium from pounds of material. Losses of sample plutonium are always a possibility, but with addition of tracer, these losses are positively known. Thus, if the entire sample was lost due to undetected processing irregularities, the sample was reported as lost and not as zero dpm of plutonium.

The sample activities measured in this program are not germane to this paper, since samples came from both exposed and unexposed

animals. Some animal organs were found to contain levels of plutonium below our limits of detection, which is nominally 0.05 dpm. The lowest samples showed activity from 0.05 to 0.10 dpm.

Variations in yields introduced no special problems, since each sample was individually yielded. However, 50 to 70% was considered most desirable for obtaining optimum sensitivity. The fluctuations in yields emphasize the inaccuracies which would have resulted had the work been done without tracer. Also, in retrospect, an adequate tracer-free plutonium assay procedure would have been very difficult to develop.

#### **ACKNOWLEDGEMENTS**

The valuable contributions of our scientists in this program are too numerous to credit. However, the authors wish to thank Dr. C. E. Gleit, D. W. Billings, C. E. Ensor, and Mrs. D. E. Hawkinson, for their exceptional assistance.

#### CHEMICAL PROCEDURE

- 1. Remove the frozen sample from its polyethylene bag and allow it to thaw for a few minutes under a heat lamp. If the sample shape or size is such that it will not fit in the bottom half of a fourliter beaker, cut the sample into appropriate sections and place each section in a separate beaker. Rinse the plastic bag with HNO<sub>3</sub> and add the washings to the beaker.
- 2. Add enough H<sub>2</sub>SO<sub>4</sub> to completely cover the sample. Pipet an appropriate amount of <sup>236</sup>Pu tracer (within a factor of five of the expected sample activity but a minimum of 15 dpm) into each beaker and add approximately 5 grams K<sub>2</sub>SO<sub>4</sub> and 2 to 3 drops Hg metal. Spray the sample with Dow-Corning Antifoam A silicone defoamer.
- 3. Attach the stem of an inverted 6-inch funnel to a ring stand and clamp and lower the funnel mouth into the beaker. Secure a few inches above the sample.
- 4. Digest the sample gently with low heat until a black tarry mixture is obtained. Increase the heat gradually and reflux until the mixture is a clear solution. The tarry mixture will turn to a black jelly, a black liquid, a red liquid, and finally, a clear solution. Raise or lower the funnel during dissolution to control the reflux action. Wash down any carbonacecus material on the beaker and funnel walls with H<sub>2</sub>SO<sub>4</sub>.
- 5. Evaporate the H<sub>2</sub>SO<sub>4</sub> until salts start forming. Remove the funnel and add HNO<sub>3</sub> cautiously to cool the solution. If the sample had been divided into sections for the dissolution, combine the sections into one beaker and evaporate to low volume. Cool and transfer the sample with water to a large polyethlene bottle

- or a 2-liter acid bottle. Dilute to approximately 500 ml.
- 6. Add 10 ml sat. NH<sub>2</sub>OH·HCl to the diluted sample and about 50 ml CHCl<sub>3</sub>. Stir with a mechanical stirrer for 1 minute at high speed. Add 75 to 100 ml 6% aqueous solution of cupferron and stir again at high speed for 5 minutes.
- 7. Transfer the mixture into four 250-ml polyethlene bottles and centrifuge to separate the phases. Transfer the CHCl<sub>3</sub> extracts with a transfer pipet to a 400-ml beaker, taking care not to disturb the interfaces. Repeat the extraction, without addition of more cupferron, until the CHCl<sub>3</sub> phases are colorless.
- 8. Boil the CHCl<sub>3</sub> collections to low volume (approximately 3 ml) and allow contents to cool. Rinse down the sides of the beaker with HNO<sub>3</sub> and boil to approximately 3 ml. Rinse the sides with furning nitric and boil down again (avoid dryness). Add 10 ml turning nitric and 25-30 ml HClO<sub>4</sub>. Cautiously heat until an exothermic HClO<sub>4</sub> reaction begins. Remove the beaker from the bot plate and allow the reaction to go to completion.
- Fume the HClO<sub>4</sub> to low volume. Gool the solution and transfer with water washes to a 40-ml centrifuge cone.
- 10. Add 10 mg Fe unless the sample is known to contain that much or more. Add 19N NaOH until the solution is basic to precipitate metal ions. Add 5 ml sat. Na<sub>2</sub>CO<sub>3</sub> solution and digest the precipitate in a hot water bath for 10 minutes. Centrifuge and decant the supernate to waste. Dissolve the precipitate in HNO<sub>3</sub> and dilute to approximately 30 ml.
- 11. Make the solution basic with NH<sub>4</sub>OH and digest the precipitate in a hot water bath for 10 minutes. Centrifuge and decant the supernate to waste. Wash the precipitate twice with 10-ml portions of H<sub>2</sub>O containing 1 drop NH<sub>4</sub>OH.

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- Dissolve the precipitate in a minimum of HNO<sub>3</sub> and add 10 ml
   6N HNO<sub>3</sub>.
- 13. Prepare a 100 to 200 mesh Dowex 1-X 10 resin column by adding approximately 1/2 inch of resin to a tubulated glass column, 12 mm I.D. and 85 mm in length, containing a Dacron wool plug at the bottom. Insert another plug at the top and pre-condition the column with 40 ml 6N HNO3.
- 14. Pour the solution from step 12 onto the column. Wash the centrifuge tube with 20 ml 6N HNO3, followed by 20 ml HCl and add the washings to the column.
- 15. Allow the column to drain and elute the plutonium into a 50-ml beaker with 30 ml of freshly prepared HCl-NH<sub>4</sub>I (approximately 50 mg NH<sub>4</sub>I per 30 ml HCl).
- 16. Evaporate the cluate to approximately 5 ml and add 5 ml HNO<sub>3</sub> to destroy I<sup>\*</sup>. Add 3 ml HClO<sub>4</sub> and evaporate to wet dryness.
- 17. Take up the residue in 1 ml HCl and evaporate to dryness. Do not bake. Rotate the beaker to insure complete dryness. Add 2 ml HCl, boil to 1 ml, and transfer to a propared electroplating cell (note a). Rinse the beaker with two 1/2-ml HCl washes and one 1/2-ml water wash. Transfer each wash to the plating cell (note b).
  - a. The platinum disc and anode must be freed of any grease film by rinsing several times with acetone and alcohol. Write the sample identification on the back of the disc. Ignite to red heat in a Fisher burner flame. The electroplating cell must be clean and free of any foreign material. Check for leakage before use.
  - b. Keep the plating solution at minimum volume during this transfer and also during the titration.

- 18. Add drop methyl red indicator. Add NH<sub>4</sub>OH dropwise until the indicator shows the solution to be basic (yellow). Add 2N HCl dropwise until the solution is just acid. Add 1 drop in excess.
- 19. Place the sample on a Sargent-Slomin electrolytic analyzer.

  Adjust the rotating anode to approximately 1/4 inch above the platinum disc. Plate for 10 minutes at a starting current of 2.5 amp and approximately 5 volts. The current may fluctuate during the plating period. Check occasionally and adjust the current to maintain 2.6 amp throughout the plating operation.
- 20. At the end of the electroplating period, add 1 ml NH<sub>4</sub>OH. Stir for 15 seconds. Turn off the current and stirrer. Remove the anode from the plating solution.
- 21. Immediately transfer the plating solution into the beaker used for evaporation. Rinse the inside of the plating cell 3 times with water washes. Combine the washes with the plating solution in the beaker.
- 22. Dismantle the plating cell and remove the platinum disc. Rinse with alcohol and ignite the disc to red heat.
- 23. Place disc in a lined and labeled tin box and submit for alpha pulse height analysis.

Sample Weight Range	Tissue *	Number of Samples	Pu Recovery Range	Pu Recovery Average
Ounces			4	1,
0.2-0.3	Hilar Node (D)	12	44-74	61
0, 3- 0.4	Hilar Node (S)	9	21-62	43
0.2-0.9	Hilar Node (B)	12	33-77	60
0.4- 0.8	Phar. Mucosa (D)	15	37-75	65
0.7-0.8	Nasal Mucosa (D)	2	50-79	65
1, 1- 1, 6	Traches (D)	9	39-85	58
1, 4- 2, 3	Kidney (D)	14	24-88	51
1, 7- 3, 2	Phar. Mucosa (B)	6	32-68	49
2, 4- 3, 6	Lungs (D)	9	28-64	49
3. 3- 4. 0	Kidney (S)	6	40-72	55
4.8-6.4	Liver (S)	17	26-80	47
5. 3= 8. 5	Controls	ŽÓ	24-74	52
6, 1- 7, 3	Bone (S)	18	24-69	42
7, 8-14.	Liver (D)	10	30-77	50
Pounds			7.	*
0, 9- 1, 1	Trachea (B)	5	42-70	56
1.0- 2.5	Kidney (B)	16	21-80	49
1.9-3.3	Bone (B)	24	23-75	54
3.6-10.2	Lung (B)	3	34-75	52
4.4- 8,9	Liver (B)	10	37-57	48
	TOTAL	223	21-88	52

<sup>\*</sup> D = Dog, S = Sheep, B = Burro

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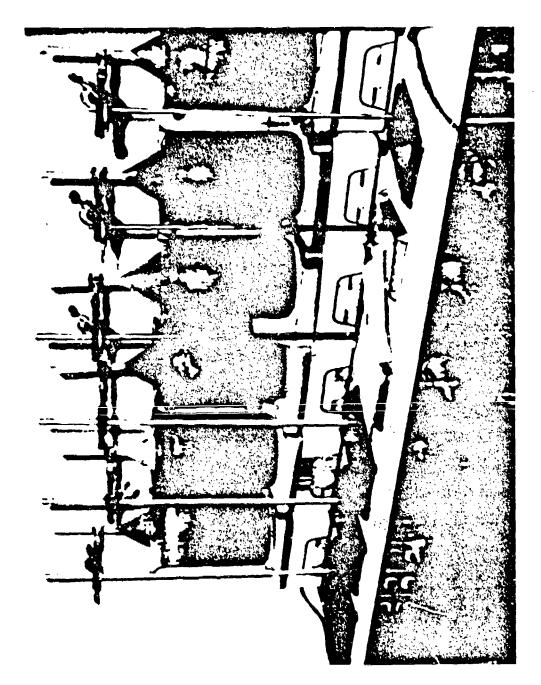


Figure D.1 Reflux apparatus for biological sample. (Traceriab photo)

```
Frozen Sample
            Thaw
           Section
           HIS C.
           Tracer
           K,SQ, (Solid)
Diss.
           Hg metal catalyst
            Defoamer
            Digest with heat
            Reflux
                                                          Organic Matter
       Clear Solution
           Evaporate to salts
           HNO,
            Combine sections
Sep'n.
            Transfer to large polyethlene bottle
            Satd. NH2OH · HCI
            CHCl; - Cupferron extraction
            Repeat extraction - Transfer to beaker
                                                      -> Insoluble Salts
       Metal Ions in Organic Extractant
            Evaporate to low volume
            HNO, and f-HNO, evaporations
            HClO, - exothermic reaction
            H2O - Transfer to centrifuge cone
                                                      -> Organic Extractant
       Metal Ions in Aqueous Solution
            Fe<sup>+</sup>,
            19N NaQH
            Said. Na<sub>2</sub>CO<sub>3</sub>
                                                       -> Uranium Carbonate Comple:
       Metal Hydroxides and Carbonates
            HNO,
            NH, OH
                                                       → Soluble Salts
       Metal Hydroxides and Carbonates
            6N HNO,
            Dowex 1-X10 Resin column
            6N HNO, - HC1
            HCl - NH4I elution
                                                        -> Metal Complexes
```

Figure D.2 Chemical dissolution, separation, and purification sequential scheme.

```
Pu+1.4
     Evaporate to low volume
     HNO, and HClO.
     Evaporate to wet dryness
     HC1
     Evaporate to dryness
     HCl, evaporate to lml
     Transfer to electroplating cell containing platinum disc
Pu<sup>†3, 4</sup> in plating cell
     Methyl Red Indicator
     HC HK
    2N HCl - ph adjustment
Trans z onto an electrolytic analyser
Cell in electrolytic analyser
     Anode adjustment - 1/4 inch above disc
Current - 2.5 amps., 5 volts
     Plate 10 - 20 minutes
     ИНОН
     Remove anode
     H<sub>2</sub>O wash
     Remove platinum disc
     EtOH
     Ignite
```

Figure D.3 Plutonium electroplating sequential scheme.

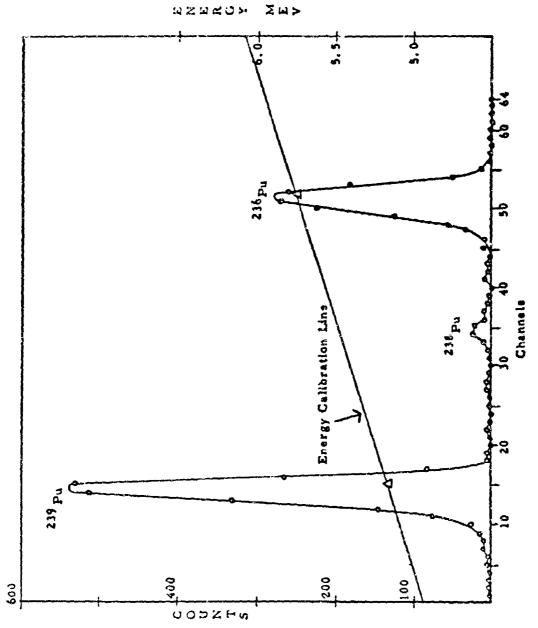


Figure D.4 Typical spectra biological sample (burro liver).

# APPENDIX E DATA TABLES FOR PU<sup>136</sup>, PU<sup>136</sup> AND URANIUM (U<sub>1</sub>O<sub>1</sub>) IN PHYSICAL, BIOLOGICAL, TRACER, AND QUALITY CONTROL SAMPLES

#### KEY TO PHYSICAL SAMPLE TYPE

TLW Anal, Desig.	Sample Type
CCD	Casella Disc
CCF	Casella Filter
CAD	Andersen Disc
CAF	Andersen Filter
CTA	Total Air Sample
CTD	Total Air Sample Disposable
CSA	Sequential Air Sample
CDS	Deposition Sample
CQC	Soil Samples (Quality Control)
<b>cws</b>	Water Samples
CAC	Aluminum Collectors
cvs	Vegetation (Sagebrush)
CSF	Soil Fractions
CBS	Balloon Wire Swipes

## KEY TO BIOLOGICAL SAMPLE TYPE

TLW Anal Desig.		
	Sample Type	4
RDB	Dog Bone	
RDK	" Kidney	
RDL	" Liver	
RDR	" Lung	
RDH	" Hilar Node	
RDT	" Trachea	
RDS	" G. I. Tract	
RDP	" Pharyngeal Mucosa	•
RDN	" Nasal Mucosa	
RSB	Sheep Bone	•
RSK	" Kidney	
RSL	" Liver	
RSR	" Lung	
RSH	" Hilar Node	
RST	" Trachea	
RSS	" G. I. Tract	
RSN	" Nasal Mucosa	
RSU	" Urine	
RSF	" Feces	٠

## KEY TO BIOLOGICAL SAMPLE TYPE (2)

TLW Anal, Design,	Sample Type
RBB	Burro Bone
RBK	" Kidney
RBL	" Liver
RBR	" Lung
RBH	" Hilar Node
RBT	" Trachea
RBS	" G. I. Tract
RBP	" Pharyngeal Mucosa
RBN	" Nasal Mucosa

TABLE E.1 RADIOCHEMICAL ANALYSIS OF ROLLER COASTER PHYSICAL SAMPLES, DOUBLE THACKS
ARC LOCATION 1 LW 1 LW 60-239,240

i		-								
ANAL JHON	<b>%</b> %	1.7E 03 9.6E 00 3.9E 00	# <del> </del>	1.CE 02	1. SE 03 3. CE-C2	4. 36-01 6. 66-01	10-31*6		2.6E 00 3.8E-01 1.55-01 9.0E-02	3. EE-01 6. SE-01 4. EE-01 1. EE-01
COUNT TIME	2C 2C	200	2 2 2	20 20	200	O 0 6	20°C	2 <b>4</b> 6	# 2 4 8 0 0 0 0 0 0 0 0	20C 2CC 2CC 50C
Y IELC IR *RE WORK )	81.2	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	0 41 41 0 41 41	71.3	42.4		0 4 4 6 17 2 8 6 17 2 8 6 17 2 8 18 18 18 18 18 18 18 18 18 18 18 18 1	10 10 10 10 10 10 10 10 10 10 10 10 10 1	87 - 51 80 	40 00 00 
UR AN IUM (FICRO GHAMS)		•	1.13		14.7	33.5	0.172 0.168	2.86 0.0610 1.44 0.665		
FU-239,240 ACTIVITY (DFP)	3.08 ±0.07E 07 1.29 ±0.03E 07	1.2940.03E 08 2.9640.08E 07 3.4641.38E 00	.7040.07E .4240.06E .1640.03E	.49 #0.12E	.43 t0.20E	.7740.24E	.08 t0 .03E	5.77 #0.01E 03 1.70 #0.23E 00 1.92 #0.08E 02 3.92 #0.09E 02	.27 40.13E .84 40.13E .66 40.22E .10 40.03E	2.18 t0.21E 00 9.88 t0.24E 02 2.49 t0.08E 01 3.80 t0.26E 00
TLN ANALYSI S NO.	CAC- 376 377	378 379 CD S-1719	CCO-2160 2206 CAC- 380	381 2069	382 CAD-2164	2210 CCD-2165	2168	2175 2219 CTA-2176 CCO-2216	1835 1836 1837 1838	CCF-1839 CCD-1855 1856 1857
TLW COLLECTION NO.	\$614 9615	8(19	~ ~		2 520-A	8 2522-A	~ (	5698-A 8 9459-A 9456-B	-19	5 9677-1 2 3
LOCATION	ں ب	AJ-C7 BL-C9 Bn-14	066 066 070	060	0 % 0	064 068 684	2 C C C	000000000000000000000000000000000000000	0 6 0 - 1 2 6 4	0 25 0 1 0 1 0 1 0
ARC	79		∢	ا ن	0		E.			9

Note: All Physical and Biological activity requits are given for the total sample except the "A" samples (identified in Table E.13) and transferred from Project 2, 6c for radiochemical analyses. These are deficient by the approximate amounts listed in Table E.13. Data were not combined, since in 2.6c, processing particles were not precisely analyzed or an unknown faction of the sample was removed.

\*\*\*

New data this report

1								
ARC	LOCATION	TLW COLLECTION NO.	TLW W ANALYSES	PU-239,240 ACTIVITY	URAN IUM (HICRO	Y IELC (R=1E	COUNT	AHAL JHON
•	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				CHARD	MURK	1	
ຶ່	9-050	9-11-4	CC0-1858	2.3540.09£ 01		40.6	200	6. F-01
	ĸ۸	50	7	.1441.		- C - C - C - C - C - C - C - C - C - C	200	A. (F-0)
	052-1	9666-1	CCD-1846	5.61 t0.19E 02		96.4	; ;	8. 7E - 01
	~	8	1847	.3046.		87.5	100	2.25-01
	m	•	~	.6940.			20C	5. (F-01
	<b>47</b>	<b>S</b>	CCF-1849	.35 #0		80.6	300	6.46.00
	1-250	9667-1	7	2 40.08 €		82.8	20C	4. CE-01
	7	7	1881	38		٠ د د د	200	7.36 00
	<b>~</b>	m	1852	4.42 to.22E 00		69.1	300	5. E-01
	₹ (	•		9.99 #1.00 E-01		5.20	366	8
	<b>5</b>	···		1.07 to. 10 £ 00		£7.C	300	
	052-1	1-8176	CCD-1860	3.0140.06E 03		65.6	2 C	, m
	~	~	1981	4.9240.14E 01		75.2	100	4. SE-C1
	P1 -	m	1862	1.6340.09E 01		17.75	2CC	4. 1E-C1
	•	•		3.1940.27E 00		24.7	200	2. 1E-C1
	<b>.</b>	·	CCF-1864	3.23 to. 13£ 00		64.3	336	1.45-01
	1-450	1-5936		1.1440.03E 03		74.5	70	9. FE-C1
	~	<b></b>	1842	1.57 t0.02E 02		£0.€	200	6. 1E-C1
	<b>~</b> ) .	m ·	1943	1.02 to.04 E 02		3.51	74	4.56-01
	•	. س	1844	1.03 to.06E 01		24.0	2CC	3.25-01
	n.	s,	CCF-1845	1.52 to.05 € 01		61.1	2CC	2. 4E-CE
	<b>1-6</b> 00	1-9636	CCD-1.885	2.91 to .09 € 01		7.44	2C C	2. JE-C1
	~ .	7	988			46.6	ęç	6. (E-CL
	η.	m	1887	.97		66.6	(F)	1.6. 00
	•		_	1.16 tc.05E 01		72.2	2C C	4E-
	•	<b>•</b>		.50		16.3	306	5.46-01
	9 5	9663	C TO-1840	-		33.6	<b>U</b>	1. 36 CO
	1-050	1-0826				2.33	3C	1.26 00
	~	~	1866	1.3910.046 02	٠	8C.5	74	6. (E-C1

TAB	TABLE E.1 (CC	(CONTINUED)			                 	1 0 1 1	!	• • • • • • • • • • • • • • • • • • • •
ARC	LOCATION	COLLECTION NO.	TLN ANALYSIS NG.	FU-239,240 ACTIVITY (0FP)	URAN IUM (MICRO GRAMS)	Y JELC I R "RE MOŘK J	COUNT TIME	ANAL /HON
; ;	*	• • • • • • • • • • • • • • • • • • •	• • • • • • • • • • • • • • • • • • •	! ! ! ! ! ! ! ! ! ! ! ! ! ! ! ! ! ! !	\$ 6 6 7 7 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	T	 	
ی	6-9-0	9680-3	CC0-1867	6.24 t0.15E 01		36.5	2CC	3E-
,	١	**	1868			36.1	30C	£6-
	. 40		CCF-1869			44	<b>300</b>	3. EE-01
		8-09-5	CCD-2170	4.84 t0.17E 02	4.44	~	46	Æ
		æ	2214	5.01 #G. 10E 01	0.308	€	20C	
		9461-A	7	1.76 t0.05E 03	10.8	47.5	3 C	
	060-1	1-4576	CC0-1880	3.51 to. 10E 03		W)	<u>۲</u>	
		~	1881	1.87 t0.05E 02		÷	4 (	
	· m		1882	8.7840.29€ 01		Š	ęc	
	•	*	1883	2.25 to.40E 01		02.8	<b>2</b> 00	4. 76-01
	· <b>v</b> n		CCF-1884	1.98 +0.076 01		0	2C C	
	062-1		0281-000	2.6010.05E 03		~	3C	
	7	~	181	1.61 10.03E 03		Q,	36	
	m	•	1872	1.3010.025 03		G,	40	
	•	4	1873	3.5940.09E 02		Ġ.	3 C	1. CE 00
	· <b>4</b> 1	<b>'</b>	CCF-1874	3.79 to.12E 02		_	22	
	064		CC0-2169	4.69 #0.07E 03	3.74	4	<b>6</b> C	
			2217	6.4210.136 01	0.273	ው	30Z	
	064-1	_	1875	4.2310.07E 03		~,	4	A. EE-01
		~	1876	.52 to.06 E		87.2	<b>7</b>	¥.
	•	<u></u>	1817	.53 to.08E		41	<b>1</b> C	
	•	•	1878	6.12 to .20 E 02		~	) <b>1</b>	4. SE-01
	***	**	CCF-1879	.10 to. 19E		19.1	36	3. 56-01
7	060	2123-A	CC0-2161	.9610.076	3.11	æ.	4	G
	0,0	œ	2207	.62 10.12E	1.48	m	<b>4</b> C	6. IE-01
	064		COS-1720	.03 10.128		~	300	8. (E-04
_	055-1		CC0-1825	.03 tO. 18E		<b>~</b>	~	7. 1E-01
	7	~	1826			e,	721	3. IE-01
	•	· m	1827	4.4110.136 01		41 41 61	C	2.56-01

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TABI	TABLE E.1 (CO)	(CONTINUED)				i 1 1	* • • • • • • • • • • • • • • • • • • •	+7 +
ARC	LOCATION	TLW COLLECTION	1LW ANALYSIS	PU-239,240 ACTIVITY	URAN IUM	Y IELC IR*RE	COUNT	ANAL JHON
!	: : : :	NO.		(260)	GRAMS )	MORK )	† 	
_	7-5-0		ġ	.4010.04E 0		•	400	. 16
•	- <b>L</b>			.27 #0.06 E O			200	, ję.
	1-250	1-6236	CC0-1820	2.0340.05E 03		85.6	20	7.26-01
	. ~		)	.78 to.11E 0		-	20	• (E-
	, er	•	1822	.65 10.03E 0		*	30 <b>2</b>	. 2E-
	1 4	•	1 823	.93 to. 12E 0		ň	20C	. 2E-
	· uri			.25 to .03E 0			2 C	. SE-
	057-1		CC0-1930	.04 t0.22E 0		ë	3C	
				.79 to. 16E 0		•	26	E
	, ,,	· •	1832	.14 to.04E 0		<u>:</u>	e c	. CE-
	•	· •	1833	.95 to. 18E 0		8	100	4.46-01
	· w			.95 to . 14E 0		¥	100	. 3E -
	650	-	C 14-2174	.40:0:036 0	ċ	31.4	30	
	061	_		.25 t0.04E 0	~	8	ξC	
	061	9668-A	6	.54 t0.03E 0	2.02	•	ر د	
7	850	2612-A	C40-2152	.26 t0.10E 0	Ψ,	_	Ų.	1. E-01
,	8 9 0	<b>6</b> 0	~	.95 to.11E 0	'n	4	Ų	2. TE-01
ب.		6643	CDS-1723	.88 to.11 E 0		e1.1	20	•
	034-3			.04 10.46E 0		40.6	400	1. CE-02
	•		1725	.35 tO.08 E O		49.7	400	. 3E
	'n		1726	.48 10.04E 0		21.3	300	
	046-1		1727	.38 t0.12E 0		*	30¢	
	~		1728	.54 t0.04E 0			26	£.
	- (**)		1729	.5340.15E 0			٥٢	'n.
	4		1730	.5410.18E 0		¥	36	
	•		1731	.89 tO.03E 0		•	<b>V</b>	. te
	010		1732	30 40.08E 0	· Are	39.4	40	1. te 00
	010-1		1733	.77 t0.17E 0	, s <sup>i</sup> c	•	<b>V</b>	æ
	~		1734	1.17 to . USE OL		48.3	400	

LEC TION ANALYSIS A NO.	ARC	LOCATION	7 × ×	11.1	-239,2	UPANIUM	YIELC	TKUCO	ANAL /MON	NO
070-3  8C49  CDS-1735  1C8  4  1734  1734  1735  1736  1737  174  1751		1 1 1 1 1 1	COLLEC :10N NO.	AN	ACT (VITY (OFP)	CRAMS )	MORK )	THE	***************************************	į
108	_	6-010	8649	73			71.4	366	2. 16	ជ
108	ı	• • <b>•</b>	•	~			48.5	400	3.36	5
108		. w		_			9.18	<b>308</b>	3. CE - (	- 03
030		i	BC19	-	1.28 to.21E 00		6.8.5	<b>30</b> 2		0
032-1 284-1 CCD-126 -3 3	z		2829	~	8.00 t6.00 E-02		40.3	<b>)</b> ) 6		00
2 127 -2 3 128 -2 4 4 129 1 129 -1 2 129 -1 2 129 -1 2 2 33 4 -0 3 3 3 4 -0 3 3 3 4 -0 4 4 337 1 6 6 6 338 1 7 CAF - 339 1 7 CAF - 339 1 7 CAF - 339 1 8 2 8 3 7 1 2 8 3 7 1 2 8 6 0 0 2 8 7 2 2 40 3 3 3 3 1 3 3 1 4 5 6 CCF - 135 3 5 5 CCF - 135 3 6 6 6 3 3 8 6 0 7 CAF - 3 8 6 0 7 CAF - 3 8 6 0 8 3 3 7 1 8 5 CCF - 13 5 3 9 3 3 1 3 3 1 7 CAF - 3 8 6 0 8 3 3 7 1 8 6 6 6 7 1 8 6 0 8 7 7 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		032-1	2684-1		-3.00 #6.00 E-02		76.4	302	CA 1. CE	0
2 CCF - 130 - 2 2 CCF - 130 - 2 2 CCF - 130 - 2 3 3 34 - 0 4 4 335 - 2 5 CCF - 130 - 2 3 3 3 4 0 4 5 337 - 1 5 CAF - 339 - 1 7 CAF - 339 - 1 7 CAF - 339 - 1 8 6 6 7 A - 2 4 2 8 2 8 2 6 7 A - 2 4 0 13 2 8 7 - 1 13 3 3 1 1 3 1 2 1 2 CCF - 135 3 3 1 4 5 CCF - 135 3 3 1 4 6 6 7 A - 2 4 0 13 2 8 7 1 1 13 3 2 2 13 3 3 3 1 1 3 1 13 4 6 7 6 7 6 7 8 3 1 13 4 7 7 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		~		_	-2.00#3.00E-U2		11.2	<b>302</b>		00
5 CCF - 130 2 E99-1 CAD - 334 2 2 3 3 34 4 4 6 6 139 5 CCF - 130 2 2 83		m	m	128	-2.0016.00E-02		75.3	<b>20C</b>		Ç
2 CCF - 130		4	•	129	1.00 16.00 E-02		76.8	<b>466</b>		CO
2 2 699-1 CAD-334 -0 2 3 3 336 3 4 4 337 4 5 6 6 338 7 CAF-339 1 2 2 8 3 C TA-2 42 2 2 8 6 C TA-2 40 2 2 8 6 C TA-2 40 3 3 3 1 3 3 2 4 4 5 5 C C F-1 3 5 2 2 6 8 9-1 C AD-3 2 8 2 2 2 6 8 9-1 C AD-3 2 8 3 3 3 3 3 3 1 3 5 4 5 5 C C F-1 3 5 5 5 6 C F-1 3 5 7 7 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		· <b>4</b> 0	· <b>v</b>	<b>-</b>			13.1	1000	2.26	00
2 335 2 3 3 336 4 4 4 337 1 6 6 8 338 1 7 CAF- 339 1 2 2 2 3 C TA- 2 4 2 2 2 2 6 C TA- 2 4 0 2 2 2 6 C TA- 2 4 0 3 3 1 3 3 2 4 4 4 6 C C C C T T T T T T T T T T T T T T T		034-1	-663	, W	-0.50 tl. 00 E-01		72.4	96		- 04
4 4 337 11 2 2 8 3 3 4 6 6 6 3 3 8 7 7 2 8 3 3 7 1 1 1 2 8 3 2 2 8 3 7 1 1 2 8 6 7 1 4 7 2 7 2 8 6 7 1 4 7 2 8 6 7 1 4 7 1 1 3 7 1 1 1 3 7 1 1 1 1 1 1 1 1 1 1		~	7	335	2.4040.905-01		56.5	300	CA 2. EE	- 02
4 4 337 1 6 4 337 1 2 2 3 3 C 1 A - 2 4 2 2 2 2 2 3 C 1 O - 3 8 6 0 2 2 2 6 C 1 A - 2 4 0 3 2 2 2 6 C 1 A - 2 4 0 3 2 2 2 6 C 1 A - 2 4 0 3 3 3 3 1 3 3 2 4 4 4 6 C C P - 1 3 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		•	m	336	0.0040.05E 00		£1.5	300		- 03
2833 CAF- 339 2833 CTA- 242 2832 CTO- 386 2826 CTA- 240 2 2836 CTA- 240 2 2897-1 CCD- 131 3 3 133 4 4 6 CCF- 135 5 CCF- 135 5 CCF- 135 5 CCF- 135 5 CCF- 135 7 139 7 4 6 331		•	4	337			65.2	306		- 62
2833 C1A- 339 2832 C1A- 242 2826 C1A- 240 2826 C1A- 240 2 2887-1 CCD- 131 3 3 133 4 4 4 134 5 CCF- 135 5 CCF- 135 5 CCF- 135 7 134 7 4 4 134 7 4 5 139 7 7 7 8 8 9 9 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		•	40	338	1.0010.606-01		71.7	300	٠.	-03
2633 C TA - 242 2632 C TO - 386 2826 C TA - 240 2 2 2 2 3 133 3 3 133 4 4 4 4 134 5 C C - 135 5 C C - 135 5 C C - 135 7 2 8 9 - 1 C A D - 328 7 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3		~	7	m	1.5040.90E-01		¥	20C	5	
2832 CTD-386 2826 CTA-240 2 2 87-1 CCD-131 3 3 3 133 4 4 4 4 134 5 CCF-135 -1 2889-1 CAD-328 2 2 329 3 3 330 4 4 5 330		950	2833	7	2.48 tO.12E 00		51.7	226	?	
2826 C14-240 2 2 2 2 131 3 3 3 3 133 4 4 4 134 5 CCF-135 -1 2889-1 CAD-328 4 4 5 331 4 4 5 331		040	2632		0.9041.20E-01		æ	<b>500</b>	<b>5.</b> (E	
2 2 8 7 - 1 CCD - 131 1 1 1 2 2 3 3 3 4 4 4 5 CCF - 135 3 2 2 2 5 CCF - 135 3 3 3 3 3 3 4 5 4 4 4 3 3 1 1 1 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5		045	2826	~	3.99 to. 326 00		•	<b>50C</b>	3.36	
2 132 1 4 4 4 133 2 5 CCF 135 3 -1 2689-1 CAO-328 2 2 2 3 3 330 7 4 4 331 1		1-550	2 8 8 7-1		.01 40.01			20	2. SE	
3 133 2 4 4 134 1 5 CCF-135 3 -1 2 2 89-1 CAD-328 2 2 2 3 3 3 3 3 3 3 3 4 4 4 3 3 1 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5		7	7		.13 to.04E		ç	4CC	7	
4 4 134 1 5 CCF- 135 3 -1 2 2 89-1 CAD- 328 2 2 2 3 3 3 3 3 3 3 4 4 4 331 1		m	~		.11 to. 14 E			704	۲.	
5 CCF- 135 3 -1 2 E89-1 CA0- 328 2 2 2 3 129 7 3 3 3 130 7 4 4 331 1		4	•				0	30Z	<u>.</u>	
-1 2689-1 CAD-328 2 2 2 32 329 7 3 330 7 4 4 331 1		ĸ	Ś		.00 44		19.5	30C	-	•
2 3 330 7 4 531 1 1 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5		046-1	2 6 8 9 - 1	~			O	300	CA 1. CE	
31 1		~	7	~	.00 +5		74.4	724	-	
31 1		· ~	<b>C</b>	330	C# 0 5.		•	704	CA 1. (E	
32		*	•	331	1.05 to.30E 00		38.7	400	CA 1. (E	
		<b>-0</b>	49	132	5.0011.006-01		32.3	4CC	CA 1. CE	CO

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TAB	FABLE E.1 (CONTIN	NTINUED)		9	; ; ; ; ;	1	1			
ARC	LUCATION	16 NO . 16 NO .	TLW ANALYSES NO.	PU-239,240 ACTIVITY (DFP)	URAN IUH (P [CRQ GRAMS)	Y IELC I R=RE HORK )	COUNT TIME	ANA	ANAL /MON	
!	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	:	*	***************************************	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	i 4 4 1 1 1 1 1	· · · · · · · · · · · · · · · · · · ·		! ! !	•
<del>,</del>	1-950	689-7	,	.00 t0.80 E-0		64.0	300	5		
	C 4 8	836	ı	.07 t0.04E		₹0.€	100	~	• EE-	
	0.50	2 8 5 8 - 1	141 -023	.41 t0.04E		5	Ç	1-1		
	~	7		2 t0 . 13E		15.1	30C	14	. 2E	
	i (rī	m	143	.03 t0.24E		'n	)CC	5		
	4	•	144	.48 10.22E		46.1	100	CA		
	· •	· <b>5</b>	CF-	.23 10.15E		48.5	) ) (	J		
	050-1	3576-1	CCD- 146	12 10.05E		71.8	Ç	<u>۲</u>		
	~			.28 tO. 14E		19.1	20C	_		
	m	· ~	148	.87 10.08		13.4	308	_	. 1E 00	
	4	•	149	.35 10.21E		42.0	734	5		
	ď	· w	CFL	46 t0.35E		82.6	30°C	_		
	052-1	2885-1	CAD- 322	.65 10. 12E		\$3.65	30 <b>2</b>	~		
	~			.35 10.19E		,	ec.	,	•	
	m	· m	324	3310.068		65.3	300	•		
	*	•	325	.79 10.49E		11.5	100	•	•	
	•	40	326	.1040.23€		74.3	300	_	•	
	_			.7610.18E		£4.5	308	,,,	1. if-c1	
	750	639	C IA- 254	.50 10.20€		63.3	100	•	•	
	0.66-1	2894-1	,	.15 10.01E		0.09	<b>3</b> C	•••	٠	
	~	~		.95 to. 15E		~	200	•		
	m	~	138	.36 to.08 E			328			
	+	•	139	.07 tO. 27 E		~	30 <b>C</b>	_	Ç	
	*	5	CF-	.53 10.338		ပ	30C	•	00 33 ·	
	062-1	2641-1	99 -000	84 10.12E		69.0	~		). EE CO	
	~	~	67	.64 10.02E		~	328	_	U	
	m	m	6.8	.4910.16E		~	400	_	. ≥ 00 00	
	•	~	69	.8710.04E		~	JCC	_	0	
	ĸ٦	50	CCF- 70	.05 10.13€		•	)) <b>?</b>		10-31 ·	

TABI	TABLE E.1 (CONTIN				1 0 0 0 0 1 1 0 0				
ARC	LOCATION	7. X	76.6	PU-239,240	URANIUM	Y JEL C	CO U!! T	ANAL JHOY	
		COLLEC 11 ON		ACTIVITY		*	I IM		
			NC.	(052)	GRJHS)	TORK)			į
† † †	6 7 6 7 7	7 1 1 1 1 1 1 1 1 1 1							
z	064-1	2628-1	CAD- 310	3.28 to. 12E 01		14.7	<b>5</b> 00	O	0
:		~		.5310.05			ž	ပ	0
	. "	, ec	312	.7240.04E		14.1	301	Ų	U
	٠.	•	313	6 10.03		61.5	2	1.26 0	0
	• ~	• •	316	.60 to. 17 E		81.1	2C í	ш С	_
	, ~		CAF- 315	.00 to. 17E		58.7	<b>30</b> 0	m 0	0
	046	005	C 1A- 247	.49 10.20E		75.t	7.	1	_
	066	8642	CO S-1007			£9.5	×		0
	0.68	2837-A	CCD-2163	.43 to. 11 E	2.81	78.5	4	•	_
	068-3	M	2127	1.47 10.04 6 02	0.189	15.5	) ) (	0	ပ
	•	₩	2128	.96 40.126	0.835	11.1	<u>)</u>		_
	***	. 47	CF	.21 tO. 15E	0 515	50.1	<b>3</b> CC	6. 1E-C	_
	070-1	2831-1	CAD- 316	.30 to.07E	•	62.4	<b>9</b> 0	0	0
	7	)		.91 40.31 £		11.1	¥	01	_
	, (**	•	316	87 to.09 E		14.1	7	1. 1E C	0
	•	•	319	.06 10.04 €		15.6	<b>4</b> C	31	0
	•	•	320	.98 to. 16E		~	3		_
	. ~	•		.70 to.09E		<b>49.</b>	) <b>(</b>		0
	012	6	C TA - 245	u		710.7	ĭ	3. E-C1	_
	074-1	2630-1		.24 tO. 14E			) ) (	3.1E 00	0
	~	1		.41 to. 16 E			<b>9</b> CC		0
	•	(47)	73	#0.07E		5.52	) <b>+</b>		0
	, <b>-</b>	•	~	.67 40.22E		# 6 P	<b>3</b> 5	Ę,	_
	٠ 🕶	. ac.		40.22E		73.C	<b>3</b> 06		. مــ
	078		C14- 246	2 40 . 10 E		11.4	) )	). te-0	5
	1-080	2 12 7-1		.9540.096		£ 1 · 3			٥.
	~	~	9	.94 to. 11E				4. 25-	<b>~</b> (
	<b>(F7)</b>	~	63	2 10 . 22 €		7 - 2			э.
	•	•	<b>\$</b>	2.38 to. 10 E 00		"	2	3. (8-	_

TAB	TABLE E.1 (CO)	(CONTINUED)	1	1	9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		1			
ARC	LOCATION	TLY COLLECTION		TLN ANALYSIS	FU-239,240 ACTIVITY	UABN IUN EP ICRO	Y IELC I R=RE	COUNT TIME	ANA	ANAL MON	
	1	.ON		•	(0FF)	GRAMS)	WORK 1	;		1	•
2	£ 1000	3627-5	- 500	\$ 9	3.6310.128 00		82.6	2001	•	9. (E - 01	
	7 1 4 5 5	2 C C E - 1	- 6	} -	5 .00 14 . CO E -02		61 61 60	1000	_	. CE-02	
	• <b>~</b>	~	)	۰ ۲	4.9510.15E 00		18.1	1000	,_	7. CE-01	
	, ~			· m	6.4210,18£ 00		74.2	100	•	10-33:	
	•	•		4	9.1010.606-01		15.5	1000			
	- 41	. <b>w</b> .	CCF-	'n	8.97 to. 22E 00		80.9	1000	_	• 3E 00	
	020	2014	C 10-	383	7.0011.00E-01		65.5	400	~	1. LE 00	
	0 1 2	\$ 5 5 E		407	5.58 10.15E 04		10.4	2	•	1. 2E CO	
	032-1	2082-1	-000	46	2 - 70 +1 - 90 6 -01		30.78	200	_ უ	• (E C0	
		•	,	) N-	2.5010.21E 00		82.1	<b>5</b> 00	₹	1. (E 00	
	,	, ,		· 4	1 - 90 to - 30 8 - 01		73.3	306	•	3. (6-02	
	• •	•		65	1.20 11.30 6-01		21.3	3CC	•	3. (E-C2	
	* 41	**		50	4.1011.00 6-01	0.420	51.5	<b>500</b>	~	1. (E-03	
	014	8645		408	4.3410.11E 04		11.9	<u>۲</u>	•	10-31	
	024-1	2019-1	-000	11	3.00 14.00 E-02		9.19	1000	, -	7. (E-03	
	~	7		12	3.0610.136 00		64.0	100	 3		
	. ~	· ~			1.5010.40E-01		64.6	1001	•	•	
	•	•		*	0.6012.608-01		31.6	1000	_ ქ		
	•	•		15	2.1010.406-01		66.8	1001	•	7	
	1-520	2CB0-1	CAD	286	2.1510.076 01		86.5	300	<u>ა</u>	. 16 Cì	
		~		287	1.3310.066 01		77.4	300	_		
	الله و	-		208	5.0016.00 6-02		13.6	300	<u>ა</u>		
	1 4	•		233	.4110.108		46.6	<b>3</b> 36	<u>ح</u>		
	• •	. 40		230			59.1	300	_ ქ	10-3)	
	, ~	. ~		162	3.0012.501-01		40.0	300	_ J		
	0.16	2643		233	184,010.45		1.4.2	<b>307</b>	,1		
	90	80.15	CD 5-	409			71.2	><	_	10 31 .	
	8.0	•		010	1.2210.036 05		41.1	۶۲	_		
	0.28-1	1-6752	-000	13	6.7010.411 00		40.5	) ) (	_	00 €.	
	, , ,										

TABI	TABLE E.1 (CO)	(CONTINUED)			9 9 9 9 9 9	1		
ARC	LOCATION	TLW TLW COLLECTION	TLW ANALYSIS	FU-239,240 ACTIVITY	URAN TUM THECRO	Y IELC (Rare	COUNT	ANAL MON
1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	NO.	NO.	(OPF)	GRAMS 1	WORK )		
a.	038-2	2619-2	CCD- 42	3.3010.086 01		45.1	u	
	· ~	, m		4 10.526	0.221	41.7	3CC	6. (E-
	4	4	55	.1041.208-		7-84		
	'n	5		.52 to.17E		63.7	J	
	040-1	2C77-1	CAD- 280	.22 to.07E		80.2		
	7	~		.26 tO.21 E			100	9. SE-01
	C	m	282	.4310.146		58.7		
	4	4	283	.2040.536		44.2	S	
	•	9	284	.44 t0.21€		£8.1	U	5. (E-C2
	~	7		.23 40.228		70.5		6. (E-C1
	042			.93 to. 14E		75.4	ပ	5. 1E-01
	045	8645	COS- 411	.05 tU.02E		11.5	36	4.4E CC
	044		415	10.21E		₹3.€	75	1. E CO
	044-1	2 C 8 9 - 1	15 -000	31E		14.5	100	3. (E
	7		25	.36 tO.11E		16.1	1000	4.36
	m	m	53	390		50.1	<b>30</b> 2	3. (6
	•	*	54			67.3	<b>307</b>	
	**	<b>v</b> n				£9.6	<b>308</b>	1. E-C1
	1-950	2090-1	CAO- 304			15.1	¥	
	~	~		•••		63.4	) ) (	
	M	•	306	.40 t0.04E		72.4	7	<b>O</b>
	~	•	307	301.04.32.		66.5	<b>3</b> 00	9.26
	•	•	308	•••		35.58	20C	۶. ج
	_	7		.46 10.49€		59.3	<b>300</b>	8.58
	C16-A	8643	CDS-1054	10.16E		12.1	×	1.48
	Ø			.57 10.246		1.11	<u> </u>	5.46
	0 4 8		m	10.01E		11.0	36	7. 6
	050	9645	COS- 414	5.5510.15E 04		2.59	<b>5</b> 2	3.46 60
	1-750	_	~	10.0%E		40.4	ĭ	٠.

TAB	TABLE E.1 (CO	(CONTINUED)				1	•	0 0 0 0 0 0	_
ARC	LOCATION	TLW COLLECTION NO.	TEM ANALYSES NO.	PU-239,240 ACTIVITY (0FP)	UPANIUM (MICRO GRAMS)	Y IELC (R=RE MORK)	COUNT TIME	ANAL MON	
1			*	• • • • • • • • • • • • • • • • • • • •	•			, , , , , , , , ,	
ے	052-2	2087-2	CAD- 299	.14 to.04 E O		78.4	٦٢		
	· •			.34 to .33E 0		14.6	٢		
	•		301	.06 40.128 0		46.8	306	1.18 00	
	•	÷	305	.02 10.08E G		60.48	300		
	~	~		.72 t0.08E 0		0	20C		
	950	2019	C14- 232	.22 tO . 13E O		•	>		
		8645	7	.32 to.13£ 0		4	٢		
	60	•		.18 to. 16 E O		ä	21		
	950		416	8 E O		60.1	<u>ب</u>		
	1-950	2054-1	CC0- 36	7E 0		89.4	200		
	~			46 0	•	~	308		
	· ~	6	38	5E 0		6	300	1.26 00	
	*	*	39	5F 0		~	30 <b>2</b>		
	*	*	CCF- 40	4.02 to. 126 01		30.6	300	4.16-01	
	058	8645	CDS- 417	8E 0		•	<u> </u>		
	058-1	2084-1		9E 0		41	Š	, (A	
	~	٠.٠		38 0		ŝ	37	2. žE 00	
	(17	m	567	5E 0		67.3	"		
	•	•	295	6E 0		÷	400	J, (E	
	• •	4)	296	4E 0		*1	200	CA 4,4E 01	
•	2	~		9 6		٦,	20C	•	
	040	2081		0 30		11.1	3 C	٠	
	1-070	2053-1	CC0- 31	.45 10.01 E O		~	100		
	~			.3640.046 0		82.3	30 <b>2</b>	۳,	
	•	•	33	.95 to.09E 0		30.5	ect	۳.	
	•	*	34	.8710.43E 0		¥	3C C		
	***	5	CF-	.03 10.10E	016.0	22.4	200	÷	
	062	8645	0.5-1	0 390.0110.		54.1	ĭ		
	042-1	2020-1	91 -000	.50 10.07E 0		64.6	100	1.46 00	

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TAB	TABLE E.I (CCNTTNUED)	MINUED)	1			† † † † † † † † † † † † † † † † † † †	1 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	1			
ARC	ARC LCCATION	COLLECTION	TLM	¥	FU-239,240 ACTIVITY	URANJUR I P I CRO	Y 1ELC (A*RE WORK)	COUNT	ANA	ANAL /MON	
•		•									•
a,	062-2	2620-2	-622	17	.1440.026		75.1	309	-	7-16 0	_
ŗ	•			<b>6</b> 0	2.7040.05E 01		4	100C	7	0 32.	~
	) <del>«</del>	•		19	.58 40.156		71.9	1000	<b>,,,4</b>	. 1E 0	_
	· Kn		CCF.	20	1.06 #0.07E 01		35.9	300	-	. ie c	_
	064-1		CAO-	258	.02 t0,02 E		75.4	5¢	•	٠ ا	~ .
	~	~		257	.03 #0.036		8	Š	<b>40</b>	. 16-0	_
	M	~		258	.45 #0.27		4	<b>3</b>	€	. 25-0	_
	•	4		259	.27 tO. 12E		ີ	100	_	• 4E - C	_
	•	•		260	.4610.426		51.5	20C	•	0-3)	
	~		CAF-	261	.40 40.07 E		•	<b>5</b> 0 <b>c</b>	~	. (E-0	-
	066	413	- 41 J	239	2.38 t0.07E 02		9	30	m	• EE-0	_
	0.68	C45	<b>COS</b> -	814	.96 40 - 14'		€9.€	2	4 <b>4</b> 4		~
	068-1	2643-1	-000	12	1840.0481.		٠	20C	~		$\sim$
	~			22	.39 40 . 176		~	<b>60C</b>	<b>-</b>		·
	e en v	m		23	1.37 to.08E 01	0.860	•	20C	e=4		Ÿ
	•	*		2.4	3.88 to.13£ 00		•	1000	•	4	_
	٠ •		CCF-	25	7.64 to . 19E 00		85°¥	1000	~		_
	213	685	C 1A-	237	5.3340.168 01		÷	၁ ၂ ၂	m ·		~
	074	683	C 10-	385	1.03 #0.02E 02		<b>~</b>	20C	•		_
	074-1	2647-1	-023	26	2.3042.60E-01		٠	200	•	• (E - C	•
	~	~		2.7	4.00 +8.00 E-02		74.5	300	بر ح	ָשַ י	~
	•	19		2.8	4.60 t0.80 E-01		÷	20C	 ~	₩.	_
	•	*		53	0.9011.106.0		~	<b>502</b>	4		$\boldsymbol{\underline{-}}$
	**		CCF-	30	1.26 to. 13E 00		•	300	_	1 · (E - C	
	0.36-1	2644-1	CA0-	292	3.54 to.06E 03		-	100	لاس		_
	~	174		263	4.17 10.15 01		₩	<b>1</b> 0¢	Φ.		_
	er?	m		264	1.98 to. 106 ol		~	<b>3</b> 0 <b>C</b>	<b>(</b> **		_
	· •	4		265	.38		41.1	<b>3</b> 2 <b>7</b>		•	
	• <b>•</b> 0	•		266	7 40.08 E			306	<b>-</b>	. (E 0	~

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TAB	TABLE E.1 (CO)	(CONTINUED)			0 0 0 0 0 0 0 0 0	0 0 0 0 0	1		1 1 1 1 2
ARC	LOCATION	אסי ורא ורא	TLW ANALYSI S NO.	FU-239,240 ACTIVITY (DPF)	UN BN IUM UN ICRU CRAMS D	Y IEL C I R = R E WORK )	COUNT TIME		ANAL JKON
	† † † † † † † † † † † † † † † † † † †	! ! ! ! !		1 1 1 1 1 1 1 1	• • • • • • • • • • • • • • • • • • •		: : : :		
٩	1-910	2 C 4 4 - 1	AF- 26	.03 #0.20		£ 6 . 4	<b>5</b> 00		1. CE-C1
	078	2078	- 2	.42 t0.18E 0		73.3	9 U		5. (E-01
	062-1	C48-	A0- 2	.41 tO.08 E		~	20		1.46 00
	~		569	4.70 to. 30 E 00		•	<b>5</b> 00	చ	5. CE 00
	-	•	270	.90 to . 15 E			100		6. SE-01
	•	•	271	6.20#0.63E 00		25.5	200	3	6. (E 00
	•9	•	272	.75 40.		~	300		2. CE 00
	_	_	7	.64 10,		64°C	300		2. (E-01
	930	2027	~	.83 40.		72.6	100		4 . 4E - C2
	0.86-1	2015-1	\$ -000	8.65 to. 13E 01		75.7	1001		2. EE 00
	~		1	.35 #0.		52.7	10cc		2.CE 00
	m	~	<b>6</b> 0	.9340,		73.3	1000	ζ.	3. CE 00
	*	•	σ	.00 10.		75.4	1000	_	1. CE-01
	٧n	'n		.03 40.		84.0	locc		7. (E-C1
	068-1	1-1527	CAD- 274	6.90 t0.90 E-01		93.0	300	3	1. CE 00
	~	7	~	3.88 to.33E 00		20.0	302		1 - SE - C2
	m	m	276	5.85 t0.30 00		80.3	300		6. (E CO
	4	4	277	1.35 +0.13 € 00		78.4	300	చ	1. (E 00
	•	•	278	.40 10.		1.51	300		1. (E CO
	-	~	CAF- 279	1.0740.346 00		24.3	<b>30</b> €		8. CE-01
	068-1	2091-1		0.50 #1.00 E-01		65.3	))\	3	1. CE CO
	~		57	.0047.		75.0	104		1. (E- C2
	æ	٣	58	.31 40.		74.2	400		3. (E-01
	•	4	53	.50 40,		76.2	400	చ	1. (E 00
	w:	80	CF-	300.1401.		81.4	3C C		1. (E CO
	050	2088		.30 10.		6.6.8	<b>30</b> 2		2. (E-01
	011	4020	1-50	3711.036		82.5	<b>4</b>		1.46 00
œ	000	2529	5	.55 10.27		48.6	<b>300</b>		5. CC-01
	01%	2530	3.08	10 341.0158.5		11.5	30°C		5. te-cl

TABI	TABLE E.1 (CO)	(CONTINUED)			4 4 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	1 1 2 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1 1 1		8 9 9 1
ARC	LOCATION	TCW COLLECTION NO.	TIN ANALYSI NO.		FU-239,240 AC11VITY (DFF)	UPAN IUN (PICRO GRAMS)	Y 1ELC IR*RE WORK 1	COUNT TIME	ANAL JHOR
!	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			**********	# # # # # # # # # # # # # # # # # # #	  -  -  -  -  -		
Q	0.26-1	1-1856	-073	16	4.65 t0.05E 02		78.0	200	
٤	,	2	) )	77	1.02 #0.01E 02		71.3	300	2. žE C0
	· ~	· en		18	8.1640.24E 01		74.5	300	
	•	4		19	8.9440.46E 00	0.800	23.0	200	
	· w	· w	CCF-	80	2.30 #0.05E 01		76.5	)) <b>3</b>	<b>4</b>
	0.50		C 14-	248	4.3240.11E 02		71.6	36	4
	0.56	7778	CDS-	393	7.5340.18E 04		62.5	21	<u>ب</u>
	034-1		CAD	364	1.8240.05E 02		8 J . E	30	
	,		!	365	2.6840.08£ 02		70.7	36	
	•	m		366	1.95#0.06E 02		32.2	36	
	•	· •		367	1.0140.038 02		70.4	<b>1</b> cc	
	· •C	<b>.</b>		368	4.1340.138 01		57.1	<b>704</b>	
	-	-	CAF	696	4.5940.17E 01		40.E	<b>3</b> CC	
	٠,	8644	-503	394	4.2240.118 04		70.4	<u>ح</u>	
	0 2 8			395	4.9340.07E 04		85.3	3 C	<u>ب</u>
	040	2538	C 10-	389	1.2340.028 03		30.3	Ç	. Æ-
	040	BC44	<b>COS</b> -	396	4.1640.06E 04	0.620	13.5	) (	<u>.</u>
	040-1	2 5 4 4 - 1	C 40-	346	1.97 t0.06E 01		19.1	<b>4</b> 00	ب د د
		~		347	1.27 to.04E 02		11.1	ٔ د	<b>.</b>
	•	e		348	1.33 to.06E 02		2° ° °	<b>3</b> 0C	بر ا
	-	4		349	6.5540.26E 01		£0.8	100	۳:
	•	₩		350	2.7140.12E 01		46.1	2 C C	۳;
	_	~	CAF-	351	3.0240.09£ 01		72.1	20 00 00	• ZE-
	042	5628	C 1A-	252	1.4240.038 02		17.4	)   	
	042	6644	C0 S-	397	3.2540.08E 04		W . W .	) 	
	044			396	2.5240.06E 04		25.4	<u>.</u>	
	0.46			399	2.12 to.05E 04			) (	
	0 4 50	5 6 9 3	C 14-	283	6.0340.18E 01		7.07	ب د	10-27-6
	048	8044	CD S-	400	1.1240.036 04		9.2.4	<b>,</b>	( , JE V

TABI	TABLE E.1 (CONTINU	(TINUED)			9 8 8 9 9	1	; ; ;	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4
ARC	LOCATION	TLW COLLECTION	TLW ANALYSI S	PU-239,240 ACTIVITY	CRANIUM FYICRO	Y IELC (R=RE WORK)	COUNT 11ME	ANAL JHON
	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1							***************************************
œ	0-0-0	2536-1	18 -022	6.9740.17E 00		81.0	1000	
:	7	~		.7940.06E		8C.4	300	4. 56
	· m	•	63	.1440.03E		83.5	) ) )	CA 1.1E 01
	•	4	48	.02 #0.04 €		3.08	<b>)</b> ) 5	2.2E
	'n	*	CCF- 85	1.47 tO.15E 00		76.C	<b>3</b> 00	CA 1. (E CO
	0.52-1	2548-1	40- 3	.20 #1. 40E-		23.6	400	1. Œ
	~	7	~	.74 tO.16E		46.5	400	<b>7.</b> (E
	m	m	354	.7840.17E		40.2	400	CA 2. (F. CC
	•	•	355	40.14E		45.C	<b>3</b> 00	1. CE
	•	·	356	1+00.		41.E	<b>4</b> 00	1. (£-
		~	CAF- 357	-80 40 • 6CE-		15.5	<b>7</b> 9 <b>5</b>	1. CE
	450	2546-A	~	.08€	01.0	<b>t</b> e.1	7	
	410	60	2	1.16 #0.03E 02	4.275	64.3	Ç	6. 5E-01
	440	8044	COS- 401	1.1940.025 04		46.7	<u>۲</u>	
	950		405			4.03	36	
	8,0		1008	.68 10.27 €		63.4	ĭ	
-	0 58- N	2551-1	CA0- 358	€ 0		<b>30°C</b>	) (	9. E OG
	~		359	0		80.2	Š	
	•	m	360	0		_	) ) (	6. 16 60
	4	•	361	.46 t0 . 05 E		. 4 . 7	2CC	5. ZE CC
	¥	•	362	.92 to . 29 E		£7.£	<b>3</b> C(	#
	•	~	n	.8440.208		74.6	<b>4</b> CC	
	0 7 0	2452	-1	.94 to.12E		76.7	ĭ.	3. (E-01
	070	8644		7.4410.17E 03			76	1. 16
	062		404	.83 tO.22E 0			) 2 7	E
	062-1	2540-1	CC0- 86	.07 to.03E 0		73.0	) ) (	
	~	~	10	.18 t0.05E 0				ָ פּ
	<b>.</b>	m	8	10.03E		74.2	<u>ک</u>	1. EF CC
	∢		60	6.97 to. 23E 01		-	107	2.4£ 00

TABI	TABLE E.1 (CONTINUED)	TINUED)	-			1	1	4 4 9 9 9 9
ARC	ARC LOCATION	7	7 L h	PU-239,240	AUI NAU	YIELD	COUNT	ANAL JHON
		COLLECTION NO.		(00k)	GRAMS)	MORK)	3411	
œ	062-5	2540-5		.1340.148 0		72.c	300	
:	066	3578	CTA- 235	3 ‡0		74.3		4.36-01
	0 € 6 - A	8044		.87 t0.20E 0		C1	36	
	60		739	.36 to. 19E 0		*		
	068	ᠵ	<u>_</u>	-07 tO-15E 0	0.125	6 6 . S	<b>S</b>	£-
	068	8044	COS- 406	.2940.10E D		50.4	2	
	070-1	S.	A0-	.4740.06E 0	0.0510	67.1	40	. 1E
	~	~	371	.4440.16E 0	0.0710	67.5	70	
	m	6	372	.97 #0.08 E O	0.0850	71.7	<u>چ</u>	
	4	•	373	E 0	0.421	67.3	90	
	•	9	374	.84 tO. 26E 0	0.720	66.1	2CC	
	_	_	AF-	E 0	1.12	46.6	γ	
	072	2566	ŧ	-1340.13E 0		71.2	3C	
	074-1	2560-1	-03	.88 #0.06E 0		17.4	30C	
	~		101	.06 #0.06E 0		75.3	) ] [	
	<b>(**)</b>	m	108	0		11.1	100	3.4E CO
	∢	∢	109	.22 tO.06E D		13.7	) ()	1.75 00
	€0		-	.57 to.17E 0		61.6	<b>3</b> 00	
	076	2952	CTA- 251	.07 to. 12E 0		74.4	<b>3</b> C	3.76-01
	080	2569	m	.6340.02E 0		14.5	<b>3</b> C	• tE-
	0.60-1	2561-1	-	.02E 0		11.8	<b>3</b> 00	
	~	-	112	.3440.07E 0		91.4	20	. 1E
	m	m	113	.1340.02E 0		74.4	300	1. !E 00
	4		114	.60 #0.04E 0			3001	• CE
	<b>e</b> n	'n	CCF- 115	.85 to.06 E 0		64.1	1000	• 1E
	230	2534-A	CAD-2166	.34 tO.26E D	4.67		<b>3</b> CC	. CE-
	230	60	~	40.37E 0	0.181	31.0	3	8. (è-02
	066-1	2564-1	CCO- 116	.6440.02E 0		74.5	300	• 3E
	~		-	.5740.04E 0		12.4	201	٠ بي

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ARC	LOCATION	TLW COLLECTION NO.	TLW ANALYSIS NO.	PU-239,240 ACTIVITY (DPP)	C A AN ICH C P I C R O G P A P S J	Y LEL C (R=RE WORK)	COUNT TIME	ANAL MON
<b>a</b>	F - 9 4 0	£-745 C	CCD- 118	4		2,0	305	
:	•	4	611	36 +0 - 10 E		10.3	300	1 - EE CC
	. au	· <b>K</b> N		.32 to.09		82.6	306	
	0 68-1	2937-1	CAD- 340	.87 tO. 14E		62.2	<b>3</b> 0 <b>C</b>	
	7	7	341	.33 40.06E		62.4	<b>3</b> 0 <b>6</b>	
	m	6	345	.53 #0.63E		40.4	<b>3</b> CC	
	•	◀	343	.67 #0.21E		70.3	) ) (	
	•	•	344	.10 to.32E		15.2	Ą	
	~	7		.85 to.37E		12.2	400	
	050	2545	CTA- 250		0.222	45.6	ž	
	250	2567				46.8	2	•
	1-250	2553-1				67.5	) ) 	
	~	7	16			73.4	100	2. fe co
	m	m	96			1.08	300	3. (F CC
	*	4	66			83.5	<b>306</b>	1. 3€ CC
	×۸	<b>5</b> 0	•			105.	<b>3</b> 26	1.76 00
	058-1	2557-1	101 -023			77.6	1000	1.2E CO
	~	~			1.11	~	<b>308</b>	1. (E CC
	m	m	103			0	<b>30</b> 0	1.€ 60
	•	4	104			•	<b>30</b> 2	1.26 00
	ŧ۸	*	CCF- 105			47.8	<b>30</b> 0	1. 3E 00
	1-431	2550-1				41	306	1. 16 CC
	~	7				62.5	300	1. TE 90
	m	m	93			83.0	) ()	2. EE 00
	4	*	96			80.6	<b>302</b>	3. CE
	<b>s</b> n	•				83.1	) ) (	CA 1. EE 01
	169-1	2568-1	CCD- 121	1.2940.136 00	2.06	42.4	200	1. CE
	~	~	122			56.5	) ) (	ш
	<b>m</b>	ED.	123	1.70+0.08E 01		59.6	)) <b>?</b>	1.5€ 00

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1	TABL	TABLE E.1 (CO	(CONTINUED)		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			
	ARC	LOCATION	· •	TLN ANALYSI S	FU-239,240 AC1 [VITY	UR AN IUM	Y SELD (R*RE	COUNT 114E	ANAL MICH
!		1	·ON		(440)		4   4   5		
	α	106-4	2568-4		.8340.18E		•	300	. CE
	ì	<b>.</b>	S	CCF- 125	92		6.9°E	<b>3</b> CC	1. £E 00
	8 AL	L 5, P 17	2507-A		.00 40.038	•	•	Ų <b>V</b>	÷.
	1	L 5, P 17	60		.01 #0.02E	1.07	•	7,	ij.
		17.99	2443-A		.95 40.07E	7	•	300	<u>س</u>
		18,921	-15	CC0-2157	.33 tO.02E	~	•	300	H.
		1.8, P.21			.08 #0.15E	Ç.	37.68	7	¥.
*	884	L 6, P 13	2482-A	2159	.13 #0.03E	1.44	•	) ) (	'n
	1			2205	.46 #0.06 E	0.166	ŗ	£C	÷.
	0.4	CHR-14	9106	C VS-2076	.9940.35E		•	<b>3</b> CC	
		91			.47 #0.40E		90.4	<b>3</b> C C	
		<b>Z</b> A		2018	.50 t0.34E		•	1000	
		28		2019	.35 to.37E		15.5	<b>30</b> 0	
		4		2080	.57 t0. 24E		•	) 20.	
		<b>9</b>		2081	.1640.368			<b>3</b> 0 <b>7</b>	
		14	9720	2082	.9340.05E		•	<b>3</b> CC	
		91		2083	.58 ±0.13E		14.7	200	
		2A		2084	.78 #0.08 E			၂ ၂	
		<b>82</b>		2085	.0740.226		•	100	
		44		2086	.69 #0.09E		~	) ] [	
		48		2087	.14 to . 09 E			200	
		11A		2088	.78 to. 11 E		٠	<b>3</b> 00	
		118	- · `.	<b>508</b> 3	.1140.06E		46.6	) ) (	
		12A		2090	.45 to . 13E		76.5	<b>5</b> 00	
		120		1602 .	.91 to.31E		36.6	) ) (	
		S 1K - 10	10012	CDS-1078	.6341.158		19.3	30 <b>2</b>	2. (E-
		104	•	1079	.2012.10E-		4.06	) ) (	
		11		1080	.6341.09E		m . m	<u>ر</u>	1 - 1E - C2
		1 1 A		1001	7.60 t6. 10 E-0 J		14.4	36	

\*Sample inadvertently combined with TB-F2 2305B in chemistry.

TABI	TABLE E.1 (CONTINUED)	NTINUED)				1	1 1 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	i
ARC	LOCATION	12 K	7Ch ANALYSIS	FU-239,240	UR AN IUM	Y JELC (R*RE	COUNT TIME	ANAL JMON	
		NO.		(DFP)	GRAMS)	WORK )		0 4 9 1 0	i
				1 1040 405 00		C. T.	26	5. CF - C3	
Ý	51K-12	7111	7807-663			4.4	ָבָיב <u>ָ</u>	CA 1. (F-02	
	124		1084	1.28 to.04E OI		63.1	200	2. (E-	
	VE 1		1085			*	100	CA 1. (E-C2	
	7 -		1086	1.70 #0.30 E 00		83.4	100	1. CE-	
	144		1097	8.70 +1.80 E-01		G	) ) (	CA 1. (E-02	
	15		1130	4.86 to.30E 00		5.59	<b>3</b> CC	<del>-</del> 3) •	
	16	1 CC13	1131	9.2042.90E-01		70.5	۲	1. (E CO	
	164	1	1132	2.1540.198 00		13.7	<b>3</b> 00	4. (E-	
	~		1133	3.6910.14E 01		ċ	100	CA 3.4E 01	
	174		1134	6.3140.42E 00		47.7	<b>3</b> 00	36	
	18		1135	3.2540.23E 00		.;	) ) (		
	1.84		1136	1.67 #C. 06E 01			300	4	
	161		1137	2.65 t0.10 01		~	)) <b>?</b>	1	
	194		1138	6.0940.22E 01		S	100	بير	
	20	1 ( C C S	1017	1.7240.06€ 02		71.3	<b>3</b> 00	<u></u>	
	36	10001	1076	1.44 10.05E 02		•	<b>&gt;</b> 2	7	
	71	10010	1056	7.96 t0.25E 03		Q.	36	Ē	
	12		1057	6.49 to. 20 E 03		œ	<b>&gt;</b>	Ŧ.	
	73	•	. 1058	6.9440.22E 03		73.1	,	Ä	
	74	• .	1059	5.29 to. 17£ 03		78.6	36	느	
	75	•	1060	6.60 #0.23E 03	,	72.4	3 C	1	
	25		1001	3.07 to. 11E 03	•	74.7	36	딒	
	11		1062	3.8140.136 03	•	16.4	<b>~</b>	w	
	7.0		1063	3.6740.138 03		14.1	36	۳	
	6		1064	4.1840.14E 03		75.5	ž	<u>.</u>	
	00		1065	4.1140.146 03		8c.3	<b>&gt;</b>	3. (E CO	
	6		1066	2.2510.098 02		71.5	~	-3. Æ-	
	97		1067	2.9540.106 02		74.3	<b>&gt;</b>		

TABI	TABLE E.1 (C	(CONTINUED)	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	***************************************		111111111111111111111111111111111111111		
ARC	LOCATION	10 10 10 10 10 10 10 10 10 10 10 10 10 1	TEN ANDIYSIS	FU-239,240 ACTIVITY	UR BN IUM	Y JELC IR BE	COUNT TIME	ANAL /HON
			NO.	1940)	GRAMS 1	WORK )	1	***
) 				•			7.0	1,7 4 6
٧0	STK-63	10101	CD 2-1068	.98 to . 15t			. ·	֓֞֝֝֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֡֓֓֓֡֓֡֓֓֓֡֓֡֓֡֓֡
	<b>5</b> 2		1069	.1240.25E		9	7	ָ נ נ
	85		1070	.33 t0.04 E		65.5	<b>3</b> C	4
	98		101	.3240.03€		71.7	20	æ
	. 4		1072	.47 40.04E		16.4	26	
	250	1,000	1073	1.03 40.03 6 04		73.4	20	1. (f co
	251	•	1074	.41 40.16E		10.6	76	
	252		1075	.20 to . 16 E		19.1	20	
PCMR	P CMR 2- 1- E	NCNE	C SF - 1 890		7.24	13.1	));	3. (E-C3
	۱ سر •	<b>!</b>	_		8.31	91.5	30 <b>2</b>	1. CE-03
	<b>,</b> ~		1892	5.25 to. 12E 02		43.6	100	2. CE-02
	· a.		1893		31.7	4.1.4	<b>3</b> CC	1. 16-01
			1894		0.11	4C-1	3CC	3. (E-C2
	10		1 6 9 5	10.12E	25.6	21.1	30 <b>c</b>	4. (E-C2
			1896	.6340.09E	6.95	4.5.4	9	4. (E-02
	2-5		1897		31.6	56	3 C C	2. (E-03
	• <b>•</b> ••		1898	.0840.036	19.3	2.62	36	1. 2E CC
	. ~		1899		31.5	2005	20	1.36 00
	·		1 900		2.50	76.4	<b>3</b> CC	3. 36 - 01
	v		1061		2.34	84.6	76	8.4E-01
	10		1902	.8240.43E	7.41	66.1	308	4. (E-C2
	11		1903	.58 40.08 E	3.45	84.3	<b>3</b> 0 <b>c</b>	1. te - c1
	4		1904	.17 tO.06E	15.9	45.5	30 <b>2</b>	3. 1E 00
			1905	6.4010.298 01	13.0	•	) ) <b>?</b>	Ē,
	_		1 906	.43 t0.21E	28.8	£0.3	) ) )	1.56-61
	· <b>4</b> 2		1907	.02 40.116	4.78	21.7	<b>308</b>	<u>.</u>
	v		1908	358	47.2	7.55	<b>308</b>	2. CE-02
	21		1909	1	1.41	27.5	<b>3</b> 66	10 - 36 - 61
	7		1910	8.3410.438 01	4.87	53.4	))?	5. te-01

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NONE   CSF-1911   S-54 10.41E   O   2.08   31.7   200   1.40   S-64 10.24E   O   2.08   31.7   200   S-64   S-64 10.24E   O   2.08   31.7   200   S-64   S		16 F	TLW ANALYS!	-239,24 CTIVIIY	8 44 TU	1EL R=9	COUNT 1 146	ANAL JANA
NONE   CSF-1911   S.54 10.41E 00   2.08   71.7   200   1.46   1.91   1.45 10.31E 00   2.08   71.7   200   1.46   1.90	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	. !	*ON	100	NAME I	81		
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1915   1918	F CMR 2- 6- 6	202	141-70	0 314.01424	9	-	·	
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1918   2.5740.06E 02   1.62   16.5   200   1.1E-0   1919   1.4740.06E 01   11.40   1	1 11		7	.9143.596-0	7	ċ	u	G
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10	• •		92	.64 t0.46E 0	.02	72	u	-33
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1925 8.3040.276 01 19.9 14.6 9CC 3.8F-C 1925 1.3440.476 01 15.0 75.6 2CC 3.8F-C 1927 1.2440.46 02 15.0 75.6 2CC 3.8F-C 10 19.2 19.2 19.2 19.2 19.2 19.2 19.2 19.2	-L		92	.52 to. 40E 0	-	۲.	u	. (F-
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1932   8.7240.25E 01   3.35   73.4   200   1.7E 0   1.934   2.28   40.12E 01   2.53   45.2   200   1.95   1.95   1.934   2.28   40.12E 01   8.41   2.3.7   500   1.95   3.26-0   3.26-0   1.935   1.3140.04E 02   2.73   20.4   500   1.4E 0   3.26-0   1.31   8.24   200   3.26-0   3.26-0   1.31   8.24   200   3.26-0   3.26-0   1.31   8.24   200   3.26-0   1.60   1.60   1.60   1.60   1.60   1.31   8.24   200   1.60			93	.7940.736-0	.94	ċ	u	<del>.</del> .
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10 1935 1.3140.04E 02 2.73 2C.4 5CC 1.4E C 10.7 1936 9.9940.33E 01 10.7 29.5 2CC 3.2E-C 11.31 1938 1.7540.04E 02 1.31 82.4 2CC 3.2E-C 12.5 12.5 12.9	•		93	.28 #0.12E D	۲.	•	u	£-
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E LOCATION TLN TLN NO.  COLLECTION ANALYSIS  NO. NO. NO.  LOCATION ANALYSIS  LOCATION ANA	1 1				4 6:4 1 10/11:	
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5 5 5 7 7 CAF- 20 8123 CDS-	*		1.13	<b>3</b> 0 <b>2</b>	5. (E-C2	
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000	1.71 40.04 8		72.1	<b>~</b>	2. (E 00	
,	2.0240.05E		<b>65.</b>	<u>۲</u>	2. 1E CO	
•	1.7340.178 05		7.08	<b>5</b>	5.4E CC	

	ANAL /MON	ñ	3. (E CO	پې		3. (E CO	, Se C	. SE- C	. if			٠	(E	1	2. ft-c1	w	2. £6-01		Ę.		ĥ	æ	Ę.	1E-	•	Ę,	1. 36 01	• <del>.</del> £	÷.	
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	TLW ANALYSIS NO.	CDS-1166	1147	1148	1149	1150	1111	1739	1152	$\sim$	CD S-1153	_	CC0-1602		1604	1605	Ų	CD 5-1155		1157	1158	1159	1160	14-1	1591-000		1653	1654	91-	_
(CONTINUED)		B123	;			8124		8688	8124	l •			2568-1	)	'n	•	•	8124						3526	3402-1	)	) em		•	3\$00
TABLE E.2 (CO)	LOCATION	j 11	•		0 2 5	"	17	1	191	-	451	77	4 42 40	,	, en	•	· <b>6</b> 7	45	-	~	1	0.24		4	+		, ~	•	• •	000
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TABI	IABLE E.2 (CONTIN	ONTENUED)			1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	• • • • • • • • • • • • • • • • • • • •	1		į
AAC	AAC LOCATION	11 W COLLEC 11 ON NO.	TLW ANALYSES NC.	PU-239,240 ACTIVITY IDFP)	URAN IUN (PICRO GRAMS)	Y IELC (R*RE WORK)	COUNT	ANAL MON	
1	!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!								
_	050	3492	CTA- 425	4.45 to.33E 00		3.6	734	4. (F	0
,	022-1	2676-1	CAD- 717	1.10 to. 70 E-01		76.1	<b>3</b> CC	_	္ဌ
	•	~	21.8	6.00#4.00 E-02		76.0	300	_	_
	• 11	\$ <b>4</b> 43	719	4.00 t6.00 E-02		66.7	308	CA 1. (E 00	0
	4	•	120	4.30 tl. 30 E-01		45.2	2C C	•	_
	- 40	. 40	721	1.10 to. 80 E-01		5.4.5	300	•	_
	, ~	•	CAF- 722	1.80 to. 70 E-01		11.6	<b>300</b>	CA 5. (E-0	70
	0.24-1	2663-1	CCD- 181	4.70 +1.00 E-01		39.4	300	<b>1.</b> (E	င္ပ
		• 653.1		0.4011.005-01		26.3	704	1. (E-	~
	4 (1	8 647	183	1.80 +0.80 E-01		57.5	704	1. (E-	_
	3 🕶	•	\$ <b>0</b> ~	0.0040.05E 00		61.3	400	1. (E-	5
	r sn	· •		1.90 to. 70 E-01		81.C	20C	2. EE 0	o
	0.26-1	2656-1	LCD-1623	4.7643.57E-01		4.00	4	1. CE 0	0
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	ı en	, m	1625	1.6840.18E 00		75.C	<b>30</b> 2	4-26-0	_
	•	•	1626	5.7045.70E-02		74.2	<b>308</b>	7. (6-0	6
	. w	. <b>K</b> U	-	2.78#1.04E-01		0 6 6 . C	2CC	4. CE-0	_
	0.28-1	2475-1	CAD- 711	1.11 #0.03E 03		66.6	<b>3</b> C	6. ¿E	ပ
	~	~		0.5041.00£-01		45.7	) ) (	CA 1. (E 0	0
	•	•	713	2.70#1.106-01		65.1	<b>30</b> 0	J. (£	0
	) <b>A</b>	<b>₹</b>	714	7.00 \$6.00 E-02		19.6	<b>3</b> 00	 E	0
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	) P-	. ~		.10+11.20E		69.6	<b>5</b> CC	CA 1. (E 0	0
	0.22-1	2457-1	CC0- 166	2.1240.04E 02		16.8	302	<b>4.</b> (€ −0	_
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		1119	168	.6041.208		49.0	306	1. CE	0
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	*	41	17	•		45.0	FCC	. Œ.	~
	1-5:0	2613-1	CAO- 699	.6640.		36.0	<b>3</b> 0 <b>C</b>	2. E 0	0

	induit est (continued)	(TOOLS)		**************************************					******
ARC	ARC LOCATION	16 T T T T T T T T T T T T T T T T T T T	TIL ANALYSIS NC.	PU-239,240 ACTIVITY (DFP)	CP ICRO	Y JELC (R*RE MORK)	COUNT TIME		ANAL MON
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7	034-2	2613-2	CAD- 700	0.80 11.00 6-01		73.1	3CC	5	3. CE-01
	•	m	101	4.0045.00E-02		73,	<b>4</b> CC	5	1. (E 00
	•	•	702	1.0040.606-01		62.	40	5	1. CE CO
	÷	÷	703	9.00 #4.00 E-02		9.99	) <b>∀</b>	3	1. CE CO
	<b> -</b>	7	CAF- 704	6.10+1.80E-01		79.€	100	Į,	2. CE-02
	026	3489	C 1A-1361	7.87 to. 32E 01		26.4	<b>306</b>		1. 26 00
	638-1	2658-1	CCD- 171	6.00 to. 70 E-01		\$4.5	77	5	1. (E-C1
	~	7	172	2.00+1.30E-01		36.3	) ) (	3	1. (E 00
	•	m	173			69.3	30C	3	1. CE OC
	•	4	174	1.44 to. 20 E 00		46.3	200		8. (E-01
	*	<b>K</b>	CCF- 175	4.1048.106-02		58.1	<b>30</b> 0	5	2. (E-03
	1-053	2674-1		6.35 to.21E 01		75.0	) ) (		4. SE 00
	~	7		1.60 #0.50 €-01		11.1	400	ರ	1. CE 00
	<b>F279</b>	<b>(1</b> )	707	1.80 to. 30 E-01		61.3	1000	5	1. CE 00
	₩.	•	108	5.00 14.00 E-02		72.0	300	5	1. (E CO
	÷	•		1.20 +0.40 €-01	•	71.5	200	5	1. G 00
	_	~	CAF- 710	2.20 +0.70 €-01			<b>3</b> CC	3	5. (E-C2
	045	3450		7.15 to. 19 6 02		11.3	7		1. CE CO
	045-1	2584-1		1.7640.19€ 00		12.3	200	5	2. (E CO
	~	~	724	1.76 to.26E 00		35.6	) ) (	ರ	
	m	~	725	2.30 +1.60E-01		36.5	<b>3</b> 00	3	
	•	~	126	4.50 to. 90 E-01		65.0	) ) (	3	w
	•	₩	727	2.50 +1.00 E-01		69.6	300	5	•
	p.	~		2.6041.30E-01		72.3	130	3	1. (E CO
	1-550	2661-1	CCO- 176	5.40 to. 32 E 00		25.7	300	3	
	~	~	177	2.20 to. 80 E-01		41.3	300	ರ	1. (E CO
	~	1-7	178	1.7011.106-01		41.5	<b>300</b>	Z	•
	*	~	179	7.2041.30E-01		36.6	) ) (	J	1.€ 00
	'n	'n	CCF- 180	1.50 t0.80 E-01		62.2	30C	3	1. CE 00

rabi	ABLE E.2 (CC	(CONTINUED)			9 9 9 9 9 9		; 4 8 9		1 4 2
, KC	LOCATION	COLLE	ANALYSI S NO.	FU-239,240 ACTIVITY (DPF)	C S P S C S C S C S P S C S C S C S C S	Y IELC (R*RE HORK)	COUNT TIME	ANAL /MON	Z .
		· • • • • • • • • • • • • • • • • • • •	1 1 1 1 1 1 1 1 1 1			 			!
_	0.48	9.	.A- 42	.70 to. 80 E-C		*	Ç	•	10-
_	650	50	10-156	.62 0.58E 0		<b>,</b>	O	•	၀
	022-1	2653-1	89	.13 to.13E 0		-	U		10-
_	~		99	.4011.10E-0		•	) <b>)</b> \$	-	0
	1 17		8	.20 40.60 E-0		ä	S	۶.	
	) -g	•		.00 #4.00 E-0		ő	O	7	
	· •	. 4	Ο.	.00 #4.00 E-0		14	Ç	\$	
	, r-	•	AF- 69	.6010.50E-0		•,	C	•	13-
	024-1	2443-1	ø	.00 17.00 E-0		~	J		
		) ;	10	.9041.00 E-0		++3	30£		00
	) (F	ा स्था	•	. 30 40.60 E-0			U	177	
	<b>.</b>	***	V	.70 \$1.30 E-0		<b>:</b>	U	÷	
	· en	· 40	CF- 16	.95 #0.18E 0		÷	C	•	-03
	028-1	2654-1	O.	.6640.07E 0		~		•	7
	7		69	.80 #0.20 E-0		<b>~</b> ;	Ü	_:	-CI
	, eri	· m	Q.	.60 10.50 E-0		~	ب	<b>:</b>	93
	• •	4	σ	.60 10.90 E-0		÷	U	-	90
	• •3	٠ س	O.	0-305.04040		'n,	Ċ	•	00
	~	·~	CAF- 698	00 46 . 0		15.7	308		00
	030	<b>\$15</b>	42	.00 +1.00 €-0			ن		10 -
	0.20-2	8127-2	DS-174	.66 to.05 E C		~			ÇO
	· ~	!	174	.10 to 10 E		•;	2	•	00
	•	**	74	.\$3 to.03E 0		•		٠	ဝ
	•	•	3	.5640.218 0		•			ပ္ပ
	032-3	2432-1	3	.31 to.usE		ä		•	၀၁
	ry ;	,	161	.16 12.08 6-0		_:		٠	-01
	, 44		Ģ	.2512.136-0		ü		•	50-
	• •₹	*	J	.88 11 . 77 E		19.6		•	
	w	ĸ	5	.7710.756-0		÷		<b>₹.</b> (€	- 03
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ARC EO									
	EOCATION	COLLECTION NO.	TLW ANALYSIS NG.	FU-239,240 ACTIVITY (DF!!)	URANIUM (FICRO GRAMS)	Y JELC (R.RE WORK)	COUNT I IME	ANAL JHON	<b>8</b> 0
	! !	*	: : : : : : : : : : : : : : : : : : : :		• • • • • • • • • • • • • • • • • • •	! ! !	,   		
ر 03	024-1	2633-1	CCO- 151	1.2040.80E-01		61 61	400	m.	<b>~</b> ;
1	. ~	~		2.00 t6.00 E-02		62.1	<b>3</b> 0 <b>C</b>	CA 5. CE	-03
	· (~)	'n	153	2.00 +1.20 5-01		3 <b>.</b> 9.	400	CA 3. (E	-05
	<b>-</b>	*	154	1.01 50.40€ 00		12.5	400	-:	
	. <b>F</b> L	· <b>v</b>		.40 to .80 E		37.6	<b>3</b> 00	CA 5, (E	-C
60	<b>.</b>	1.5		.301.106		36.3	400	CA 1. CE	8
60	8-1	634-1	951 -000	1.10 +0.06 01		49.€	<b>30C</b>	1.46	00
	. ~	~		.00 40.09		55.5	<b>3</b> 0 <b>C</b>	1. CE	00
	, (41	1 647	158	.1013.008		48.6	<b>308</b>	CA 1. (E	္ဌ
	٠ -	1 €7	159	.00 49.00		41.1	30C	CA 1. (E	00
	· 47	~		.00.49.00		63.1	30C	CA 1. CE	8
0.4	040-1	2647-1	CAD- 681	.00 to.06 E			<b>308</b>	CA 1.1E	ច
)	· ~	~		0.00 10.06 60		80.0X	<b>9</b> 00	1. CE	0
	•	~	683	.00 #4 .00 E		11.4	<b>3</b> 000	CA 1. CE	0
	2	•	489	2.00 #4.00 £-02		66.3	<b>9</b> 0%	<b>:</b>	00
	•	4	685	1.00 43.00 6-02		7.00	<b>3</b> 0	CA 1. (E	ខ
	_	~	AF-	1.2810.27E 00		6 P	) ) (	9. (E	ຽ
40	· ~	3508	٠	.98 40.16 €		10.0	) ) (	1.	3
045	2-2	8127-2	1-50	.73 to.53E		141 41	300	J. 76	0
	m	<b>E</b>	1747	0		44.6	) ) (	7. CE	-03
	•	•	1.748	3.88 to. 10f 02		£0.5	) ) (	3. 56	~
	Ś	•	1749	.5a to. 11 E		11.1	300	3. EE	5
40	044-1	2645-1	8191-000	0		63.1	2CC	1. 4E	5
,	~	13		.7343.45E		41.0	<b>)</b>	1. (6	8
	•	•	1620	.67 42.75 E-0		7	<b>)</b>	1. (E	S
	•	•	1621	. 17E		55.5	<b>3</b> 0 <b>C</b>	(E	
	· •	~	•	4.1914.798-01		4.44	<b>V</b>		10-
40	C46-1	8127	COS-1500	5.18 to . 57 € 00		14.1	2cc	3.26	
•	14	8127-2	1221	.0010.246 0		62.	308	•	

TAB	TABLE E.2 (CONTINU	ONTINUED)			0 6 8 8 8 8 8	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		-	;	
ARC	ARC LOCATION	16.6	#1L	FU-239,240	NOT WE WO	3731 4	COUNT		ANAL /MON	
		COLLEC 11 C4	•	ACT # V 2 T Y (OP P. )	GRAMS )	MORK 5	K	1		:
	• • • • • • • • • • • • • • • • • • •	- - - - - - - - - - - - - - - - - - -	2 6 7 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	0 7 8 7 Y 8 8 8 8 8 9 8 9 9 8 10 7 7						
د	046-3	8127-3	CD 5-1 752	.17 10.43		36.4	306	_	. (E-03	_
	-	•	-	.30 to. 29E		65.1	<b>3</b> CC	14	, <u>;</u> £	_
	· •••	· •	1754	.68 10.08		-	<b>30</b> 6	~	<del>.</del>	~
	8-940	125	1169	9.7740.278 02		•	ęç			
	2. 2.	3307	<b>TA</b> -	.73 to.16E 0		•	100	3	35.	_
	048	•	C TD-1662	.32 tO.17 £ 0			)) <b>?</b>	~	35.	_
	0.0	127-	0 5-1	.24 #0.53E 0		66.5	30C	,-	• 2€	_
	~		-	.65 tO. 37E. 0		46.4	<b>302</b>	.4	35.	_
	) (C)	; e41	1757	.79 to.17 6 0			306	~	33.	0
	•	•	1758	.8040.17E 0		A. 8.	1400	•	. EE	_
	* 19E7	**	1759	.85 tO.88E 0			30Z	_	. E	0
z	070	695	TA-	1.08 +0.18 500		41.4	<b>302</b>	7	• CE 00	0
:	022-1		CAD- 742	.0049.005-0		A. C.	3cc	2	. (A Q	0
	~			4.2042.106-01		33.6	<b>3</b> 28	2	ě.	0
	· (41	~	144	1.10 to. 30 E-01		61.3	300	3	· CE-0	_
	•	•	345	1.1040.805-01		65.3	30 <b>c</b>	_ უ	S. O	0
	•		146	2.0041.00 €-01		40.5	704	5	. (E-0	~4
	-	_		1.9041.106-01		76.2	30 <b>2</b>	5	. (E-C	~
	026	3572				42.C	<b>302</b>		3	_
	026-1	3336-1	CCD- 216	•		30.€	400	Ck 2	3	~
	7	) 		.9041		91.1	322	7	• (E	0
	· (**	· 647	218	•		14.2	<b>3</b> CC	<u>ح</u>		_
	•	•	219	•		10.6	<b>30C</b>	5	, (E-	~
	•	•	Cf.	00 16.		72.3	<b>3</b> 2C	•	• {E	_
	028-1	3346-1	CAD-1646	67 40		16.0	<b>5</b> CC		•	et d
	~	1	_	Ň		29.5	300	•	٦.	
	•	~	1642	•		€2.6	<b>302</b>	~	• !E-	_
	•	•	1643	78 +0. 99 8		59.4	20C	•	•	_
	ه.	•	1644	.07 10.61		11.1	308	•	• (E-	_
	,									

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ARC	ARC LOCATION	1L W COLLECTION NO.	ANALYSI S NO.	FU-239,240 &CTIVITY (DFP)	URANIUM (FICRO GRAMS)	Y JELC I R * R E WORK J	TIME	ANAL	ANAL JHON
z	028-7	37.6-7	CAF-1645	1.20#1.206-01		49.0	2C C	2.	(E-03
:	0 20		7	1.71 +0.05 8 02			300		(E-03
	032-1	334-1	CCD- 211	9.00 \$6.00 E-02		49.5	<b>3</b> CC	CA 2.	2. (E-02
	æ	~	212	2.5041.00E-01		37.2	) () ()	•	CE-02
	· ~	m	213	3.70 +2.90 6-01		34.4	<b>4</b> CC	-	1.66 00
	•	4	514	8.10 11.50 6-01		45.1	328	4	(E-C2
	. RJ		N	6.6011.00E-01		80.1	3C C	-	(E-05
	036	573	CIA- 435	1.16 to.04E 02		85.5	36	*	10-35
	040	S 42	97	9.15 to.726 00		25.B	2CC	Ψ.	3€ 00
	040-1	3349-1	CAO- 748	2.00 to. 70 E-01		49.8	704	CA 1.	íE CO
	~	~	149	3.00 47.00 E-02		24.6	<b>4</b> CC	•	\$ (E-05
	· m	m	150	0.3041.006-01		35°6	300		(F 30
	•	•	151	2.50 #2.10 E-01		19.2	<b>4</b> C¢	CA 1.	(E 00
	<b>4</b> 0	Ų	152	1.00 40.70 5-01		\$4.4	300		10-3)
	~			4.00 #1.40 E-01		70.1	96	_	(E-03
	045	571	C 14- 433	4.2044.206-01		51.1	<b>3</b> 00	5	(E 00
	046-1			1.51 to.07E G1		•	3 C C	•••	£E 00
	~	~		1.60 +1.60 E-01			<b>4</b> C	•	(E-C2
	~	•	156	0.20 \$1.00 E-01		37.7	306	<b></b>	00 3)
	*	•	151	9.00 t9.00 E-02		-	30C	~	. (E-02
	•0	Ý	158	9.00 tb.00E-02		4	<b>400</b>	•	£-01
	~		CAF- 759	2.00 14.00 E-02		~	3C C	∵ 3	(E-03
	040	558	C 1A- 431	5.20 11.30 E-01		36.2	3C C	_	(E 00
0	045	8126	CC 11-500	1.32 to.03E 04		65.3	2		
	046-A		11.71			•	Ç		
a.	273	3662	14-1	.1510.236	2060.0	7. °C	308	÷	). 26
	0.50	3667	C 10-1315	.4310.128	0.0880	69.1	906		. SE-C1
	1-520	1-226	5691-000	.6211.46E		11.6	<b>3</b> 0 <b>%</b>	Š	€-01
						•			

TAB	TABLE E.2 (CONTILUED)	NTTL.UED)			\$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$	; 6 9 9	;			
AAC	L0CA113N	TLW COLLECTION NO.	TEW ANALYSI S NG.	FU-239,240 ACTIVITY (DFP)	URAN 1UM I M I CRO GRAMS )	Y SEL D I R "RE WORK S	COUNT TIME	₩ W ₩	ANAL JHON	
					† • • • • • • • • • • • • • • • • • • •					
۵,	6-520	3322-3	CCD-1637	.37 tO. /BE		u	ر در		1	
	+	4		.1641.30E		<b>u</b> 1	30Z	_	· 16-	
	•	ĸ	7	.03 to . 69 E-0		~	<b>302</b>	•	ë.	
	026	993	C1A- 420	.15 tO. 19E O		-	.200	UMF	- 34.	
	030	9		.16 #0.18E 0		٣1	<b>30</b> 0	_	• (E-	
	030	6628	7	3040.406		~	500	_	. CE-	
	024-1	318-	CC0- 201	.1340.09E 0		O	30°	_	• EE	
	• •			.50 to. 13E 0		72.C	400	•	• <b>(E</b> ~	
		m	203	.60 41. 30 8-0		O	704		. (E	
	4	45	204	.70 +1.00E		4	30 <b>2</b>	2		
	· vn	<b>S</b>	CCF- 205	2.80 +1.10 6-01		44.C	<b>30</b> 6	<u>ა</u>	00 3) •	
	036	090		.37 40.07E		81.2	30	4		
	0.40-1	3316-1		,73 t0.09 E		19.1	<b>3</b> C	74		
	~		197	.85 t0.08 E		64.7	<b>308</b>	•		
	las.	m	1 98	.04 t0 . 14 E		35.4	<b>4</b> CC	(-1		
	∢	<b>-5</b>	1 99	301.1107.		52.4	30C	2		
	'n	v.		.80 +11.30 E		28.2	<b>308</b>	- ئ	• (E 00	
	045	C71	,	.6140.03		11.1	<b>.</b>	•		
	044-1	3329-1	CCD- 206	.43 #0.22E	0.235	62.6	) }	_		
	~	7	201	.37 40.136	•	45.4	100	~		
	m	m	802	.62 40.24E	•	60.B	30€	3		
	•	4	508	.40 tl. 70 E	•	3.8.5	) ) }	3	•	
	W)	s.		301 * 1 * 00*	•	m;	<b>30</b> 2	<u>ح</u>		
	0.48	3C73	C 10-1628	.76 to.12E		69.5	<b>30</b> 2	•	- 35 •	
	750	3072		.64 tO . 16 E	0.142	62.7	100	•	• (E-	
	040	3665	_	.36 10.48 €		€0.€	) ) )	,	• EE-	
	51.1	3676	10-1	17 10		27.5	<b>3</b> 00	_	1. SE-C1	
BAL	11,96	5294	CB 5-1442	.25 10.07E		85.6	<u>۲</u>	~	Ä,	
	t. 11, P.22		1644	9		83.6	20	_	, 1E do	

TAE	TABLE E.2 (CONTIN	(TINUED)			1	1	1 1 1 1	***************************************
ARC	LOCATION	COLLEC TI ON	TLE ANALYSIS NO.	PU-239,240 ACTIVITY	UR AN SUM I F I CRO	Y JELC IRORE	COUNT TIME	ANAL JHON
			; ; ; ; ; ; ; ;				; ! !	
BAL	_	5284	CBS-1434	.1240.09E 0		14.1	20	
	_		1436	.06 tO. 14E 0		~	26	36
	L 12, P 9		1438	.05E		19.1	36	
	£ 12,019		1440	.7310.096 0		84.4	36	
	3,99	5298	1648	.92 t0.09 E		84.2	<b>,</b>	• tE
	4.P 3	29	1446	.28 t0.03E		68.1	26	. Es
	1.15,917-1	~	CCD-1656	.43 t0.08 E		72.1	<b>)</b>	• £E
	2	2	1657	.93 40.07E		. H.	203	. E-
	•	m	1658	.28 to .21 g		£. 2.	30°C	• 4ñ
	4	<b>√</b>	1659	0.0 9.		54.5	<b>30</b> 2	2. EE-01
	80		CF.	.7 0.06E		45.4	<b>200</b>	• 4E
	L 18, F 21		CC0-2180	.6940.05E	3.70*	15.4	¥	
	1.18,P21-3	m	2130	07 tO. 12E	0.582	75.5	36	
	4	4	2331	.87 10.07 E	4.	16.6	30C	
	\$		CCF-2132	.75 10.43E	0.0103	63.2	30C	
	L 19,P9	1C13-A		.79 to.13E	16	02.68	300	
	L 25,P 9 ·	3C38-A	21.79	.21 t0.00E	108,	£2.1	<b>5</b> 80	
	129,89	-995	CCD-2181	.41 to.01E	6.20	36.4	301	
	£-6d.627	~	2133	360.	0.0305	10°C	ر دور	
	4	4		.38 tO.36E		69.1	302	
	S	S	1	.68 tO.83E	0.00100	75.5	300	
M08	KM-C07-1	3597-1	7	.30 t0.02E		•	26	
	~	2	222	.00 to.06 E			30C	
	~	m	223	1.3240.086 01		34.6	704	
	•	•	224	.5044.90E		22.5.	300	1. (E
	<b>1</b> 0	5		.00.49.00.			2C.C	1.6
	1-11)	3405-1	2	7.7510.01E 02		81.6		€. € 00
	7	~	181	4.60 t0.90 E-01		10.4		
	<b>E</b>	•	1 88	1.9011.106-1	-	63.6		CA 5. (E-02

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TAB	TABLE E.2 (CONTINUED)	NTINUED)	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					-		i
ARC	ARC LOCA: 10N	COLLECTION	Z	FU-239,240 ACTIVITY	CP BN IUR	Y JELE IR BE	COUNT	Z	ANAL JHON	
1	9 9	-QN	NO.	(DFF)	GRAMS)	MOKA				i
A. C.	KM-C13-4	3.05-4	07:	1.2340.15€ 00		64.6	200	3	0. (E B	O
)	<b>1</b>	. wa	CCF- 190	-0.5041.006-01		29.1	<b>4</b> 0¢	ర	•	~
	C12-1	3000-1	_	5.6640.19E G1		22.5	302		4.7E C	0
	2	· ·		2.30#1.00E-01		41.0	<b>300</b>	3	1. (E 90	0
	1	m	193	7.5044.90E-01		11.1	<b>30</b> 0	చ		0
	4	4	194	1.3040.90E-01		54.6	<b>3</b> 02	3	1. (E 0	0
	· 40	<b>L</b>		1.5040.806-01		65.0	<b>30C</b>	Z	1. CE C	္ပ
	C13-1	3(09-1	CAD- 729	64 40.04E	0.279	64.6	36		. 35	00
	2			3.77 to.27£ 00	0.0650	54.5	306		2. 76-01	
	i en	i (41)	731	30 +1 . 20 E-		11.9	20C	3	2.15-0	_
	4	4	732	00 44.00 E-0		71.5	306	2	1.16-01	_
	4	•	133	2.00#3.00E-02		75.2	) ) ( (	5	2. CE-C	_
		~		1.7041.30E-01	0.181	27.7	20C		3. CE-0	~
	C14-1	3C10-1	CAD- 735			73.0	36		8.1E 0	00
	~		736	1.6940.045 03		17.8	300			~
	m	(P)	737	4.00 #7.00 E-02		64.3	321	చ		0
	*	•	738	0.00 #0.06 E 00		3° C	100		1. (E C	00
	9	Ą	140	6.0049.00E-02		76.2	100	2		00
	~	~		0.0040.05E 00		76.0	30C	3	3 - 3) • 1	<b>6</b> 2
	C & 6-1	1-6556	CCD- 226	1.80 #0.06E 02		76.1	) ) 			ပ
	~	-		.3041.10E		41.6	20C	3		Ċ
	~	~	228	3.00 \$2.60 £-01		24.1	306	ರ		0
	4	4	553	2.4041.00E-01		62.1	30C	2		0
	8,	S	CCF- 230	-2.00 44.00 E-02				<b>I</b>	1. (E C	G
70	CHR-A3A	9101	C VS-2092	.12 #0.61E		27.5R	<b>3</b> C			
	A 38		$\sim$	.0140.20E 5						
	A SA		2094	1.21 to.04E 01			201			
	A 48		2095				<b>)</b> )}			
	₹ <b>V</b>		2097	3.3440.208 00			200			

TAB	TABLE E.2 (CONTINUED)	(TINUED)	# # # # # # # # # # # # # # # # # # #	• • • • • • • • • • • • • • • • • • •				***************************************	
) d V	LOCATION	COLLECTICN NO.	₹ 4	FU-239,240 AC11917Y IDFP)	UMPNIUM (PICRO GRAMS)	Y IEL C FR * RE WORK 1	11KE	ANAL MON	
	: : : : : : : : : : : : : : : : : : : :		• • • • • • • • • • • • • • • • • • •	•					
40	CHR-434	9121	C VS-2098			76 - 3			
,	15 ¥ 1112		. ~			10.5	1001		
	\ <del>\</del>		23.00			47.7	2		
	¥ 4€		2101			71.1	20.		
	. d . d . d		2102			12.4	2		
	( d)		2103			٤١.٦	)3 <b>?</b>	•	
	41x - A13	4656	CDS-1161			11.4	221	2.4E 00	
	418		. ~			40 P	¥	5. (F C0	
	7 W		1163			** 29	) <u>)</u>	4. EF	
			1164			60.2	22	1. {{ c1	
	2 5		1165			e 1.5	) ) 	1. (5 0)	
	- C		1166	.12 40.08E		t t . t	100	3. 1E 01	
	2 6		1167	.77 10.09 E		34.2	10 <b>C</b>	3. 16 01	
		•	1168	.80 #0. 11 F			) ) 	4. FE 01	
27.0	010 0100-0-01-5	NUN	9561-553	.57 10. 168	8.32	4.4	<b>ာ</b>	7. (1-03	
נ د	7 - T V - 7 V		, –	.22 40.20E	5.52		308	1.(6-02	
	<b>,</b>		1948	.61 40.49E	24.6		) ) (	3. (E - C3	
	• •		1949	324 40. 12E	15.5	80.3	)) <b>?</b>	2. (E-02	
	, v		1950	6.8610.66E 00	3.86	•	<b>2</b> C C	1. (f-c2	
	٠		1961	50	14.6	35.5	308	5. (E-03	
	-		1952	7.1940.378 01	56.6	~	) ) (	5. (E-C)	
	A 3-5		1953	6.1340,25E 01	39.8	÷	<b>3</b> 0 <b>€</b>		
	7 ×		1954	8.2140.94E 00	26.1	•	<b>3</b> 2 <b>2</b>	F-	
	) r		1955	4.5540.108.02	102.		))Z	۳.	
	- •		1956	1.9640.06E 01	6.54	÷	<b>2</b> 00	•	
	<b>9 Q</b>		1 957	1.31 40.07 8 01	3.31	41	) )		
	, <u>.</u>		1958	1.55 to.07E 01	17.6	•	)) <b>&gt;</b>	٠	
	-		1959	3.86 10.14 01	0	£2.3	) ) (	•	
	A 4- 50		1960	1.25 to. 19E 01	26.5	34.8	<b>3</b> 20	•	

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TABLE F.2 (CONTINU	NTINUED)		4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			
ARC LOCATION COL	CULLEC 11 0	TLE ANALYSIS NG.	FU-239,240 ACTIVITY (DFP)	UP BN TUM LY ICAG GR BYS 3	Y TELC (R.RE WORK)	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	ANAL MON
1:11:1:1:1:1:1:1:1:1:1:1:1:1:1:1:1:1:1	,	C SF - 1961	5.8140.45£ 00	1.4.7	91.0	300	5. (6-62
		1967	4.1740.24E 00	16.4	69.5	) ()	10 - 32 - 1
- •		1963	1,00,00,06	19.9	80.1	)) <b>?</b>	1.46 00
		1 364	2.6611.775-01	1.31	19.5	<b>3</b> 0 <b>2</b>	5. 36-01
~		1965	3.84 10.23 € 00	4.20	86.6	200	3. 2E-01
1 5 1 5 4		1966	8.3842.21E 00	25.7	64.2	100	1. {E-C1
•		1967	2.6910.358 01	24.0	45.1	)) <b>?</b>	10-36-61
9 (~		1968	5.71+3.26E UO	97.2	86.8	40	6. (E-C2
• «		1969	6.6940.24E 01	22.7	86.8	.2CC	1.16 00
Φ,		1970	1.7740.095 01	36.9	64.5	<b>3</b> 00	4. žE-CI
C		1261	1.1440.078 01	16.5	83.3	200	2. (E- C;
) ·					7 67	200	3 36.01

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CONFIDENTIAL

New data this report.

TAB	LE E.S RA	DIOCHEMICAL	, ANALYSIS OF	TABLE E.S RADIOCHEMICAL ANALYSIS OF ROLLER COASTER PHYSICAL SAMPLES, CLEAN SLATE	YSICAL SAMPLI	S, CLEA	SLATE	=	•
ARC	LOCATION	COLLECTION	TIN ANALYSIS	FU-239,240 ACTIVITY	UPBN LUM CPICRO	Y JELE (R*RE MORK)	COUNT	ANAL JPON	<b>*</b>
		;		<b>L</b> !					
Ç	21 - 16	5 8 4 2	CAC-1420	.10 to. 17 E		1.53	36	1.16	00
J P		9842-1	205	9.8410.27£ 06		94 0	36		
	•	)	2059	.89 40.186		w	3 č		
	) <b>V</b>	e71	2066	.97 tO. 22E		19.3	)?		
	<b>*</b>	~	2067			11.4	7		
	• 6	•	2060	.03 40.03 8		46.6	2¢		
			~	.4940.16E			2		
	- 2	202	~	. 20 €		34.1	Ç,		
	1		N	10.22E		32.4	<b>5</b> 0		
	7		~			46.6	<b>3</b> ¢		
			~	.49 40.178		16.6	26	,	
	BH- C 2	4.C82-	C 14-2		9.90	2.8.	~	2. 4E	ó
	30-C4	242	AC-1	1.7440.046 06		19.E	<b>5</b> C	,	,
	18-92	2303-4	12-00	.12 tO. 19E	24.6	04.48	<b>3</b> 00	6. 3E	ဗွ
•	910	116-	TA-21		2.95	S. C. S.	<b>)</b> 4		,
	<b>\$</b> 10	2286-4	CCD-2183	1.2643.026 04	17.0	11.5	30¢	4. 2E	Ö
•	010	45	AC-1	.3940.08E			ž	,	
,	***	2371-A	C0-2	.48 to. 28 E	52.1 •		30	1.16	00
	044-3	• • •	~	.97 +0.22E 0	0. 446 •		<b>&gt;</b>	5. (E	ပ္ပ
	•	•	2125	.38 t0.06 E	1.06		76	2. EE	ប្ដ
	•	•	~	.16 tO.06 E O	0.373 *	51.4	<b>3</b> 0 <b>C</b>	2. JE	ပ္ပ
		e12-1	C 14-2197	1.0540.036 03	4.15	29.3	<u>بر</u>		(
	040	2370-A	~	.73 to.07 £ 0	2.36	<b>→</b>	72		2
	040	***	~	.21 10.02E 0	00.0	<b>.</b>	700	*	5
	06.8	2369-A	2187	1104611	A6.6		<b>~</b>	•	
	068-3		-	0	1.13		2	3. 15	9
	•	*	-	-91 +0- 15-	2. 1c +		) 	Z. 4E	00
	*	<b>1</b> 0	-21	4 E O	1.05	<b>6</b> 1 (	<b>≍</b> ∶	1. :	0
u	010	9842	AC-14	3.51 to.08 E 06		-	<u>ب</u>		

TABI	TABLE E.3 (CONTINUE	N'THUED)			**************************************	1	1 2 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		į
ARC	LOCATION	כסרר	A	9U-239,240 ACTIVITY	URAN 1UK (FICRD	> ~ )	COUNT TIME	ANAL /MON	_
1	; ; ; ;		NO.		~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	K			-
Ų	040	9752	CAC-1429	.6740.07E 0		46.6	20		
,	0 0	•		82 to.04 E		11.3		E	00
	8 70	6111	CDS-1173	.36 to. 09 E O		_			2
	010			.92 to.06 E		0		₹.	õ
	010	~	AC-34	.094U.16E D	٠,	83.3	<b>3</b> C	,	
	012	8111		0 380.0168.	•	*		یږ	0
	4/0	,		.691 06E D		4		<u>بر</u>	2
	910		111	.12 to.04E 0		S.		. 7E	0
	910		1170	. 79 ta. 02E 0	. •	•		1. ¿€ (	9
	010		-	.65 to.04E 0		~		• (E	ö
	080	•	AC-1	.03 10.08 E 0		÷			
	062	9111	C05-1140	.56 to.03E 0		_		. 3E	8
	065		~	.4610.03E 0		•		J.	0
	0.66		1182	.30 to.03E 0		~			0
	0 6 8		1103	.11 to.02E D		¥		• ?£	0
	050			.00 to.02E 0		41		. Æ	0
	CSC	~	AC-1	0 340.0114.		4			
	C 5 3	9111	2	.90 to .22E 0		_		• •E	0
	150		1186	.38 to.22E 0		0		, 1E	Ö.
	950		1107	.61 40.19 6 0		47		- 56-	<b>=</b>
	0.50		1186	.67 to. 13E 0		~		te	0
	221		1189	.5210.126 0		•		¥:	S (
	103		1 90	.02 to.12E 0		•		<u>.</u>	2
	104		1611	.12 to .07 E O		~		• <b>t</b> w	<u>e</u>
	106		1 1 92	.03 to.08 E 0.		e,		• IE	0
	100		1193	380.0119.		~		• <u>:</u>	
	110		1194	.05 to.05E 0		0		. JE	0
	112		1 2 9 5	.65 10.046 0		~		• (E-	<b>=</b>
	114		9611	.7910.05€		~		- 3:	=

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TABL	TABLE E.3 (CONTINUED)	NTINUED)				; ; ;		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
ARC	LOCATION	TEK COLLECTION	71.W AMALYSIS	FU-239,240 ACTIVITY	URANIUM	Y IEL C I R*RE	COUNT	ANAL MON	
!	6 6 7 2 8 8	NO.	NO.	(OPF)	GRAMS)	ORK I			
٥	666-1	2228-1	CCD- 441	14.		72.5	ပ	4. CE	
ì	7	7		2041.		66.2	30C	CA 2. 5E-01	
	· ল	(F)	443	.40 40.90E			Ų	<b>1.</b> CE	
	•	•	***	٠,			C	-3: •2	
	'n	·w	4	00 *6 * 00 *			S	E	
	010	6112	COS-1197			62.3		7. SE CO	
	012		-	.33 to.04				. ž£	
	014		1199	.66 40.11				F	
	910		1200	3 40 - 13				1. te 01	
	910		1021	.76 40.13		r;	36	1. CE 01	
	020		1202	.89.00.08		"	¥	<b>.</b>	
	050-1	3180-1	CA0- 910	3.61 t0.09E 04	14.6	72.C	36	1.56 61	
	~		116	.40 40.07	1.14		36	¥.	
	· M	m	915	.78 40. 101	.49	Œ,	100	• <del>(</del> E-	
	*	4	913	.95 #0.63	.15	•	) ) (	• (E-	
	w	9	<b>\$16</b>	.10 +2 . 20	0.252	64.E	100	2. (E-	
	~	~	AF-	041.90	.45		<b>300</b>	CA 3. (E-02	
	022-1	2227-1	CCD- 436	.42 10.04		9.59	2	3. 78 00	
	7	~	437	.92 +0-22		12.3	7	5. (E 0C	
	~1	~	438	.68 to .09			266	3. (E	
	#	*	439	.20 41.20		•	) ) (	CA 1. (E 00	
	'n	*	CF-	.7041.50		•	100	2. CE-	
	220	8112	CDS-1203	1 40.09		¥	<u>۲</u>	5. E CO	
	¥20		1204	.30 10.07		76.6	76	4.40	
	030	4163-A	~	.39 40.11	0.340	_	) ) (	¥	•
	970	6112	05-12	. \$2 40.02		~	¥	¥.	
	010		1206	.28 to.03E		71.6	2	3. ee cc	
	2:0		1207	~		4.0.4	2	E	
	032-1	2232-1	CCD- 456	9.67 tO. 26E 02		m,	<b>3</b> C	W.	

[ABI	ABLE E.3 (CONTINU	NTINUED)		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	!			•	
) ()	LOCATION		TEN ANALYSE S NO.	FU-239,240 ACTIVITY (DPM)	URAN IUM I M I CAD GRAMS )	Y JELC (R#RE WORK)	COUNT	ANA	ANAL /MON	
1	- - - - - - - - - - - - - - - - - - -					,		,	ļ	
_	0:2-2	2-2822	CC0- 457	.0440.07E 0		£ 6 - 2	200	- 1	, i	
	~	~		.75 to.22E 0		16.8	) ) (		• EF	
	*	•	459	.81 to.35 E O		75.8	300	<u>3</u>	. CE-01	
	•	•	F- 4	.10+1.10E-0		75.5	<b>3</b> 00		• (E-	
	024	162-	0-21	.55 tO.08 E O	968.0	51.6	2CC	•	• (E-	
	**	=	5-1	.55 to.19£ 0		71.5	36	~	. JE	
	0.24-1	2233-1	*	.2440.14F		62.5	300	~	H	
	7		46	.33 tO.03E 0		82.5	200	<b>.</b>	ָּהָ יַּה	
	(P)	m	4	.71 to.05E 0		17.6	4 C C		F	
	*	*	494	.6440.25E 0		€0°€	2C C	3	E	
	**	<b>'</b>	- 46	.15 to.13E 0		64.4	100	~	• tĒ	
	024-3	3182-3	CA0-2142	.27 10.22E 0	0.0122	56.7	2C C	_	- 3E •	
	4		~	.89 to . 05 E O	1.15	71.7	J J J		بپ	
	•0	49	2144	.55 t0.56 E-2	.67	66.1	<b>305</b>	<b>(</b> ,	. <del>(</del> F-	
	^	~	2145	.2540.038 0	0.502	e1.c	) ) }	,~	• <del>(</del> E-	
	036	164	14-127	4041.		5.1.v	200	_	1. (£ C0	
	036	8112	6021-503	.44 tO.14E 0		61.3	20	174	. 4E	
	036-1	4151-1	54-1	.00 t0.19 E 0		66.C	2CC	~	• 2E-	
	~		1459	.28 to.20E		54.2	2C C		• ¿E-	
	æç	•	1460	.7440.20E 0		4.11.4	3CC	۲ ح	¥.	
	<b>₹</b>	*	1461	.6740.16E 0		73.5	200		. 4E-	
	¥	•	1462	.07 43.		83.6	4	5		
	•	•	1462	.98 46.		84.8	200		• (E-	
	~	-	1464	.0941.04E-0		£7.5	300	,-		
	e)	•	1465	.75 14.60 E		£1.6	7	5	• (E-	
	ው	6	1466	.74 to. 37E		74.7	306	_	• (E-	
	2	70	1467	.65 40.09		72.5	3CC	14	2.4E 00	
		6112	2-5	.23 tO. 15E		£9.¢	3.5		• £	
	0:8-1	1-1222-	CCO- 451	.2 ! 4		£0.4	<b>5</b>	<b>7</b> *1	H	

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TAB	TABLE E.3 (CO	(CONTENUED)	1				,	•
ARC	LOCATION	COLLECTION NO.	TLE ANALYSIS NO.	FU-239,240 ACTIVITY (05P)	UR AN TUH (V TCRO GR AMS )	Y IELC I R #RE WORK I	COUNT	ANAL
1				•••••••				
c	0 - 8 - 0	2233-2	CCD- 452	1.6540.06E 01		£6.8	) ) <del>\</del>	-
•	) )			3.70 \$0.14E 01		61.3	200	7
	· «	•	454	2.80 to. 12E 01		56.0	<b>30</b> 2	7.
	· 61	. <b>16</b> 7	CCF- 435	5.44 to.33E 00		74.1	<b>30</b> 0	.÷
	040	8112	CDS 511	5.65 to. 13E 05		£8.E	۲	<u>۴</u>
	040-1	3183-1	CA0- 916	3.3140.08E 02		72.4	<b>4</b>	•
	~	~		2.5940.06E 01		66.4	<b>300</b>	
	, en	· M	916	1.5140.05E 01		76.1	<b>4</b> CC	÷
	•	**	616	1.25 to. 15E 00		5.8.5	300	•
	· •0	· •	920	5.0041.10E-01		58.5	<b>302</b>	۲ ۲
	~	~	CAF- 921	4.3041.50E-01		76.5	2	_:
	042	4165	C14-1275	9.18 to. 20E 01		13.5	707	÷
	04.0	8112	CD 5-1212	4.9140.102 05		86.0	>	<b>:</b>
	04.5		1213	4.6410.138 05		( ) ( ) ( )	3C	2.
	044-3	2230-1	955 -000	1.01 +0.03 € 03		15.1	2 €	<b>~</b>
	~	~	447	6.0440.19€ 01		75.1	308	ë.
	•		448	4.27#0.28E 00		75.8	30 <b>č</b>	÷
	· •	**	655	7.70+1.106-01		71.5	200	<b>S</b> 2
	•	**	CCF- 450	1.60 t0.90 E-01		76.C	<b>30</b> 2	CA 6.
	0.46	8112	CG 5-1214	3.85 t0.07£ 05		53.7	3 C	æ.
	0.48	4147	C SA-1468	1.0710.036 02		34.6	<b>302</b>	C 3.
	8 4 0	4134	C 70-1279	1.97 to.CSE 03		81.2	×	o.
	046	4166	C 14-1276	9.10 10:20 E 01		74.0	100	٠.
	0.48	8112	CDS-1215	3.64 t0.08E 05		J0.C	3 C	~
	050		1216	3.31 to.08 E 05		4 B . E	ž	÷
	7,0		1217	2.61 10.06E 05		29.7	36	÷
	052-1	3186-1	CAD- 928	6.1340.176 01		49.6	<b>4</b> CC	~
	~	7	626	3.56 to. 10£ 01		16.2	400	٠.
	m	•	930	1.2710.08E 01		25.4	<b>)</b> ) <b>†</b>	÷

TAB	TABLE E.3 (CC	(CONTINUED)		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0 0 0 0 0 0 0 0					i
ARC	LOCATION	COLLECTION NO.	TLW ANALYSIS NO.	FU-239,240 ACTIVITY (DPF)	URBNIUM IPICRO GRANSI	Y JELC FR*RE KORK	COUNT	1	ANAL JHON	i
!	1 1 1 1 1 1 1 1 1 1	:	• • • • • • • • • • • • • • • • • • • •	• • • • • • • • • • • • • • • • • • • •		; ; ; ;	, 1 1 1 1			i
c	4-6-0	3186-4	CAD- 931	.10 40.		72.8	909	-		
•	4		<b>.</b>	94 #0. 14E		82.7	223	-	Ĥ,	
	<b>-</b>	, ~	CAF- 933	70 +0. BOE		0	400	7	Œ,	
	750		7	34 10.		43	100	•		
	4 4 0	8112	C15-1218	.48 #0.06E 0		60.2	36			
	7 4	1110		.4940.05E 0		~	3 C	• •		
	1 - 9 - 0	2238-1	CCD- 486	.76 to.07E		67.8	2	•	3.3£ 01	
	• ^		<b>}</b>	.70 #0.04E 0		64.9	20C	.,		
		4 (**	4.88	98 40 - 19 6 0		41.4	300	•	3E	
	۱ ،	1 -	489	.62 t0 . 13E		61.5	30Z			
	ישי	• 447	CCF - 490	1.41 to.11E 00		75.4	3 3 <b>4</b> C C	5	1	
	, 89 69	112		.29 to.06 E	•	S	36	•	Ä	
	0.50	3185-1	A50-	.32 40.06E 0		~	7			
	~			.51 40.22E 0		~	100		بڻ	
	1 ~~	1 (17)	924	.2740.04E 0		Y	400	•		
	•	•	925	.75 t0.06E			40C		ш	
	• •	. 43	926	.7640.26E D		æ	2CC		•	
	, ~	. ~		.46 #0.18 E O		81.6	20C	•		
	. 070	9	C TA-1278	.70 to. 70 E-0			304			
	0.66-1	2239-1		4.00 to. 70 E-01		•	300	5	1. CE-C!	
	~	) )	4	.60 to. 90 E-0		•	300	చ	1. E-C1	
	) <b>(</b> **	•	ு	.7041.10E-0		•	300	3		
	•	•	464	.00 55.			400	3	7. (E-02	
	• •	· •••	CF- 4	.00 #5.00 E-0			400	5	1. (E 00	
	017	6112	•	.90 10.05E 0		•	<b>3</b>	•	1. (E 00	
	042-1	2234-1	<b>♦</b> -03	.3010c 70 E-0		•	30C	Š	1. (E 00	
	~	~	,	00 *6 * 00		£ € • 3	30C	3	10-33.1	
	· ***	m	468	.00 18.		٠	<b>308</b>	3	00 3) 1	
	) <b>~</b>	•	469	.3910.16E O		£ . £ 4	<b>3</b> 00	<b>5</b>	. G CO	

TABL	E E.3	(TINUED)			**************************************		1		•	
	ARC LOCATION	TEN COLLECTION	TLW AMALYSIS	PU-239,240 ACTIVITY	URAN IUM ( M I CRO	Y TEL C I R * R E	COUNT		AHAL MON	ž
		NO.	NO.	(0.6.1)	CR AMS 1	WORK )	!	•	! !	;
۵	062-5	*0	CCF- 470	2.6040.70E-01		83.2	308	3	1. (E-	5
	064		CDS-1222	1.85 to.04E 05		71.6	30		1.16	S
	1-570	_	CAD- 934	2.48 tO. 10 E O1		19.6	300		6. 2E	00
	7	7	938	1.10 +2.10 E-01		66.9	پ چ	చ	1. CE	00
	m	m	936	1.20 +1.20 E-01		20.0	) †	3	1. CF	00
	4	•	937	1.50#1.50E-01			7	5	1. (E	0
	÷	<b>.</b>	938	4.89#0.23E 00		41.5	) (1) (2)		4. SE	8
	-		CAF- 939	9.00 #5.00 E-02		77.0	300		1. (E	8
	970	158	TA-1	1.21 to.13E 00		75.4	200		3. CE-	5
	066-A	8112	COS-1223	1.51 #0.03E 05		3.0€	), ,		2. 1E	00
	€0		~	1.5840.036 05		68.1	3C		2. (E	00
	0 6 8		1225	1.53 #0.05E 05		3.8.4	36		1. EE	00
	068-1	2237-1	CCO- 481	3.22 t0.70 £-01		41.4	300	3	1. CE	00
	~	7	482	00 41 . 00 E-0		63.E	<b>3</b> CC	3	ا. ش	Ö
	m		483	0		45.C	1CCC		3. (E	00
	•	•	+8+			45.8	<b>3</b> ℃	5	1. (E	00
	· w		CCF- 485	9		72.0	<b>30</b> 0	3	1.66-	~
	070	8112	COS-1226	1.4840.036 05		J- 8-9	3.C		1 · EE	ပ္ပ
	012		1221	0		62.8	7		1.36	8
	920		E 10-1280	2.19 t0.05E 03		81.9	<b>5</b> C		1. EE	8
	074	~	CDS-1228	.5240.03E 0		62.4	×		<b>l</b> . 1E	8
	1-410	2-1	114 -022	0		62.3	36		2. SE	္ပ
	~	~	472	е О		65.5	2		2. i£	8
	M	•	473	10.04E 0		70.5	<b>308</b>		33 . 2	8
	•	•	+14	2.0040.086 01		72.5	3CC		J. 7E	00
	ĸ			10.05E 0		91.5	) ) (		2.(6-	5
	076-A	8112	COS-1229	10.04E 0		10.5	<b>3</b> C		1 · 1E	10
	€3		1230	44 10.04E		30.0	30		1. IE	8
	0 7 8	4162	C 1A-1273	2	56.9	£7.8	<b>3</b> 6		6. (E-	ត

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TABI	ABLE E.3 (CONTIN	NTINUED)			1 1 1 1 1 1 1 1	1		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
ARC	LOCATION	COLLEC FION	TLW ANALYSIS NO.	PU-239,240 ACTIVITY (DFF)	URAN IUM (FICRO GRAPS)	Y 1EL C I R *R E WORK )	C0041 114E	ANAL JHON	
	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		***	•	• • • • • • • • • • • • • • • • • • • •	 	 	:
٥	9.0	8112	C05-1231	.27 #0.03E 0		v	<b>3</b> C	4E 0	0
,	٠.		12	.2440.02E 0		¥		0	O
	) Œ		. ~	.2740.03E 0		-		7E C	ں
	066		1234	F 0		š		7E O	0
	6	2236-1	· •	.4440.11E 0		<b>:</b>		0	o
	,		4	.5040.21E 0			~	3E C	ပ
	יח ו		478	.21 40.03E 0		;	J	U	ပ
	۱ 🐗	· •	419	.76 40.20E 0		2		4E 0	_
	r <b>4</b> 7	• ••	CF - 4	.18 40.06 0		£.	S	(£-0	_
	23		CDS-1236	.00 #0.18E 0		11.1		. SE C	Ç
	· v		12	.49 to.17E U		æ		U	ပ
	· v		C	.91 to.15E 0		ç		. M. C	a
	· v		~	.54 10.10E 0		41		<u>ب</u>	0
	٠.		~	.5740.12E 0		¥		Ç	U
	, <b>U</b>		~	.17 to.12E 0		æ		بي	J
	·	•	~	.62 to.09 E O		41.6		1E 0	0
				.11 to.07 E 0		~		. 2E O	0
	108		1244	.7040.066 0				Q	0
	· –	-	~	.61 #0.07E 0		4		¥	0
	-		ď	.33 #0.06E 0		7		0	0
	-	5	10-12	.93 to.11E 0	20.1	~		6-0	_
	-	1	N	.91 t0.04£ 0		ው		3 - 3E •	
4	-	0113	12	.23 #0.11E 0				U	0
,	-	: 	~	.85 t0.07E 0		~		. 3E 0	0
	-		~	.82 t0.09 E 0				. TE 0	0
	~		~	.40 40.148 0		•		. EE O	_
	17		~	.35 t0.09E 0		Ġ		3.2E C	o.
	026		1253	43 40.06		_	×	. 46.0	0
	020		~	.28 t0.03E 0		13.8		5. (6 0	0

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TAB	TABLE E.3 (CONTINUED)	NTINUED)	1		1 2 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	i 1 1 1		1
ARC	LOCATION	COLLECTION	TL WANALYSIS	FU-239,240 ACTIVITY	UR AN JUM	Y IELC I R*RE	COUNT TIME	ANAL THON
!		*0N:	NO.	(08%)	GRAMS )	ORK I		
u	0.0	2113	<b>LO</b>	.21 40.14E 0		•	36	. 1E 0
,	0.22	•	1256	5.78 t0.13E 05		14.1	<b>5</b> C	2.26 00
	024		25	.44 #0.10E 0		•	<b>5</b> 2	• <b>6</b> E 0
	0.26-A		25	.81 #0.08E 0		•	), (	. 4E C
	<b>6</b> 2		25	.6740.078 0		•	<b>9</b>	• 4E 0
	0.28		26	.40 #0.06E D		٠	O 61	35.0
	044-1	9640-1	23	.46 tO. 14E 0	.87	٠	<b>5</b> C	
	~		223	.92 t0.05E 0	0.161	٠	3CC	ပ မ
	141	•	23	.91 *0.22E 0	• 10	•	<b>5</b> CC	
	•	•	~	.46 tO. 10E 0		٠	<b>5</b> 0 <b>C</b>	. SE 0
	· <b>4</b> 0	ι.	23	.42 40.22E U	.24	•	BCC	. 4E O
	044-1	9641-1	23	.16 40.12 €	6.02	•	101;	. žE C
	~		23	.72 #0.11E 0	-13	٠	)) <b>?</b>	۳,
	, eri	m	23	.38 t0.07E	0.0550.0		1000	, 4E 0
	*	•	23	.51 to.06E 0	(3	•	1000	. £E-0
	· 67	'n	23	.65 #0.13 E 0	50.2	75.6	1000	• <b>E</b> -C
	C48-1	9638-1	22	.37 to. 12E 0		•	<b>300</b>	
	~	7	22	.05 tO.01 E O	•	~	20C	
	(P)	6	22	.8340.026	6	Š	100	1. EE 00
	•	•	22	.66 40.06E 0	0	÷	10CC	•
	**	87	22	.41 t0.03E 0	-15	÷.	1000	- 4E-
	048-1	1-6696	22	.49 to. 11E	0.335 •		<b>5</b>	
	~	~	22	.11 to.03E 0	.21	÷	ت م	1
	<b>m</b>	•	22	.84 #0.05E 0	.62	φ.	U	. ff.
	•	•	22	.56 t0.15E 0	•	ċ	302	9
	'n	*	222	.43 t0.30 E 0	2.	ċ	0	• (E-0
Ŀ	920	8119	~	.59 10.11 8		0.00	2	1. 26 00
	1-920	2173-1	CO-157	.16 t0.03E 0			<b>3</b> C	
	7	7	5	.14,10.			<b>y</b>	•

NEW DATA THIS REPORT

	1,4	FU-239,240	UPANIUM	YIELD	COULT	ANAL MON
	N ANALYSIS NO.	ACT1V!TY (0PP)	CP ICRO	(R*RE MORK)	TIME	
,					,,,,	,
113-3	6/67-077	317 · 0 · 41 ·			77	ָי יָנ
Λ.	1580	.2940.05E		711.6	30 <b>2</b>	¥
5	CCF-1581	.56 #1.00E		70°C	<b>3</b> 00	1. (E-C2
642	CAC-1423	.61 tO.06E		16.8	36	
9119	1921-503	.1646.068		14.1	3.5	
	1762	.34 40.03E	٠.	71.3	100	1.36 03
	1763	.0240.12F	•	75.3	<b>5</b>	
	1764	.03 10.03E		65.2	<b>3</b> C	. 1E
	1762	.1740.03E		8.58	<b>5</b>	
	1766	.92 to . 09 E		63.2	3 C	3; t
646	CAC-1424	8.50 tO. 18 E 06		76.2	<b>3</b> 2	
6119	COS-1767	.76 40.05 €		71.6	4	-3
	1768	.18 #0.05 E		£8.3	<b>3</b> C	£
	1769	.75 to. 16E		£6.2	36	1.76 00
	1770	32 tO. 18E		11.1	<b>3</b> C	É.
9646	CAC-1425	.54 to.03E		19.3	<b>3</b> C	
1119	CD S-1771	301.0496.		62.3	36	9. ce - cl
	1112	.0340.11E		20.5	300	
	1773	.50 tc. 06 E		59.5	3 C	j.
1646	(ディー1426	.01 40.02 €		17.3	26	
6119	COS-1774	360.C146.		48.7	<b>3</b> CC	2. SE C2
	1775	.38 50.07E		51.4	200	ZE-
942	CAC-1427	.76 tO.13E		76.3	<b>5</b> C	
	~	.37 40.08 8		66.7	26	
1-952	CC0-1597	.5340.126		100	<b>∪</b>	<b>.</b>
~	1598	.38 tO.06 E		63.3	3 C C	3. 'E 60
m	1599	.44 10.07E		24.1	) ()	<del>.</del> 6
•	1600	.02 #0.03E		58.3	) ) <b>?</b>	. te
•	CCF-1601	.20 to . 16 E		50 · F	200	- 35 -

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TABI	TABLE E.3 (CONTINUED)	N'ITNUED)					1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
ARC	ARC LOCATION	TLW COLLECTION	TLW ANALYSIS	PU-239,240 ACTIVITY	URAN JUM (FICRO	Y IELC (R*RE	COUNT TIME	ANAL /HOH
1	1 1 1 1	•0N	NO.	(DFP)	CK AMS 1	Z	1	
7	022-1	2205-1	CCD-1587	1.08 #0.04E 01		83.6	200	
	~		1588	1.88 to.07E 01		62.4	<b>30</b> 0	÷.
	1 (47)	· (4)	1589	.20 #0		81.5	20C	
	•	•	1590	.0041.75E		47.2	<b>500</b>	<del>-</del> 9) •
	- 50	· w	7	.43 #2.66E-0		œ	¥	E
	0.26		-	.46 #0.08E 0			<b>308</b>	. žž.
	054-1	22C1-1	CCD-1582	.26 10.06 E D		72.1	<b>4</b>	
	~			.89 #0.09 E O		Ð	20C	. 46
	· ~	m	1584	.80 #0.29E 0		•	30Z	• <del>[</del> E
	•	•	1585			0	30 <b>C</b>	1.(6 00
	•	·w	7	.71 #1.20 E-0		_	30C	• (E-
ل.	045		C TA-1665	.68 #0, 17E 0		71.5	<b>4</b>	4.18-
<b>;</b>	0.48	4017	_	.81 t0.20E 0		÷	ű	
	0.60-1	2222-1	CC0-1592	.03 10.126 0		¥	<b>3</b> C	<u>1</u> E
	~	-	1593	.05 t0.19E 0		4.69	20C	æ
	m	•	1594	.02 t0.18E 0		44	<b>5</b> 00	
	4	•	1595	.01 #0.04E 0		.57	400	Ť
	ĸ	*	7	.81 +0.09 € 0		_	<b>30</b> 2	Ť
BAL	L 1,P17	4C22-A	C TA-2192	.20 44.65 6-0	0.593	45.7	40	6. 26-
	L 2, P 5-1	2310-1		.00 43.00 E-0		4.19	300	J. CE
	~	7	245	.80 +1. 20 E-U		47.8	30¢	. Œ
	~	•	543	.88 tO.24E 0		13.61	306	• (E
	•	4	544	.40 t0.90E-0		41.3	306	7. E
	*	· Kn	CCF- 545	.80 #1.00 E-0		49.0	) 32	2. (E-
	L 2, P13-1	2314-1		.48 t0.01E 0	2.63	71.6	<u>۲</u>	ų
	7	~	547	.3140.02E		_	36	<b>.</b>
	•	m	548	.1210.076 0	2.64	13.1	Ç	<b>.</b>
	•	**	549	.29 to.08 E		œ	<b>1</b> 00	
	<b>5</b>	*	CCF- 550	2.8810.116 01		69.3	<b>3</b> 00	1.16 00

TAB.	TABLE E.3 (CONTINUED)	TINUED)								:
ARC	ARC LOCATION	TLW COLLECTION NO.	TLW ANALYSIS NO.	FU-239,240 ACTIVITY (DFM)	URANIUM (FICRO GRANS)	Y IELC (R#RE WORK)	COUNT TIME	X X	ANAL JHON	-
!			; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;	* * * * * * * * * * * * * * * * * * * *						)
BAL	13.09	2312-A	CCD-2185	4.24 to.09 E 03	.55	74.3	<u>ح</u> د			
1	1 2, 9 9-3		2115	3.01 to.10E 02	•	19.8	3 C		8. 16-(	5
	•	4	2116	3.09 to.11 6 01	0.841	19.6	זכנ		1. (6-)	6
	· <b>4</b> 7	· 10	CF.	1.3940.046 02	•	12.1	<b>30</b> 0		1.46	8
	13,917-1	2307-1	CC0- 531	7.50 #1.10 E-01		₹8.€	40C		1. CE .	00
	2			3.00 #5.00 E-02		C: **	400		1. CE-	20
		•	533	4.70 t0.90 E-01		42.1	400		1.66	8
	· •	•	534	2.48 #0.23 € 03		39.4	400	5	2. (E	00
	· 10		GF.	1.9011.106-01		44.C			5. (6-	02
	1 2, 9 22	318	CBS-1456	5.63 #0.12E 03		88.6			. ië	ဝ္ပ
	6 d • 5 -	5317		2.52 to .05 E 04		83.2			1. EE	00
	L 4, P 21	4C24-A	C 1A-2193	7.76 t0.23E 02	* 09L ' 0	55.4	36			
	1 4, P 22	317	8 S-1	.12 t0.04E		81.9			3.46	0
	[ 5, P l- 1	2322-1	CCD- 561	.12 t0.05E		67.1	1000		3.56	8
				391.016€		1:51	308		1.46	ပ္ပ
	. ~		563	.52 #0.41E		€0.€	308	3		00
		•	564	.17 to. 11E		19.6				00
	* 417	•				74.2	100		J . 4E	00
	16.911	5316	CBS-1557	.63 t0.06 E		78.3	3 C			ပ္ပ
	9	•	-	.06 to. 11 E		71.0			Ψ̈	0
	(6, 911	5315	1450	.89 10.04E		19.8	<b>5</b> C			00
	4		1451	.66 40.08 E		85.2	۶ د			၀
	( 6, P.13-1	2308-1	CCD- 536	.07 t0.21E		81.2	36		5. IE	00
		, ! !		.76 to. 28 E		76.2	100			8
	9 647	, m	538	.2010.136		65.3	<b>3</b> 00			00
	•	4	539	.44 10. 16E		14.3	306	చ		5
	* 67	•	,	.58 10.30€		71.4	) ) (		1.66	8
	L 6, P 21-1	2321-1	955 -022	.85 to.05 E		11.1	<b>&gt;</b>			8
		7		2.7310.096 02		72.4	<b>3</b> C			00

\*New Data this report

TAB	TABLE E.3 (CONTINUED)	(TINUED)							
ARC	ARC LOCATION 11 COLLEC	11 M COLLEC 71 ON NO.	TLE ANALYSIS NG.	FU-239,240 ACTIVITY (0FM)	UP AN JUH (FICRO GRAMS)	Y IELC IR=RE WORK I	COUNT TIME	ANAL JHON	
	0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		,						
6 J	L 6, P 21-3	2321-3	CCD- 558	8.89 to. 30 E 01		£ 5 . £	ر اد	ָ נו	
	•	4	559	.14 40.28E		71.5	100	. 2E	
	41	S	CCF- 560	.9940.19E		62.4	400	CA 3.(E 00	
	17.09	4C11-A	-21	342.0101.	1.16	42.1	100	• 1E	
	17,917	4010		0.00 to.04E 00		72.4	400	1.(E 00	
	L E, P 13-1	2343-1	119 -000	.02 #0.02E		10.1	400	2.58 01	
	7	~		.57 40.28E		\$9.C	704	1.46 00	
		· (1)	613	.76 t0.14E			400	4. CE - C1	
	**	4	<b>519</b>	٠,٠		81.0	) []	CA 1. (E 00	
	S	r.		2.1041.40F-01		57.1	100	1. CE	
	L E, P 21-1	2336-1	265 -022	.20 #0.15E		67.1	3CC	CA 1.CE 00	
	7	7	597	8.59 to.30 E 00		44.0	<b>3</b> 26	<b>8. £</b>	
	E	m	965	•		56.8	3CC	1.06	
	•	∢*	599	.00 +8 .00		83.6	36	CA 1. (E-02	
	'n	W)	CF-	5.00 49.00 E-02		66.5	<b>)</b> 5	1. (E-	
	1-60'57	2339-1	909 -000	2.0044.00£-02		£1.2	) ) (	1. (E 00	
	7	લ્ય	209	1.30 +1.90 E-01		41.2	100	1. CE 00	
	~	n	809	1.60:1.20 6-01		66.5	100	1.CE 00	
	4	4	609	0.00 to.06 50		61.5	100	1. (E JO	
	'n	S	CCF- 610	2.3010.806-01		67.6	100	CA 5. (E-C2	
	1-5,917-1	2338-1	,	-0.4041.20E-01		80.4	<u>و</u>	1.(6-	
	2	_	209	0.00 #0.09 E 00		67.7	<b>J</b> 6	1. (E-	
	m	~	603	1.20 to.70E-01		19.5	3CC	3. CE-	
	4	<b>J</b>	<b>\$</b> 0 <b>9</b>	-1.00 +1.30 E-U1		27.1	30C	1.06 00	
	ĸ	L	CCF- 605	0.00 40.036 00		81.8	20C	1. (E 00	
	L 10,P 5	4014	1-11	1.2040.706-01		80.2	306	1. (E 00	
	L 1C, P 13		1266	5.9010.146 01		65.3	<b>300</b>	7. 4E-01	
	L 16, P 21	401	1267	2.11:0.185 00		***	400	2. 1E CO	
	L 11,P17-1	233	165 -000	2.4041.20 € -131		58.5	30 <b>C</b>	C4 1.(£ 00	

TAB	TABLE E.3 (CONTINU	INUED)		0 8 8 8 8 9 9 8 8 8 8 8 8 8 8 8 8 8 8 8		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1 1	3 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	
AKC	ARC LOCATION CO	1LW COLLECTION	N Y	FU-239,240 ACTIVITY	UPZNIUM (PICRO	Y IELC (R=RE	COUNT TIME	ANAL JHON	
1		NO.	• C V	(000)	CKAMS J	<b>5</b> '		1	
8.41	. 11.017-2	2335-2	CCD- 592	.30 11.10		65.6	2C C	1. (6	
7		•	) }	00 47		3.55	30 <b>2</b>	1.08-	
	**	4	594	.0014.00E-0		66.4	<b>500</b>	CA 1.CE 00	
	· <b>v</b> n	'n	1				<b>300</b>	1.(6-	
	L 12,P 13-1	2334-1	CCO- 586	.54 t0.04E 0		83.6	<b>500</b>		
	7		583	40.34E 0			<b>3</b> 25		
	•	m	588	.30 tl. 30 E-0			<b>300</b>	2. (E-	
	•	<b>4</b>	589	43 to.21 E 0		42.1	300	<b>1.</b> (E	
		· 40	CF-	59 t0.20E 0		53.4	<b>302</b>	<b>1.</b> (E	
	(12.821-1	2331-1	CCD- 576	22 tO.07E		46.3	<b>5</b> CC	CA 1.2E 01	
	7	•		38 10. 18E 0		57.6	2C C	J. (E	
		ורח	578	25 to. 17E 0		2.4.5	302	1	
	• •	•	579	.07:0.07E 0		10.8	306	1.36-	
	•	ul	CF.	40 10.30		65.5	<b>3</b> 36	•	
	1.13.05	2	┐	.1840.09E 0		٠	<b>)</b> ) 5	2.76	
	L 14,P1-1	2328-1	-03	.24 tO . 20E 0		•	704		
	~	1		.8910.30E 0			<b>3</b> 0 <b>0</b>	7. (?-	
	, ,	1 (**	568	.20 #1.00 E-0		73.6	30 <b>C</b>		
	~	ধ	569	.9340.2		53.8	<b>50C</b>	5.(6-	
	<b>4</b> 7	5	,	.1740.			) ) )		
	1 14.95-1	2329-1	-00	-300.44.006-		5. 9.3	<b>3</b> 04	J. (E	
	~	,		.00 15.00 E-0		35.6	400	<b>.</b> (E	
	, ,	· m	513	.00.19.00		24.0	<b>30</b> 2	1. CE	
	4	<b>4</b>	574	.2110.276 0		2.55	) ) ?	). (E	
	· <b>v</b> :	· w	C.F.	3011.		. 8.	<b>30C</b>	 (E	
	1.15.217-1	2332-1	185 -000	.0042.		61.7	ეე5	7. (E-	
	~		585	1.4010.106-01		66.2	)) <b>?</b>	A 3.(E-	
			583	.60:1.80		39.4	<b>3</b> 00	J. €	
	- 4	4	584	.6015.		70.5	<b>30C</b>	CA 1.(E-C2	

TAI	TABLE E.3 (CONTINUED)	INDED)			1 1 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	•	
. A	AAC LOCATION CO	16 W COLLECTION 19.	1LW ANALYSIS NO.	FU-239,240 ACTIVITY IDPP)	URAN 1UH (r ICRO Grams)	Y 1ELC (R#RE WORK)	COUNT TIME	AHAL MON
i	• • • • • • • • • • • • • • • • • • • •	† † † †	: : ! ! ! ! ! ! ! ! ! ! ! ! ! ! ! ! ! !		7 7 1 8 7 7 8 8 8 8 8 8 8 8 8 8 8 8 8 8		! ! !	7 6 7 7 8 8 8 1 8
C AL		2332-5		5.00 48.00 6-02		\$6.5	<b>300</b>	1. (E CO
	1.17,99-1	2363-1	CCO- 676			61.5	30Z	÷
	7	7		0.30 11.30 6-01		51.7	100	2.
	· m	m	678	4.00 16.00 E-02		70.5	<b>500</b>	-
	-SF	4	619	1.50 40.90 6-01		188.6	<b>3</b> CC	-
	•	'n	CCF- 680	6.50#1.70E-01		41.5	<b>30</b> 2	-
	1.18.95-1	2359-i	CCJ- 666	1.20 #1.20 E-01		45.1	<b>5</b> 00	_:
	~	7		0.0010.116 00		41.7	<b>5</b> 0 <b>C</b>	
	· ~	m	899	1.1440.146 00		75.5	200	<u>.</u>
	•	4	699	1.9041.106-01		59.0	100	'n
	Š	S	CCF- 670	1.5041.006-01		5.5	) ) (	CA 1. (E CO
	L 18, P 21-1	2361-1	,	6.00 t9.00 E-02		63.4	<b>5</b> 00	<b>:</b>
	~	7		4.00 17.00 E-02		21.0	<b>500</b>	
	•	· M	673	5.30 11.20 6-01		43.4	400	÷
	*	4	674	1.40+11.00 €-01		29.6	<b>၁</b> ၁၄	1. (E CO
	•	Ņ	CCF- 675	0.0040.076 00		31.6	<b>506</b>	1.06 00
	. 68'517	4130	_	4.0010.406-01		78.4	306	1. CE CO
	L 20, P 5- 1	2354-1	949 -022	4.57 to. 30 £ 00		83°C	<b>5</b> 00	5. (E-01
	2	2		6.55 to.27E 00		85.7	<b>5</b> CC	8. (E-01
	1		619	4.7340.316 00		12.3	<b>300</b>	<b>:</b>
	<b>J</b>	4	649	7.00 #7.00 E-02		90°C	<b>30</b> 8	CA 2. (E-C2
	· 40	v		1.4040.706-01		68.1	20C	-
	L 20, P 13-1	2355-1	CCO- 651	7.6740.238 01		82.£	20C	4
	7	~		3.1040.90E-01		63.6	2 C C	CA 1. (E-01
		, m	653	1.7010.196 00		56.5	<b>3</b> 00	6. (E - Cl
	•	•	654	8.0018.00E-02		73.8	<b>30</b> 0	
	· 40	'n		1.4010.706-01		70.5	20C	3. CE-
	L 21.P1-1	2358-1	199 -023	1.40 to. 70 E-01		69.5	<b>30</b> 2	3. (E-
	2	~		1.20 +1.00 €-01		4.64	<b>5</b> 00	ċ

TAB		TINUED)					† † †	0 0 0 0 0 0 0 0 0 0
ARC	LOCATION	COLLECTION 16.	TLW ANALYSIS NG.	FU-239,240 ACTIVITY (OFP)	URAN IUM (P [CRO GRAMS)	Y IELC (R*RE YORK)	COUNT TIME	ANAL MON
1		: : :	; ; ; ; ; ; ;	* * * * * * * * * * * * * * * * * * *	T 1 1 1 1 1 1 1	† † † † †		) ( ) ( )
8 ∆(	L 21,P1-3	2358-3	CC0- 663	.00 46.00 E-0		59.C	<b>308</b>	5. CE-
	*	4		.00 +7 .00		67.1	<b>302</b>	; · (E-
	s.	Ś	CCF- 665	2.00 t5.00 E-02		52.3	200	1. CE
	L 21,P17-1	2357-1		.20 40.90		9.75	<b>50C</b>	• (E
	~	7	159	.60 40.90		65.6	<b>302</b>	٠
	3	m	658	.4011.30		5.54	100	1.06-
	*	•	659	.00 #5 .00			<b>3</b> 0 <b>2</b>	1. (E
	5	5	,	06.0404.		62.5	30 <b>£</b>	6. CE-
	L 22,P 13	4005	C 14-1263	.44 10.17		80.5	300	2. 4E
	L 23,P 1-1	2353-1		.00 +9.00		51.1	<b>307</b>	<b>i.</b>
	7	7	642	.00 +5 .00		68.6	<b>302</b>	1. Œ
	m	<b>m</b>	643	.10 40.60		11.5	30 <b>c</b>	1.6
	4	4	949	3.10#1.00E-01		5.33	<b>308</b>	t
	41	2	1	.0017.001		€0.€	) ) ?	<b>1.</b> (6
	L 25,P9	4005	7	.87 40. 191		19.3	300	2. SE
	L 26,P13-i	2316-1	168 -000	3.30 #1.10 E-01		65.1	<b>3</b> 02	CA 1. CE 00
	~	~	555	0.00 40.08 E 00		€2.₽	<b>32</b> 2	1.08
	<b>~</b>	m	553	6.00 16.00 E-02		63.3	300	1.(8-
	4	4		1.40 10.70 8-01			400	3, (8-
	r	vı	CCF- 555	2.00 to.60 E-01		£0.5	400	2. CE-
	1-21,617-1	2346-1		94 00.			<b>5</b> CC	1. (£
	7	~	617	Š		ċ	) ) 1	1. CE
	m	e	819	.00.14.00E		5.54	904	1. (E
	*	4		1.2011.206-01		£1.7	100	CA 1.CE 00
	8	ĸ		t0.80E-		61.3	) ) )	1.06-0
	L 28,P13	4017	126	w			)) <b>?</b>	1 . ¿E
	L 29,P1-1	1-0582		10.176		£7.6	<b>5</b> 00	1. CE C
	~	7	627	0+1.00 E-0			) ) (	CA 3. CE-02
	64	m	6.28	4.02 10.20 00		30.1	<b>4</b> CC	5. (E-C2

TAB	TABLE E.3 (CONTIN	(TINUED)				1	1	
ARC	ARC LOCATION	TLK COLLECTICN NO.	TEM ANALYSIS NO.	PU-239,240 ACTIVITY (DFP)	URANIUM (FICRO GRAMS)	Y 1ELC (R*RE WORK)	COUNT TIME	ANAL /HON
!	* * * * * * * * * * * * * * * * * * * *		* ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;	8 1 0 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	• • • • • • • • • • • • • • • • • • • •			1 1 1
BAL	129,P1-4	2350-4	629 -000	4.38 to. 37 E 00		72.B	<b>30</b> C	'n
	ın	'n	CCF- 630	.00 18.00		78.5	<b>30</b> 0	2. CE-
	1-60'627	2351-1		.77 to.22E		74.1	308	-3)
	2	2		.04 to . 18 E		52°C	<b>307</b>	2.56-01
	m	6	633	.4740.216		72.6	ეე <b>?</b>	<del>.</del>
	~	4	634	1.70 t0.70 E-01		62.8	<b>308</b>	<b>:</b>
	5	5		8.0016.00E-02		61.C	308	. (F
	1-210'627	2352-1	969 -000	4.00 #4.00 E-02		14.7	4CC	6. (E-
	?	2	637	3.00 #7.00 E-02		48.1	<b>406</b>	1.08
	m	<b>m</b>	638	3.00 14.00 E-02		3.61	<b>3</b> 24	1.06
	•	4	639	3.00 t3.00 E-02		4.53	134 4CC	<b>1.</b> (E
	~	ĸ	CCF- 640	0.0010.076 00		46.1	30Z	CA 0. CE 00
	L 30, P 21-1	2349-1	CCO- 621	1.80 #1.20 €-01		25.1	10	<b>1.</b> (6
	2		622	3.00 t6.00 E-02		16.4	100	1.(6
	•	~	623	2.5041.306-01		47.4	20C	. (E
	•	•	524	5.30 15.00 E-92		16.3	<b>302</b>	1.(6-
	•	S	CCF- 625	1.30 10.00 6-01		82.C	221	1. (6-
	131,49	4 ( 1 8	7	8.7940.715 00		£ 6 . 2	) ) (	1. CE-C2
OBAL	11,91	4C56	1575	9.68 t0.16  03	11.5	64.3	¥	€. (E - C1
			CCO-2186	7.49 10.28E 02	٠.	1.52	<b>Y</b>	2.4€ 00
	11,91-3		2118	2.0112.016-01	٠.	16.6	Ę	5. (E-02
	<b>.</b>	4	2119	5.57 to. 32£ 00	1.59	£1.2	325	1.4€ 60
	v	S	CCF-2120	1.25 11.25 6-01	ċ	82.E	ξÇ	1.26-01
	14,P5	4 C 2 3	7	3.1411.116-01		63.6	30Z	۳
	( 7, P1	4015	1572	1.4912.236-01	0.208	. B.	100	1. (E CL
	L 13,01	4001	1570	6.3710.37E 00	0.257	1. B.	300	w.
	L13,P17	4009	1251	-0.2911,136-01	0.193	28.6	100	٣
	1.19,917	4131	1576	4.2013.156-01	0.133	27.6	221	1.06 01
	122,05	4004	1568	4.0311.376-01	0.454	51.6	) ) (	1. (8 01

TABLE E.3 (CONTINUED)	NTINUED)		***************************************	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1		
ARC LOCATION	11 NO 11 ON	ILA ANALYSI S	FU-239,240 ACTIVITY	URAN IUM (PICAD	Y IELC (R=RE	COUNT TIME	ANAL /HON
	٠٥٠	NC.	Q 1	4	WORK 1	; ; ;	
88AL L 22.P 21	4006	14-1	. 22E 0	2.1	-:	200	<u> </u>
1 28	4020		.80 #1.19E-		1.53	20C	1. (E C1
E 1C-5-5		C SF -1 980	.40 #0.23E 0	158.	<b>:</b>		æ
43		1981	44 +0.04E 0	668.	4)	30	æ
~		1982	.0940.198 0	95.3			Æ
w		1 983	.67 to. 11E 0	74.1	٧ì		¥
v		1984	.05 #0.13E 0	143.			÷
10		1985	.86 tO.25E 0	14.3		40	Ę.
		1986	.75 t0.03E 0	16.1		3 C	£E-
14-5		2001	11 to. 13E 0	383.	~1		<b>.</b>
		2002	56 t0.04E 0	.761.	٩.	7 %	#
) <b>F</b> ~		2003	.7940,16E 0	2.20.	~	3 C	æ
• •		2005	240.05E 0	19.4	4	<b>)</b> 6	3,5
· Or		2005	.27 tO.03E U	.44	σ	4 (	<b>3</b>
01		2006	.7810.178 0	0.683	_	3.5	ų
- 1		2007	4	. 166	32.0		<u>3</u> E-
MG8 DM-C1-1	2277-1	115 -000	.4910.25E 0		¥	~	ξĒ
2		215	.98 to.16E 3		æ	Ü	ĬĘ.
m	æ	513	.40 10 .04E 0		1		3. 4E
4	4	514	.24 10.17E 0		0	C	. CE
so.	V.	CF-	.65 t0.32E 0		Ġ.	ũ	<b>g.</b> (F
1-23	1-6122	CC3- 521	.32 to .23E 0		¥	Ñ	<u></u>
2	7	525	.12 to .03E 0		3.0	3 C C	<u>.</u>
· M	~	523	.16 tO.15E 0		<b>;</b>		A
*	*	524	.0110.036 0		;	ū	- 3E
•	*	CCF- 325	.7410.05E 0		÷	J	• (E
C3-1	2274-1	-05	5 40.15E 0		66.1	)? ?	3. 36 60
~	7	164	.3010.26€		<u>.</u>		<b>.</b>
•	•	4.28	10.5210.016.01		£2.8	20C	1. CE CO

\* - NEW DATA THIS REPORT

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TAB	TABLE E.3 (CONTINI	N TINUED)	; ; ; ;	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					
A P. C.	ARC LOCATION	H 1	i	FU-239,240	HOL MA RU	Y 1EL C	COUNT	ANAL JHON	
		כסרו	ANAL YSI S NC.	ACT1VITY (DFP)	CRAMS )	MCRK)	<u> </u>		
1		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	T		*				
	DH-C3-4	2274-4		3.4041.10E-01		52.5	<b>3</b> CC	£	
1	•	<b>5</b>		8.0048.00E-02		4.75	<b>30</b> 2	•	
	7	2278-1	. 216 -000	4.51 #0.00E 02		75.6	76		
	~	ı		1.82 to. 12E Ol		36.8	<b>300</b>	4. te cc	
			518	3.31 to.08 E 02		11.3	100	5. IE 00	
	<b>1</b> 4	, 4	519	5.8040.70E-01		66.E	300	w	
	. 47	· •		4.77 t0.79 E 00		5.15	20,5	4. (E	
	, <u>,</u>	2275-1	CCD- 501	7.42 +0.22E 02		78.6	32	بې	
	``		502	2.26 t0.06E 02		80.1	100	3. (E CO	
	1 600	(F)	503	1.67 #0.05E 02		75.3	100	بھ	
	1	1 <b>4</b> 7	504	6.80 #1.10 E-01		01.6	<b>300</b>	CA 1. (E CO	
	* ***	. RU		1.30 to 60 E-01		1.5.1	325	1. CE	
	C 6- 1	2280-1	620- 526	8.71 to.25 E 02		82.1	<b>3</b> 2	ш	
	7			3.36 t0.12E 01		4.5	<b>4</b> CC	ш	
	, m	· m	528	8.70 to.30 E 01		6 2 ° C	<b>3</b> 00		
	-	•	529	7.74 to.41E 00		40.1	<b>)</b> ) <b>†</b>	6. (E-	
				2.70 t0.80 E-01		A. 6. C.	300	<b></b>	
	C7-1	2276-1	905 -000	7.00 10.22E 02		76.7	٦٢	3. 3E CO	
	~	2		4.4140.13E 02		74.2	<b>5</b> C	ш	
	· m	m	508	1.5240.05£ 02		é t • C	100	3. ¿E	
	•	*	209	5.50 t0.80 E-01		52.4	S S S	L	
	·	5		8.50 t0.90 E-01		66.6	400	♣, (Ē-	
	17-1	3211-1	CAD- 940	5.1440.136 03		12.C	~	2.1E 00	
	7	7		6.76 #0.21E 01		<b>90.4</b>	3 C C	1.4E CC	
	i (**	· ~	345	2.10 #0.09E 01		40.¢	) ) (	1.(E 00	
	•	· •	943	1.36 tG. 14E 00		65.3	300	•	
	_	~		1.0540.136 00		14.1	300	4. (E-C2	
	1.5	3212-1	CAD- 945	1.27 40.03E 03		15.6	36		
	7	~	946	5.20 to. 35E 00		64.0	3CC	5. fc-01	

ا لا لا	LOCATION	11 M COLLEC 11 CN NO.	TEN ANALYSIS NC•	FU-239,240 ACTIVITY (OFF)	UP AN TUM LY ICRO GRAMS)	Y IELC (R=RE WORK)	COUNT TIME	ANAL JHON
¥ 0	HO	3212-3	CAD-	3.03 #0.30E 00		¥	20C	3. (E
2	3	4	948	1.20 + 2.40 & -01		48.6	30	CA 1. (E 00
	• •	· •	646	1.70 #3.40 E-01		(1)	Š	2. (E-
	• ~	. ~	AFF	.40#1.		•	100	2. (E-
	DM-CEN	581		.14 #0.		4	30C	. 4E
	DP-12	2272-A	CD-2	.4340.04E	23.3*	G)	30	• EE
		•	~	8.7043.27E 01	0.243	~	10C	• 66
	•	4		5.08 to.16E 01	ċ	1,4	307	• <b>6</b> E
		'n	Ę,	8.20 *1.07 E-01		_	300	. Œ-
0 م	CHR-81A	9768	C VS-1469	7.29 to.58E 00		~	<b>30</b> 2	
	818		1470	.53 t0.04E		4	200	
	B 2A		1471	2.78 t0.22E 00		14	) ) ? (	
	8 28		1472	.0140.516		•	<b>30</b> 0	
	8 34		1473	.39 #1.02 E-		36.0	<b>60C</b>	
	B 38		1474	.49 t0.43E 0		16.2	1000	
	B 1A	9722	2104	2.10 t0.06£ 04		÷	)CC	
	918		2105	.11 #0.06 E O		÷	900	
	BZA		2106	.85 to.19E 0		•	321	6. CE-02
	8 28		2107	.29 to.05E 0		ď	<b>20C</b>	
	8 34		2108	.16 #0.03E 0		•	266	
	8 38		2109	.34 t0.03E 0		÷	3CC	
PCMR	PCMR 2-83-5	NONE	C SF-1973	.98 tO.49E 0	58.0	5	<b>302</b>	. 4E
,	• • •	<b>!</b>		.86 to. 40E 0	31.5	13	300	· CE-
			1975	.62 #0.08 E 0	O.	ü	30 <b>C</b>	•
	• •		1976	.82 to.06 E O			ÇÇ	2. EE OI
			1977	.33 t0.07E 0	•	÷	200	• EE
	10		1978	.57 t0.04E 0	0.954	<b>;</b>	<b>30</b> 0	. ž
	)					•		4

. NEW DATA THIS REPORT

TABLE E.4 RAI	DIOCHEMICAL	ANALYSIS OF	TABLE E.4 RADIOCHEMICAL ANALYSIS OF ROLLER COASTER PHYSICAL SAMPLES, CLEAN SLATE	ISICAL SAMPL	ES, CLEAN	SLATE	Ħ	:
ARC LOCATION	TLW 11 NO.	TLW ANALYSIS NO.	FU-239,240 ACTIVITY (DFM)	UR BN IUM (FICRO GRAMS)	Y IELC (R*RE HORK)	COUNT	ANAL /MON	
1	 	1		# *				Ι.
62-1 TA-C1-1	5259-1	CCD- 905	1.17 #0.03 € 06		86.1	32	- C	_
7	7	906	940.05E 0		68.3	3 C	2.5E 0	_
6	9	106	740.25E 0		62.C	1 C	6. SE 0	_
•	4	806	2.84 #0.06E 03		73.5	26	1.26 01	_
in	5	CCF- 909	940.09E 0		\$6.1R	200	4. 3E Q	_
G2-2 TA-C2-1	5258-1	006 -000	0 #0.04E 0	742.	76.4	<b>3</b> 2		_
~	7	106	4940.04E 0	955.	76.4	26	1. EE 01	_
<b>m</b>	•	306	1.75 to.04E 05	55.2	£1.C	<b>5</b> C	1.26 01	
*	4		0	•	66.P	75	B. CE CO	0
•	<b>5</b>	CCF- 904	5.97 tO.18E 01	0.292	64.5	300		0
62-4 TB-C2-1	5256-1		2.25 t0.05E 05		74.5	20	2.28 01	_
~	~	968			62.7	2	1.56 01	
•	m	897			78.3	2	1.76 01	_
•	4	868	5.90 t0.17 € 02		13.6	100	1.26 01	_
'n	<b>V</b> 1	CCF- 899			73.4	<b>1</b> 00	8.36-0	_
C2 BC-C3	5112	C TD-1294	1.49 t0.07E 01		79.0	100	3. 26-0	_
	4597-1	CAD-1041	4.30 to. 13E 02		14.0	721	9.36 00	0
7	7	1042	0		₹0.€	<b>3</b> 00		0
9	E	1043	2.87 +0.08 E 02		8C . 2	300	7.26 01	_
•	4	1044	4 40.08 E 0		£0.3	308	1	_
•	<b>.</b>	1045	0		66.9	2	2. tE	0
<b>L</b>	_	CAF-1046	o		93.2	<b>302</b>	CA 7. (E-0)	د
C 3	5113	C TD-1295	1.06 #0.10 E 00		67.0	300	2.16-01	_
. 63	5115	1296	6.3640.23E 01		19.1	<u>ې</u>	1. CE 00	0
10-1	4 5 92-1	CAD-1011	5.00 t5.00 E-02		£0.8	300		0
~	7	1012	2.9110.236 00		76.7	<b>30</b> C	<b>2.</b> EE	0
•	m	1013	2.1010.80 6-01		91.6	<b>308</b>	CA 2. 16-01	_
~	•	1014	9.37 10.34 5 00		26.4	1366	9. (6	0
•9	•	1015	5.00 46.00 E-02		¢1.1	300	1. (E C	0

TAB	TABLE E.4 (CONTINUED)	NTINUED)			1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				į	i
ARC	LOCATION	TLW COLLECTION NO.	TEM TEM ANALYSIS NO.	FU-239,240 ACTIVITY (0P#)	URAN JUM (P. ZCRO GRAMS 1	Y JELC (R.RE MORK)	COUNT	ANAL	NAL /MON	;
	• • • • • • • • • • • • • • • • • • •	i	) . 			•	ć		1	•
79	8C-1C-7	ç	AF-1	.30 #1.20 E-0		41.5	777	<b>5</b>	اِوا	<b>,</b>
; )	-	-	10-1	.7940.25E 0		73.5	<b>3</b> 6	<b>:</b>	Ħ	0
	7	8165-A	COS-1399	.45 ±0.07 E 0		48.6	<b>3</b> C	•	ä	0
	2 -	` `	· -	.51 ±0.07 E 0				m	2E-	_
	74.0		2004	7940.07F 0		19.9	20		3E	0
			1401	6.6540.216 03			2	•	(E-01	
			1402	47 t0 . 09 E 0		64.0	2 C	*	3	0
	7 7		1 403	.42 ±0.05 E 0		•	3 C	'n	3,	0
	5 -		1604	.06 #0.04 E O		82.3	36	ě	y	0
	2 1 7 1 B	6114	10-1	11 40.08E 0		5	2 C	•	1E	0
	10-10	4694	CAN-1035	40 #11 00 E-0		72.5	20C	۶.	3	_
	,	•	• -	7041		•	200	•	1E-	
	÷ F		1037	.75 to .27E 0		•	30c	ë.	<b></b>	,
	1 4	4	1038	.24 to . 14E 0		77.2	30 <b>2</b>	₹ 1.	36	00
	· •	· •	1039	.56 to. 30E 0		74.6	20C	÷	<b>1</b> E	0
	, ~	, ~	AF-1	.80 +0.16E 0		R1 .	30Z	<b>-</b> :	ξE	-
	r 1	4581-1	1	.7240.10E 0		64.0	36	٠	æ	0
	• ~	;	}	.62 t3.04E 0		11.9	100	2.	æ	0
	) (~	, en	832	.1240.07E 0		17.1	100	3.	#	0
	•	•	833	.2940.076 0		47.2	2C C		žĒ	0
		· vr	CF	.0640.28E 0		34.4	20 C	CA 2.	æ	CI
	( 3-1	4 572-1		.61 tO.07 E O		63.3	3¢		£	_
	. ~		;	.38 tO.03E 0			3 C	•		~
	•	•	827	.26 +0.17E 0		19.5	<b>)</b>		Æ	_
	• •	•	828	.63 to.04E 0		0.8×	3C		Ħ	0
	•	- 127	- 40	.25 to .22 E O		48.2	20C	3.	Ą.	0
	(3-1)	1-565 7	CAD-1029	.85 to . 16 E O	369.		<b>3</b> C	÷	<b>.</b>	0
	. ~	•	_	.56 to .24E 0	4.83	64.2	<b>3</b> C	8	w	0
	, m	· m	1031	.5140.09	0.268	4.2	7	-	9	0

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TABLE E.4 (CONTINUED)
A
CAD-1032
1033
AF-1
_
CCD- 840
84.1
842
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1020
1021.
CAF-1022
AD-1
1024
1025
1026
-
C 10-1301
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05-140

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TABLE	

ANAL JHON	2.1.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2
COUNT TIME	
Y IELC (R=RE MORK)	
URAN 1UM ( MICRO GRAMS 1	2.54* 2.54* 0.0250 0.517 0.140 1.36 4.27*
PU-239.240 ACTIVITY {DFP}	5.08 +0.11E 07 6.04 +0.11E 07 6.04 +0.11E 06 5.69 +0.10E 06 3.59 +0.08E 05 3.59 +0.08E 03 3.70 +0.09E 03 1.89 +0.09E 03 1.89 +0.09E 03 2.17 +0.19E 01 4.49 +0.19E 01 6.51 +0.19E 01 6.51 +0.16E 02 1.33 +0.09E 02 1.33 +0.09E 02 1.34 +0.09E 02 1.36 +0.09E 02 1.36 +0.09E 02 1.36 +0.09E 03 1.36 +0.09E 03 1.37 +0.09E 03 1.38 +0.09E 03
TLW ANALYSI S NG.	CDS-1406 1407 1409 1358 1358 1358 CCD- 821 822 CCF- 824 CAD-1009 CAD-2201 CDS-1360 CAD-2201 CCF- 835 CCF- 835 C
COLLECTION NO.	8153 4569-1 4569-1 4583-1 4581-3 4581-3 4587-3 4587-3 8153 8153
LOCATION	# # # 222222
ARC	75

\*New data this report

TAB	TABLE E.4 (CONTINUED)	NTINUED)	3 3 8 8	* * * * * * * * * * * * * * * * * * *	1 1 1 1 1 1			
ARC	ARC LOCATION	COLLECTION NO.	TLN ANALYSI S NO.	FU-239,240 ACTIVITY [DFF]	UP BN IU4 (* 1CRO GRAHS)	Y IELC (R=RE WORK)	COUNT TIME	ANAL MON
				7240.19F G		4,000	20	
79	מש - עם		5051-503	0 101 000		4.08		
	7-83	٠, ر	LCU- 845	1.1640.28F 02		72.4		4. CF 00
	<b>V</b> (1	4 (*	6.40	.27 to.15E D		62.6	72	
	1 4	. 4	848	46 40 . 20 E O		74.2	20	
	· •	. <b>r</b> u		.24 #0.03E 0		78.2	<b>3</b> CC	
	C.8-1	4586-1	CAD- 999	.6940.09E 0		71.6	)) <b>?</b>	
	7		_	.64 #0.14 @ 0		9·69	100	
		· M	1001	.32 t0.23£ 0		711.2	301	
	· •	•	1002	.5040.05E 0		B7.8	100	
	••	w	1003	.05 to.03E 0		71.6	3CC	
	~		CAF-1004	.49 t0.05E 0		27.7	<b>33</b> 5	
	6)		C TO-1302	.12 #0.05E 0		75.3	<b>&gt;</b>	
	5)		CDS-1364	.69 #0.16E 0		9.89	<b>5</b> C	
	20	4-813-A	CCD-2199	.00 #0.03E 0	4.45*	55. F. S. B.	š	
	10		CDS-1365	.96 to.13E 0		11.1	4	
	10-3	~	CC0-2139	.13 #0.07E 0	1.73	81.3	20C	
	4		2140	.86 #0.08 E O	80.9	76.7	) ) 	
	5	S	CCF-2141	.7340.07E 0	0.637	£1.1	300	
	11		C TD-1299	.34 #0.03E 0		83.6	<b>3</b> C	
			CDS-1366	.95 t0.06E 0		40.6	3C	
	12-1		CCD- 815	.08 to. 10 E 0		11.5	30	
	7	~	816	.1540.025 0		67.57	36	
	i mi	m	817	.40 40.02E 0		£ 6 . P	<b>¥</b>	
	*		918	0		76.7	40	
	**		CCF- 819	.6840.226 0		35.6	20C	
	13		C T0-1300	.2540.07E D		76.5	<b>)</b> 2	
	13	8153	CD 5-1368	0 40.01E 0		4.80	35	
	7.		1369	.75 #0.17 E O		0.08	<u>ح</u>	

\*New data this report

	•	! ! !	8	ဝ	10	10	00	င္ပ	00	00	00	00	ဝ္ပ	00	00	ပ္ပ	ဝ္ပ	00	00	ರ	00	00	0	៰	9	00	Ş	00	00	Q
ANAL JHON			• CE	•	•	•	•	•	•	•							•		•		•				•			٠	٠	•
ANA			Ľ	e i	~	<u>ن</u> ــ	~	7	4	ď	ĸ	'n	m	m	m	m	m	พ้	-	j	×٠	۳.	<del>z</del> i	نہ	ø	Ň	~	Ä	ż	2
COUNT	11ME	† • • • • • •				ي م								3C	30	20	3 C	3C	3 C	<b>2</b> CC	3C	3C	<b>5</b> C	2 C	2 C	3C	<b>3</b> C	3 C	2 C	2C
Y IEI C	FARE WORK )		27.8	27.5	66.3	72.7	65.E	19.5	43.1	75.0	11.1	£9.8	54.4	75.C	•	60.c	82.0	71.4	72.4	4.69	8. 5.	17.9	49.5	68.1	63.6	10.8	7.75	\$1.5	66.3	
ALL VA GL	CP ICRO	  -  -							٠,	•			•																	
			70	07	10	20	90	90	0.5	. 90	90	60	0 5	0 5	90	90	90	90	<b>†</b> 0	01	90	1.0	07	01		0 5				
07	<u> </u>		m		98	.04E	. 16E	4	[]	2	17	9	12	.08€	S	4	2	0	360.	5	08	9	60	•03E	91	~	3	2	0	
2,940	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		*	6 ‡0	9	68 40.	0 ‡ 0	5 +0	7 +0	2 10	5 ‡0	6 +0	4 +0	940	5 40	9 t0	0	3	3 10	1 10	S	•	5 to	0	9 10	1 +0	2 40	1 0	0 0	04.40
)-11d	ACT [V]		•	•	•	2.6	•	•	•	•	•	•	•	•	•	•		•		•	٠	•		•	•		•		•	
	5 ]		0	~		*	Ń	۰	~	<b>م</b> ې	Ġ.	0	_	2		<b>.</b>	S	9	-	٠		2	<u>س</u>	•	s	9	_	•	<b>~</b>	<b>C</b>
3	ANALYSE NO.		05-137	_	134	1344	134	134	134	134	134	135	135	135	135	135	135	135	135	177	137	137	137	137	137	137	137	137	137	4
•		1	ວ																											
(1) 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	EC TI ON	!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!	1153																	15	1153									
CONTINUED)	כסרו		30																	₩.	∞									
	5		•	0		~	m	4	•	•	<b>G</b> C)	σ.	o	_	~	<b>~1</b>	4	'n	9	a)		5.1	5.2	5.3		7.0-1			7.2	
	LUCA   104	1	BH-1	80- C		Ü	U	U	U	U	· u	. <b>.</b>		-		-	7	<b>-</b>		J	CH-C	ပ	U	U	U	U	Ç	U		
700	ر «		<b>62</b>																											

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TABI	TABLE E.4 (CO)	(CONTINUED)		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	 	1 1 1 1 1	1 1 1	į
ARC	LOCATION	בסררבכ נוסא אריע	TLW ANALYSIS	£U-239,240 ACTIVITY	URBNIUM (MICRO	Y JELC IR*RE MORK J	COU41 118	ANAL MON	<del>z</del>
!		• • • • • • • • • • • • • • • • • • • •	•00				1		
62	CM- CS. C	8153	œ	.8840.16E 0		•	20	3. EE	60
!			138	.16 40.03E 0		÷	4 (		
	6.63		æ	.35 #0.03E 0		ij	2 C		ပ္ပ
	69.4		8	.81 #0.09 E O		65.C	<b>&gt;</b>		ဌ
	11.0		ຕ	.0240.12E 0		ö	3 C		00
			ထ	.20 #0.13E D		φ.	3 C		၀၀
	-		8	.2140.04E 0			¥	•	ဌ
	1		Ø	.6240.03E 0		•	<b>3</b> C	•	00
	-		σ	.81 40.14E 0		"	<b>5</b> C		8
	13.0		1391	6.0040.17E 03		5. 22	<b>3</b> 2	1.35-	2
	(11	8152	3	.0710.06E 0		æ	ĭ		ŝ
	CS		31	.39 #0.06E C		41	<b>3</b> C		
	S		31	.98 to.13E 0		E	2 C	3. 16	ဝ္ပ
			31	.00 40.07E 0		<b>W</b> 1	3 C	2. TE	00
			32	3 4C * 01 9 4 .			<b>5</b> C	2. 2E	ပ္ပ
	67.0		~	.96 fü. 13E 0	•	16.1	3 C	1. EE	င္ပ
			32	.31 40.08 E O		¥	<b>3</b> C	1.7	00
	C1.2		32	0 360.01039.		•	32	1. EE	္ပ
	67.3		32	.99 #0.05E D		61	3.0	<b>2.</b> €€	0
	C7.4		32	.12 #0.06 E O		Ġ.	<b>5</b> C	2. 5E	8
	C9.C-A	₩.	32	.33 40.02E G		41	2 C	1.8	9
	6-0.53	60	32	.2340.02E D		4	<b>5</b> C	1.75	00
	6		32	.4340.03E 0		<b>w</b> 1	26	1. EE	Ç
	6		32	.60 10.04E 0		~	40	<b>1.</b> %	00
			4	.15 to.02E 9		S	<b>3</b> C	<b>7.</b> €€	S
	65.4		33	.90 to.05E 0		0	36	1. 1E	ဝ္ပ
	11.0		4	.74 #0.04E 0		~	<b>V</b>	3. CE	0
			33	.4240.02E 0		0	Ų	•	0
	11.2		33	.16 40.26 9		40.5	<b>5</b> C	<b>5.</b> 36	00

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	ANAL /HON	2.7.1.3.4.1.1.7.4.2.1.1.2.1.1.2.1.1.2.1.2.1.2.1.2.1.2.1
1 1 1 1 1	COUNT	
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	Y IELC (R=RE WORK)	
	URAN IUM (FICRO GRAMS)	221.
	FU-239,240 ACT1VITY (OFP)	9.6110.20E 04 2.37 t0.15E 04 2.94 t0.06E 04 2.17 t0.06E 04 7.72 t0.21E 03 2.50 t0.08E 03 3.09 t0.07E 02 1.82 t0.08E 05 1.82 t0.08E 05 1.25 t0.08E 05 1.25 t0.08E 05 1.25 t0.08E 05 1.25 t0.08E 03 1.14 t0.08E 03 1.14 t0.08E 03 1.14 t0.08E 03 1.14 t0.08E 03 1.14 t0.08E 03 1.14 t0.08E 03 1.15 t0.08E 03 1.16 t0.08E 03 1.17 t0.08E 03 1.18 t0.08E 03 1.18 t0.08E 03 1.19 t0.08E 03 1.19 t0.08E 03 1.11 t0.08E 03
	TEM ANALYSI S NO.	CDS-1335 1336 1337 1339 1339 1340 CAD-1286 CAD-951 CCD-770 CCF-771 771 772 CAD-956 CAD-956 CAD-956 CAD-956 CAD-957 771 772 771 772 773 774 773 774 774 777 777 777 777 777
(CONTINUED)	1LW COLLEC TI ON NO.	8152 8152 3258-1 4557-1 5163 3259-1 4562-1
TABLE E.4 (CON	LOCATION	CC
TABL	ARC	79 ₹

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TABI	TABLE E.4 (CONTINUED)	NTINUED)			9 9 9 9 9		1 1 1	
A A C	LOCATION	1LW 1LW NO.	AXA	6U-239,240 ACTIVITY (DFP)	URANIUM IVICRO GRAMS 1	Y IELC (R*KE WORK)	COUNT TIME	ANAL /MON
	• • • • • • • • • • • • • • • • • • • •		1 1 1 1 1 1 1 1		; ; ; ; ; ;			
•	018-4	4562-4	-03	.00 4C.01E 0		66.3	7	•
	· w7	  -	CF-	.7940.14E 0			2 C C	• 1E
	0.30-1	4563-1	008 -000	.50 #0.06E 0		5.61	3 C	ĥ
	7			.86 #0.08 E O		71.6	26	• ĉE
	, (**	•	802	.4310.02E 0		27.3	1300	끧
	· •	· ~	803	.86 to.11E 0		65.2	20C	u.
		· 45	CF-			34.5	20C	35
	030	4-516-A	CAD-2200	.07 t0.03E 0		54.8	3C	
	0 4 0		~	.60 t0.26E 0	010.	72.5	30 <b>2</b>	
	1 <b>4</b> 7	4	2147	.5940.11E 0	0.0155	52.1	300	Ĥ
	پ -	. 43	2148	.67 t0.08 E 0	.011	14.8	<b>3</b> 00	. (E
		~	AF-2	.5340.06E 0	9	78.2	20C	1
	042-1	3300-1	CAD- 963	.66 t0.04E 0		66.5	3 C	
	~	1		1.9240.05E 02		66.7	9,0	1.56 00
	, (41	· (**)	596	.48 tú.24E 0		69.6	30C	
	· •\$	4	996	.12 to.17E 0		15.2	20C	
	· •0	• •0	196	. 10.13E 0		61.2	40C	
	-	~	AF-	.25 to.08 E O		57.1	404	بق
	042-1	4561-1	062 -033	.02 t0.17E 0			3 C	¥
	7			.12 tO.02E 0		71.2	30	. IE
	· «1	•	192	.0740.035 0			۶۲	. 3E
	<b>4</b>	•	193	.85 tO.09E 0			20	Fi
				.59 to.11E 0		15.0	400	t
	640	161	10-1	. 59 to. 100 0		75.0	3	• 2E
	0.4-1	4515-1	٠,	0 360.0111.			30 <b>C</b>	
		·		.15 10.03E 0			<b>366</b>	
	, ~		971	.4110.22E 0		71.3	30C	36
	<b>.</b>	-\$	972	.37 10. 18 0		52.7	<b>5</b> 0 <b>C</b>	
	0	ري. ۱	973	.6910.16E 0		17.5	1001	
	,							

TABL	TABLE E.4 (CONTINU	NTDNUED)	# # # # # # # # # # # # # # # # # # #					
ARC	ARC LOCATION	COLLEC TI ON	TLW ANALYSIS	FU-239,240 ACTIVITY	URANIUM	Y IELE [R*RE	COUNT	ANAL /HON
;		NO.		(DPF)	GRAMS )	WORK 1	1	
-	2-550	53	CAF- 974	.22		17.2	1000	
:	054-1	4558-1	CCD- 715	.48 +0.06		71.5	20	, ii
	~			.41 #0.24E		66.8	3 C	٠ <u>.</u>
	m	m	111	4.82 to.12E 02		<u>-</u>	) ) )	
	4	4	778	.81 #0.20E		34.0	<b>5</b> 00	E
	'n	<b>v</b> n		.39 to.06E		*	20C	۳.
	0 6 0	5164	C TD-1289	.04 40.03€		•	<b>5</b> C	. 3E
	046-1	4559-1		.53 t0.04E			) <b>?</b>	W
	~		781	.95 40.08 E		~	10C	. ff
	m	m	782	.87 tO. 26E			<b>50</b> 6	
	*	4	783	10.10E		£ . 5	<b>302</b>	#
	v	S	CF-	•0•			3CC	• (E-
	1-990	4578-1	CA0- 987	.39 tO. 15E			<b>5</b> C	• <del>(</del> E
	~	~	988	7.76 to.27E 01		67.5	<u>5</u>	1.4€ 00
	m	m	686	341.0116.			3CC	. (f.
	4	4	990	.7010.16		•	400	E
	¥	49	166	.71 tO. 10E	•	0	20C	• <u>i</u> E-
	<b>~</b>	~		.38 to . 09 E			400	, îŧ
	012	5167	C 10-1291	. 93 to. 14E			<b>5</b>	#
	018-1	4565-1		.54 10.03E		•		ب <u>ن</u>
	~	~	806	.30 to.07£		•	ŝ	Ψ.
	m	m	807	.36 tO.26E		•	Ç	٠. بو
	4	4	808	.34 10.02E			300	Æ
	'n	'n		3 90 · 01 96 ·			J	¥.
	078-1	1-6254	CAD- 993	.26 t0.04E		•	3 2	• <u>I</u> E
	~	~	466	2 40.			100	
	m	•	566	380.0141.		57.6	1000	
	<b>.</b>	•	966	.2510.04E			1001	
	¥	ų	166	.3510.16€			704	

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ARC.	ARC LOCATION	ורא כסררבכ 11 מא NO•	ANALYSES NO.	FU-239,240 ACTIVITY (DFP)	UPARIUM IVICRO GRANSI	Y IELE IRAR WORK I	COURT 11HE	Anal Mon
•	0.18-7	4		1.7940.066.01		3.8.8	466	1.16
r	. 790	157	7	46640.058		78.0		2. !E
	050-1	4566-1		.05E		41.7	35	3. 7E
	. ~			.38 10.02E		71.3		1. CE
	. (4)	· M	815	.13 to.08 E		16.2	<b>.</b>	3. 4
	•	•	613	.75 #0.05E		14.0	<u>ي</u>	2.16
	· vn	•		.98 tO.09 E		64.0	- <del>-</del>	1 · 1E
	050-1	4517-1	CAD- 981	.47 40.12E		۲۰۰۲	رب د	4. 28
	~	. •	Q.	1.78 40.05 € 02		67.6	<u>ب</u>	<u>.</u> .
	m	•	983	. 04E		70.6	<b>4</b>	٠
	• <b>•</b> \$	•	486	. 22 E		12.1	<b>5</b> 0 <b>0</b>	
	•	· •	985	.01 10.48 E		¿c.s	30 <b>2</b>	•
	· <b>-</b> -	_	ο.	.07 #0.09 E		24.7	1001	¥:
	. 4		-12	.07 #0.09E		60.3	30	٠
	2.2	4-664-1	-21	.58 to. 19E	3.64	31.6	9	1.38
	162-3		212	.08 10. 128	1.42	12.1	100	
	· •	•	7	.03 (0.03 €	€.	11.6	166	2. E
	· <b>•</b> 7	· •••	ŗ	.03 40.07 8	5.69	3.55	ည် <b>က</b>	1. 4E
	108	\$162-A	~	.37 to.33E	2,76*	24.0	) <del>+</del>	
	1 14- 1	4560-1	,	.35 to.05E		40.4	<b>3</b> C	
			_	.92 10.08 E		71.7	<b>,</b>	4. (E CC
	1 74		187	.18 10.04 E			) ) (	¥
	•	•	788	.1340.20E			306	. ff
	•	· <b>w</b>	CF- 1	.64 to.07 E			308	¥
	1 1 1 4 - 1	1-9157	CAD- 975	.3740.216			36	, tf
			926	.16 40.03 €		55.4	100	¥
	, ~	, ~	977	.9740.23E		10.7	30°C	
	<b>¬</b> ≺	•	920	4940.216		5.91	100	. ĭf-
				֡				

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TAB	TABLE E.4 (CC	(CONTINUED)					1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
ARC	LOCATION	COLLE	TEN JWALYSIS NC.	FU-239,240 ACTIVITY (GFF)	URBNIUM (P. ICRO GRAMS)	Y JELC (R*RE WORK)	COUNT TIME	ANAL MON	4
				0 3 70 0 7 10		4 0 4	J 0 7	77	
<	- I C I -	` .	1 20	0 307.00.00			) C	֓֞֝֝֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֡֝֡֓֓֓֡֓֓֡֓֓֓֡֝֡֓֡֓֡֓֡֝֡֓֡֓֡֝֡֓֡֡֡֝֡֡֡֡֓֡֡֝֡֡֡֝֡֡֡֡֡֡	
Œ	125	3267-1	C 10-1690			70.7	7 7	4. 7E CO	
3		139	-12	.86 ±0.03 € 0			100	<u>.</u>	
	068	2	-139	.0240.146 0		46.4	3 C	36	
	C 5 2	<u> </u>	-171	.61 #0.06E 0			<b>30</b> 0	. 4E	
	C 5 2	2	-139	.22 #0.07E 0			3.5	H	
	650		_	.0040.05E 0			36	• 2E	
	100		1395	.03 40.02 60		Š	) ć	<u> </u>	
	10.45 10.45		1396	.03 40.20 E 0		۲.	<b>5</b> C	. 3E	
	108		1397	.15 to .08 E O		26.2	<b>)</b> }	• (£-	
0	012	27	C SA-1478	0 361.018E 0		യ	20C	1.36	
	028-1	5279-1		.50 10.116		10.e	<b>30C</b>	2. 1E	
	~		1479	.44 40.20E 0		4	<b>300</b>	. 4E	
	m	•	1480	.6710,318 0		80.5	<b>308</b>	5. 16	
	4	•	1481	.62 40. 20 E		78.3	30 <b>C</b>	2. EE	
	'n	'n	1482	.76 #1		82.1	30 <b>C</b>	7.88-	
	•	•0	1483	.98 11. 10 6-0		_	308	8.	
	~	1	1584	.02 t0.99 E-0		÷	332		
	•	æ	1485	.43 t0.87E		;	30 <b>2</b>		
	<b>o</b>	5	1486	.44 *1.12E		•	<b>3</b> 0 <b>2</b>	- 46-	
	10	01	1487	.83 40.37E 0			30 <b>C</b>	7. EE	
	11	7.4	2155	.23 #0.07E 0		Ç	<b>30</b> 0	5. (E	
	30	861	05-1	.08 to.07 E O		36.0	300	. CE	
	034-1	4554-1	CC0-1688	.45 tO.04 E O		22.1	<b>4</b>	. JE	
	~		1689	.6040.62E 0		Φ	<u>ک</u>	. E.	
	6	6	1690	.9710.2			3C C	10-37-1	
	4	•	1691	.2240.02E 0		*	¥	£.	
	Ś	S	CCF-1692	.26 11.13 6-0		¥	<b>300</b>	6. (E-Ck	

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(CONTINUED)
TABLE E.4

ANAL /MON	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2
COUNT TIME	
Y IELC (R=RE WORK)	
UR AN IUM (P. ICRO GRAMS)	
FU-239,240 ACTIVITY (0FP)	0.90#1.81E-01 1.01#0.03E 03 6.87#0.36E 00 8.80#0.76E-02 8.80#0.17E 00 0.00#0.17E 00 8.55#5.70E-02 2.25#0.17E 00 8.15#0.41E 00 8.56#0.43E 00 1.33#0.04E 01 4.79#0.12E 02 1.51!0.05E 03 1.21!0.05E 03 1.21#0.05E 03 1.21#0.05E 03 1.21#0.05E 03 1.38#0.05E 03 1.46#0.21E 00 8.99#1.82E-01 8.99#1.82E-01 8.99#1.82E-01
TEM TEM NO.	C SA-2156 C TA-1717 C SA-1488 1490 1491 1492 1493 1494 1495 1495 1796 1781 1781 1781 1781 1781 1781 1781 178
16 H COLLEC 71 ON NO.	4151-11 5277-1 5277-1 2 3 4 4 8163 8163 8163 8163
LOCATION	625 625 626 626 626 627 627 627 627 627
ARC	C 4.

2. (E-01 1. 2E-02 2. (E-03 2. (E-03 2. (E-03 2. (E-03 1. (E-03 1. (E-03 1. (E-03 2.66 00 2.16 00 1.96 00 4.66-01 7.66-01 220 3. (6-01 1. CE-COUNT TIME Y TEL E I R R R E WORK ) URANIUM 17 ICRO GRANSI 2.0640.06E 5.1340.20E 1.2940.02E 5.3140.15E 1.9540.06E 2.3340.07E 3.0840.10E 4.6040.17E 8.22 to.29E 7.36 t0.20E 1.06 t0.02E 1.78 t0.05E 1.15 #0.02E 2.09 #0.03E 3.52 #0.12E 1.96 #0.05E 9.65 #0.26E 4.9910.16E 4.5510.08E 5.4410.11E 1.91 #0.06 E 1.07 #0.02E 4.84 ±0.14E 3.87 #0.12E 1.9140.06 2.66 tO. 13E FU-239,240 ACTIVITY (DFF) TLH ANALYSIS NO. CD S-1789 TLW COLLECTION 4 693-1 TABLE E.4 (CONTINUED) 8163 ARC LOCATION 066-2 1-9:0 052-1 0.00-1 1-510 0 58-1 4

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TABI	TABLE E.4 (CONTINUE)	NTINUED)				1	1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
ARC	LOCATION	1LW	ANALYSIS	FU-239,240 ACTIVITY	URANIUM	Y JELC (R*RE	COU'17 TIME	ANAL MON
	* * * * * * * * * * * * * * * * * * *	: . !	CN	(0 6 4)	GRAMS )	ORX		
u	052-4	8163	CD S-1813	.84 #0.21E C		22.1	306	6. EE C2
•	<b>4</b>	,	1814	6140.		3.55	100	
	162	5	C 1A-1710	.77 #0.12E 0		57.6	100	3. fE-
	104-1	8163	CDS-1815	.75 40.09E 0		9.99	100	. EE 0
	2	)	1816	.3640.18E 0		10.1	<b>3</b> CC	
	ı m		1817	.27 #0.26E 0		12.2	<b>3</b> CC	
	•		1818	.82 #0.23E 0		_	<b>3</b> CC	
	· <b>6</b> 4		1819	0 390.0169.		16.4	2	
c	050	·	1778	.70 to.17E 0		4.4	) 3 8	3. 7E 01
. 1	0.50	5036	IA-	.05 #0.08 E O		•	γ	2.4E CO
•	0 20-1	38-	CC0-1668	.48 tO.04E 0		~	<b>3</b> 0 <b>2</b>	4. (E CO
	. ~	١	1	26 #0.04E 0		•	30C	#: ·
	ı m	m	1670	.6940.08E 0			<b>3</b> CC	3:
	•	•	1671	.31 to.08 E 0		69.6	<b>3</b> CC	. 2E
	. 41	· #1	CF-1	.00 #0. 29 E O		•	ž	1. CE- C2
7	0.66-1	4514-1	CC0-1683	1040		•	<b>3</b> 00	. Æ
,			-	.06 #0.15E 0		0	<b>3</b> 00	#:
	1 (41)	m	1685	.81 to.09E 0			300	1.36 60
	•	•	~	.77 #0.04E 0		94.6	200	• 2E
	•	•	CCF-1687	.9940.08E 0		57.5	<b>30</b> 0	¥.
	050		7	.81 +0.16E 0		4.55	231	A.
¥	0.16			.9440.04E 0		42.1	) ) 	3.66
:	030		1713	.01 #0.14E 0		ę.	<b>3</b> 00	
,	0 6		COS-1779	.11 to.09 E 0		ċ	100	2. CE-02
	0 60		C TA-1312	82 40.06 E O		~	400	쁘
	9 9 0		1314	.57 to. 18E 0		æ	<u>ي</u>	- 36 -
	9.0	5115	C 10-1283	.86 +0		19.5	30	•
	0.66-1	1	092 -000	11 to . 38 E O		34.4	704	. Œ
	~		192	.00 tu.16E		29.6	704	1. 2E 00

TABL	TABLE E.4 (CON	(CONTINUED)					1		į
ARC	LOCATION	11 P COLLEC 11 ON	TEH ANALYSIS	FU-239,240 ACTIVITY	CPICAD	Y IELE (R*RE	COUNT TIME	ANAL MON	
		NO.	NO.	(440)	/ I	έ i			į
_	6-30	£ -7757	CCD- 762	.72 #0.11		44.6	40C	. 1E	0
,	**	4		.15 #0.05E 0		24.6	400	. CE-	5
	r er	· ufi	ı	.64 #0.35E		28.€	4CC	Ē.	5
	050		C 14-1714	34 10.33E		40-6	<b>3</b> 0 <b>c</b>	8-46-	_
	058-1	4	•	-80 +0.50 E-		15.2	<b>7</b> 00	E	8
	. ~	•		.61 #0.32E		9-19	<b>3</b> 00	6. CE-01	_
	l eu	m	767			45.6	<b>3</b> CC	. CE-	ត
	•	₹	768	.58 t0.18E		78.1	<b>3</b> 00	E	~
	· <b>4</b> 17	· •	CCF- 769	.43 #0.22E		74.3	) 01	3. CE-	~
	100	117	-12	0.00 #0.226 00		64.8	40	E	0
	102	5112	C 1A-1313	.44 t0.20E			<b>30</b> 6	3. (E-	50
	114	5107		.9043.50E-		ċ	¥		8
884	L 2, P 13	5370	CB S-2037	.16 #0.03E		78.5	<b>4</b>	7. 2E - 01	_
; !	L 3, P 2		7	.14 #0.02E		;	<b>3</b> C		
	1 3, P.1		2039	.28 +0.02E		68.2	4	ų.	ဝ
E1C-128	128-5	NCNE	C SF - 1987	.0240.03E	94.1	28.1	S	H	C
)	4	•	1988	.81 #0.05E	2	37.2		æ	0
	~		1989	.2240.03	24.9		36		8
	•		1990	.9740.	4.5	•	7	Į,	0
	Ç		1661	.77 #0. LdE	1.30	÷	<b>3</b> C	ų	Q
	10	•	1992	.67 40,038	55.5	Š	Ş	۳,	8
	1		1993	.05 tO. 15E	$\sim$	e;		ᆵ	_
	13-5		1994	.09 t0.05 E	160.	27.5	<b>)</b> 4	1.56 01	_
	•		1995	.57 to. 13E	•	<u>.</u>	<u>پر</u>		
	-		1996	.66 #0.07 E	5	50.3	¥	2. IE 01	_
	•		1997	.97 to. 06E	Q.	26.8	36		
	· (J*		1998	.0310.22E	101.	*	30	8. fE 0	8
	10		6661	.07 tO.11E	õ		ž	• 4E	0
	=		2000	.9940.31E	247.	21.6	40	1.2ECO	0

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TABLE E.4 (CONTINUED)	ONTINUED)						8 6 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	
ARC LOCATION	COLLECTION NO.	TLW TLW NC.	FU-239,240 ACTIVITY (DFM)	URAN IUM Fricro Grams)	Y IEL C I R=RE MORK I	COUNT	ANAL JHON	1
				•	•		ų	
E 1C-14A- 5	NONE	C SF - 2 GOB	. 83 4U.USE C	9	2 4 5	) (	,	
~		2010	.07 40.20E 0	•	26.9	26	پ	
w w		2009	6.44 #0.14E 05	896.	40.	40	'n	
		2011	.13 #0.07E	112.	31.5	96	#	
, (	•	2012	7	51.9	68.6	<b>Y</b>	1, 25 01	
. 5		2013	.97 #0.14E	37.6	28.5	40	E.	
2 =		2014	.24 t0.03E		15.0	36	3	
HOB DM-C1-1	5005-1	CCD- 875	0		52.6	<b>3</b> CC	:	
	7		.00 #0.		28.5	206	1. (E	
<i>•</i> •••	. m	877	.0048.00E-0		56.7	<b>20</b> €	1. (E	
•	•	878	.2041.20E-0		39.4	<b>300</b>		
•	- 47	CF-	.00 19.00 E-0		50.3	<b>300</b>	1. CE	
C2-1	5006-1	CCD- 880	.20 #1. 20 E-0		16.8	<b>30</b> C		
2	7	1	.1041.		45.5	20C	CA 1. CE 00	
,	ı (M	882	.5040.		5 to 0	<b>308</b>	1. CE	
•	•	883	.3340.		46.1	2C C	2. 2E	
•	·		2011.30E-		62.2	<b>307</b>	2. SE-01	
C3-1	5000-1	CCD- 855	.71 40.		66.7	4 C C	3.5	
~					43.2	<b>4</b> CC	1. CE	
	60	857	1.2040.70E-01		64.6	400		
•	•	858			59.6	400	1. Œ	
· 161	· ••		.2041.		27.5	400	<b>1.</b> CE	
1-73	5002-1	CCD- 860	.40 40.		34°C	400		
	7		.00 46 .00E		74.5	) •		
	•	962	2.70 #1.50 E-01		£1.5	) C C	1. CE	
•	•	863	.9041.70		55.1	100	CA 2. 5E-01	
*	· w	CCF- 864	.00 43.		5¢.2	) }	۲. (۴	
1-53	1-4008	CCD- 870	1.80 to.08E 01		31.9	100	2. 3E CO	
•	•	1	5011.60		35.0	3CC	CA 1. (E 00	

NEW DATA THIS REPORT

TABİ	TABLE E.4 (CO	(CONTINUED)						1	1	
A&C	LOCATION	TLW COLLECTION NO.	TLW ANALYSI S NO.	FU-239,240 ACTIVITY (CFP)	URAN JUM (P. J.CRO GRAMS)	Y JELC (R=RE WORK)	COUNT TIME	Ž <b>4</b>	ANAL /MON	
!		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	• • • • • • • • • • • • • • • • • • •		* • • • • • • • • • • • • • • • • • • •		, , ,		• • • •	!
M 08	DM- C5-3	5.04-3	CC3- 872	9.40 #1.60 E-01		26.2	4CC	3	1.68 00	0
		4		.9041.906-0		31.5	<b>5</b> 00	5	4	_
	•	'n	CCF- 874	6.00 #9.00 E-02		64.4	<b>30</b> 2	5		a
	. <del>- 1</del> - 1	5CC7-1	CCD- 885	.90 #1 . 2GE		•	400		•	_
	~	7		.22 #0		50.5	20C	ರ		0
	· m	'n	887	.00 #0. 10 E		•	<b>5</b> 00	5	1. (E 00	0
	•	4	888	.40 #1.50E		47.6	<b>30</b> 0	చ	1. CE 00	0
	· 44	· <b>5</b> 7	ı	.20 #1.40 E-0		£°. 0,	<b>30</b> 0	₹	<b>.</b> . ∈ Ç	ø
	C 3- 1	5003-1	-03	.93 40 . 17E 0		72.5	20C		2.46 01	<b>,</b>
		7	1	.54 40.		83.5	20C			0
	, m	ım	867	.7010.11E G		82.6	4C C		2, SE 00	0
	•	*	868	.52 t0.28E		56.5	704			0
	· w	· vn	CCF- 869	9 40. 19 E		50.4	<b>50C</b>			_
	CB-1	5014-1	CAD-1047			69.1	3 C			0
	~	7	-	ш			100			_
	· m	•	1049	.00 to.07E		34.6	<b>308</b>		1	_
	**	*	1 050	4		61 61 61	300			_
	•	Û	1021	10.26E		70.1				~
	_	٢	7	u		8 8 a		\$	1. EE 00	0
	1-11	5cc1-1	CCD-1698				4			0
	7	~	1699	1.48 to.05E 02		9.48	) ) (		2.76 0.	o
	· M	m	1 700	1.97 to.07 € 01		63.¢	30E			_
	•	4	1011	.49 to. 18 E		39.6	300			0
	· <b>s</b> n	'n	7	2.66 t0.23£ 00			<b>300</b>		1	_
	15-1	4599-1	CCD- 850	.23 +0.03 8		81.5	30			o
	7	7		.58 #0.03E			e C			0
	(A)	•	852	7.08 to.45E 00		5.42	<b>50C</b>		1. EE 00	0
	•	•	853	.17 #0.27E		€1.4	<b>30</b> 0	చ	3. (E C	0
	· vo	v	CCF- 854	9.1011.408-01		A. A.	3 C C	3	1. (E Q	0

	•		
•	•		

TAB	TABLE E.4 (CONTINU	NTINUED)		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		1		;
ARC	LOCATION	TLW COLLECTION	ANALYSIS	FU-239,240 ACTIVITY	URBN IUM I F I CRO GR AMS 1	Y 1ELC (R*RE MORK 1	COUNT TIME	ANAL JHON	
	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		!						1
<b>8</b> 0	DM-16-1	5009-1	068 -023	.34 #0.		15.6	20	.16 0	0
	7		68	5 #0.03		75.5	100	1. EE 0	0
	, m	·M		.65 #0.		3.92	400		
	•	4		.70 #1. 70E-		40.2	<b>30</b> 0	. CE	9
	'n	ĸ	8	0.60 #1.80 E-01		29.6	<b>307</b>	CA 1. (E 00	0
	18-11-1	5023-1	CCD-1703	.47 #0		46.1	<b>30</b> 2		_
			17	0 # 0		36.2	30C		
	<u> </u>	•	1705	1.18 #0.58E-01		51.5	300	1	_
	*	4	1706	3 ‡0		31.5	366		~
	v	ν.	CF-17	.10#5.		45.4	300	-	~
40	CSI-J-CC0	S	C 70-1303	.36 +0		80.2	3 C	• 2£	_
	W-CC0	Ś	~	.7640		76.5	<b>5</b> C		0
	1-0c0	S	1307	.3340		84.5	<b>3</b> C	E	_
	M- CC0	S	1306	.73 +0		17.4	<b>3</b> C	. Æ-	
	033 -N	S	1305	.32 tO.26E 0		74.4	26	• 4E	_
	900	ς,	1310	.45 t0.07E D		11.C	300	#	~
	012	5216	1309	3.2040.126 02		34.6	<b>3</b> 2	E	_
	910	S	1308	.52 to .14E 0		m	10C	8. EF-01	
	CMR-C1A	6	C VS-1502	.84 t0.44E		6	<b>30</b> 0		
	C 18		1503	.07 40.		÷	707		
	C 2A		1504	.65 tO.31E		÷	200	6. EE CO	_
	C 28		1505	.04E		69.6	<b>20C</b>		
	C 3A		1506	.136		10.9	<b>3</b> 0 <b>2</b>		
	C 38		1507	7.14 to.29E 01		14.4	200		
	CIA	9723	1475	•		C 5 . E	1000		
	C 18		1476	7.42 t0.22E 02		40.5	<b>500</b>		
	C 24		1477	.76 +0.26 €		14.4	<b>60</b> C		
	C 28		1499	46 10.0		24.ER	<b>5</b> 00		
	C 34		005]	.26 to.05E		19.5	3 3 3		

6.86 #11.05 #0.00	(GTOWN NOO) FIR THEY						
CHR-C3B 9723 CVS-1501 6.86 #1.23 E 01 35.  6	COLLEC NO	TLE N AMALYSIS NG.	FU-239,240 ACTIVITY (DPF)	URBNIUK (FICRO GRAMS)	Y IELC IRERE MORK)	COUNT TIME	ANAL IPON
R Z-C1-5  NONE  CST-2015  9.3340.61E 00  7.2017  9.4240.52E 01  7.2018  9.3340.61E 00  7.2019  3.2640.13E 02  9.2020  3.0540.11E 01  2.022  1.6240.12E 01  42.640.12E 01  42.740.09E 02  2.024  1.1640.09E 01  2.025  2.0240.09E 01  2.024  1.1840.09E 01  2.024  1.1840.09E 01  2.025  2.0240.09E 02  1.1840.09E 01  2.026  1.1840.09E 01  2.026  1.1840.09E 01  2.027  3.8940.34E 00  1.1540.04E 02  1.4540.32E 01  1.5640.32E 01  1.5640.32E 01  1.5640.32E 01  1.5640.32E 00  1.6640.32E 00  1.664	CH8 - C38 912	C VS-1501	6.86#1.23£ 01		03.5R		
2016 9.33#0.61E 00 7. 2017 2017 9.42#0.52E 01 28. 2019 3.26#0.13E 02 9. 2019 3.26#0.10E 01 0. 2020 3.02#0.11E 01 46. 2022 1.62#0.12E 01 46. 2023 2.22#0.29E 01 42. 2024 1.18#0.09E 01 29. 2026 1.18#0.09E 02 14. 2027 3.89#0.34E 00 136. 1.2029 3.60#0.32E 01 136. 2030 8.25#4.13E 00 105. 2031 0.00#5.66E 00 105. 2033 1.77#0.21E 00 14.	NON STORY OF STREET	C ST - 2015	.05 #0.08 E	35.2	55.6	2CC	$\overline{z}$
2017 9.42 ±0.52 € 01 28. 2018 4.36 ±0.13 € 02 9. 2019 3.26 ±0.10 € 01 0. 2020 3.05 ±0.11 € 01 2. 2021 3.02 ±0.11 € 01 2. 2022 1.62 ±0.12 € 01 45. 2024 1.18 ±0.09 € 01 42. 2025 2.92 ±0.09 € 02 3. 2026 1.18 ±0.09 € 02 14. 2027 3.89 ±0.09 € 02 14. 2028 2.05 ±0.09 € 02 14. 2029 3.60 ±0.32 € 01 136. 2030 8.25 ±4.13 € 00 195. 2031 0.00 ±5.66 € 00 105. 2033 1.77 ±0.21 € 01 33.		2016	.33 #0.61E	7.64	83.6	30 <b>C</b>	2. CE-02
2018 4.3640.13E 02 9. 2019 3.2640.10E 01 0. 2020 3.0540.11E 01 2. 2021 3.0240.11E 01 46. 2023 2.2240.29E 01 46. 2024 1.1840.09E 01 29. 2025 2.9240.09E 02 3. 2026 1.1540.09E 02 3. 2027 3.8940.34E 00 11. 2028 2.1640.32E 01 136. 2030 8.2544.13E 00 105. 2031 0.0045.66E 00 105. 2033 1.7740.21E 00 14.	. ~	2017	.42 t0.52E	28.4	•	3 C C	3
2020 3.26 ± 0.10 ∈ 0.1 0.2 2.2 2.2 ± 0.11 ∈ 0.1 2.2 2.2 2 ± 0.11 ∈ 0.1 2.2 2.2 ± 0.12 ∈ 0.1 45.2 2.2 ± 0.29 ∈ 0.1 2.2 ± 0.29 ∈ 0.1 2.2 ± 0.29 ∈ 0.1 2.2 ± 0.29 ∈ 0.1 2.2 ± 0.29 ∈ 0.1 2.2 ± 0.29 ∈ 0.1 2.2 ± 0.29 ∈ 0.2 2.2 ± 0.29 ∈ 0.2 2.2 ± 0.29 ∈ 0.2 2.2 ± 0.29 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 2.2 ± 0.20 ∈ 0.2 €	- 40	2018	.36 t0 . 13 E	9.12	47.1	<b>300</b>	35
10 2020 3.0540.11E 01 2.2222 2.2240.11E 01 46.2022 1.6240.12E 01 46.2023 2.2240.29E 01 42.2024 2.2240.29E 01 29.2024 2.0240.09E 02 2.025 2.9240.09E 02 3.6224 2.025 2.9240.09E 02 3.6224 2.0224 2.0226 2.0224 2.0226 2.0224 2.0226 2.02226 2.022226 2.02226 2.02226 2.022226 2.022226 2.02222222222	, o	2019	.26 to. 10 E	0.358	•	30C	3
2021 3.02 ±0.11E 01 2022 1.62 ±0.12E 01 2023 2.22 ±0.29E 01 2024 1.18 ±0.09E 01 2025 2.92 ±0.09E 01 2026 1.15 ±0.04E 02 2027 3.89 ±0.34E 00 2028 2.16 ±0.32E 01 2030 8.25 ±4.13E 00 2031 0.00 ±5.66E 00 2032 2.66 ±0.28E 00 2033 1.77 ±0.21E 00 2034 3.84 ±0.17E 01	01	2020	.05 40.116	2.14	J• 08	30C	H
2022 1.62#0.12E 01 46 2023 2.22#0.29E 01 42 2024 1.18#0.09E 01 29 2025 2.92#0.09E 01 29 2026 1.15#0.04E 02 14 10 2027 3.89#0.34E 00 14 2029 3.60#0.32E 01 136 2030 8.25#4.13E 00 16 2031 0.00#5.66E 00 105 2033 1.77#0.21E 00 7 3.84#0.17E 01 33		2021	.02 #0. 11E	4.03	81.7	30 <b>C</b>	Ή.
2024 1.18 ±0.29 € 01 42 2024 1.18 ±0.09 € 01 29 2025 2.92 ±0.09 € 02 3 2026 1.15 ±0.04 € 02 14 10 2027 3.89 ±0.34 € 00 136 2028 2.16 ±0.32 € 01 136 2029 3.60 ±0.32 € 00 15 2030 8.25 ±4.13 € 00 165 2031 0.00 ±5.66 € 00 105 2032 2.66 ±0.28 € 00 14 2033 1.77 ±0.21 € 00 7	C2-5	2022	.62 tO. 12E	0.94	41.5	30 <b>C</b>	#
2024 1.1840.09E 01 29 2025 2.9240.09E 02 3 2026 1.1540.04E 02 14 10 2027 3.8940.34E 00 14 2028 2.1640.32E 01 136 2029 3.6040.32E 00 15 2030 8.2544.13E 00 105 2031 0.0045.66E 00 105 2033 1.7740.21E 00 7	1 <b>- 40</b>	2023	.22 40.29E	42.1	30.5	30C	3
2025 2.9240.09E 02 2026 1.1540.04E 02 1 2027 3.8940.34E 00 2028 2.1640.32E 01 13 2029 3.6040.32E 00 1 2030 8.2544.13E 00 1 2031 0.0045.66E 00 10 2032 2.6640.28E 00 1 2033 1.7740.21E 00 3	. (-	2024	.1840.098	29.4	48.9	<b>2</b> 0C	¥.
2026 1.15 ±0.04 € 02 1 10 2027 3.89 ±0.34 € 00 2028 2.16 ±0.32 € 01 13 2029 3.60 ±0.32 € 00 1 2030 8.25 ±4.13 € 00 1 2031 0.00 ±5.66 € 00 10 2032 2.66 ±0.28 € 00 10 2033 1.77 ±0.21 € 00 3	· <b>a</b> s	2025	.92 tO.09 E	3.40	42.2	<b>30</b> 0	Ä
10 2027 3.8940.34£ 00 2028 2.1640.32£ 01 13 2029 3.6040.32£ 01 13 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	•	2026	.15 #0.04E	14.7	87.7	100	#
2028 2.1640.32E 01 13 2029 3.6040.32E 00 1 2030 8.2544.13E 00 1 2031 0.0045.66E 00 10 2032 2.6640.28E 00 1 2033 1.7740.21E 00 3	10	2027	.8940.348	1.28	83.6	<b>30</b> C	Ÿ.
2029 3.6040.32E 00 11 2030 8.2544.13E 00 11 2031 0.0045.66E 00 10 2032 2.6640.28E 00 10 2033 1.7740.21E 00 10	, sed	2028	.16 40.328	136.	24.E	<b>3</b> 0 <b>C</b>	2
2030 8.2544.13E 00 1 2031 0.0045.66E 00 10 2032 2.6640.28E 00 1 2033 1.7740.21E 00 7 2034 3.8440.17E 01 3		2029	.60 t0.32E	15.5	86.4	30Z	3
0.0045.66€ 00 10 2.6640.28E 00 1 1.7740.21E 00 3.8440.17E 01 3	, <b>v</b>	2030	.25#4.13€	19.9	68.7	40	8
2.6640.28E 00 1. 1.7740.21E 00 3.8440.17E 01 3	•	2031	.00 #5.66E	105.	25.C	<b>7</b> C	w
1.77 #0.21E 00 3.84 #0.17E 01 3	•	2032	.66 40. 28 E	14.8	82.5	<b>5</b> 00	7
3.84 t0.17E 01 3	, σ-	2033	.77 #0.21E	7.80	89.2	<b>2</b> CC	7. £E-62
	07	2034	.84 tO.17E	33.0	23.6	200	1. CE CC
3.41 40.146 01 90	) ~	20.0	34140.145	6.06	42.6	200	8 · (E - C2

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, +7

LEFT FEHUR ROB 635 1.4 025. 2.10#8.60E-02 400 33.2 0.567 KIDNEY ROK 449 2.1 025. 0.00#0.25E 00 60 37.3 0.273 C. LIVER ROL 392 9.1 025. 1.38#0.13E 00 1000 24.48 0.050 LUNG ROR 480 2.9 025. 3.36#0.20E 00 59.5 2.88 HILAR NODE ROH 447 0.3 025. 2.40#6.10E-02 500 48.5 0.734  LEFT FEMUR ROB 657 1.4 025. 0.00#0.17E 00 40 85.4 KIDNEY ROK 150 1.8 025. 2.53#6.93E-01 400 59.2 LUNG ROR 311 3.5 025. 5.03#0.16E 01 400 55.8 0.182 RIVER ROT 141 1.4 025. 3.82#0.32E 00 1000 12.58 TRACHEA ROT 141 1.4 025. 4.57#0.15E 02 500 56.5 F. MUCO SA RON 327 0.6 025. 5.27#0.36E 00 500 38.6

. NEW DATA THIS REPORT

TABLE	E.5 (CONTINUED)	۵) م	1	1		1	1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
AV IMAL NO.	SAMPLE TYPE	TLW NO.	X.W.	WET WEIGHT	PU 239, 240 ACTIVITY (OPM)	COUNT TIME	Y 1 EL O ( R *R E - MORK )	URAYIUM (MICRO GRAMS)
1024 - 1	LEFT FEMURE KIDNEY		! •	4	1.23#2.47E-01 4.74#5.93E-02	400	57.5	
	. LUNG HILAR NOJE 7 TRACHEA 8 G. I. TRACT 9 P. MUCOSA	RDR 334 RDA 66 RDT 80 RDS 627 RDS 627	4 4 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	625. 025. 025. 025.	5.39*0.34ē 00 1.50*5.80E-02 2.10*1.20E-01 8.56*0.31ē 01 5.22*6.95E-02	\$00 200 40 200	27.8R 64.9 39.4 77.2 61.9	0.490
10229-11	S JAJJIEG				1.23#0.78E-01 0.00#0.20E 00 0.00#0.04E 00 2.25#0.80E-01 4.58#0.17E 01 4.15#0.27E 00 2.13#0.09E 01 2.55#0.08E 02	600 400 300 1000 500	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	0.567
1035 - 10035 - 10040 -	Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z				1.8411.47E-01 2.21#0.10E 01 0.00#0.20E 00 3.60#6.00E-02 2.51#0.74E-01 5.17#0.12E 01 8.26#1.06E-01 1.04#0.62E 02 1.12#0.10E 00	300 600 600 600 100 100 100	200.2 200.2	0.550

LOST IN DISS.

REMARKS

TAGGED . VOMITED!

LOST 18 0155.

. NEW DATA THIS REPORT

TABLE 1	(c)	Ω)				1			
AV!YAL	SAMPLE TYPE	TLW NO.	3 9	NET WE 1 GHT	PU 239, 240 ACTIVITY	COUNT 11HE	Y 1 EL D ( R * R E -	URANIUR (MICRO GRANS)	REHARKS
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1							
1040 - 1	LEFT FEMUR		1.4		0.85#1.69E-01	0	83.7		
*	KIDNEY		2.1		1.22 to.98E-01	200	48.2		
	LIVER		6.5		5.2840.636-01	900	41.8		
:	100		3.1		5.58 # 1.29 E-01	300	46.3	0.928	
:	HILAN NOJE		0.3		1.31+1.156-01	<b>5</b> 00	72.2		
	TRACHEA		1.1		1.63 # 0.68 E-0'	400	59.4		
•	6. 1. TAACT		5.5		2.69#0.08E U.5	30	41.1		·
•	P. MUCOSA		4.0		1.11#0.128 00	300	76.9		VUMI I INCLUDED
10		RDN 318	0.9	025.	7.00 40.33E 00	1000	25.3R		
1041 - 1	FET SENIE	RDB 641	1.6	075.	4.9344.936-01	0,	38.3		
7	KIONEY	ROK 78	2.4	025.	3.28 11. 79 6-01	200	39.6		
۳ • •	+ 1 VE \	ROL 148	12.9						LOST 1H DISS.
:	L WG		4.0		1.37+0.05E OL	400	63.8	0.073	
:	HILAN NODE		0.0		0.00+0.07E 00	200	69.2		
	TAACHEA		2.2		1.04+J.06E 01	200	6.10		
:	G. 1. TRACT		0.0		5.35+0.176 02	90	23.3		
6	P. MUCOSA		0.0		4.80+3.80E-02	400	74.2		
01	N. MUCOSA		0.0		3.83+3.836-01	70	31.7		
1045 - 2 KIDNEY	KIDNEY	RDK 158	3.4	.220	4.21#4.21E-01	9	\$0.5	0.041	C.5. 11
1046 - 1	SET SENIE		1.6		1.2840.856-01	400	36.9		C.5. 11
	KIDNEY	ROK 451	2.1	025.	1.14+2.275-01	9	41.6	0.166	c.s. =
:	HILAN NODE		0.3		0.0040.10E 00	100	58.7	0.411	C.5. 11

•	TABLE	Ξ.	Ω	,	1	1				1 1 1 1 1	1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
•	AVIVAL ND.	SAMPLE	TLX NO.	† †	¥ %	X 10 X 1	PU 239, 240 ACTIVITY (DPH)	COUNT TIME	Y I EL O I R = R E - WORK )	URANIUM IMICRO GRAMS)	ARK
iÄ	1050 -	1 LEFT FEHUR 2 KIDNEY	808 66 80K 8	:	1.8	025.	0.00+0.25E 00 2.36+4.72E-02	40 200	35.5 70.6		
	::	3 LIVER			0°3		0.54#1.08E-01 2.55#0.24E 00	00 <b>9</b>	26 . 18 22 . 68	0.032	
	::	S HILAN NODE			1 10		1.01#1.01E-01 7.56#0.42E 00	300 200	55.0 68.0		
		8 6. 1. TRACT			<u>_</u>		1.64±0.04E 04	50	54.0		
	::	P. HUCOSA		168 309	<b>د</b> ه	.570 025.	1.64#0.07E 01 2.49#0.07E 02	200 500	70.9 36.2		
•					c		00 966 0100 0	4	1 67		
~	1054		20 X	2 6	. 4	025.	-2.30#9.20E-02	200	24.3		
	:	3 L I VE 3		06	8.5		8.50 #8.50 E-02	300	58.7		
	:			0	2.2		3.63 #0.26E 00	700	26.2		
	:			60	0.2		1.24#0.12E 00	000	48.88		
	:	7 TAACHEA		E 1	1.0		1.16#0.18E 00	1000	0, 11		
	:	ؿ		œ	4.2		3.26 #0.06E 03	200	m ·		
	:	9 P. MUCOSA		145	0.3	025.	6.14#1.17E-01	1000	12.9R		
	~	10 N. MUCDS4		80	0.1		4.14#1.32E-01	<b>4</b> 00	n		
7	1060 -	1 LEFT FEHUR			1.1	025.	1.10#2.216-01	9	57.0		
)	:	KIDNEY	RDX	10	5.0	.570	0.00 #0.05E 00	200	6.84		
	:	3 Liver			0			,	•		COST IN UIS
	:	רטאט			m (		9.9541.346-01	009	c- <b>17</b>		
	-				m ·		5.00+6.30E-02	004	· · ·		
	=	RACHEA			، و		0.6141.846-01	001	1.4		
	=	ئ			σ,		5.73#0.63E 0C	007	¥		
	=	9 P. MUCOSA			n ·		0.00+0.04E 00	000			
	<del>-</del>	10 N. MUCUSA			10		10-261.0478.1	0	?		

. NEW DATA THIS REPORT

- <del>-</del> -	JAPE TYPE	TLW NO.	WET WEIGHT	PU 239, 240 ACTIVITY (DPH)	COUNT TIME	Y 18.0 18.86- MORK 1	URANIUM IMICRO GRAMSI	g.	EMARKS
- 2	(	ROB 617	1.5 025.	ነ የ	700	09.9R	0.345	5: 1	 
-		39	.7 02	.89+3.89E-0	<b>4</b>	•	.08	.s.	_
:	3 LIVE?	42	70 1.	.0130.14E D	400	60	0.110	.5. 1	_
•		48	3.8 02	2.88 to . 23 E 00	200	37.0		.5.	-
=	S HILAN NODE	4.4	.2 02	.92#1.37E-0	100	_	0.311	.5. 1	
1067 -	1 LEFT FEMUR	8 64	20 0.	<b>G</b> 7	90	0.09			
:		X = X	4 02	.82 #1.03E-0	200	57.5			
:	3 L1VE1	ROL 218	7.3 025.	2.14 #0.83E-01	600	39.1			
:	1 UNG	R 31	70 5.	.05 #0.05E 0	900	37.4	0.039		
-	5 HILAN NOJE	۲ ۲	30 E.	.60 #5.10 E-0	300	74.3			
:	TRACHEA	91 1	₹3 05	.7949.795-0	200	30.2			
•	ၒ	\$ 61	570 E.	.85 #0.65E 0	200	26.8			
	۲,	8	70 5.	.00 #0 .71 E-Q	300	8.69			
<u>.</u>	10 N. HUCOSA	30	\$ 078	.90 # 5 . 60 E -0	400	63.6			
- 6931	1 LEFT FEMUR	9 66	N	1.04#2.096-01	40	67.9R			
•	K TON	K 13	9					LOST :N	215
:	3 LIVER	12 7	o	.69+0-94	006	25.9			
•	5 <u>8</u> 2	RDR 291	3.1 025.	E 0	200	40.8			
•	SHILAS NODE	<u>ج</u>	3	.14 +0.16	300	51.5			
-	TZACHEA	80	4	.42 t0.71E-0	400	54.58			
•	•	\$ 63	~	.15+0.03E 0	100	17.9			
•	ď	P 1.1	5	.01 #0.84E-D	200	70.1			
<u>-</u>	O N. MUCOSA	× 32	~	.00 #0.09E U	<b>00</b>	28.5			
1073 -	1 LEFT FEMUR	19 9	.7 02	.18+1.875-0	100	12.2R	.43		
•		* **	70 2"	.1441.716-0	100	45.1	==	.5.	
•		ROL 427	13,4 025.	3.4011.32E-01	400	37.3	0.695	C.S. 11	
•		× 48	°1 0	.94 tO.21E 0	200	54.0	1:1	3:	
-	1166.								

<u>.</u>

TABLE E.5 (CONTINUED)

ANIMAL NJ.	SAMPLE		13. WE 1 GH T	PU 239, 240 ACTIVITY (DPH)	COUNT TIME	Y 1 ELD (R = R E - HOPK )	URANIUM (MICRO GRANS)	REMARKS
107	LEFT FEMUR KIDNEY LIVER LUNG HILAR NODE TRACHEA G. I. TRACT G. I. TRACT P. MUCOSA	RDB 640 RDK 113 RDL 203 RDR 314 RDJ 151 RDS 629 RDP 75 RDP 75	1.4 025. 1.8 025. 7.5 025. 0.3 025. 1.7 025. 1.7 085. 0.5 025.	1.01#2.02E-01 2.90#4.80E-02 6.16#1.11E-01 1.85#0.85E-01 0.00#0.04E 00 2.32#0.84E 00 5.90#4.90E-02	500 500 300 300 400	54.1 27.68 60.2 52.4 72.4	·	רס\$ד נא 2155.
10.81 - 1.82	LEFT FEHUR KIDNEY LIVER LUNG HILAR NODE HILAR NODE TRACHEA G. I. TRACT G. I. TRACT P. MUZOSA	RDB 637 RDL 209 RDR 315 RDH 315 RDH 161 RDS 615 RDP 198	1.7 025. 2.1 025. 11.0 025. 3.6 025. 0.3 025. 1.6 025. 6.0 025. 0.4 025.	2.77*1.23E-01 1.78*1.19E-01 5.38*0.54E-01 4.91*0.18E 01 9.00*9.00E-02 7.74*0.45E 00 7.61*0.19E 02 1.16*0.31E-01 2.54*0.32E 00	400 200 200 400 400 400 900	23.0 56.3 53.2 53.2 66.1 17.5 81.6	0.516	
10.87 - 2.2 - 2.4 + 4.4 - 2.5 -	LEFT FEMUR KIDNEY LIVER LUNG HILAR NODE TRACHER G. 1. TRACT G. 1. TRACT P. MUCGSA N. MUCGSA	RDB 664 RDK 227 RDR 313 RD4 59 RD4 162 RD5 621 RD9 69	1.5 025. 2.3 025. 10.0 025. 0.2 025. 1.7 025. 5.4 025. 5.4 025.	4.06*4.06E-01 0.00*0.31E 00 4.41*0.64E-01 6.01*0.17E 01 5.60*2.80E-01 1.59*0.19E 00 2.75*0.05E 03 1.90*1.90E-01 2.25*0.47E-01	600 500 600 500 100 200 200 900	20 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0.319	

. NEW DATA THIS REPORT

TABLE E.5 (CONTINUED)

	• CN	TYPE	NO.	* .	¥	WE I GHT	PU 239, 240 ACTIVITY (DPK)	INE	1 1 EL U ( R = R E - MOR( )	CRAYION CRICRO CRAYS)	REMARKS
1694		LEFT FEMUR	RO8 XOX	638	2.8	025.	5.07#1.45E-01 2.37#0.47E-01	400	29.3	,   	1 1 1 1 1 1 1 1 1 1 1 1
•	1 177	LIVER	ROL	210	12.4		1.10#0.296-01	200	76.9		
:	4	DNO T	RDR	342	.5		1.06#0.03E 02	400	43.1	0.098	
:	S	HILA? NOJE	ROH	144	0.3		0.00 # 6.00 E 00	40	9.89		
:	_	TAACHEA	10%	171	5.6		2.50 # 0.09 E 01	900	21.4		
:	80	G. 1. 134CT	R.) S	019	8.0		2.3840.07E 02	20	55.1		
:	σ		ROP	89	4.0		1.24#0.16E 00	200	6.49		
:	10	N. MUCOSA	RON	344	1.0	.820	1.53#0.07E 01	1000	18.68		
1096 -		LEFT FEMUR	ROB	651	1.3		0.00#0.18E 00	4	78.5		
:	~	K I UNE Y	<b>80</b> ×	452	2.6		5.59#1.ABE-01	1000	15.28		FDUND 2/20/64
:	M	L 1 VE 3	RDL	226	10.8		3.33#0.93E-01	400	59.1		
:	4	L WG	ROR	339	3.5		1.77 #0.64E-01	009	66.5		
:	'n	HILAN NODE	80 ±	84	0.5		-3.80 # 7.60 E-02	200	43.7		
:	~	TRACHER	RU T	167	2.2		4.40#2.80E-02	600	85.1		
:	∞	G. 1. TRACT	S CS	620	2.5		4.80 +0.57E 00	200	18.0		TAGSED "LARSE"
:	σ	P. MUCOSA	ROP	.51	0.1		7.00 #8.80 E-02	200	67.7		
:	10	N. MUCOSA	<b>X</b> 0 <b>N</b>	319	0.8	•\$70	1.59+0.17E 00	400	42.5		
1097 -	-	LEFT FEMUR	ROB	624	1.5		1.69+2.54E-01	9	80.8	0.435	
:	~	K 10NE Y	#0#	444	2.2		1.76 + 0.88 E-01	400	53.6		
=		LIVE	ROL	397	10.2		8.00 # 1.08 E -01	009	28.8	0.074	
:	4	1040	ROR.	483	3.3	025	3.04#0.21E 00	200	47.3	0.299	C.S. 11
=	•	HILAN NOSE	<b>8</b> 04	442	0.5		-0.55 tl. 10 E-01	100	51.7	0.214	

. NEW DATA THIS REPORT

TALLE E.6 (CONTINUED)

URAN IUM REHARKS IM ICRO GRAMS 1	0.311 LOST IN DISS. TAGGED 'BLOOD IN TRACHEA'	0.251	0.524 C.S. II 0.789 TAGGED *CONTENTS EXPOSED*
YIELD (R:RE-	34.5 86.6 46.7 29.7 51.9 17.9 59.8	668 668 668 668 668 668 668 668 668 668	6 4 1 2 2 2 2 3 4 3 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
COUNT	000 000 000 000 000 000 000 000 000 00	400 600 600 600 400 700 700 700	600 1000 300 400 40 200 200 600
PU 239, 240 ACTIVITY (DPH)	2.05*4.11E-01 9.10*5.50E-02 6.61*0.66E-01 1.09*0.0¢E 01 3.11*1.09E-01 1.17*0.0¢E 03 3.86*0.10E 01 9.49*0.27E 01	0.00#0.50E 00 1.14#2.28E-01 4.00#0.64E-01 3.97#0.19E 00 8.88#0.98E-01 2.01#0.16E 00 7.08#0.23E 02 1.07#0.05E 01	4.70+0.236 00 2.87+2.87E-01 4.03+1.68E-01 3.79+0.21E 00 1.03+2.06E-01 6.08+1.33E-01 8.33+0.17E 03 9.64+1.45E-01
NET WEIGHT	1.7 025, 2.1 025, 1.2 0 625, 0.2 025, 1.6 025, 7.0 025, 7.0 025, 0.4 025,	1.3 025. 1.7 025. 13.3 025. 0.2 025. 1.5 025. 6.6 025. 6.6 025.	3.8 025. 1.0 025. 7.9 025. 2.4 025. 0.2 025. 1.1 025. 5.3 075. 0.4 025.
TLW NO.	RDB 642 RDK 164 RDL 205 RDR 332 RDH 152 RD T 197 RD S 626 RD 65	RDB 665 RDK 138 RDL 134 RD4 310 RD1 173 RD 5 623 RDP 76	ROR 337 ROK 170 ROL 91 ROL 91 ROH 157 ROT 143 ROS 614 ROP 85
SAMPLE TYPE	LEFT FEMUR KIDNEY LIVER LUNG HILAR NODE TRACHEA G. I. TRACT P. HUCOSA	LEFT FEMUR KIDNEY LIVER LUNG HILAR NODE TRACHEA. G. I. TRACT P. MUCOSA	LUNG LEFT FEMUR KIDNEY LIVE LUNG HILAY MODE TAACHEA G. I. TAACT P. MUCOSA N. MUCOSA
AVIHAL ND.		232	1119 - 4

\* HEM DATA THIS LEPORT

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REMARKS	LOSF IN 01 SS.	C.S. 11 C.S. 11 LOST IN DISS	C.S. 11
URANIUM (MICRO GRAMS)	0.251	0.888	0.334
YIELD (R#RE- WORK)	20 20 20 20 20 20 20 20 20 20 20 20 20 2	36.1	0.99
COUNT T IME	400 400 100 100 1000	400	200
PU 239, 240 ACTIVITY (DPH)	2.38 # 2.38 E - 01 9.80 # 5.40 E - 02 3.18 # 0.21 E 00 0.00 # 0.07 E 00 6.10 # 6.10 E - 02 2.38 # 0.14 E 01 1.41 # 0.40 E - 01 2.80 # 0.51 E - 01	1.22 to.16 E 00	5.40#7.20E-02
WE IGHT	1.5 025. 10.8 025. 10.8 025. 6.1 025. 1.8 025. 1.8 025. 0.6 025.	1.1 025.	0.2 025.
TLW NO.	RDB 654 RDK 112 RDL 219 RDR 333 RDH 60 RDT 172 RDS 611 RDP 72	RDB 636	
SAMPLE TY?E	1132 - 1 LEFT FEHUR 2 KIDNEY 3 LIVER 4 LUNG 5 HILAR NODE 7 TRACHEA 8 G. 1. TRACT 9 P. HUCOSA	1134 - 1 LEFT FEMUR	HILAN NODE
ANIMAL NO.	1132 - 1	1134 - 1	: :

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TABLE E.5 (CONTINUED)

NEW DATA THIS REPORT

G. I. TRACT P. MUCD SA V. MUCD SA

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LEFT FEMUR KIDNEY 0.781

400 400 400 400 400 400

0.0040.32E 00 4.5043.60E-02 1.2640.63E-01 5.8240.80E-01 0.6041.00E-01 0.0040.05E 00 2.5840.07E 03 3.1945.32E-02

025. 025. 025. 025. 025. 025. 025.

653 166 220 335 212 202 154

R008 R008 R008 R008 R008 R008

> LIVER LUNG HILAR NODE TRACHEA

	SAMPLE TYPE	TLW NO.		WE I GI	<b>T</b>	PU 239, 240 ACTIVITY (DPM)	COUNT TIME	Y 1 B. D (R = RE- WORK)	URANIUM IMICRO GRAMS)	REMAGKS
LEFT	FENUR	1	517	5.8	.570	1.94+0.55E-01	006	45.4		
KIDNEY	>		514	•••	025.	4.54#4.548-01	40	46.8		
IVE			504	1.6	LBS.	3.18#0.826-01	800	35.2		
LUNG			115	12.4	025.	4.01#0.84E-01	400	9.04		
HILAS	NOSE	RS-	213	0.2	.520	2.50#3.10E-02	009	80.3		
LEFT	LEFT FEMUR		599	7.8	025.	1.15 \$0.82 6-01		36.1		
X I D NE	<b>&gt;</b>		590	3.1	S	1.22#0.70 6-01	400	45.2		
LIVER			574	1.3	LRS.	4.68 t0.30 E DO	٠.	36.8R		
200			542	15.2	.220	2.40 #0.10E'01	200	35.7	2.41	
HILA?	NODE		605	0.3	025.	0.00 \$0.54E 00	04	26.3		
TAACHEA			592	3.5	.820	1.28 #0.51 E-01		55.3		
_	. TRACT		667	14.9	L85.	9.07 +0.30 € 01	800	03.88		
¥.	MUCO SA	R SN	965	3.1	.220	2.73 to.08 E 02	1000	19.5R		TAGGED . VOR. I TED.
K I DINE Y	<del>_</del>		910	•	.820	.23#2.45E-0	40	57.7		
LEFT	LEFT FEMUR	R 58	655	5.5	.570	1.57#0.59E-01	200	53.4		
LIVE			507	•	1.45.	.01 #0.54 E-0	006	44.8		
1 U.1G			995	•	025.	.03 #0.13E 0	400	47.4		
HILAS	360N		615	•	.570	.30 £ 3.80 E-0	009	65.4		
LEFT	FEMUR		466	\$	025.	.3047.906	600	29.8		
K 1045	K TOYEY		370	٠. و.	025.	1.10#2.20E-01	40	4.49		
LIVER			358	Ų,	LAS.	.9640.79	400	55.7		
L U.S			556	Ġ	.570	.09#0.116	200			
HILES	1006	RST	287	*	.570	.15+3.92E	400	60.2		
LEFT	LEFT FEMUR		486	6.5	025.	2.15+2.15E-01	40	65.7		
KIDNEY	>		250	3.9	0.25.	4.12+8.24E-72	100	8.89		
1 1 46.4		A SE	272	1.2	Las.	3.3310.806-01	300	19.1		
1000			257	-,-	Liss.	5.16+0.59E-01	200	47.8		

. YEW DATA THIS REPORT

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AVIMAL NO.	SAMPLE	TLW NO.	7 %	WET WE 1 GHT	PU 239, 240 ACTIVITY (CPH)	COUNT TIME	Y I EL D I R = R E - WORK I	URANIUM IMICRO GRAMSI	æ	REMARK S
2013 - 11	LEFT FEMUR KIDNEY LIVER LUNG	RSB 532 RSK 369 RSL 275 RSK 545 RSH 280	5. 15.	4 025. 6 025. 6 LBS. 7 025. 4 025.	2.31#0.46E-01 3.37#3.37E-01 2.92#0.18E DO 9.30#0.30E 01 7.09:4.26E-01	500 40 500 400 70	2 6 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8			
2015 - 1	LEFT FEMUR KIDNEY LIVER LUVG HILAR NODE	RSB 457 RSK 246 RSL 360 RSR 554 RSH 277		4 025. 4 025. 1 L85. 6 025. 3 025.	2.9641.29E-01 1.62#0.13E 00 0.00#0.15E 00 6.20#0.38E 00 1.91#1.91E-01	1000 900 60 400 200	15.3 32.1 69.3 26.7			
2019 - 1	LEFT FEMUR KIDNEY LIVER LUVG HILAR NODE	RSB 407 RSK 436 RSL 419 RSR 477 RSH 398		1 025. 7 025. 6 LAS. 4 025. 3 025.	0.00#0.18E 00 2.65#0.61E-01 7.34#1.16E-01 3.93#0.31E 00 0.69#1.38E-01	100 300 200 500 100	32.0 59.8 67.5 28.1	0.648 0.344 1.22 0.805	, , , , , , , , , , , , , , , , , , ,	
2027 - 1	LEFT FEMUR KIDNEY LIVEN LUNG HILAR NOJE	8	55 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	9 025. 9 025. 4 LBS. 0 LBS. 3 025.	5.29#0.99E-01 1.97#0.79E-01 7.30#0.93E-01 1.02#0.04E 01 2.01#1.34E-01	1000 900 400 400 400	15.58 40.0 69.0 79.6 21.2			
2028 - 2030 - 5	HILAN NODE HIEFT FEMUR KIDNEY LIVEN LUNG HILAN NODE	RSA 34 RSH 11 RSB 53 RSK 36 RSK 36 RSK 36 RSK 36	E E E E E E E E E E E E E E E E E E E	3 025. 2 025. 3 025. 9 025. 4 025.	1.08+0.04E 01 -1.80+3.50E-02 2.07+2.07E-01 4.30+8.59E-02 3.55+0.84E-01 5.12+0.29E 00 2.49+0.92E-01	200 200 44 500 600 600 600 600	58.9 57.3 57.3 57.3 57.3 54.1	0.092	C.S. 1	

. NEW DATA THIS REPORT

TABLE E.6 (CONTINUED)

-CN	SAMPLE	TLW ND.	э,	HE I	WEIGHT	PU 239, 240 ACTIVITY (DPH)	COUNT TIME	V18.0 (R*RE-	URAN IUM (M ICRO GRAMS)	REMARKS
2031	U.S. INF	222	51 489 427	2.5	LBS.	.53#0.09E .88#0.05E	20 20	15.9		A 4
:::	255 255 255 255 255 255 255 255 255 255	2225	454 266 353	25.0	LBS.	.61#0.03E .98#0.15E	7007	62 54 20 54 20 54 20 54		. ~ ~ ~
	UNINE UNINE FECES	R S S U R S S U R S S F R S F	5236 5236 5255 5255	0.58	LBS. LBS. LBS.	2.54*0.12E 01 2.83*0.05E 03 1.91*0.04E 03 1.42*0.03E 03	300 300 300	15.05 15.08 7.05 16.08 7.08 7.08		21 JUNE 22 JUNE 16 MAY 17 MAY
2032	1 LEFT FEMUR 2 KIDNEY 3 LIVER 4 LUNG 5 HILAR NODE	R SK R SK R SL R SK	408 473 433 433	19.00	025. 185. 025. 025.	2.31*2.31E-01 4.10*4.10E-02 2.01*0.21E 00 4.30*0.28E 00 0.00*0.09E 00	100 300 500 800 100	36.8 61.3 27.7 25.6 61.4	0.470 0.939 0.599 0.288	
2036	URINE URINE URINE URINE FECES	A R S K K S K C C C C C C C C C C C C C C C	461 238 297 524	14 mm 0 4 4 m 0 m	LBS. LBS. LBS. LBS.	1.81+0.05E 02 5.61+0.08E 02 1.48+0.06E 02 4.78+0.02E 03 9.63+0.25E 02	800 400 400 400	61.9 07.48 04.48 16.1		23 HAY 20 JUNE 21 JUNE 22 JUNE 16 HAY
2039 - 1	1 LEFT FEMUR 2 KIONEY 3 LIVER 4 LUNG URINE	R S S K S S S S S S S S S S S S S S S S	5 5 4 5 5 5 4 5 5 5 4 5 5 5 4 5 5 5 4 5	1.9	025. 025. 185. 185. 185.	0.00+0.12E 00 1.91+1.27E-01 3.12+0.26E 00 2.18+0.05E 02 3.02+0.11E 01	1000 300 400 400 200	13.98 46.8 36.5 76.1		16 4AY

. NEW DATA THIS REPORT

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CONFIDENTIAL

TABLE E.6	E.6 (CONTINUED)	(05						
AN EMAL NO.	SAMPLE	TLH NO.	WE I GHT	PU 239, 240 ACTIVITY (DPH)	COUNT	YIELO (R=RE- MORK)	URAN LUM (M ICRO GRAMS)	
2050		•	•		001	18.5	4.68	3.
	A LIVER				9	32.1		ם נ
::	4 LUVG	RSR 474	15.5 025.	9.05+0.21E 01	500	63.5	0.028	5
	í.				2	1.35	974.0	;
- 2502	1 LEFT FEMUR				70	42.9		
:		85K 588	3.5 025,		40	69.5		
:					400	34.1		
:					006	19.8	2.13	
•					40	56.6		
:		RSI			400	41.8		
:	ۍ	RSS			200	35.5		
-	O N. MUCOSE	R SN			700	21.28		
2057	UR INE				200	19.2		16
:	UZ INE	RSU 534	2.2 LBS.	4.6740.15E	100	63.2		23
:	FECES	RSF 500	0.9 185.	1.50+0.046 03	200	09.0R		16
- 0902	LEFT FEMUR				100	53.1		
					9	1.49		
•	3 L!VE?	RSL 242	1.6 (85.	5.28 #1.06 E-01	400	33.5		
:					200	74.0		
:	S HILAY NODE				<b>4</b> 00	0.89		
- 4902		R 58 465			300	39.6		
•					400	83.3		
•	3 LIVER	RSL 366	1.2 185.	3.18+3.185-01	90	48.6		
	25				1000	14.68		
•	S HILAR NOJE				200	56.3		

16 HAY 23 HAY 16 4AY

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REHARKS

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TABLE E.6 (CONTINUED)

	AD. IYPE	NO.	<b>1</b>	ие тен т	ACTIVITY (DPH)	T IME	Y 1 EL D (R = R E - HORK)	CRAMS)	NE NEW N.S
18	UNINE	RSU 414	2.9	1.85.	1.46#0.04E 03	30	48.5	i 1 1 1 1 1 1 1 1	•
•	021VE		•	, 19 19	0	100	15.7	. •	
:	37.5		٠	185	0	200	16.1	. •	
:	FECES		0.8	L85.	0	90	30.2		16 MAY
2602	U2 [15		0.0	LBS.	2.34#0.05E 04	0.7	46.7	-	,
:	UNINE		1.4	<b>.</b> 85.	4.68#0.10E 03	9	33.2	-	7 4 4 4
:	<b>LA 1</b> 'AE	RSU 523	4.3	185.	3.46#0.10E 02	4	52.0	- ••	21 MAY
:	U3 INE		3.4	185.		)	} !	22	MAY LOST IN DISS.
:	UNINE		1.5	LBS.	1.78#0.048 02	100	50.5	''	:
	UNINE		3.8	L85.	1.0250.03E 02	200	14.7R	. —	18 JUNE
	UR INE		3.0	LBS.	2.96 # 0.14 E 01	200	19.5	-	3K17 61
<b>:</b>	UNINE		5.6	183.	2.63#0.07E 02	1000	2 .9R	21.	JUNE BREAKAGE LOS
<b>:</b> 23	325		3.3	L85.	4.64#0.15E 01	200	26.2		22 JUNE
=	FECES		0.3	L85.	1.28 ¢0.02E 02	200	58.6		L6 MAY
=	FECES		0.5	<b>.</b> 65.	9.27 #0.22E 02	40	54.8	- <b>-</b>	17 MAY
2093 -	1 LEFT FEMUR			025.	7.42#1.42E-01	1000	10.78		
:	2 KIDVEY				0.76*1.52E-01	09	62.2		
:			1.2		6.81*1.19E-01	200	26.4		
:	+ LUNG				1-12+0-15E 00	500	32.88		
:	3 HILAN NODE	RS4 284		.\$20	0.00+0.27E 00	04	51.5		
2095 -	1 LEFT FEMUR		5.7		2.07#1.665-01	300	24.4		
:			3.2		0.00+0.136 00	9	77.4	•	
:	3 L I VE 1	RSL 273	1.3		5.55+0.78E-01	004	6.84		
:			13.4		4.39+0.34E 00	400	33.7		
:	S HILAR NOJE		0.3	ozs.	0.00+0.17E 00	200	21.0		
2097	37) 70	SU.5	-	185.		9	•	•	~
=	FECES	RSF 536	e. 0	LOS.	7.97+0.41E 00	300	56.0R	•	× × ×

ANIMAL SAMPLE TLW NEGHT ACTIVITY TIME (RARE-URANIUM TOWN)  2100 - 1 LEFT FEMUR RSB 411 5-1 025. 2-10+3-15E-01 100 27.0 1.76 C.5  2101 - 1 LEFT FEMUR RSB 411 5-1 025. 2-10+3-15E-01 400 38.4 0.340 C.5  2110 - 1 LEFT FEMUR RSB 411 5-1 025. 2-10+3-15E-01 400 38.4 0.340 C.5  2110 - 1 LEFT FEMUR RSB 412 5-1 025. 2-50+3-70E-02 400 63.6 0.18 C.5  2110 - 1 LEFT FEMUR RSB 429 5-7 025. 1-20+2-10E-01 400 20.8 0.18 C.5  2110 - 1 LEFT FEMUR RSB 429 5-7 025. 1-20+2-10E-01 90 60.3  2111											
2100 - 1 LEFT FEWUR RSS 412		ANINA MO.			HE I	GHT	24 ITY	COUNT TIME	1 EL O R * R E O R K )	URAY1UM (M1CRO GRAMS1	REMARKS RR
110 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		2100	THE PERSON IN	17 83		0.55	10#3.15	100	27.0	1.7	
1				7		2	. A2 & 1 . D2	100	4.86	34	.5.
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## 2 KIDNEY			-	80	8.	.570	.20\$2.416	100	23.6		.5
111 UAINE RSW 475 12.3 0ZS. 5.08 +0.21		-		×			.5241.046	90	60.3		
## Comparison		:		: 5		•	.7940.77	200	0.84		.5.
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2111 U3142 RSU 529 1.9 185. 2.3040.04E 04 60 55.0 12.48 1. U31NE RSU 237 5.1 L35. 3.3040.06E 02 300 12.48 117 1. U31NE RSU 247 5.1 L35. 3.3040.06E 02 300 12.48 119 1. U31NE RSU 248 5.0 L85. 5.5040.10E 02 200 29.28 119 1. U31NE RSU 241 2.6 L05. 2.8940.09E 03 30 65.8 1. U31NE RSE 501 0.2 L05. 4.4540.14E 01 200 30.4 1. FECES RSF 491 0.4 L85. 1.1840.03E 02 200 17.4 2 K10NEY RSE 386 5.6 025. 1.6540.51E-01 400 68.8 2 K10NEY RSE 191 3.3 025. 2.3440.91E-01 400 60.5 1 JUNE RSE 380 0.8 L85. 1.6540.61E 00 40.5 1 JUNE RSE 380 0.8 L85. 1.6540.01.00 40.5 1 JUNE RSE 570 13.7 L05. 1.6540.03E 02 600 60.68 1 JAACHEA RSE 570 13.7 L05. 1.6540.03E 02 600 60.68 1 JAACHEA RSE 380 0.8 L85. 1.6540.03E 02 600 60.68 1 JAACHEA RSE 570 13.7 L05. 1.6540.03E 02 600 60.68 1 JAACHEA RSE 381 2.9 025. R.0541.19E-01 1000 16.98		:	HILAR	E,	Ö	**	.19 #2.37 E-0	9	59.7		~;
11. U31NE RSU 469 1.8 LRS. 5.1640.16E 03 100 07.9R 17 11. U31NE RSU 268 5.0 LBS. 5.5640.10E 02 200 29.2R 19 11. U31NE RSU 268 5.0 LBS. 1.2440.05E 02 30 65.8 11. U31NE RSU 268 5.0 LBS. 1.2440.09E 03 30 65.8 11. U31NE RSU 261 2.2 LRS. 1.2440.09E 03 30 65.8 11. FECES RSF 491 0.4 LBS. 1.1840.09E 03 30.4 11. EFT FEMUR RSB 386 5.6 025. 1.6540.91E-01 400 68.8 12. LLEFT FEMUR RSB 386 5.6 10840.09E 01 300 40.6R 13. LLNG RSK 380 0.8 LPS. 1.0840.09E 01 300 40.6R 14. LUNG RSK 380 0.2 025. 2.9440.91E-01 400 34.4 15. AACHEA RST 187 3.8 025. 2.9441.18E-01 400 34.4 16. G. 1. TAACHEA RST 187 3.8 025. 2.9441.18E-01 400 34.4 17. TAACHEA RST 187 3.8 025. 2.9441.18E-01 400 34.4 18. G. 1. TAACT RSS 670 13.7 LBS. 1.6540.03E 02 800 08.6R 19. U10 V. VICOSA RSW 381 2.9 025. R.OSA11.19E-01 1000 16.9R	43	1111	***	2 1 2		60	.3040.046	9	55.0		7 4
UAINE   RSU 288   5.0   185.   3.30 +0.06E 02   300   12.44   19	20	: :		4 110	-		.1640.16E 0	100			# ~
UNINE RSU 268- 5.0 LBS. 5.5C#C.10E 02 200 29.2R 19 UNINE RSU 349 2.2 LBS. 1.24+0.05E 02 30 65.8 UNINE RSU 241 2.6 LBS. 1.24+0.05E 02 30.4 FECES RSF 501 0.2 LBS. 4.45#0.14E 01 200 30.4 FECES RSF 491 0.4 LBS. 1.18#0.03E 02 200 17.4 2 KJONEY RSK 191 3.3 DZS. 2.34+0.91E-01 400 68.8 4 LUVG RSK 191 3.3 DZS. 2.34+0.91E-01 400 80.5 4 LUVG RSK 380 0.8 LBS. 1.08#0.06E 01 300 40.6R 5 HILAA NDDE RSH 185 0.2 DZS. 0.0C#0.61E 00 40 23.2 7 TAACHEA RST 187 3.8 DZS. 2.94+1.18E-01 400 34.4 8 G. 1. TAACH RSS 570 13.7 LBS. 1.65*00.03E 02 800 08.6R 10 V. WUJCHSA RSW 381 2.9 DZS. R.OS#1.19E-01 1000 16.9R		:	UA INE	SU 2	'n	جد :	.30+0.06E	300			7
UNINE RSU 349 2.2 LBS. L.2400.05E 02 30 65.8 2.8 UNINE RSU 241 2.6 L0S. 2.8940.09E 03 50 19.2R 22 LBS		:	US 1NE	2.05	÷		.5C+0.10E 0	200			7
UNINE RSU 241 2.6 LOS. 2.8940.09E G3 50 19.2M 22 FECES RSF 501 0.2 LOS. 4.4540.14E 01 200 30.4 FECES RSF 491 0.4 LBS. 1.1840.03E 02 200 17.4 17.4 -1 LEFT FEMUR RSB 386 5.6 DZS. 1.6540.51E-01 400 68.8 2 KJONEY RSK 191 3.3 DZS. 2.3440.91E-01 400 57.4 3 LIVER RSL 376 1.2 LBS. 6.1640.00E-01 400 80.5 4 LUVG RSK 380 0.8 LBS. 1.0840.06E 01 300 40.6R 5 HILAR NODE RSH 185 0.2 DZS. 0.0040.61E 00 40 23.2 7 TRACHEA RST 187 3.8 DZS. 2.9441.18E-01 400 34.4 8 G. 1. TRACH RSS 670 13.7 LBS. 1.6540.03E 02 800 08.6R 10 V. WUCOSA RSN 381 2.9 DZS. R.0541.19E-01 1000 16.9R		•	03.1%	50.3	۲.	45	.2440.05E 0	30			_
FECES RSF 501 0.2 LOS. 4.45#0.14E 01 200 30.4 10		:	57170	2 ns	~	400	.89 to.09 E C	20			~
FECES RSF 491 0.4 LBS. 1.18#0.03E 02 200 17.4  - 1 LEFT FEMUR RSB 386 5.6 02S. 1.65#0.51E-01 400 68.8  2 KJONEY RSK 191 3.3 0ZS. 2.34#0.91E-01 400 57.4  3 LIVER RSL 376 1.2 LBS. 6.16#0.80E-01 400 80.5  4 LUVG RSK 380 0.8 LPS. 1.08#0.06E 01 300 40.6R  5 HILAR NODE RSH 185 0.2 0ZS. 0.0C#0.61E 00 40 23.2  7 74ACHEA RST 187 3.8 0ZS. 2.94#1.18E-01 400 34.4  8 G. 1. 74ACT RSS 670 13.7 LBS. 1.65#0.03E 02 800 08.6R  10 V. WUZUSA RSW 381 2.9 0ZS. R.05#1.19E-01 1000 16.9R		:	FECES	SF 5	Ċ	105.	.45 #0.14E 0	200			Z Z
- I LEFT FEMUR RSB 386 5.6 025. 1.65#0.51E-01 400 68 2 KIDNEY RSK 191 3.3 DZS. 2.34#0.91E-01 400 87 3 LIVER RSL 376 1.2 LBS. 6.15#0.80E-01 400 80 4 LUVG RSK 380 0.8 LBS. 1.08#0.06E 01 300 40 5 HILAR NODE RSH 185 0.2 DZS. 0.0C#0.61E 00 40 23 7 TRACHEA RST 187 3.8 0ZS. 2.94#1.18E-01 400 34 8 G. 1. TMACT RSS 670 13.7 LBS. 1.65#0.03E 02 800 08 10 V. WICHSA RSH 381 2.9 DZS. 8.0S#1.19E-01 1000 16		:	FECES	SF 4	ċ	185.	.1810.036 0	200	•		x ~
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LIVE? RSL 376 1.2 LBS. 6.16+0.AUE-01 400 80 LUVG RSK 380 0.8 LMS. 1.08+0.06E 01 300 40 HILA? NODE RSH 185 0.2 0.2. 0.0G+0.61E 00 40 23 TAACHEA RST 187 3.8 0.25. 2.94+1.18E-01 400 34 G. 1. TAACT RSS 670 13.7 LBS. 1.65+0.03E 0.2 800 08 V. WUCOSA RSW 381 2.9 0.25. 8.05+1.19E-01 1000 16		•		SK 19	m	. \$ 70	.34 t0.91 E-0	400			
LUVG HILAR NODE HSH 185 0.2 0.25. 0.00 +0.06 E 01 300 +0 HILAR NODE HSH 185 0.2 0.25. 0.00 +0.05 E 00 TAACHEA AST 187 3.8 0.25. 2.9 + 1.18 E -01 +0.0 3 + G. 1. TAACT RSS 670 13.7 LBS. 1.65 +0.03 E 0.2 800 08 Y. WICHSA RSW 381 2.9 0.25. 8.05 +1.19 E -01 1000 16		:		SL 37	-	.05.	.15 tO.AUE-0	400			
HILAN NODE KSH 185 0.2 0.25. 0.00 to .61E 00 40 23 TAACHEA RST 187 3.8 0.25. 2.94 to 18E-01 400 34 G. 1. TAACT RSS 670 13.7 LBS. 1.65 to .03E 0.2 800 08 Y. WICHSA RSW 381 2.9 0.25. 8.05 to .19E-01 1000 16		:		SK 38	ċ	L P. S.	.08 to.05E 0	300	40.64		
TAACHEA RST 187 3.8 025. 2.9441.18E-01 400 34 G. 1. TAACT RSS 670 13.7 LBS. 1.6540.03E U2 800 08 Y. WICHSA RSW 381 2.9 025. R.OS41.19E-01 1000 16		:		84 FS	ċ	÷	.0C +0.61E 0	9	23.5		
G. 1. PA4CT RSS 670 13.7 LBS. 1.65+0.03E UZ 800 08 Y. WILDSA RSH 381 2.9 025. 8.05+1.19E-01 1000 16		:		ST 18	<u>.</u>	025.	.941.186-0	400	34.4		
4. WITOSA RSH 381 2.9 025. R.OS+1.19E-01 1000 16		•	-	55 67	13.	105.	0 360.0489.	800	89.80		
		:	÷	SN 38	~	.\$70	.0541.19E-0	1000	16.9R		

TABLE E.8	E. 8	1						
AVINA AUS.	SAMPL		WET WEIGHT	PU 239, 240 ACTIVITY (DPM)	COUNT TIME	VIELD (R.RE- KORK)	URAN IUN IN ICRO GRANS)	REMARKS
2119 -	1	# 1	! }	1	99	0.64	0 4 4 5 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	9 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
:::	S LIVES 4 LUNG 5 HILAR NODE	* * * * * * * * * * * * * * * * * * *	14.5	5.84#0.37E 1.04#0.52E	1000	144 16.1 54.1		
2124 -	1 LEFT FEHUR 2 KIDNEY 3 LIVER	JK R SB 569 R SK 515 R SL 506	5.3 025. 3.8 025. 1.6 LRS.	. 1.15*2.31E-0; . 0.00*0.22E 00 . 1.16*0.16E 00	9 4 0 0 0 0	40.9 63.2		
::	4 LUNG HODE	R SR S H	1.1	1.40 \$0.22E 7.20 \$0.72E	009 900	23.6R 71.1		
240	1 LEFT FEHUR 2 KIDNEY 3 LIVER 4 LUNG	R SS 416 R SK 445 R SL 417 R SR 478	5.8 025. 3.6 025. 1.6 LBS.	8.3046.70E-02 1.0742.13E-01 1.7540.17E 00 3.6040.13E 01	400 40 1000 600	28 5 4 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	2.31 C.S. 11 0.423 C.S. 6.5. 0.316 C.S.	11 2 PIECES ONLY
2128	S HILAR NOJE LEFT FEHUR Z KIONEY 3 LEVER 4 LUNG S HILAR NODE	A A A A A A A A A A A A A A A A A A A	5.2 3.2 1.1 15.7		600 500 400 400	58 -0 115 -0 12 -0 13 -0 14 -0 15 -0	2 9 2 -	
2129	1 LEFT FEHUR 2 KIONEY 3 LIVER 4 LUNG 5 HILAR NOSE	R R S S S S S S S S S S S S S S S S S S	6.3 325. 6.9 025. 1.9 tes. 1.1 tes. 0.3 025.	0.00*0.22£ 00 0.98*1.14£-01 9.47*4.21E-01 9.20*1.20£-01	100 100 100 300 500	24.5 34.7 26.98 61.68		

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7	SAYP	TLW NO.	WET WEIGHT	PU 239, 240 ACTIVITY (BPH)	COUNT TIME	Y 1 EL D (R *RE- WOW)	URAN1UM (M fCRO URANS)	REMARI
7	1 LEFT FEHUR 2 KIDNEY 3 LIVER 4 LUNG 5 HILAR NOJE 7 TAACHEA 6 G. I. TRACT 0 N. HUCOSA 1 LEFT FEHUR 2 KIDNEY 3 LIVER	R SB 601 R SC 575 R SC 575 R SC 575 R SC 540 R SC 568 R SC 568	3.2 025. 16.1 025. 16.1 025. 11.9 025. 3.0 025. 3.0 025. 4.9 025. 4.9 025.	0.0040.28E 00 1.7543.50E-01 2.6340.37E-01 3.7240.11E 01 7.9045.20E-02 1.6740.06E 01 5.3040.38E 01 5.9740.40E 00 4.2640.17E 00 4.2640.17E 00 4.2640.19E-01	100 100 100 100 100 100 100 100 100 100	6 4 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	0.297	
	S HILA1 NODE UNINE UNINE UNINE UNINE UNINE VAINE				40 20 300 300 900 900	25 22 4 4 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2		16 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4
2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	LIVER JUNE UNINE UNINE UNINE FECES	RSE 356 RSL 345 RSU 468 RSU 32 RSU 404 RSU 29 RSF 537	6.3 025. 1.3 L85. 1.4 L85. 3.6 L85. 2.9 L85. 1.0 L85. 1.0 L85.		600 400 200 200 1000 90	44.04 W W W W W W W W W W W W W W W W W W W		C.S. 11 C.S. 11 12 HAY 22 JUNE 22 JUNE 12 HAY

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REMARKS		IEO RY ELIMINATION TAGGED •VOMÍTED•	•
URANIUM IMICRO GRAMS)	0.342	0.196 10ENT IF I E0 7AG	
Y 1 EL D ( R = R E - MORK )	25.6 64.0 64.2 57.1 57.1 59.3 59.3 10.8 10.8 10.8 10.8	0.00.00.00.00.00.00.00.00.00.00.00.00.0	) (
	1,000 800 800 800 1,000 1,000 1,000 1,000	100 500 500 1000 1000 1000 1000 1000 10	
PU 239, 240 ACTIVITY (DPH)	2.15#0.14E 00 0.00#0.72E 00 5.53#0.72E-01 6.90#0.95£-01 1.24#2.48E 31 0.00#0.62E 00 6.60#5.60E-02 2.11#0.53E-01 5.88#0.19E 01 8.40#3.40E-02 7.65#0.35E 00 1.36#0.02E 02	1.69#2.53E-01 1.32#2.63E-01 5.80#1.20E-01 5.19#0.27E 00 0.00#0.14E 00 0.00#0.16E 01 4.62#1.76E-01 7.68#0.26E-01 7.19#0.25E 01 6.05#0.25E 01 6.05#0.25E 01 6.05#0.25E 01 7.19#0.25E 01 7.19#0.25E 01 6.05#0.12E 01 7.75#0.12E 01	
WET WEIGHT	5.2 025. 1.2 L65. 1.2 025. 0.2 025. 5.9 025. 3.0 025. 1.0 L85. 1.0 L85. 0.2 025. 3.6 L85. 3.6 L85.	5.1 025. 1.3 185. 16.2 025. 0.2 025. 13.0 185. 3.0 025. 1.3 185. 1.3 185. 1.3 185. 1.4 185. 1.4 185. 1.5 2 025.	•
TLK NO.	RSB 580 RSK 512 RSL 505 RSR 565 RSA 391 RSL 374 RSL 374 RSL 374 RST 189 RST 189 RST 189	R S S R S S R S S R S R S R S R S R S R	•
SAMPLE TYPE	LEFT FEMUR KIDNEY LIVES LUNG HILAS NOJE HILAS NOJE LIVES LIVES LIVES HILAS NOJE HILAS NOJE HILAS NOJE HILAS NOJE HILAS NOJE HILAS NOJE HILAS NOJE HILAS NOJE HILAS NOJE	LEFT FEMUR KIONEY LIVEN LUNG HILAN NODE TACHEA G. 1. TACT N. MUCOSA LUNG LIVEN LIVEN TACHEA TACHEA G. 1. TRACT	•
- CX	22.79 - 2.		

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SAMPLE TLW WET PU 239, 240 COUNT YIELD URANIUM TYPE NO. WEIGHT ACTIVITY TIME (R=KE- (MICRO ICPM) MOR4] CRAMS URINE RSU 265 4.6 LBS. 3.60+0.13E 01 500 14.2R DATE URINE RSU 535 1.5 LBS. 1.28+0.03E 04 20 62.8 URINE RSU 465 2-1 LBS. 4.43+0.10E 03 30 56.1 URINE RSU 463 5.8 LBS. 1.04+0.03E 03 20 10.1R FECES RSF 464 0.7 LBS. 5.54+0.16E 01 300 27.7R FECES RSF 550 2.7 LBS. 3.19+0.04E 03 200 21.0R FECES RSF 550 2.7 LBS. 1.00+0.02E 02 200 23.0R	TABLE	E.6	UED)					
URINE RSU 265 4.6 LBS. 3.60+0.13E 01 500 14.2R DATE URINE RSU 535 1.5 LBS. 1.28+0.03E 04 20 62.8 LRSU 4.5 2.1 LBS. 4.43+0.10E 03 30 56.1 RSU 403 5.8 LBS. 1.04+0.03E 03 20 10.1R RSU 403 5.8 LBS. 1.04+0.03E 03 20 10.1R RSU 30 4.5 LBS. 6.62+0.19E 01 700 34.°° FECES RSF 464 0.7 LBS. 5.54+0.16E 01 300 27.7R FECES RSF 551 1.1 LBS. 3.19+0.04E 03 200 21.0R FECES RSF 540 5.8 LBS. 1.00+0.02E 02 200 23.8R	AVIVAL NO.	1 1	TLW NO.	WEIGHT	PU 239, 240 ACTIVITY (CPH)	COUNT	YIRD (R=RE-	
RSU 535 1.5 LBS. 1.28 +0.03 E 04  20  62.6 RSU 455 2.1 LBS. 4.43 +0.10 E 03  30  56.1 RSU 403 5.8 LBS. 1.04 +0.03 E 03  20 10.1R RSU 30 4.5 LBS. 6.62 +0.19 E 01 700 34.7 RSF 464 0.7 LBS. 5.54 +0.16 E 01 300 27.7R RSF 550 2.7 LBS. 3.19 +0.04 E 03 200 23.8R RSF 550 2.7 LBS. 1.00 +0.02 E 02 200 23.8R	.0.	UN INE	, ~	4.6 LBS.		500	14.28	C TOTAL TOTAL STREET,
RSU 455 2-1 LBS. 4-43+0-10E 03 30 56-1 RSU 403 5-8 LBS. 1-04+0-03E 03 20 10-1R RSU 30 4-5 LBS. 6-62+0-19E 01 700 34-7 RSF 464 0-7 LBS. 5-54+0-16E 01 300 27-7R RSF 551 1-1 LBS. 3-19+0-04E 03 200 21-0R RSF 550 2-7 LBS. 1-00+0-02E 02 200 23-8R ASF 240 5-8 LBS.	:	U3 INE	RSU 535	1.5 1.85.		200	A2.8	2 - 1010 101 101 101 101 101 101 101 101
RSU 403 5.6 LBS. E.04+0.03E 03 20 10.1R RSU 30 4.5 LBS. 6.62+0.19E 01 700 34.7 RSF 464 0.7 LBS. 5.54+0.16E 01 300 27.7R RSF 551 1.1 LBS. 3.19+0.04E 03 200 21.0R RSF 550 2.7 LBS. 1.00+0.02E 02 200 23.8R ASF 240 5.8 LBS.	:	US INE	SU 4	2.1 LBS.		90	1, 40	> 4 5
RSU 30 4.5 LBS. 6.62#0.19E 01 700 34.7f RSF 464 0.7 LBS. 5.54#0.16E 01 300 27.7R RSF 551 1.1 LBS. 3.19#0.04E 03 200 21.0R RSF 550 2.7 LBS. 1.00#0.02E 02 200 23.8R ASF 240 5.8 LBS.	:	U3 1 V.	4	5.6 1.85.		2 2	10 TB	
RSF 464 0.7 LBS. 5.54#0.16£ 01 300 27.7R RSF 551 1.1 LBS. 3.19#0.04£ 03 200 21.0R RSF 550 2.7 LBS. 1.00#0.02E 02 200 23.8R RSF 240 5.8 LBS.	:	UN INE		4.5 LBS.		200	36.08	20 C C
RSF 551 1.1 LBS. 3.19#0.04£ 03 200 21.0R RSF 550 2.7 LBS. 1.00#0.02E 02 200 23.8R RSF 240 5.8 LBS.	:	FECES	RSF 464	0.7 185.		300	27 78	
ASF 550 2.7 LBS. 1.00#0.02E 02 200 23.8R ASF 240 5.8 LBS.	:	FECES	RSF 551	1.1 185.		200	20° ( 6	>4 E
S ASF 240 5.8 LBS.	:	FECES	ASF 550	2.7 185.		000	21.82	
	:	FECES	RSF 240	5.8 LBS.		2		D + 30 LOST IN DISS.

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FECES

FE

TABLE E.7 (CONTINUED)

REHARKS		FOUND 11/29/64	TAGGED X258-8
URAN IUM (MICRO GRANS)			
Y 13.0 (R=RE- HORK)	18.1 18.6 37.6 25.0	13.00 42.00 6.00 6.00 6.00 6.00 6.00 6.00 6.00	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4
COUNT TIME	200 1000 200 1000 900	200 200 200 200	200 200 200 1000 1000 1000 200 200 200 2
PU 239, 240 ACTIVITY (DPM)	1.7040.46E 00 1.0641.50E-01 1.60+0.06E 01 1.64+0.04E 01 3.92+0.79E-01	4.72#0.49E 00 6.86#0.27E 01 2.46#0.21E 00 1.59#0.12E 00	1.59*0.22£ 00 4.29*0.13£ 01 -0.74*1.48£-01 1.39*0.14£ C0 1.97*5.10£-01 1.31*0.05£ 01 6.68*0.35£ 00 -0.70*4.20£-02 1.68*0.39£ 00 0.29*1.50£-01 2.67*0.12£ 01 7.20\$\$.00€-02
WET WEIGHT	2.3 LBS. 1.7 LBS. 8.3 LBS. 3.3 LBS. 1.0 02S.	2.0 LBS. 5.2 LBS. 0.4 02S. 2.2 LBS.	185. 185. 185. 185. 185. 185. 185. 185.
TLW ND.	RBB 120 RBK 10 RBL 17 RBR 459 RBH 3	RBK 41 RBL 103 RB4 46 RB8 225	
AVIHAL SAMPLE ND. TYPE	LEFT FEMUR KIDNEY LIVER LUNG HILAR NOJE		TACHEA G. I. TRACT P. MUSOSA LEFT FEHUR KIDNEY LUNG HILAR NODE LEFT FEHUR KIONEY LIVER LUNG
AVIMAL ND.	3023	1 2029	

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3039 - 1 LEFT FEMUR 1 2 KIONEY 2 KIONEY 3 LIVER 1 5 HILAR NOJE 1 6 A 1GHT FEMUR 2 KIONEY 3041 - 1 LEFT FEMUR 3041 - 1 LEFT FEMUR 3043 - 1 LEFT FEMUR 3043 - 1 LEFT FEMUR 3 LIVER 3 LIVER 3 LIVER 3 LIVER 3 LIVER 4 LUNG 1 5 HILAR NOJE 3 LIVER 3 LIVER 4 LUNG 1 5 HILAR NOJE 3 LIVER 4 LUNG 1 6 HILAR NOJE 3 LIVER 5 HILAR NOJE 6 HILAR NOJE 7 A LUNG 7 A LUNG 7 A LUNG 8 HILAR NOJE	EMUR RBB RBC NOJE RBH FEMUR RBB TAACT RBS TRACT RBS TRACT RBS TRACT RBS TRACT RBS	2	MET MET 1.9 LBS 2.0 LBS 2.5 LBS 2.5 LBS 2.5 CS 5.5	GHT LBS. LBS. LBS.	PU 239, 240 ACTIVITY (DPH)	COUNT	Y 1 EL O ( R *R E-	UR AN TUM	REMARK S
- 1 - 1 - 1 - 1 - 2 - 2 - 2 - 2 - 2 - 2	* * * + + + * * * * * * * * * * * * * *	.   4 6 6 7 7 7 7 8 7 8 8 8 8 8 8 8 8 8 8 8 8	7	# 88.5 85.5 85.5 85.5	ACTIVITY (DPM)	T ME	XXX		
L LEFT L SALINGS	t } 1	SHEN BRANCHORN		8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		,	MORK )	GRAMS 1	
1	ec	SEEN BAAGANGE	0 8 40 0 6 6 6 6	85. 85.	.85#3.14E-0	200	22.6		
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	<b>α ⊢⊢</b>	SHEN BAFBANGE		.85.	0.44*1.10E-01	200	53.7		
	cc +-	SHEN BARBAND		.85.	.52#0.07E 0	200	44.2		
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	æ <b>⊢ ⊢</b>	SEEN UPFS	NO @ @ NO		.28 #0.06E 0	300	19.4	0.610	
1 LEFT A LUNG A	<b>K</b> ++	SHEN BAFF	0 6 6 6 6 0	. 57	.91 *0.50 E	1000	59.3		
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	<b></b>	SHEN BARS	995	.85.	8.33#2.48E-01	1000	12.6		
1	<b></b>	SHEN BAR	<b>8</b> N 40	.85.				FOUND	Ë
1 LEFT A LUNG A	<b>-</b>	3 WHW 5	6.5	.88.	3.62#0.11E 01	300	32.3		NUMBER DUPLICATED
- 1 LEFT A HILAR - 1 LUNGA - 1		2 2 2 2	•	.BS.	0	200	29.3		LICATE
		2 1 1 2		.520	.93 *1.30 E-	300	51.0		
1 I I I I I I I I I I I I I I I I I I I		2 1	0	A.	99#0.36F 0	200	26.1		
		2 10		85.	.34#0.17E 0	300	46.7		
		5		85.	.09+0.14E	200	18.2		
1 1 1 1 1 1 mm+m4t			2.7 L	LBS.		300	39.1		
1 1 managa		-	m	.57(	.00*6.00E-0	1000	4.19		
1 Nutu	88	294	5	85.	1.68+0.15E 00	300	84.8		
1 W45 20045041	88 X	258	~	.85	4.00#0.60E-01	500	59.9		TAGGED X258-2
1 WWAWAL	RB L	301	5.1 L	LBS.	4.60#0.226 01	1000	04.7A		
1 N 2004840		581	4	.85	2.53#0.08E 03	300	18.9		
1 ~~~~~~~	RB +	233.	<b>E</b>	.57(	5.50#1.20E-01	400	47.0		
14 m 4 m 4 h		208	<b>@</b>	.8S.	1.70#0.14E 00	004	68.1		
W 4 NV 40 L	88 K	96	8	LBS.	1.00 #0.75E-01	004	50.3		
4 10 4 5		106	•	.85.	9.43+0.34E 01	900	10.4		
w 4 r		495	•	.85.	2.47#0.05E 02	200	25.18	0.344	
9 1		æ	~	125.	7.60 \$7.60 E-02	200	62.4		
•	883	252	3.1 L	LBS.	1.85 to . 18E 00	200	15.4		
•		9.6	0	.85.	7.85*0.33E OL	200	42.1	•	
Ġ	AB S	647	•	.85.	9.50 to.22E 01	200	46.2R		
** 10 4. MUCOSA	R8 N	596	8	.570	1.13#0.08E 01	1000	08.2R		
3053 - 1 LEFT FEHUR	TUR RBB	611	3.2 L	185.	1.71+0.186 00	1000	16.9R		

TABLE E.7 (CONTINUED)

D URANIUM REMARKS E- (MICRO ) GRANSI	R RENJAKED THICE	ec ec ec	TAGGED X258-4	ææ	σ≼
Y 18.0 (A=RE- HORC.)	61.9 25.5 02.08 25.1	04000	28.7 12.0 67.5 28.9	65.4 62.9 09.1R 16.1R 40.8	19.1 04.58 50.5 7.6.7
CDUNT T IME	400 700 1000 200 600	200 500 500 700 700 700 700	400 1000 900 200	4 4 5 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	200 200 400
PU 239, 240 ACTIVITY (DPM)	2.01#0.16E 00 3.49#0.73E-01 7.49#0.50E 01 2.94#0.09E 03 4.77#0.52E-01		1.35*0.07E 01 5.38*0.37E 00 5.59*2.80E-02 5.13*0.20E 01	1.46#0.12E 00 8.64#0.34E 00 1.26#0.09E 01 5.98#0.48E 00	1.60+0.29E 00 5.91+0.31E 01 9.22+0.51E 00 4.00#5.00E-02
WET WEIGHT	2.0 LBS. 2.0 LBS. 8.5 LBS. 3.6 LBS. 0.5 OZS.		5.0 t85. 3.1 t85. 0.9 025. 3.3 t85.	2.7 LBS. 1.6 LBS. 8.6 LBS. 3.6 LBS.	1.3 L8S. 5.1 L9S. 3.7 L9S. 0.3 QZS.
TLW NO.	RB 263 RB 131 RBL 127 RBR 533 RBH 126	8×-4× 8×	RBL 253 RBR 602 RBH 228 RBH 582	RBB 322 RBK 256 RBL 222 RBR 560 RBH 229	RBK 254 781 251 RB4 585 784 231
SAMPLE TYPE	1 LEFT FENUR 2 KIDNEY 3 LIVER 4 LUNG 5 HILAR NODE		3 L1VER 4 LUVG 5 HILA1 NODE 4 LUNG	1 LEFT FEHUR 2 KIDNEY 3 LIVER 4 LUNG 5 HILAR NODE	Z KIDVEY 3 LIVER 4 LUVG 5 HILAR 1006
AVINAL NO.	3055 - 1	3074 - 23	3 000	102	3103 - 201E

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TABLE E.7 (CONTINUED)

¥I NA CV	SAMPLE TYPE	⊢ Z	WE IGHT	<u>=</u>	PU 239, 240 ACTIVITY (DPM)	COUNT TIME	Y 18.0 (R #RE- NOR()	URANIUM (HICRO GRAMS)	REHARK S
3105 - 12	LEFT FEHUR KIDNEY LIYER LUNG HILAR NODE	R8B 261 R8K 128 R8L 195 RER 527 RBH 123	2.5 L8 1.9 L8 4.8 L8	185. 1 185. 2 185. 2 185. 6	1.67*0.22E 00 2.50*1.10E-01 3.40*0.13E 01 6.20*0.23E 01 1.90*7.60E-02	200 400 200 200 200	46.0 43.9 10.58 28.9 62.0		
3108 - 1 11 2 11 3 3109 - 3	LEFT FEMUR KIONEY LIVER HILAR NODE	R88 352 R8K 255 R8L 289 R8H 232	2.8 105 1.3 185 7.4 185 0.6 025 5.0 185		2.55+0.22E 00 4.42+0.49E-01 8.18+0.21E 01 1.92+0.44E-01 5.06+0.16E 01	400 300 400 1000	37.6 75.1R 42.8R 77.8	TAGGED LIVER.	IVER, BUT IS LUNG
m 1111 1111 1111 1111 1111 1111 1111 1	LEFT FEMUR KIONEY LIVEY LUNG HILAN NODE AIGHT FEMUR TACHEA G. I. TAACT P. MUCGSA	RBB 221 RBK 42 RBL 101 RBR 493 RBB 259 RB 599 RBS 656 RBP 44	3.00 LB	LBS: LBS: LBS: LBS: LBS: LBS: LBS: LBS:	1.94+0.13E 00 3.11+1.12E-01 5.96+0.25E 01 5.28+0.14E 01 0.84+1.18E-01 1.75+0.17E 00 1.74+0.06E 01 2.78+0.10E 01 1.00+1.30E-01	800 2000 2000 2000 2000 2000	44.45 44	0.389	
1 1			~ 0 & 9 i 8		6.01+3.34E-01 9.81+2.90E-01 3.55+0.09E 01 1.84+0.06E 01 4.10+8.00E-02	1000 400 1000 400	117.7 06.9 21.5 55.9 67.9		
* * * * * * * * * * * * * * * * * * *	3 : 1 VE 4 4 LUVG 5 HILA? NOJE 0ATA THIS REPOR	RBL 293 КВR 600 КВН 235	3.2 LA 0.9 Q2	LAS. 025.	5.64+0.23E 01 1.10+0.03E 02 0.00+0.16E 00	1000 300 60	93.48 59.6 59.6	2	TAGGED X258-4

REMARKS																					NUMBER DUPLICATED	A DUPLICATE								
URANIUM (MICRO GRANS)																														
YIBLD (R=RE- WORK)	50.3	50 .7	29.4	21.0R	10.6R	67.6	66.3	6.44	26.46	30.05	72.3	06.0R	45.7	_	10.2	$\overline{}$	58.5	~	œ	36.2	M	~	~	-	-	0	9	4.19	•	•
COUNT 7 1ME	200	400	700	1000	1000	200	500	904	9 6	000	400	1000	200	1000	1000	300	1000	200	200	200	200	900	200	1000	200	200	300	200	000	?
PU 239, 240 ACTIVITY (DPM)	1.62#0.14E 00	.84#0.17E	.51 #0.28 E	.69#0.17E	.61#0.49E	7.68#1.19E-01	.89±0.21F 0	. 48±0.11F	. 40 ±0 - 17 F 0	42#0.22F 0	2.1844.356-02	1.05*0.100 00	.08#0.17E	.09#0.14E	.53#0.10E	.35#0.05E	8.7043.00E-02	.79#1.17E-	.35 to.16E	.36 +0.33E	.89+1.56E-	.6940.15E	.03#0.36E	.06+0.14E	.58 to. 13£	.81#0.10E	.10#3.80E-	2.01#0.06E 01	A140.27F	3.7.
ET I GH	2.2 LBS.	~	.9 185	.4 LBS	.8 1.85	.220 4.0	4	4	, Y		0.4 025.	8.3 025.	~	م	0	~	0.3 025.	~	•			0	2 18	9 LES	4 1.85	3 LAS	570 5	2.6 1.65.	2 0	2
	RBB 194	26	12	18	55	RBH 124	36	, ,	7	י ה ס	RBH 230	RBN 295	15	~	σ	48	RBH 25	18	σ	54	m	~	-			4		RBB 179	t	
YPE.	1 LEFT FEAUR	1 LEFT FEMUR			9N0 1	5 HILAN NODE	1 FET FERID			1 LIVE	S HILAR NOJE	O N. MUCOSA	1 LEFT FEMUR	2 KIDNEY		9101 4	5 HILAR NODE	•		ئ	ď	9 P. MUCOSA	LEFT FEHUR							
1 <del>2</del> 2	3129 -	1		:	•	:	3166 -				•	01- 951E 250	3147 -	:	:	:	:	:	:	:	•	•	3146 - 1	:	:	:		:	-	•

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TABLE E.8 ESTIMATED ACTIVITY EXPENDITURE OF PROJECT 2.6c "A" SAMPLES IN PARTICLE ANALYSIS Doubletrack Clean Slate 1 Event 2,45E 03 - 3,17E 03 1,09E 02 - 2,29E 02 3. 93E 02 - 6. 21E 02 3.89E 02- 1.26E 03 2. 74E 02 -2. 16E 03 2, 18E 02 1,43± 0,03E 01 0,00E 00 1.39 ± 0.15E 00 Activity (DPM) Pu-239, 240 7.88E 01 2. 72E 02 1.28E 03 0, 00E 00 5.57E 01 O. 00E 00 2.01E 01 2. 28E 02 1.74E 02 4. 90E 02 8.91E 01 4.05E 02 6.16E 02 0.000.0 2. 50E 01 \* 34 Analysis No. CCD-2160 CAD-2164 CCD-2170 CCD-2165 CTA-2173 CAD-2162 -2168 -2175 CCD-2169 CCD-2172 CCD-2163 CCD-2180 CTA -2171 2161 CTA-2174 CAD-2166 CCD-2177 CCD-2157 CTA-2178 2179 CCD-2181 CTA-2176 -2167 CTA-2158 TLW Collection No. 3013A 3038A 2922A 9699A 9661A 9656A 9691A 9668A 2946A 2934A 2907A 2526A 2920A 9624A 9698A 9660A 2723A **96694** 2812A 2837A 2443A 2151A 3466A 2482A 3449A L25, P9 (2) L29, P9(2) L19, P9 (2) L8, P21(2) L7, P9(11) L6, P13 L18, P21 LS, P17 058(1) 068 058(2) Location 890 058 058 058 190 090 650 058 190 190 054 082 B 8a1 ARC Bal 881 ひとませいけいます

(1) Data determined by precision counting. Values without an error assignment determined by 2 Tr counting. (2) A range value indicates an unknown fraction of the sample has been removed.

TABLE E.8 (CONTINUED)

ARC	Location	71.W	TI.W	Pu-239, 240	
		Collection No.	Analysis No.	Activity (DPM)	Lvent
Na Na	01	4082A	CTA-2194	5, 91501	Clean Slate II
18	P2(2)	2305A	CCD-2184	3.475 01 - 2.44502	•
<	036(1)	4116A	CTA-2195	5.53 ± 0, 19E 01	•
<	054(1)	2286A	CCD-2183	5.41 + 0. 19501	•
. 62	044(1)	2371A	-2189	4.47 ± 0.21E 00	1
<b>6</b>	054(1)	4812A	CTA-2197	1.26 + 0.06E 01	•
<b>•</b>	060(1)	2370A	CCD-2188	8.71 ± 0.30E 00	•
•	(1)890	2369A	-2187	6, 00 ± 0, 195 01	•
Ω	030	41634	CIA-2196	00 300 0	
۵	034	3182A	CAD-2190	0.000.00	•
Bal	L1, P17	4022A	CTA-2192	0, 00% 00	•
Bel	L3, P9(1)	2312A	CCD-2185	5.67 ± 0, 12E 01	•
Bal	1.4, P21	4024A	CTA-2193	0° 00E 00	•
Bal	L7, P9	4011A	-2191	00 300 0	•
8 8a?	L1, P1	2366A	CCD-2186	6.29 ± 0.34E 00	•
Mob	DP-12	2272A	CCD-2182	7.21 ± 0.36E 00	•
BX	90	4987A	CAD-2201	0.000.00	Clean Slate III
×	10	5184A	CTA-2203	0.0000	•
BM	(1)01	4973A	CCD-2199	1. 26 ± 0. 03E 00	•
<	030	4974A	CAD- 2200	0.00E 00	•
: <b>«</b>	102(1)	4964A	-2198	3.45 ± 0.13E 01	*
<	108	\$162A	CTA-2202	0.000	

TABLE E.9 PLUTONIUM AND URANIUM ANALYBES OF ROLLER COASTER DISTILLED WATER SAMPLES (1)

			\$ 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	***************************************	
R. C. Sample No.	K-030-3133	K-030-3134	L-018-3128	1-030-3129	1-642-3127
		0. 700	0.825		0, 750
Sample pH	5.1	5.3		5.3	5.0
Centrif. Supernate	1.34x107	2.80x10°	0.00	4.83x10 <sup>3</sup>	4.50×101
Leach Operation (2)	•	•		•	
Millipore Filtrate	1.17 ± 0.02×194	1.99 ± 0.04x103		$4.45 \pm 0.14 \times 10^{3}$	
173 min. Leach Fill.	2. 76×104	6.72x103		5.05x104	
400 min. Leach Fil.	5.70×10 <sup>3</sup>	3.92x103		1.88×104	
968 min. Leach Fill.	5.70×10 <sup>3</sup>	3.08×10 <sup>3</sup>		1.49×104	
1273 min. Leach Fil.	1.20x10 <sup>3</sup>	1.12×103		7, 26×10 <sup>3</sup>	
2896 min. Leach Fill.	5.40×10 <sup>3</sup>	2.24x10 <sup>3</sup>		2, 15×104	
Extraction (3)	2.8 ± 0.06x104	7.87 ± 0.15x103		4.03 + 0.10x104	
Water plus Crud			$1.05 \pm 0.04 \times 10^{2}$		6.98 ± 0.42x102
Bottle Wash(4)			9.43 ± 0.94x100		2.35 ± 0.05x101
Uranium (Mill. Filtrate) (5) 0.593	5) 0, 593	9.07		1.40	0.79\$
Uranium (Bot. Leach)(5)			0.965		1.04
Uranium (Water + Crud)(5)	<b>?</b>		1.08		1.30
•					
		i			

(1) All Pu values are given as dpm/tot, sample vol. Values without an error assignment are stippled samples.

Separate aliquot auccessively leached with 0.1N HCl volumes of 25 ml with intermittent attrring, and filtering. (2)

<sup>(3)</sup> Extraction performed on separate aliquot at listed sample pH.

<sup>(4)</sup> Bottles washed with hot HNO3-HCI and IN HNO3-HF.

All uranium values are given as ug U/total sample volume.

TABLE E.10 PLUTONIUM AND URANTIM ANALYSES OF ROLLER COASTER DISTILLED WATER SAMPLES (1) CLEAN BLATE II

			• • • • • • • • • • • • • • • • • • • •	, , , , , , , , , , , , , , , , , , ,	•
R.C. Sample No. D-0	D-010-4175	D-018-4176	D-026-4177	D-034-4178	D-042-4179
Sample Vol. (Lit.)	0.225	0.450	0.450	0.250	0.450
Sample pH	5,3	5.3		5.3	,
Centrif. Supernate	1.55×10 <sup>3</sup>	7.47×10 <sup>3</sup>	1.02×104	4.43×104***	2.97×10 <sup>3</sup>
Total Sample Act (2)	•	9.28×105	1.32×10 <sup>6</sup>	5.52×10 <sup>5</sup>	2.97×10 <sup>5</sup>
Millipore Filtrate	$3.58 \pm 0.09 \times 10^{2}$	60	$8.29 \pm .21 \times 10^3$	3.94 ± 0.12×104	2.93 * . lix10 <sup>3</sup>
% of Tot. Samp.			0.07	7,14	66.0
173 min. Leach Fil.	1. 7.36×10 <sup>3</sup>		1.30×10 <sup>5</sup>	2.93×104	5. 36×107
% of Tot. Samp.	Act.		9.85	5,31	. B. D
400 min. Leach Fi	1. 1.64×10 <sup>3</sup>		5.18×10 <sup>4</sup>	9.10×10 <sup>3</sup>	1.73×107
% of Tot. Samp. Act.	Act.		3.93	1,65	
968 min. Leach Fil.	1. 1.10×10 <sup>3</sup>		6.90×104	1.07×104	1.44×10*
% of Tot. Samp. Act.	ť		5. 25	1.94	4,85
-	711. 5.94×10 <sup>2</sup>		2.94×104	5.4x10 <sup>3</sup>	7.45×10³,
g % of Tot. Samp. Act.	Act.		2, 23	0.98	2.51
77	Fil. 1.12×10 <sup>3</sup>	5.45×104	9.33×104	2. l×104	1.63x104
% of 'fot. Samp. Act.	Act.	5.87	7.07	3.87	5.49
Millipore Filter		$5.29 \pm 0.14 \times 10^{5}$	9.38 ± C.25x105	$4.38 \pm 0.13 \times 10^{5}$	1.86 ± 0.05x10 <sup>3</sup>
% of Tot. Samp. Act.	Act.		11.11		
Extraction (3)	$7.47 \pm 0.17 \times 10^3$		$6.34 \pm 0.35 \times 10^{5}$		1.84 ± 0.04x105
% of Tot. Samp. Act.	:	56.7	49.0	29.8	
Uranium (Mill. Filtrato) (4) 0. 5	1te) <sup>(4)</sup> 0.549		43.7	112.	68.8
Uranium (Mill. Filter) (4)	(+)	396.	890	137. 2	219.
Uranium (Ext. Aliq.)(4)	€			27.4	

(1) All Pu values are given as dpm/tot, sample vol. Values (other than tot, sample act.) without an error assignment Separate aliquot successively leached with 0, IN HCl volumes of 25 mi with intermittent stirring, and filtering. Tot. sample act. is sum of millipore filtrate and filter, and all leaches. are stippled samples. (2)

(3) Extraction performed on separate aliquot at listed sample pH.

(4) All uranium values are given as µg U/total sample volume.

TABLE E.11 PLUTONIUM AND URANIUM ANALYSES OF ROLLER COASTER DISTILLED WATER SAMPLES(1), CLEAN SLATE III

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R. C. Sample No.	A-012-5249	A-036-5248	A-060-5252	A-084-5251	A-108-5250
Sample Vol. (Lit.)		0.200	0.300	0.400	0.300
Sample pH	5.9	5.6	5.3	5,3	5.3
Centrif. Supernate	5.04×10 <sup>3</sup>	4.2×101	$5,49\times10^{2}$	1.78×10 <sup>3</sup>	1.91×10 <sup>3</sup>
Tot. Sample Act. (2)	2.24×10 <sup>6</sup>		•		***
Millipore Filtrate	$8.87 \pm 0.28 \times 10^{3}$	$4.46 \pm 0.19 \times 10^{4}$	$6.30 \pm 0.02 \times 10^{2}$	$1.41 \pm 0.05 \times 10^{3}$	1.51 ± 0.05×10
% of Tot. Samp. Act.	0,40			7	
173 min. Leach Fil.	1.29×10°	6.56×10 <sup>3</sup>	4. 44×10°	1.68×10*	1.10×10
% of Tot. Samp. Act.	5,16		• • • • • • • • • • • • • • • • • • •	<b>1</b>	*
400 min. Leach Fil.	5.02×10 <sup>±</sup>	2.88×10 <sup>3</sup>	1.56×10°	6.72×10°	3.96×10 <sup>3</sup>
% of Tot. Samp. Act.	2,24	,	•		
968 min. Leach Fil.	5.52×104	2.24×10 <sup>3</sup>	7. 20×10 <sup>2</sup>	5.92×10 <sup>2</sup>	2.76×10³
s & of Tot. Samp. Act.	2,46			1	
7 1273 min. Leach Fil.	3.87×164	1.28×10 <sup>3</sup>	7.20×10 <sup>c</sup>	2.72×10°	1.68x10
% of Tot. Samp. Act.	1,73	•			
2896 min. Leach Fil.	1.08×10 <sup>5</sup>	3.68×10 <sup>3</sup>	2.28×10°	7.04x10°	3.72×10³
% of Tot. Samp. Act.	4.82				
Millipore Filter	$1.85 \pm 0.05 \times 10^{6}$				
% of Tot. Samp. Act.	82.6				
Extraction(3) % of Tot. Samp. Act.	$1.47 \pm 0.04 \times 10^6$ 65.6	$7.66 \pm 0.19 \times 10^4$	3.89 ± 0.11×104	8.37 ± 0.20x104	4.98 ± 0.12×104
7)				•	
Uranium (Mill. Filtrate) '718.		10.6	11.2	15.4	10.5
Uranium (Mill. Filter)(1)	645.				

<sup>(1),</sup> All Pu values are given as dpm/tot. sample vol. Values (other than tot. sample act.) without an error assignment

Separate aliquot successively leached with 0. IN HCl volumes of 25 ml with intermittent attring, and filtering. 10t. sample act. is sum of millipore filtrate and filter, and all leaches. (2)

<sup>(3)</sup> Extraction performed on separate aliquot at listed sample pH.

All uranium values and given as ug U/total sample volume.

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CABLE E.11 (

R. C. Sample No.	ple No.	H-006-5219	H-030-5221	H-054-5224	H-078-5225	H-102-5227
Sample Vol. (Lit.)	ol. (L1t.)	0.525	0.500	0.725	0.500	0.700
Sample pH	<b>bys</b>		5.3	5.3		<b>)</b>
Centrif. Supernate Leach Operation (2	Centrif. Supernate Leach Operation (2)	<1.0x10 <sup>0</sup>	8.40×10 <sup>2</sup>	1.96×10 <sup>2</sup>	<1.0×10 <sup>0</sup>	<1.0×100
Millipore 173 min	Millipore Filtrate 173 min. Leach Fil.		$7.22 \pm 0.29 \times 10^2$	2. $12 \pm 0.13 \times 10^2$		
400 min	400 min. Leach Fil.		1.92×104			
968 min	968 min, Leach Fil.		1.32×104	1.52×10 <sup>2</sup>		
2896 min	14/3 min. Leach Fil. 2896 min. Leach Fil.		6.60×10 <sup>3</sup> 1.42×10 <sup>4</sup>	1.52×10 <sup>2</sup>		
Extraction(3)	(3)		$6.14 \pm 0.12 \times 10^4$	4. 18×10 <sup>3</sup>		
Water plus Crud Bottle Wash (4)	Crud 1h (4)	$1.93 \pm .09 \times 10^{2}$ $1.04 \pm 0.04 \times 10^{1}$			$1.96 \pm 0.13 \times 10^{3}$ 3.31 $\pm 0.07 \times 10^{1}$	3 7.19 ± 0.30×10 <sup>1</sup>
Uranium (	Uranium (Mill. Filtrate) (5)		Not Detectable	Not Detectable		
Grenium () Grenium ()	Uranium (Bot. Leach) <sup>(3)</sup> Uranium (Water plus Grud) <sup>(5)</sup> 2	0.175 (5) <sub>2.44</sub>			0.190 3.24	0.745

(1) All Pu values are given as dpm/tot, sample vol. Values without an error assignment are stippled samples.

Separate aliquot successively leached with 0. IN HCI volumes of 25 ml with intermittent stirring, and filtering. (3)

Extraction performed on separate aliquot at listed sample pH.

Bottles washed with hot HNO3-HCI and 1M HNO3-HF,

All uranium values are given as ug U/total sample volume. **E E** 

			1	•	•
Aliquot	Method	po	Total Tracer Act. (dpm/ml)	Ave. Tracer Std. Dev. Act. (dpm/ml) o'i	584. De
			Pu-236	Pu-236	
~	Isotapic Dilution	Dilution	25.57		
~	=	=	26.04		
€	=	=	26.02		
•	÷	=	25.38		
10	<b>a</b>	=	25.41		
ص	=	=	25.72		
•	2	=	26.13		
geò.	5	=	25.75		
6	=	=	25.62		
				25.74	+ 1.0%
10	Exhaustiv	Exhaustive Plating	24.92		ı
11	=	=	24.60		
12	=	=	25.37		
13	=	=	25.32		
			(Ave 13 Plates)	25.05	1.5%
			25.48 + 1.8%		
			1.026% diff.		

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INDLE E 13	13 RAE	RADIOCHEMI	CAL	VNALYSIS	OF ROLLER	COASTER	BIOLOGI	CAL ANALYSIS OF ROLLER COASTER BIOLOGICAL QUALITY CONTROL SAMPLES	TOULS	AMPLE	
APC LO	00 11 10 NO	COLLECTION NO.	_	TLW ANALYSES NO.	EVENT	TYPE	41. (22)	PU-239, 240 ACTIVITY [CPM]	Y 1 EL D (R=RE WORK)	COUNT T 1KE	R EM ARK \$
NON E				22	BLK/SPIKE	TISSUE BCNE FISSUE		1.05±0.03E 03 1.75±0.03E 03 1.32±0.05E 03 5.05±0.05E 03 5.05±0.05E 03 1.70±0.05E 03 1.70±0.05E 03 1.70±0.05E 03 2.31±0.05E 03 5.75±0.05E 02 6.25±0.05E 02 6.35±0.05E 02 6.35±0.05E 02 6.35±0.05E 03 7.70±0.05E 02 8.75±0.05E 02 8.35±0.05E 03 8.75±0.05E 03 8.75±0.05E 03 8.75±0.05E 03 8.75±0.05E 03 8.75±0.05E 03 8.75±0.05E 03	00000000000000000000000000000000000000		

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TABL	TABLE E.13 (CONTINUED)	(CONT	INUED)				1				
ARC	רטכענ	00 00	LOCATION RUC COLLECTION A NO.	TLW IN ANALYSIS NO.	EVENT	TYPE	HT. (02)	PU-239, 240 ACTIVITY (DPH)	Y IEL C (R=RE WORK)	VIELE COUNT (R=RE TIPE WORK)	
NONE				RQC-29 30 31	BLK / SP I KE	1155LE		3.54#0.08E 02 2.60#0.08E 02 3.76#0.09E 03		200 200 200 200 200	
			32-C 33-C	M M M				1.15+0.02E 03 3.07+0.07E 03 4.55+0.10E 02		223	
			36-0	' SS - 40 Pr PS - PS -		BONE		1.86+0.04E 03 1.25+0.03E 03		\$00 \$00 \$00 \$00 \$00 \$00 \$00 \$00 \$00 \$00	
			9 B B B C C C C C C C C C C C C C C C C			11 S¢ LE	, o o d	2.9140.06E G3 8.0540.18E 02 3.7240.08E 03	76-1 82-5 63-6	222	
			41-0 43-0	444 M CV W		BCNE		4.44+0.18C 00 1.32+0.41E 00 1.23+0.11E 00	68.6 72.6 44.4	200 200 200 200 200 200 200 200 200 200	

OPS-1  Lab Blank Urine  1  1  1  1  1  1  1  1  1  1  1  1  1	T L W Analysis	Sample Event	Sample Type	Pu-239, 240 Activity Idom	Yield R=Rework	Count	Re	Remarks
2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	- 10kr	· · · · · · · · · · · · · · · · · · ·	11-12-	1.76 ± 0.24	44.3	200	Chemist Urine	
25 11 10 11 13 14 15 16 17 18 18 19 11 11 11 12 13 14 15 16 17 18 18 18 18 18 18 18 18 18 18	1-640	A DIAIR	) 	1 08 + 0 32	14.1	200	=	•
25 11 11 12 13 14 15 16 17 18 18 19 19 11 11 11 11 11 11 11 11 11 11 11	7	: :	=	10.0010	14.0	000	2	:
55 6 7 7 10 11 13 14 15 16 17 18 18 18 19 11 11 12 13 14 15 16 17 18 18 18 18 18 18 18 18 18 18	m ·	: :	=	3 74 + 1 16	. 7	200	•	:
25 11 11 12 13 14 15 16 19 11 19 11 19 11 11 11 11 11 11 11 11	₩,	: =	=	01:11 F1:3	. 4	200	=	:
22 22 22 22 24 25 25 25 25 25 25 25 25 25 25 25 25 25	n 4	=	=	50.0 ± 50.0	36. 6 R	250	:	=
11 10 11 12 13 14 16 19 19 19 11 19 11 11 11 11 11 11 11 11	۰ د	=	:	0 63 40 40	10.7	200	=	=
9 10 11 12 13 14 16 19 19 19 11 19 11 19 11 11 11 11 11 11	~ 4	=	=	0.53 ± 0.26	21.4	200	=	=
11 12 13 14 16 16 19 19 19 11 19 11 11 11 11 11 11 11 11	<b>D</b> (	=	=	0.20 ± 0.20	35.6	121	=	=
111 112 113 114 115 119 119 119 119 119 119 119 119 119	<b>~</b> •	=	=	0.16 ± 0.16	43, 5	121	=	t
112 113 114 115 119 119 119 119 119 119 119 119 119	<u>-</u>	68 68	=	0.00 × 0.09	30.0	200	=	:
13 14 15 16 19 20 22 22 31m. Blank	7 -	=	=	$0.00 \pm 0.20$	14.5	200	=	2
115 116 119 120 120 130 14 11 11 11 11 11 11 11 11 11 11 11 11 1	- C	=	=	0.16 ± 0.16	27,3	200		=
15 16 19 20 22 22 36m. Blank	7	2	=	$0.07 \pm 0.09$	51.4 R	300	=	=
16 19 20 22 31m. Blank 11 11 11 11 11 11 11	P 4	:	=	0.19 ± 0.19	22.2	200	=	=
19 20 22 31m : : : : : : : : : : : : : : : : : : :	n 4	=	=	0.00 ± 0.02	62.0	200	=	:
22	9 6	:	z	0.06 ± 0.06	39,3	900	:	:
22	9 0	=	=	$0.22 \pm 0.22$	47.4	250	=	=
22 H H H H H H H H H H H H H H H H H H		=	=	0, 10 ± 0, 06	45.0	400	=	=
Sim. Blank	2 6	=	=	0.08 ± 0.10	50.1	220	2	<b>2</b>
				0.004 µg U**	65.0	Fluor.	**Conc/0. 2 1h	sample
	۲,			6.005	68.0	<b>:</b>		: :
	) ee	=	=	0.003	:	2	=	=
= = =	) <del>-</del>	z	=	0.005	69.0	=	: : m 0 :	2 :
2 Z	-	=	Hamburger	0.019 "	;	=	: : :	=
	7-5	=		0.010	66.5	=		: 1
7-27		2	:	0,005	;	=	. 0.5	=
E. 4	H-4	=	=	0.016 "	48.4	z	: :	=

TABL	E E.15	RADI	<b>OCHEMICAL</b>	ANALYSIS O	F ROLLER	COASTER	PLYSICA	TABLE E.15 RADIOCHEMICAL ANALYSIS OF ROLLER COASTER PLYSICAL QUALITY CONTROL SAMPLES	ROL SAN	IPLES	
ARC	L D C A T 10 V	201	TLW COLLECTION NO.	TLW ANALYSIS NO.	EVENT	TYPE	HT. (G/S)	PU-239, 240 ACTIVITY (DPM)	Y 16LO (R=RE YORK)	COUNT	A EMARK S
25.	BA-65-A 05-B	A-G	NONE COC-	1091	SPIKE	SOIL_	1	1. 26+0.03E 05 1. 59+0.05E 05		202	TD
	88-09 86-03	O- M		1098			e.83	ŀ			
	80-0	. ~		1089							
	85-04 86-04	<b>4</b> ) 4		1100							
	811-10-4	<b>∀</b> -0		1093			5.9%	6.06#1.58E 00	43.8	6	10
		P		1103			5.94	0	64.7	60	10
	61-07			9601							
	8K -08	€		1601							
	8L-01	<b>~</b>		1095							
	8H-0	<b>~</b>		1099				1	,	;	
	80-08-	5-A		1092			10.47	2.39t0.07E 02	63.5	, ,	<u>ာ</u>
		9		1102			10.41	0	<b>1.09</b>	9 9 9	10
	CM-09-A	¥-6	8164	1104	C S 1 I		50.03	0	35.0	C?	PC
		-42		1112			50.03	0	14.9	S,	A 10
	50-K)	8-6		1105			50.03	Ò	31.9	Ç,	PCE
	CM-09-82	9-82		1113	-		50.02	1.16#0.05£ 03	48.3	20	R 10
	Bn-0]	1 - A	8188	1108	C \$111		50.05	0	14.2	9	PCE
		-42		1116			50.05	1.0340.036 05	37.8	0.7	ر 0
		ပု		1110			CA50		90	Q	PCE
		<b>~</b> 2		1118			CA50		54.3	0,	م 5
		P		1109			ó	0	-	9	PcT
		-82		1117			50.04		35.7	S	5 5
	221-A	•	NONE	221-A	QUAL.	sor.			80.5	1000	DP3/34
	<b>6</b> 2	<b>6</b> C		P				.00 + 1 .00 E-0	67.5	1000	1×/ Kd0
	513-A	•		\$13-A					53.5	1000	1k/ Kd0
	7	•		<b>•</b>				-300.	4-14	1000	DP:4/ML
	CA-29-A	A-0		CA-99-A				2.3010.04E 01	9.29	1000	DP3/4L

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TO \* FOTAL DISSOLUTION
PD \* PARTIAL DISSOLUTION
RIO \* RESIDUE - TOTAL DISSOLUTION
PRE \* PARTIAL DISSOLUTION AND EXTRACTION

(CONTENUED)	******************	ION TLY
TABLE E.16		ARC LOCATION

G2 CA-99-B QUAL. SOL. 2.26#0.04E 0  CB-42-A CB-42-A S.0. S.0. S.0. S.0. S.0. S.0. S.0. S.0	ARC LOCATION TLW COLLECT NO.	LOCATION TLW COLLECTION NO.	TLW OLLECTIC NO.	TLW DV ANALYSIS ND.	EVENT	TYPE	¥1.	PU-239,240 ACTIVITY (DPH)	YIELD (R:RE WORK)	YIELD COUNT (R*RE TIME WORK)	REHARKS
CB-42-A -11-A -11-A -11-A -11-A -11-A -11-A -11-A -11-A -11-A -8 -8 -8 -8 -8 -8 -8 -8 -8 -8		A-90-47	A VON	COC-CA-99-B	OUAL.	\$01.		2.26#0.04E 01	90.6	1000	DPM/ML
-11-A -11-A -13-A -8 -8 -8 -8 -8 -8 -19-A -8 -733-A -8		CR-42-4	)	C8-42-A	· •			5.08#0.15€ 02	76.6	20	DPH/PL
-11-A -8 -8 -8 -8 -8 -8 -8 -219-A -8 -733-A								5.01+0.15E 02	73.4	0.7	DPH/HL
CC-30-A CD-93-A CD-93-A CA-58-A CA-58-A 219-A 733-A		4-11-		-11-				4.68#0.14E 02	73.6	90	DPM/HL
CC-30-A -B -B -B -B -B 219-A -B 733-A			•	( ec				4.58t0.15E 02	4.19	20	DPIX/ML
CD-93-A CA-58-A CA-58-A 219-A 733-A		4-04-07	•	A-0.F-2.2				8.59#0.28£ 01	68.3	O.	DPH/HL
-A CO-93-A -B -B -A CA-58-A -B 219-A 733-A								8.21#0.24E 01	68.2	2	DPM/HL
-8 -8 -8 -8 -8 -8 -8 -8 -8 -8 -8 -8 -8 -		4-60-00		CD-63-4				4.8410.08E 03	73.1	20	DPH/HL
-A CA-58-A -B 219-A 733-A -B				4 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5				4.8310.076 03	77.2	63	DPH/ML
219-A 219-A 733-A 733-A		- 8 S - 4 J		CA-58-A				2.1740.06£ 01	31.4	1000	DPM/PL
219-A -B 733-A								2.25#0.05E 01	37.1	1000	DPH/FL
733-A 733-A		219-4		219-A				1.00 # 1.00 £-02	56.1	200	14/840
733-A -8				; ec				2.00#2.00E-02	33.9	800	DP11/ML
82-		711-4		733-A				2.0012.00E-02	42.2	<b>2</b> 00	DPH/ML
		<b>1</b>		87				1.00#2.00 E-02	43.3	200	DP3479L

TD = TOTAL CISSOLUTION
PD = PARTIAL DISSOLUTION
RTD = RESIDUE = TOTAL DISSOLUTION
PCE = PARTIAL DISSOLUTION AND EXTRACTION

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1	TABLE E.18	RADIOCHEMICAL AN	ALYSES OF TL	W PHYSICA	LOUAL	RADIOCHEMICAL ANALYSES OF TLW PHYSICAL QUALITY CONTROL SAMPLES
TLW Analysis No.	, 6,	Sample Type	39, 24( wity	**	Count	Remarks
DWS-1	1	Reagents	# .	70.6	0001	
TIS- LL-1	Pu-236 Std	Pu-239 Tracer	20.9 ± 0.4/ml	71.6	000	Fu-239(Mass Spec)=20.8 dpm/mi
TLS-HL-1	=		1024 # 1 //II.1	65. 6	0	3COT= ::
=	=	:	1028 ± 29/ml	<b>66.4</b>	20	m/mdp 2501=
CBR-1115	Lab Blank	Floor Swipe	41	72.0	360	Floor in front of Hond #1
=======================================		Bench Swipe	<b>*</b> 0.	80.1	360	Slight Contam. removed
1120	=	Floor Swipe	ö	80.4	300	Floor in front of Hood #2
1121	:	Bench Swipe	<b>*</b> 0.	84.2	300	
1123	=	Virg Nev Soil	#	40.4	35	Preshot CS 1 Soil Sample
1124	=	= = ==	$0.76 \pm 0.30$	51.0	300	
1125	:	10 88 11	£ 0.	31.1	300	20 27 27
1126	=		3	86.3	360	High Level Tracer used
1127	: :	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	$0.05 \pm 0.14$	711.7	360	=
1128		:	12 ± 0.	84.9	200	Low " " "
1129	=	=	$0.11 \pm 0.11$	78.2	200	
RRB-1	=	:	$0,33 \pm 0,38$	38, 3	250	** ** **
7	=	:	$0.02 \pm 0.09$	47.0	250	a
· (e1	=	:	0.08 ± 0.08	43.3	240	
₩	=	•	$0.14 \pm 0.13$	58.9	240	41 68 69
'n	=	2	$0.17 \pm 0.17$	45.0	150	
•9	=	2	$0.0 \pm 0.13$	29.3	150	= = =
1	=	**	0.04 ± 0.16	29.1	240	:
80	=	<b>:</b>	$0.31 \pm 0.27$	60.9	240	** ** **
R-1	=		0.001 HZ U**	90	Fluor.	Fluor. + *Conc/Total Sample
R-2	=	•		=	=	
R-3	:	3	0,002 " "	=	3	
R-4	=	:	0.002 " "	=	=	
X-5	:	:	0.001 " "	3	:	
R-6	=	=	0.002 " "	=	:	
R-1	=	:	0.001 "	=	=	
80.≃	:	:	0.001	=	=	
· - ≃	=	=	0.007 " "	=	=	
R-10	:	=	0.002 " "	:	:	
n-11	=	=	0.001 " "	=	=	
R-12	:	=	0.007	:	=	
<b>*</b>	=	=	0.001	:	=	
	: :	2	0.002 "	:	:	

TABLE E.17 NUMBER OF ANALYSES OF BIOLOGICAL SAMPLES FOR PLUTONIUM AND URANIUM	ES OF BIO		CAL SAMPLES	Ž	PLUTONIUM	A	D URANIUM		4 4000	
SAMPLE TYPE	ა გ		SHEEP		BURKO		NO ANIMAI	 	TOTAL	
	Pu	n	Pit	Ξ	Pu	3	Pu	ח	Pu	٥
									·	
Bone	31	~	35	9	30			-11-	96	10
Kidney	31	7	34	4	27				92	=
Liver	30	9	30	7	30			-	06	2
Lung	27	20	34	12	29	62			06	35
Hilar Node	30	~	34	9	29			7	93	13
Trachea	22		80		7			-	37	
G.I. Tract	21		80		9				35	
Pharyngeal Mucosa	21	-			9				27	
	21		<b>~</b>		7				31	
Urine			61						19	
Feces			21					-	21	
R.C. Qual. Control (Tissue)							39		39	
R.C. Qual. Control (Bor.e)							<b>~</b>		•	
TLW Qual. Control (Urine Bik)							20		50	
TLW Qual. Control (Meat Bik)							# 60	<b>6</b> 0	<b>6</b>	Ð
Total	234	4	273	29	166	m	"	φ	744	83

\* Not listed in the data tables of this report

TABLE E.18 NUMBER ANALYSES OF PHYSICAL SAMPLES FOR PLUTONIUM AND URANIUM	S OF PH	YSICAL	SAMP1.	S FOR	PI.UTON	N AN	2 2 2 2	MΩ				
SAMPLE TYPE	DOUBLETRACK	TRACK	C.8.7	_	C.S. II	=	C.8.11	=	NO EVENT	_	TOTAL	
	Pu	ח	P.	ב	Pu	n	Pu	ם	Pu	+	٠, م	2
				•		•		-				
Casella Samples	262	37	129	12	314	44	197	35		ψ1 •	905	108
Anderson Samples	132	12	98	<b>с</b>	46	7	111	15			384	۲.
Total Air Samples	30	9	27	7	37	18	7				801	31
Total Air Sampler Disp.	=	~	9	7	က		32	7			25	พ
Sequential Air Samples					11		24				35	
Deposition Sample	63	-	55		104		137				389	~
Water Sample			30	æ	44	2	65	Ξ			139	32
Aluminum Collector	ထ		<b>~</b>		24						36	
Vegetation (Sagebrush)	91		7		12		13				21	
Soil Fractions	98	26	37	37	31	3	25	22			176	176
Balloon Wire Swipes			16*		14*		<b>*</b> 9			-	36.	
R.C. Qual. Control (Soil)					₩		•	-	35	-	45	
A.C. Qual. Control (Solution)									20		20	
TLW Qual, Control (Lab. Blk)									23 1	<del>-</del> -	23	*
Tracer Standardization (Sol.)									13	-	13	
Qualification Samples (Soll									82		<b>3</b> 6	
and Solubility)											-	,
Misc. (Casella's and	~20**	~ \$0**	~50** ~50** ~50**	~ \$0** ~ \$0	~ 50	~ \$0 + ₹	~50** ~50**	* 20**		}	~ 200~	, 100 ·
Andersen's)												
T A SOL	57.8	213	0.9	170	644	217	656	198	119	14   20	2607	598
	;	-							-	إ	( · · · · ·	1

Duplicate analyses performed
 Analyses performed on samples received from Ebarline Instr. Inc. and Isotopes Inc. but not listed in the data tables of this report.

## APPENDIX F EQUIPMENT AND PLOT OF TYPICAL SPECTRUM

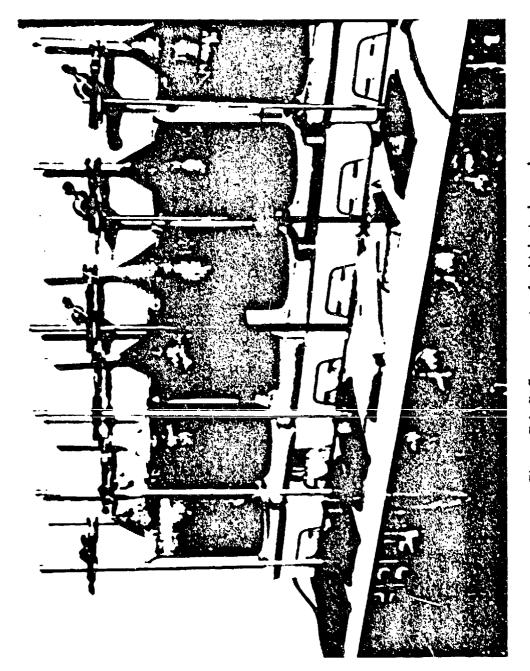


Figure F.1 Reflux apparatus for biological samplo. (Tracerlab photo)

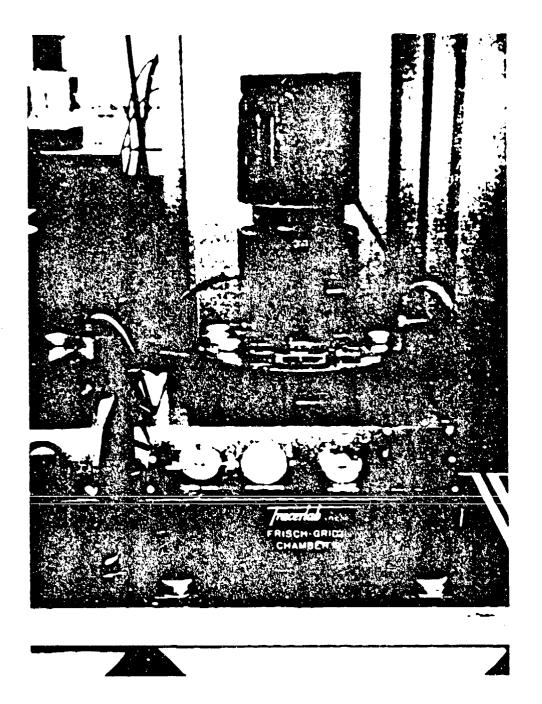


Figure F.2 Frisch-grid chambers. (Tracerlab photo)

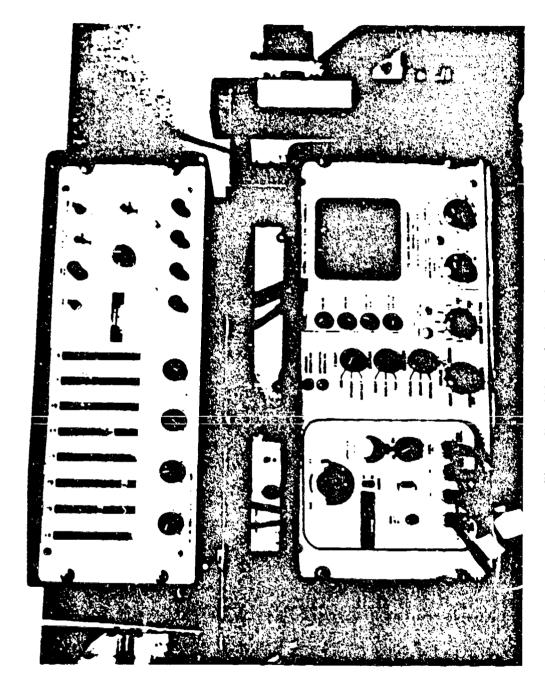


Figure F.3 TMC multichannel analyzer. (Tracerlab photo)

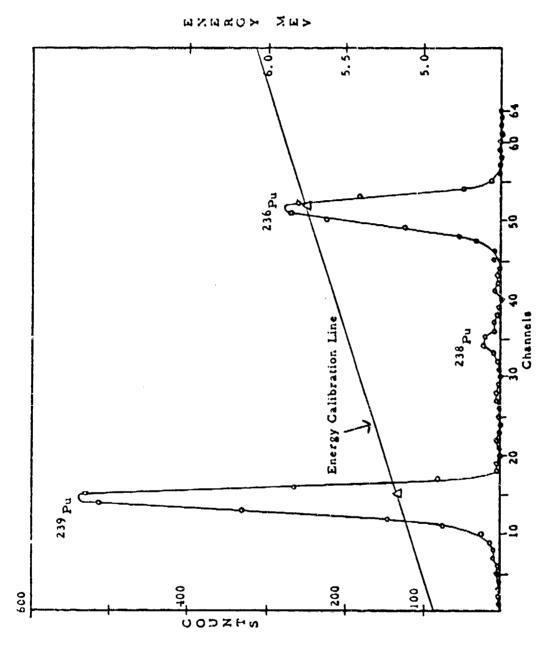


Figure F.4 Typical spectra, biological sample (burro liver).

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